# Reactivity of dolomitizing fluids and Mg source evaluation of fault-controlled dolomitization at the Benicàssim outcrop analogue (Maestrat Basin, E Spain)

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#### Abstract

The mechanisms responsible for the formation of huge volumes of dolomitized rocks associated with faults are not well understood. We present a case study for high-temperature dolomitization of an Early Cretaceous (Aptian-Albian) ramp in Benicàssim (Maestrat basin, E Spain). In this area, seismic-scale fault-controlled stratabound dolostone bodies extend over several kilometres away from large-scale faults. This work aims at evaluating different Mg sources for dolomitization, estimating the reactivity of dolomitizing fluids at variable temperature and quantifying the required versus available fluid volumes to account for the Benicàssim dolostones. Field relationships, stable <sup>13</sup>C and <sup>18</sup>O isotopes, as well as radiogenic <sup>87</sup>Sr/<sup>86</sup>Sr isotopes, indicate that dolomitization at Benicassim was produced by a high temperature fluid (> 80°C). <sup>13</sup>C and <sup>18</sup>O isotopic compositions for dolomite vary from +0.5and +2.9 ‰ V-PDB and from +21.1 and +24.3 V-SMOW, respectively. A Mg source analysis reveals that the most likely dolomitizing fluid was seawater-derived brine that interacted with underlying Triassic red beds and Paleozoic basement. Geochemical models suggest that evolved seawater can be considerably more reactive than high-salinity brines, and that the maximum reactivity occurs at about 100°C. Mass-balance calculations indicate that interstitial fluids with high pressure and/or high temperature relative to the normal geothermal gradient cannot account for the volume of dolomite at Benicàssim. Instead a pervasive fluid circulation mechanism, like thermal convection, is required to provide a sufficient volume of dolomitizing fluid, which most likely occurred during the Late Cretaceous post-rift stage of the Maestrat Basin. This study illustrates the importance of fluid budget quantification to critically evaluate genetic models for dolomitization and other diagenetic processes.

**Keywords:** hydrothermal dolomitization; stratabound dolostone; Mg source; fluid reactivity; Maestrat Basin; carbonate reservoir quality.

#### **1. Introduction**

More than half of the Earth's carbonate rocks are dolomitized (Zenger et al., 1980; Land, 1985; Warren, 2000; Machel, 2004), and some 60% of worldwide oil and 40% of gas

reserves are stored in carbonate reservoirs (Montaron, 2008). Dolomitization is one of the most important diagenetic processes that modify the petrophysical properties of carbonates (porosity and permeability), thus controlling fluid flow and oil recovery (e.g.

Allan and Wiggins, 1993; Warren, 2000). A dolostone unit can be a reservoir, but it can also act as a barrier to flow (e.g. Lucia, 2004) depending on the original properties of the host rock, the pre- and post-dolomitization diagenetic history, the reservoir geometry and the type of dolomitization. Predicting the distribution of reservoir and non-reservoir rocks in dolomitized fields using subsurface data is a difficult task, because replacement dolomitization and associated processes (dolomite cementation and calcitization/dedolomitization) cause complex diagenetic overprints. In fact, dolomitized reservoirs are multiple-porosity typically systems (Purser et al., 1994). The vast majority of dolostones (CaMg[CO<sub>3</sub>]<sub>2</sub>) are of secondary origin, so that they formed by replacement of calcite (CaCO<sub>3</sub>) in different tectonic and geochemical settings as a consequence of the interaction of host limestone or lime sediments with fluids that caused the dolomitizing reaction (e.g. Warren, 2000; Machel, 2004; Whitaker et al., 2004; Roure et al., 2005). A common type of reservoir is the one composed of structurally controlled hydrothermal dolomite (HTD; Davies and Smith, 2006). Such dolostone forms by hydrothermal fluids that are usually transported along faults and/or fractures. The term hydrothermal refers to situations in which the advected fluid is at least 5°C warmer than the reacting host rock (White, 1957). Sedimentary-exhalative and Mississippi Valley-type (MVT) Pb-Zn ore deposits are commonly associated with hydrothermal dolomite (e.g. Cathles and Smith, 1983; Anderson and Garven, 1987; Shelton et al. 1992; Corbella et al. 2004; Greg, 2004).

