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# Assessment of PM10 and heavy metals concentration in a Ceramic

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# Cluster (NE Spain) and the influence on nearby soils.

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9 Abstract

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Environmental pollution control is one of the most important goals in pollution risk 11 assessment today. The aim of this study is conducting a retrospective view of the 12 evolution of matter particulate (PM10) and the heavy metals (Cd, Ni and Pb) at different 13 14 localities (Alcora, Castellón and Onda) in the Spanish cluster ceramic in a period between January 2007 and December 2011. The study area is in the province of 15 Castellón. This province is a strategic area in the framework of European Union 16 17 Pollution control. Approximately 80% of European ceramic tiles and ceramic frits manufacturers are concentrated in two areas, forming the so-called "Ceramics Clusters"; 18 one is in Modena (Italy) and the other in Castellón (Spain). In these kind of areas, there 19 are a lot of pollutants from this industry that represent an important contribution to soil 20 contamination so it is necessary to control their air quality. In these areas atmospheric 21 22 particles are deposited in the ground through both dry and wet deposition. Soil is a major sink for heavy metals released into the environment. For this purpose the levels of 23 PM10 in ambient air and the corresponding annual and seasonal trend were calculated. 24 25 The results of the study show that the PM10 and heavy metals concentrations are below the limit values recommended by European Union Legislation for the protection of 26

human health and ecosystems in the study period. There is an important reduction of 27 28 them from 2009 in all control stations due to economic crisis and subsequent decrease of industrial activity. The atmospheric seasonal tendency of pollutants concentrations is 29 marked by the rate of industrial activity and additionally by the temperature. 30 Complementary, a comparative study of heavy metals levels in soils was performed in 31 this area. Soils with low pollution by Ni and Pb were detected, while different pollution 32 by Cd was found depending on the sampling site. Although there is an evident reduction 33 of PM10 and heavy metals levels, the results show that these pollutants have been 34 accumulated in the soil close to emission sources. 35 36 37

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39 Keywords: PM10, Heavy metals, soil pollution, ceramic cluster

### 41 **1. Introduction**

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Environmental pollution control is nowadays one of the most important goals in 43 pollution risk assessment. Metal contamination of the environment raises concern for 44 the possible impact on human health, and the existence of heavy metals in soils, of both 45 natural -inherited from the origin material (Galán et al., 2008), lithogenic or pedogenic; 46 (Tume et al., 2011)- or anthropogenic origins -due to human activities as agricultural 47 projects, water waste discharges or atmospherics emissions (Grav et al., 2003; Bech et 48 al., 2008; Hovmand et al., 2008)-, is well-recognized as a potentially important source 49 50 of human exposure (Muller and Anke, 1994; Davis et al., 2009; Chen et al., 2005).

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PM10 particles and heavy metals are emitted into the atmosphere as aerosols, mainly as 52 53 a result of human activities, and are deposited on the ground (Borgna et al., 2009). So the deposition of atmospheric particles is an important source of soil contamination in 54 areas exposed to significant pollution levels. Sakagami et al. (1982) reported that there 55 was a close relationship between heavy metal concentration in soils and those in the 56 airborne particles that fall. Other authors have found that atmospheric inputs of heavy 57 58 metals to agricultural systems can be a significant contribution to metal loading in soils. For example, Alloway. (1999) found a mean atmospheric deposition rate for Cd of 59 around 1.9 g ha<sup>-1</sup>y<sup>-1</sup>, which Nicholson et al. (1999) calculated to be 50% of the total 60 61 annual input of Cd to agricultural land in the UK. Nicholson et al. (1999) also calculated that atmospheric deposition of other metals such as Ni to be between 32-45% of the 62 total annual inputs of these metals to soils. Then, in comparison to the other forms of 63 metals inputs, potentially, atmospheric inputs have much more immediate impact on 64 agricultural systems (Gray et al., 2003). 65