The occurrence and distribution of burial fault-controlled dolostone is mainly constrained by (a) diagenetic/hydrothermal processes that dissolve/precipitate minerals and enhance/decrease porosity, (b) limestone matrix permeability defined by the sedimentary facies and pre-dolomitization diagenesis, and (c) fault network properties (e.g. length, hydraulic fault aperture, connectivity and segmentation) which often

determine fluid flow patterns. With these limitations, the resultant geometry and distribution of hydrothermal dolostones is very variable and ranges between massive irregular patches close to fault zones to stratabound bodies that extend away from faults, which act as feeding points for fluids (e.g. Davies and Smith, 2006; Wilson et al., 2007; Gasparrini et al., 2006). The principal parameters and processes controlling the transition between both end-members are still not well understood.

composition and sources of The dolomitizing fluids, as well as their driving forces, remain controversial (Braithwaite et al, 2004). Moreover, the main parameters that control dolomite stoichiometry and cation order are complex and still under debate (Dawans and Swart, 1988; Kaczmarek and Sibley, 2007; 2012). An effective replacement of calcite by dolomite requires one or more Mg sources and a physical mechanism capable of transporting sufficient amounts of it to the reaction site (Whitaker, 2004; Carmichael et al., 2008). Historically, dolomitization studies have been based on the stratigraphic, petrographic and geochemical characterization of dolostones (for reviews see Hardie, 1987; Warren, 2000; Machel, 2004), where the final outcome is normally a conceptual model illustrated by schematic cartoons. However, mass-balance calculations and numerical analyses are essential to test whether conceptual models are plausible from a hydrodynamic and hydrochemical point of view. Dolomitization models must also be hydrologic models (Machel, 2004, Whitaker et 2004) because realistic fluid-flow al., mechanisms are required to account for solute transport to and from the reaction zone. available Required vs. fluid budget quantifications must be an essential step in order to account for the fluid origin (Mg sources) and transport mechanisms (driving forces) in dolomitization case studies. Certain dolomitizing fluid properties (e.g. composition, temperature) are often calculated from geochemical analysis of rock samples. However, there are scarce examples of volume and source quantifications in dolomitization studies (e.g. Land, 1985; Kaufman, 1994; Reinhold, 1998). With the existing numerical tools it is possible to quantify the geologic processes responsible for dolomitization and gain a better understanding of them (e.g. Corbella et al., 2006; Whitaker and Xiao, 2010; Al-Helal et al., 2012;Xiao et al., 2013; Corbella et al., 2014).



**Fig. 1.** (a) Simplified map of the Iberian Peninsula showing the location of the Iberian Chain and Maestrat Basin; (b) Paleogeographic map of the Maestrat Basin during the Late Jurassic - Early Cretaceous rifting cycle showing thickness of syn-rift deposits and main fault traces. The black square indicates the location of the study area, while the black asterisks show the location of dolostones of the same age and type in the Maestrat Basin. Map modified after Salas et al. (2001).

The Benicàssim outcrop analogue (Maestrat basin, E Spain) is an outstanding field example of fault-controlled dolomitization (Martín-Martín et al., 2013; Corbella et al., 2014). In this area, seismic-scale stratabound dolostones extend for several kilometres away from largescale faults. Hydrothermal dolostones replace Early Cretaceous (Aptian-Albian) shallowmarine ramp limestones. This work aims at evaluating the possible Mg sources and quantifying the required versus available fluid volumes to account for the Benicàssim dolomitization. We first present an overview of the field area and a summary of the Benicàssim dolostone petrography and geochemistry. We then present and discuss new stable  $\delta^{13}$ C -  $\delta^{18}$ O and radiogenic  ${}^{87}$ Sr/ ${}^{86}$ Sr isotope data, which are used to model the temperature and composition of the dolomitizing fluids. The reactivity of potential

dolomitizing fluids is evaluated with geochemical models depending on fluid composition and temperature. Finally, we and compare required versus quantify available fluid budgets to account for the Benicàssim dolostone using mass-balance calculations. These methods and results allow a better understanding of the origin of fluids as well as the evaluation of fluid driving forces, helping in the construction of a conceptual model for high-temperature dolomitization.