Once contaminated, soils typically remain in this condition for protracted periods of 67 time because of sorption of metals on to particles and limited mobility (Peris et al., 2008; 68 Ferri et al., 2012). This fact produces a potential human cumulative exposure. The 69 accumulation of heavy metals in agricultural soils, including home vegetables gardens 70 may be of particular concern since consumption of vegetable grown in metal 71 contaminated soils may pose health risks for the population residing in these areas (Cui 72 et al., 2005; Intawongse and Dean, 2006). Elevated concentrations of trace elements in 73 the soil-water-plant ecological system are of great concern because of possible 74 75 influences on the food chain. (Tume et al., 2008). While essential trace metals, such as 76 Ni, are necessary for plant growth and/or human nutrition at low levels, they may also 77 be toxic to both animals and humans at high exposure. Other trace elements, for 78 example Cd and Pb, may also inadvertently enter the food chain and pose health risks to human and animals (Laughlin et al., 1999, Micó et al., 2006). Prolonged metal exposure 79 spanning neurodevelopmental periods may also increase the risk of neurodegenerative 80 conditions for elderly people (Aelion et al., 2008). Moreover, heavy metals in soils can 81 generate airborne particles which may affect the air environmental quality (Gray et al., 82 83 2003).

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The aim of this study is conducting a retrospective view of the evolution of particulate matter (PM10) and selected heavy metals (Cd, Ni and Pb) at different localities (Castellón, Alcora and Onda, fig.1) in the Spanish cluster ceramic from January 2007 to December 2011. Complementary, a comparative study of heavy metals levels in soils was performed in this area. The data were extracted from the papers of Roca-Pérez et al., (2010) and Jordán et al., (2009). 91

# 92 **2. Description of the study area**

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The study area is located in the East of Spain, in the province of Castellón. This province is a strategic area in the frame work of the European Union (EU) pollution control. Approximately 80% of European ceramic tiles and ceramic frit manufacturers are concentrated only in two areas, forming the so-called "ceramics clusters"; one is in Modena (Italy) and the other in Castellón.

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The type of climate in the study area is Mediterranean characterized by wet and mild winters, dry and warm summers, and an average temperature variation of 13.5°C. Rainfall is abundant in spring and autumn, coinciding with the dominance of western winds while summer is drier, dominated by the Azores anticyclone (Vicente et al., 2011). Yearly rainfall generally does not exceed 400 mm.

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106 This area has a complex Mediterranean atmospheric environment, with low rainfall, soil with poor vegetation coverage and frequent high particulate air-mass intrusions from the 107 Sahara (Rodríguez et al., 2002). A system of local sea breezes is also present in the 108 study area due to geographical characteristics and the proximity to the sea. These 109 periodic land-sea winds, which have been extensively studied by several authors 110 111 (Martín et al., 1991; Boix et al., 1995; Millán et al., 2001; Sanfeliu et al., 2002), govern the microclimate in this area, resulting in an overall effect of smoothing the 112 temperatures (Pogosyan, 1965). Due to this system of breezes, the concentration of 113 pollutants may be affected by emission sources located outside the study area on a daily 114 basis (Fig. 1). 115

117 The origin of air PM10 in this area is both natural and anthropogenic. The former is due 118 to the resuspension of mineral materials from the surrounding mountains with poor 119 vegetation coverage and the long-range transport of materials from North Africa 120 (Rodríguez *et al.* 2001, Pérez *et al.* 2006; Esteve *et al.* 2012). These dust intrusions 121 from North Africa influence ambient PM10 levels in the study area at around 2  $\mu/m^3$  on 122 an annual basis (Minguillón *et al*, 2009).

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Anthropogenic pollution sources originate from automobile traffic (mobile sources) and 124 125 industrial activity (fixed sources). The main industrial activity in the study area is based on producing ceramic tile (Vicente et al. 2007). This industrial sector has two types of 126 factories, one for the manufacture of tiles and the other to supply raw materials. The raw 127 materials of the tile body consist mainly of clay from sources such as opencast quarries 128 within the ceramic cluster area (Jordán et al., 2009; Sanfeliu et al., 2009). The raw 129 materials for decoration involve manufacture frits, enamels, and colour (Jordán et al., 130 2006). In the manufacture of ceramic tile, channelled and diffuse emissions from the 131 production processes and the storage, handling and transport of raw materials all 132 133 increase the concentration of particles in the air (Sanfeliu et al., 2002). However, particle emissions from the manufacture of pigments, frits and enamels probably have a 134 greater impact on the levels of heavy metals than on particle mass. (Minguillón et al., 135 136 2007). An additional important factor is that a power station, a refinery and several chemical industries are located at east of the study area (Boix et al., 2001) (Fig. 1). 137 These industries together contribute to environmental pollution in the area. Finally, 138 relevant sources of secondary PM in the area include precursor emissions of the volatile 139 organic compounds (VOC's), NOx and SO<sub>2</sub> from high temperature ceramic processes, 140

power generation, petrochemical processes and biomass combustion (Minguillón *et al.*,
2007).

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In the case of chemical pollutants in air PM10, nickel is found as a trace element in 144 petrol, and therefore its release into the atmosphere is related mainly to the combustion 145 of fossil fuels (coal and fuel oil) in electricity and heat production and also in traffic 146 147 exhausts (Pacyna et al. 1984; Ghio and Samet 1999). Nickel oxides are also widely used as components of pigments used in the ceramics industry. Concentration levels of 148 cadmium in ambient air are associated with industrial processes in the manufacturing of 149 150 frits and enamel. Emissions of cadmium are also produced in the processes of a nearby power station (Boix et al. 2001). The most important emission of lead are traffic 151 exhausts. Petrol additives contain lead (Parekh et al. 2002), which after combustion is 152 released into the atmosphere as organic lead (lead bromide and lead chlorinebromide) 153 (Pacyna 1998). With the introduction of new international laws, the use of lead in petrol 154 has been banned, and this contribution is now minimal, its use reduced to obsolete 155 means of transportation. In the ceramics industry, lead oxides are also used extensively 156 as a component of pigments. Relationships between the emissions from this sector to air 157 158 ambient levels of lead in close urban areas have been identified (Sanfeliu et al. 2002; Gómez et al. 2005). 159

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161 The study is focused on 154 km<sup>2</sup> that is the surface between the three cities studied. 162 Alcora is situated at 279 m above sea level, Onda at 194 m and Castellón at 30 m. There 163 is a high altitude difference between these cities in a small area. The distance between 164 Castellón, Alcora and Onda is 17 km.

The study area is based on Quaternary materials generated by torrents from inland massifs (Maestrazgo Range and tributaries from the Sierra de Espadán range), which have formed lagoons closer to the sea. This process is reported to have started at the end of the Tertiary period, as the Pliocene became the Pleistocene era, favoring a climate suitable for torrential erosion and subsequent transport and deposition. The main source of sediments has decreased in the last few years because of regulations of the course of the river (Vicente et al., 2007)

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Alcora is in the Miocene surrounded by Cretaceous and Quaternary. Its materials are 174 175 conglomerates, limestone, marl, clay and lignite. The materials around this city are limestone, marl, sand, clay and gravel. Onda is located in the Quaternary enclave, 176 surrounding a Jurassic area. This town is surrounded by Quaternary and Triassic. The 177 178 main materials of this city are conglomerates, limestones and dolomites, around marl, clay and limestone in the Triassic and red clays in the Quaternary. Castellón is located 179 in the Quaternary and the main materials are clays, silts and sands. All deposits are 180 Quaternary sedimentary. 181

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- **3. Materials and Methods**
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185 3.1. Atmospheric particles

186 *Sampling collection* 

A PM10 medium volume sampler model IND-LVS3 manufactured by Kleinfiltergerät
was used. This device is considered as a reference according to European regulations
(European Council Directive 2008/50/EC; UNE-EN 12341:1999), for the sampling of
PM10 particles. The technology used in the equipment consists of blowing air through

an inlet with a vacuum pump. The particulate matter was blown in through the opening 191 192 circumference between the frame and the round cover mounted on top. Within the sampler inlet the airflow was accelerated by eight impactor nozzles and then directed 193 toward the impacting surface. Particles were trapped on a permeable support consisting 194 of a 47 mm diameter filter. The device contains a temperature sensor with a radiation 195 protector that eliminates deviations in the reading caused by solar radiation in addition 196 to a pressure sensor. The sampling flow volume was 2,3 m<sup>3</sup>/h during 24 h periods. A 197 total of 4.100 PM10 samples (1.496 Castellón, 1.253 Alcora and 1.351 Onda) were 198 collected in filters from 2007 to 2010. The filters used were quartz fiber filters were 199 200 used, according to UNE-EN 12341:1999. They were made from SiO<sub>2</sub> pure base, totally free of additives. These filters allow an efficiency of separation greater than 99.5%. 201

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The samplers of the PM10 were positioned between 1,5 and 4m about the ground according to the Directive 2008/50/EC (Annex III, Microscale sitting of sampling points).