#### 2. The Benicàssim case study

#### 2.1. Geologic setting

The Maestrat Basin formed during the Late Jurassic - Early Cretaceous rifting cycle and accumulated up to 4500 m of Early Cretaceous syn-rift sediments at the basin depocentre (Salas and Casas, 1993; Salas et al., 2001)



**Fig. 2.** Geologic map of the Benicàssim area showing the spatial patterns of dolostones and their relationship with large-scale faults. UTM coordinates of upper left corner are 4448000 N - 250753 E. The Desert de Les Palmes area lies west of the map, as indicated in the Figure. Map modified from Gomez-Rivas et al. (2012) and Martín-Martín et al. (2012).

(Fig. 1). The basin was inverted during the Late Eocene - Early Oligocene into the Iberian Chain fold-and-thrust belt (Fig. 1a) (Guimerà et al., 2004). The area was also affected by a Late Oligocene - Miocene extensional period that formed the Valencia Trough and the present-day western Mediterranean basin (Roca and Guimerà, 1992).

The Benicàssim study area is located in the southern part of the Maestrat basin. constituting the eastern margin of the Desert de Les Palmes ranges (Martín-Martín et al., 2013; Fig. 2). In these ranges, the Paleozoic basement and Permian-Triassic rocks are well exposed, as well as a thin Jurassic and Lower Cretaceous cover (Simón-Gómez, 1986a; Roca and Guimerà, 1992; Roca et al., 1994; Salas et al., 2005a). The Desert de Les Palmes is characterized by NE-SW- to NNE-SSWtrending faults, which are the result of reactivation of Hercynian faults during the Early Cretaceous rifting cycle as normal faults (Salas and Guimerà, 1996). Some of these faults created the Benicàssim half graben, with a large accommodation space towards the east (Martín-Martín et al., 2013). A thick pile of Lower Cretaceous sediments (> 2000 m) was deposited in the graben, while the Desert de Les Palmes area (west of Benicàssim) remained a sedimentary high.

Two sets of large-scale faults controlled sediment deposition and later fluid circulation in the Benicàssim area: the Campello (E-Wtrending and S-dipping) and Benicàssim (NNE-SSW-trending and E-dipping) fault systems (Salas, 1987; Roca and Guimerà, 1992; Roca et al., 1994; Gomez-Rivas et al., 2012; Martín-Martín et al., 2013) (Fig. 2). The two fault systems are very likely Hercynian in origin (Roca et al., 1994) and were reactivated during the Mesozoic as normal faults. They subsequently acted as strike-slip and reverse faults during oblique-slip the Tertiary inversion and were reactivated again as normal faults during the Neogene extension episode (Simón-Gómez, 1986a; Gomez-Rivas et al., 2012).



**Fig. 3.** Synthetic lithostratigraphic log for the Benicàssim area. The stratigraphic position of the replacement dolostones in Aptian-Albian rocks of the Benassal Fm are indicated in dark grey.

#### 2.2. Rock units

Several rock units crop out in the Desert de Les Palmes area (Simón-Gómez, 1986b; Roca et al., 1994; Martín-Martín, 2004), and are also present in the Benicàssim half graben, below the Lower Cretaceous carbonates (Figs. 2, 3): (i) Carboniferous shales that constitute the top of the Hercynian basement, (ii) Permian to Lower Triassic continental red beds (Buntsandstein facies), (iii) Middle Triassic and dolostones (Muschelkalk limestones facies), (iv) a very thin unit (few meters) of Upper Triassic continental red beds (Keuper facies), (v) Jurassic partially Lower dolomitized carbonate breccias (Cortes de Fm) unconformably Tajuña that were deposited on top of underlying units, including the basement, and (vi) Middle to Upper Jurassic limestones. It is important to notice that the thicknesses of these units are variable across the field area, with averages given in Figure 3.

The stratigraphy of the Benicàssim area is shown in Figure 3, which is based on data by Simón-Gómez (1986b), Roca et al. (1994), Martín-Martín (2004) and Martín-Martín et al. (2013). The Aptian-to-Albian succession of the area represents one of the thickest carbonate deposits of this age reported from the northern Tethyan margin (Martín-Martín et al., 2013). The Benassal formation was deposited between Late Aptian and Albian times (Moreno-Bedmar et al., 2009; 2010; Garcia et al., 2013; Martín-Martín et al., 2013). Its thickness varies across foot- and hanging-wall blocks of the Benicàssim and Campello faults, reaching a maximum of 1500 m in the Orpesa Range (Fig. 3), in the hanging wall the two large faults. The Benassal Fm ramp-type shallow-marine consists of limestones with orbitolinid foraminifera, rudist bivalves, corals and algae-rich facies (Salas et al., 2005a; Tomás, 2007; Tomás et al., 2007; 2008; Bover-Arnal et al., 2009a; 2009b; Martín-Martín et al., 2010; 2013). The succession is stacked into three transgressiveregressive sequences, the transgressive lithofacies being dominated by basinal marls,

spicule mudstones to wackestones and orbitolinid wackestones, and the regressive lithofacies dominated by coral limestones, peloidal and bioclastic packstones, ooidal grainstones and rudist floatstones-rudstones (Fig. 3). The Benassal Fm is overlain by the Escucha Fm, which consists of ~ 90 m-thick clay with occasional sandy layers (e.g. Rodríguez-López et al., 2007).

#### 2.3. Dolostone distribution

Dolostones in Benicàssim form seismicscale stratabound geobodies that, at present day, extend up to 7 km away from large-scale faults and only affect the two uppermost transgressive-regressive sequences of the Benassal Fm. The largest exposed dolostones are located in the hanging walls of the Campello and Benicàssim faults (Fig. 2). The dolomitized bodies have a thickness of tens of meters, but some of them reach 150 m. Dolostone and limestone layers can be clearly distinguished in the field, due to their characteristic brownish and grey-blue outcrop colour, respectively (Fig. 4). Dolomitization fronts are very sharp and roughly parallel to although they undulate cutting bedding, bedding surfaces from centimetres to a few metres (Fig. 4b). Generally, non-replaced tight micritic facies or early-cemented graindominated facies appear intercalated within the dolomitized rocks (Martín-Martín et al., 2010; 2013). Replacement mostly affects bioclastic wackestones to packstones and peloidal packstones to grainstones (Fig. 5a). The permeability differences between layers associated with the original depositional facies and early diagenesis probably enhanced a strong lateral fluid flow along preferred layers, favouring the stratabound dolostone geometry away from faults, which must have acted as feeding points (Gomez-Rivas et al., 2010a; Martín-Martín et al., 2013; Corbella et al., 2014).

#### 2.4. Dolostone petrography and geochemistry

Dolostones are characterized by the typical burial paragenesis with the following main Gomez-Rivas, Corbella, Martín-Martín, Stafford, Teixell, Bons, Griera and Cardellach. Reactivity of dolomitizing fluids and Mg source evaluation of fault-controlled dolomitization at the Benicàssim outcrop analogue (Maestrat Basin, E Spain). *Marine and Petroleum Geology* (2014), 55, 26-42.