206

207 Gravimetric analysis

Particle concentrations levels were determined gravimetrically. This method consists of weighting the filters twice: firstly empty and then with sample. The filters must be kept for at least 48h in a special chamber. The conditions inside the chamber are 50% relative humidity and a temperature of 20°C, according to normative UNE–EN 12341:1999. Filters were weighted on an analytical balance with a precision of 0.1 mg. The PM concentration levels were determined based on the sample quantities obtained and the volume of air pumped.

216 *Chemical analysis* 

217 The levels of Cd, Ni and Pb in the PM10 samples were determined by inductively coupled plasma mass spectrometry (ICP-MS). The equipment used was an Agilent 218 model 7500CX that included a quadrupole, a collision cell and an integrated 219 autosampler. The equipment was installed in a chamber with a clean air filter unit and 220 221 an independent air conditioning system. This instrumental technique allows the Cd, Ni 222 and Pb levels to be rapidly identified after dissolution of the sample. Dissolution was achieved by acid digestion in hermetic Teflon recipients. This methodology has been 223 used by many authors (Kubilay and Saydam, 1995; Querol et al., 2000). 224

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In order to detect any possible traces of contamination-causing As, Cd, Ni or Pb contained in the reagents and quartz filter fibres, digestions with only reagents (blank reagents) and filters without a sample (blank filters) were performed. The SRM 1648 "urban particulate matter" pattern was used to validate the results. This pattern consists of particulate matter of anthropogenic origin collected in an industrialised urban atmosphere and was an adequate standard of reference for this study.

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233 3.2. Soil

#### 234 Sampling collection

The sampled soils were selected between the different samples from the papers of Roca-Pérez et al., (2010) and Jordán et al., (2009). The soils were chosen for their proximity of the industrial atmospheric emission sources and their physicochemical property diversities, especially pH (between 6.5 and 8) and carbonate content, the soil of Alcora is noncalcareous while Castellon and Onda are calcareous. In Alcora, almond trees are the main crop, followed by olive trees, vineyards, some carob trees, and a few fig trees. In Castellón and Onda, the crop of orange trees is the main. These soils have a low capacity use and a erosion risk. In addition in Onda, there are other terraced fields destined for the cultivation of carob, olive, hazel and almond trees. All studied soils are anthrosols, according to the FAO World Reference Base for Soil Resources. These soils are a type formed or heavily modified due to long-term human activity, such as from irrigation, addition of organic waste or wet-field cultivation used to create paddy fields.

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According to the soil classification of FAO/UNESCO 1998, the soils chosen of the papers of Roca-Pérez et al., (2010) and Jordán et al., (2009) are classified as:

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*Fluviosols:* Alcora and Castellón. Typical soils found of valley bottoms and
 floodplains, widely represented in the province of Castellón. These occupy the
 coastal plains and the terraces of the river courses from the inland areas to the
 coast. However, inland areas may develop on slopes fitted out by terraces. These
 soils are partly immature and they are commonly well drained. Because of their
 development from recent alluvial deposits, these soils have a significant
 thickness and a variable texture (Antolín Tomás, 1998).

*Calcisol:* Onda. These types of soils are characterized by horizons with
 secondary enrichment of calcium carbonate, favored by semi-arid conditions.
 They are present either in the coastal plain or in the mountainous interior. In
 these soils predominate washing and carbonate accumulation is associated with
 geomorphic processes of glacis formation, alluvial fan, etc. (Antolín Tomás,
 1998).

These authors collected two kilograms of topsoil samples (0-20 cm) from each sampling site, air-dried, crushed, 2 mm sieved, mixed and stored at air-dried conditions for further analysis.

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269 *Chemical analysis* 

The methodology used in soil chemical analysis was described by Roca-Pérez *et al.* (2010) and Jordán et al. (2009) in their studies. Total contents of metals were determined by ICP-MS after microwave digestion of 0.5g of representative soil sample using 9 ml nitric acid (68% w/v), 1 ml hydrogen peroxide (30% w/v), 3 ml hydrofluoric acid (48% w/v), 2 ml hydrochloric acid (37% w/v) and 5 ml deionised water in the first steps (20 min at 200°C) and 30ml boric acid (4%) in the second (5min at 170°C) (EPA, 1996).

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## 278 4. Results and discussion

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280 4.1. Atmospheric concentrations of pollutants.

Table 1 shows the assessment of PM10, Cd, Ni and Pb according the limit values established by European Legislation (European Council Directive, 2008/50/EC). The concentrations of the pollutants are below the limit values recommended by European Union Legislation for the protection of human health and ecosystems in the study period,

with the exception of daily exceedances of PM10 in 2007 in the station of Castellón.