events: calcite and dolomite cementation, host limestone replacement, post-replacement dolomite and calcite cementation, and MVT sulphide mineralization (Martín-Martín et al., 2010; 2013). Early-stage burial calcite and dolomite cementation precedes dolomitization, playing an important role in subsequent dolostone formation and distribution (Fig. 5b). Pre-replacement dolomite cements mainly affect packstones and grainstones of some Dolomitization occurred specific layers. during burial diagenesis after mechanical compaction and fracturing. Stylolites formed mostly before replacement, as well as many fractures of various scales oriented parallel to the two main faults. Most of the Benicàssim

dolostone is a replacive fabric-retentive burial dolomite with low intercrystalline porosity (Fig. 5c). Dolomite crystals are planar-s to nonplanar-a, with a cloudy appearance and polymodal size distribution, a crystal size of 50 to 600 µm and a relatively high Fe content (up to 3%). The replacive dolomite was recrystallized by successive dolomitization fluids, producing a slight increase of crystalsize and intercrystalline porosity (Fig. 5d). after dolomitization Dissolution soon generated abundant porosity in dolostones, which was followed by precipitation of dolomite rim overgrowths (Fig, 5e).



**Fig. 4.** Field images of the Benassal Fm limestone and dolostone. (a) Panoramic view of the Racó del Moro outcrop. Limestone and stratabound dolostone layers have distinct colour. Lateral field of view is  $\sim 4$  km. (b) Zooming in the rectangle of photo (a). (c) Detail of a typical limestone-dolostone front in the area. (d) Dolomitized grainstone preserving the original lamination. (e) Detail of a high-temperature vein: (1) host replacement dolostone and (2) saddle dolomite and (3) high-temperature calcite filling an open vug.

The dolostones of the Benassal Fm locally host Mississippi Valley Type ore deposits (galena and sphalerite) and Fe-oxide mineralizations (Martín-Martín et al., 2013). The mineralizations are approximately located at the intersection between the Benicàssim and Campello faults (Fig. 2), suggesting that this was the main conduit for the upwards circulation of deep fluids. According to Martín-Martín et al. (2010), acidic fluids derived from the emplacement of the MVT mineralization increased dolostone porosity by dissolution. The MVT deposits are associated with a fractured and brecciated karst system and are predated by saddle dolomite and calcite cements filling vugs and fractures (Fig. 5f). Different sets of Alpine and Neogene fractures overprint dolomitization (GomezRivas et al., 2012). Meteoric calcite cements precipitated from surface-derived fluids (Fig. 5e) and produced partial calcitization of the dolostones during uplift and subaerial exposure (Martín-Martín et al. 2013).

Dolomitization at Benicàssim was caused by high-temperature fluids (Martín-Martín et al., 2010; Gomez-Rivas et al., 2010b), as indicated by: (1) progressive  $\delta^{18}$ O decrease with dolomitization (e.g. Allan and Wiggins, 1993), (2) dominant non-planar dolomite textures (e.g. Gregg and Sibley, 1984), (3) sharp and irregular dolomitization fronts (e.g. Machel, 2004), (4) occurrence of saddle dolomite synchronous with replacement dolomite (e.g. Warren, 2000) and (5) close association with MVT deposits and high-temperature calcite cements (Gregg, 2004).



**Fig. 5.** Photomicrographs of the Benassal Fm limestone and dolostone. (a) wackestone with orbitolina and miliolids, a calcite vein cross-cuts the sample (PPL); (b) Coarse grain orbitolinid grainstone showing calcite (orange) and dolomite (grey) cement completely filling primary intergranular porosiy. Stained with Alizarin Red-S (PPL); (c) Replacive dolomite 1 showing the complete replacement of a fine grained peloidal grainstone (mimetic texture). Note the concavo-convex contacts between the original components, which suggest replacement after compaction, and the scarce intercrystalline porosity (light blue). Diffused light; (d) Replacive dolomite 2 showing abundant intercrystaline porosity. Note cloudy centres and clear borders in dolomite crystals (PPL); (e) replacive dolomite (1), dolomite cement rim overgrowth (2) and meteoric calcite cement (3) filling the remaining intracrystalline porosity (PPL); (f) saddle dolomite (1) postdated by ore-stage calcite cement (2).

Characterization of the fluid composition is limited by the poor quality of fluid inclusions in the Benicàssim dolomite crystals. Nevertheless, Nadal (2001), Grandia (2001) Grandia et al. (2003) reported and homogenization temperatures between 85 and 153 °C and melting ice temperatures of -23° to -14 °C from dolomite replacing Early Cretaceous limestones in different parts of the Maestrat basin (see Fig. 1 for location of these dolostones). According to halogen ratio data, fluid dolomitizing was evaporated the seawater that interacted with K-rich minerals, as pointed out by the high salinity and Na depletion (Grandia, 2001), probably K-feldspar from the underlying basement rocks. This hypothesis is supported by the radiogenic <sup>87</sup>Sr/<sup>86</sup>Sr signature of dolostones. Mixing of two fluids was suggested as the ore-stage minerals precipitation mechanism: a high-salinity brine at temperatures higher than 120°C and a low-salinity fluid at 40°C (Grandia, 2001; Grandia et al., 2003).

#### 2.5. Dolomitization conditions

Dolomitization in Benicàssim took place at some time between the end of the Benassal Fm

deposition (Albian) and the onset of MVT mineralization during the Paleocene, dated at 62.6±0.7 Ma using U-Pb dating of cogenetic calcite and galena (Grandia et al., 2000). More precisely, field and petrographic observations as well as geochemical data point towards a burial replacement after strong compaction and early diagenesis (see section 2.4). Thus, replacement probably occurred towards the end of the Cretaceous or in the very early Paleocene. Quantitative subsidence curves indicate that the burial depth in this time span ranged between 200 and 1000 m for the upper dolomitized Benassal carbonates (Martín-Martín et al. 2010; 2013). These depths match paleodepth values estimated by Kulzer (2011) from sedimentary (i.e. beddingparallel) stylolites. This author used the methods of Koehn et al. (2007; 2012) and (2010)Ebner et al. to analyze the morphological features of bedding-parallel stylolites and quantify the stress, compaction conditions and depth at which they formed.

Permanyer et al. (2000) calculated a geothermal gradient of 27 °C/km for the Late Jurassic to Early Cretaceous rifting period and 30 °C/km for the Late Albian of to Maastrichtian post-rifting stage in the Maestrat basin. More recently, Caja et al. (2009) estimated a geothermal gradient of 30 to 35°C/km for Early Cretaceous times. Therefore, host rock temperatures of the Benicàssim Aptian-Albian carbonates during dolomitization should have been less than 60 °C, if we only consider conductive heat transfer associated with regional the geothermal gradient. However, as mentioned (see section 2.4), the above existing geochemical data indicate that dolomitizing fluids in this area were warmer than that, up to at least 110°C.

Dolomitization reactive transport models applied to the Benicàssim case study showed that hundreds of thousands to millions of years were necessary to completely dolomitize kilometre-long limestone sections in the area (Corbella et al., 2014), assuming flow velocities on the order of meters per year. Such velocities are commonly assumed in sedimentary basins (e.g. Garven et al., 1993). Seawater-derived solutions flowing laterally through strata at such velocities can account for the stratabound dolomitization, although fluid velocity differences of at least two orders of magnitude were required to preferentially replace only some of the layers. This differential dolomitization along strata could be obliterated by a long-term dolomitizing fluid flow producing massive dolomites. Fluid and heat flow numerical simulations of the Benicàssim area (Gomez-Rivas et al., 2010a) indicate that a long-term fluid circulation mechanism, due to differences in pressure and temperature within the basin, could pump enough dolomitizing fluids at high temperature during long periods of time. On the contrary, the models suggest a pulsating flow system could not provide enough heat for the Benicàssim dolomitization, as fluids cool down quickly when flowing upwards along faults.

## 3. Methods

Two groups of methods have been used: isotope analysis and quantitative modelling. The first group encompasses the acquisition of stable C-O and radiogenic Sr isotopic data, and subsequent calculation of the isotope fractionation trajectories and estimation of dolomite formation temperatures. The second group includes modelling of the reactivity of possible dolomitizing fluids, the quantification of required fluid volumes to account for the Benicàssim dolomitization, and its subsequent comparison with those that could be produced from different sources.