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There is an important reduction of the pollutants from 2009 in all control station due to the economic crisis and subsequent decrease of industrial activity. Additionally, due to the new European Directive (2010/75/EU) on industrial emissions, the best available techniques (BAT) were implemented in the industry in order to prevent or reduceharmful impacts.

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In Modena ceramic cluster this reduction has been also observed despite the Italian ceramic industry was the first in Europe to develop and adopt techniques for the reduction of the environment impacts in the seventies (Minguillón et al., 2013). So the economic crisis has been very important in the two industrial sites.

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298 4.1.1 PM10

299 In general, PM10 levels increase during the months with high temperatures due to a decrease in precipitation (Fig. 2) with the exception of 2007 and 2008 in Castellón and 300 301 Alcora. This causes a reduction in the cleansing effect on the atmosphere (Bergametti et 302 al. 1989) and consequently a greater contaminant concentration in the ambient air. The high temperatures during these months lead to increased dryness of the terrain, which 303 304 favours the resuspension of clay-loam substrate in the area (Gómez et al., 2005). At the 305 same time, the mixing layer, or lower part of the troposphere where the pollutants are free to move through the turbulence generated in the lower layers of the atmosphere, 306 307 increases its thickness and facilitates the mixing of air masses from the North of Africa in the low layers (Kubilay and Saydam, 1995). Intrusion episodes of long-distance 308 material occur, leading to an increase in the concentration of PM10. During winter 309 310 months, temperature inversions are generated. This phenomenon occurs on clear nights, with weak or no wind, when the soil losses the heat acquired by radiation and low-lying 311 air layers are cooled faster than the upper layers of air (Wallace et al., 2010). When 312 pollutants are emitted under temperature inversion conditions, they accumulate in the 313 layers of the troposphere close to the ground. This phenomenon causes transport 314

through these layers to occur too slowly, producing an increased concentration of 315 316 pollutants (Monn et al., 1995). This accumulation of pollutants is also found in Milan due to persistent thermal inversions (Marcazzan et al, 2001). During the autumn season, 317 the lowest values of the study were detected (Fig. 2). This was due to atmospheric 318 instability, the tendency of the atmosphere to resist or enhance vertical motion or, 319 alternatively, to suppress or augment existing turbulence (Zoras *et al.*, 2006). According 320 321 to some studies, as global weather conditions change, the input frequency of air masses from North Africa is reduced (the mixed layer decreases), rainfall increases, and there is 322 a greater cleansing effect in the atmosphere (Querol et al., 2002, 2004). 323

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The seasonal evolution of the pollutants concentrations could supply valuable information about the potential origin of them. These seasonal variations are dominated by changes in meteorological conditions (Chang and Lee, 2008) and by human activities (Guangjian et al., 2009).

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330 4.1.2. Heavy Metals

The constant tendency of de cadmium concentration during the year (Fig. 3) indicates 331 332 that the contribution of this pollutant into the atmosphere is continuous from all sources, both natural and anthropogenic. The small variation is due to one of the principal 333 origins of cadmium in the study area, the use of this element in the form of oxide in the 334 335 formulation of enamels and pigments in the ceramic industry (Matthes 1985). This sector cuts its production levels in the summer months and so cadmium levels were 336 337 accordingly lower. In addition, the power plant located within the study area also emits cadmium (Boix et al. 2001), which is higher in the months of higher energy demand 338 (cold months), wherein a light cadmium concentration increase was observed. 339

341 The light seasonal tendency for nickel (Fig. 4) of high concentrations during hot months 342 and low ones during the coldest months leads to nickel concentration levels being governed by two factors: anthropogenic activities within the study area and the ambient 343 temperature. High-temperature anthropogenic activities at the power and petrochemical 344 plants, etc., release nickel into the atmosphere mainly as fly ash and not in a gaseous 345 state (Boix et al. 2001). Due to their characteristics, these fine particles behave in the 346 atmosphere like a gaseous pollutant (Wark and Warner, 2000), and so their dispersion is 347 influenced by external factors like the temperature (Vicente et al., 2011). Higher 348 349 temperatures result in greater atmospheric turbulence (Wark and Warner, 2000), bringing greater dispersion away from the emitting sources located to the east of the 350 study area. The power and petrochemical plants are located there, thus favoring the 351 352 enrichment of nickel in the particle concentration levels in farther-away zones during 353 the hottest months.

354

The general behavior of lead concentration (Fig. 5) is to reach high levels in colder 355 356 seasons and low levels during warmer seasons. In the study area, the origin is linked to 357 the use of lead in the formulation of raw materials in the ceramic sector (Matthes 1985). These oxides are volatized in high-temperature industrial processes and they condense 358 upon contact with the atmosphere. An aerosol-vapor equilibrium exists for lead 359 360 dependent upon the air temperature. Low temperatures favor the aerosol phase, while high temperatures favor the vapor phase release, explaining that the dispersion is greater 361 362 at high temperatures, and therefore lead levels are less in areas near the emitting sources. Dispersion is less at low temperatures and so greater concentrations of this 363 contaminant abound near the emitting sources (Vicente et al., 2011). Another factor 364

needing consideration is that, in summer, this sector reduces production during vacation
 periods, resulting in less emissions and lower lead concentration levels in the ambient
 air.

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369 4.2 Soil pollution.

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Heavy metal concentrations in some different environments show high variability depending on the type of environment (industrial, urban or natural). Some authors have conducted studies of soils in different environments (Galán et al., 2002; Moral et al., 2005; Tong-Bing et al., 2005; Yay et al., 2008; Maas et al., 2010) and concluded that the soils most polluted by heavy metals occur in areas affected by mining and cement factories (Soriano et al., 2012).

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Soils affected by heavy metal emissions and PM10 in the ceramic cluster are carbonate reliefs, detrital or gypsiferous materials subsequently colluvial-alluvial deposits and marsh. The presence of carbonate indicates high soil pH, which tends to precipitate the heavy metals. The Cd, and other metals have a strong tendency to be adsorbed by carbonates. The Cd mobility is medium in basic and neutral soils, low in Pb and very low in Ni (García et al., 2009).

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Table 2 shows classification of soil contamination types based on their heavy metalsconcentrations, with four categories identified.

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Table 3 shows heavy metals concentration in soil in the study area. According to the polluted soil classification (table 2), in Onda and Alcora, the soils show no pollution by nickel but in Castellón shows light pollution. In the case of cadmium, Onda has light
pollution, Alcora medium and in Castellón very extreme pollution was detected. The
soils from Alcora and Castellón are unpolluted by lead and Onda shows medium
pollution.

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According to studies by D'Elia et al., (1999) the soils of the Bologna area have standard values of lead as in Castellón and Alcora, low pollution of cadmium as in Onda, while medium pollution of nickel were detected.

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### **399 5.** Conclusions

400

In this study a retrospective view of evolution of PM10 and the heavy metals (Cd, Ni and Pb) was conducted at three cities in the Spanish ceramic cluster, from 2007 to 2011. The results show that the concentrations of studied pollutants are below the limit values of European laws for the protections of human health and ecosystems, with the exception of daily exceedances of PM10 in 2007 in the station of Castellón. An important reduction of the pollutants from 2009 in all control station are detected due to the economic crisis and subsequent decrease of industrial activity

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The atmospheric seasonal tendency of the pollutant concentrations in the study area ismarked by the rate of industrial activity and, additionally, by the ambient temperature.

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412 Additionally, a comparative study of heavy metals in soils was performed in this area 413 because although the particle deposition process cleans the atmosphere, its ultimate 414 result is the transfer of toxic atmospheric pollutants into the soil. The study of metals 415 contents in soil is currently necessary to obtain reference values and assess their
416 contamination, because the pollution from atmospheric particulate metals in the soil
417 increases their toxicity

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Soils with low pollution by nickel and lead were detected in the study area, while
different pollution by cadmium was found depending on the sampling site, Castellón
shows extreme pollution, Alcora medium and Onda low.

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Even if there is an evident reduction of PM10 and heavy metals atmospheric levels in 423 424 the study area, these pollutants have been accumulated in the soil close to emission sources for years. The persistence of these pollutants in soil is much longer than in other 425 compartments of the biosphere, especially heavy metals which can be considered 426 427 virtually permanent. So, it is very important to know and reduce the emissions in order to minimize harmful effects on human health, with particular attention to sensitive 428 429 populations, and diminish damage to the environment as a whole. In this paper, the 430 importance of soil composition has been identified to develop environmental control strategies 431

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434

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