## 3.1. Stable and radiogenic isotopes

46 samples of the host Aptian-Albian limestone, replacive dolomite, saddle dolomite as well as burial/ore-stage and meteoric calcite were selected for C and O isotopic analysis following the method of McCrea (1950). CO<sub>2</sub> was extracted by reaction of the carbonate sample with phosphoric acid at 50°C. The collected gas was analysed in a Finnigan MAT Delta S thermal ionization mass spectrometer at the Technical-scientific Laboratories of the University of Barcelona. The standard deviation of measurements was 0.03 for  $\delta^{13}$ C (‰ PDB) and 0.06 for  $\delta^{18}$ O (‰ SMOW). fractionation trajectories Isotope were calculated following the equations of Zheng and Hoefs (1993). The fractionation equations of water-calcite (O'Neil et al., 1969), H<sub>2</sub>CO<sub>3</sub>calcite (Chacko et al., 1991), water-dolomite (Zheng, 1999) and H<sub>2</sub>CO<sub>3</sub>-dolomite (Zheng, 1999) were employed.

Radiogenic Sr isotopes (<sup>87</sup>Sr/<sup>86</sup>Sr) were analyzed on 7 samples: host-rock limestone, replacive dolomite, saddle dolomite, burial/ore-stage calcite, and a meteoric calcite related to uplift and subaerial exposure. Carbonates were dissolved in 2.5N HCl at 80 °C, dried, and redissolved using 2.5N HCl. Sr aliquots were obtained using cation exchange resins, and analyzed in a Thermal VG SECTOR mass The 54 spectrometer. reproducibility of the results  $(2\sigma)$  was always better than  $\pm$  0.00004 %. Samples were the laboratories processed at of the Universidad Complutense de Madrid. The results were controlled by repetitive analysis of the NBS-987 standard, averaging 0.710260  $\pm 0.00004 (2\sigma, n=7).$ 

## 3.2. Reactivity of dolomitizing fluids

Geochemical simulations were carried out to evaluate the dolomitization capacity of different fluids at variable temperatures. The method is based on the concept of pore volume (Land, 1985), assuming that dolomitizing fluids would quickly equilibrate with the host limestone in a close system, because of the fast reaction kinetics of carbonate minerals (e.g. Sjöberg, 1976; Plummer et al. 1978), losing its dolomitization capacity. A pore volume of fluid is defined as the volume of fluid that saturates the rock's porosity. The volume of dissolved calcite in the rock and precipitated dolomite from that fluid filling the porosity is then calculated by bringing the system to equilibrium at a given temperature. The number of pore volumes calculated to

completely dolomitize the rock would be the minimum volume of fluid required, assuming 100% reaction efficiency and mole-by-mole replacement. The calculation is performed in three steps. First, original fluid chemical compositions are recalculated by equilibrating them with calcite, considering the fluid has already flowed through limestones. After that, moles of calcite and dolomite the dissolved/precipitated from the fluid from/to a certain volume of rock is quantified by bringing it to equilibrium with both minerals at a given temperature. Finally, the fluid pore volumes required to replace all calcite by dolomite at the given volume is estimated. Pore volume values can be then converted to fluid/rock ratios required to completely dolomitize a certain rock volume. The advantage of giving the results in terms of fluid/rock ratios instead of pore volumes is that they are porosity-independent.

The calculations were performed using two different approaches and two codes, to compare the obtained results. For the first approach, we used the computer code PHREEQC (Parkhurst and Appello, 1999), with the thermodynamic constants of the EQ3NR database (Wolery, 1992). For the second approach, we used the reactive RETRASO transport code (REactive TRAnsport of SOlutes; Saaltink et al. 1998). In this case, we created 1 m<sup>3</sup> models (closed systems) with certain porosity filled by the considered fluids. The moles of dissolved/precipitated calcite/dolomite were calculated by bringing the fluid to equilibrium with the host rock. RETRASO incorporates aqueous complexation, adsorption and mineral precipitation/dissolution. We ran separate sets of simulations: (1) applying thermodynamic laws, considering local equilibrium, and (2) using kinetic rates. The thermodynamic constants in RETRASO are those of the EQ3NR database (Wolery, 1992). We used dissolution/precipitation kinetic constants of  $4.64 \cdot 10^{-7}$  mol m<sup>-2</sup> s<sup>-1</sup> for calcite (Inskeep and Bloom, 1986) and  $2.20 \cdot 10^{-8}$  mol m<sup>-2</sup> s<sup>-1</sup> for dolomite (Ayora et al., 1998). The activity coefficients of aqueous species were computed with the B-dot form of the extended Debye-Hückel equation of Helgeson & Kirkham (1974), which allows the extension of the applicability to higher temperature and salinity fluids.

## 3.3. Released fluid budget calculations

The volume of fluid required to dolomitize the host rock can be compared with the volume of available fluids that could have been sourced under three different scenarios. Similar calculations, applied to ore deposits, were presented by Staude et al. (2009) and Weisheit et al. (2013). We used the model of Staude et al. (2009) to quantify fluid budgets produced by (a) release of overpressured fluids due to decompression, and (b) release of fluids due to heating.

In the first case, overpressure of the pore fluid arises from the different compressibility of fluid and solid rock. When rocks decompress, their pore volume increase is negligible. Pore-fluid pressure thus remains effectively unchanged. If pore fluids were in equilibrium before decompression, they will be out of equilibrium afterwards. Part of the fluid (the "excess volume" of Staude et al., 2009) will be released to regain equilibrium. Decompression can result from exhumation (reduction of overburden height, and hence pressure) and/or an extensional stress field that reduces the average stress, and hence pressure.

In the second case (b), an excess volume is caused by the higher thermal expansion of fluids than of the solid rock, so that heating may also result in pore fluid overpressure, which leads to release of fluids to regain equilibrium. Heating can be caused by a change in the geothermal gradient (e.g. by thinning of the crust) or by a deep heat source such as an igneous intrusion. Decompression by exhumation is instantaneous, whereas thermal equilibration is a slow process. Crustal thinning can first lead to fluid release by decompression, followed by release due to heating.

A third potential cause for fluid release is the breaking of a seal, which can lead to fluid release in order to reduce the pore-fluid pressure to regain equilibrium, if fluids are overpressured. The upward flow of the excess fraction of fluid (i.e. the volume of pore fluid that is expelled to regain equilibrium) can take porosity by porous flow if place is interconnected. Alternatively, when porosity is not connected, release of trapped fluids can and occur by the onset linkage of hydrofractures (e.g. Secor and Pollard, 1975; Bons and Van Milligen, 2001; Staude et al., 2009). This scenario assumes overpressured fluids.

The bulk modulus difference makes aqueous fluids approximately ten times more compressible than solid rocks. In order to regain equilibrium under the new conditions the pore fluid volume has to be reduced. Moreover, if a fluid in equilibrium is heated, its volume also increases due to thermal expansion. Fluid volume increase in both cases (decompression and thermal expansion) can be determined with equation 1 for small changes in pressure and temperature (Staude et al., 2009) as,

$$\Delta V = \frac{V}{V_0} = \exp^{\alpha \Delta T - \frac{\Delta P}{K}}$$
(1)

where V is the volume of pore fluid,  $V_0$  is a reference volume (m<sup>3</sup>),  $\alpha$  is the thermal expansion of the fluid (K<sup>-1</sup>), K represents compressibility expressed by its bulk modulus (Pa) and  $\alpha$  is fluid thermal expansion (K<sup>-1</sup>). As compressibility and thermal expansion of the solid rock are negligible when compared with that of water, Eq. (1) does not include those terms.

In order to calculate expelled fluid volumes according to the different scenarios, we have estimated the thickness of the different units when dolomitization took place. For that purpose, we have used the methods of Sclater and Christie (1980) and Schmoker and Halley (1982) to determine decompaction using density values and porosity/depth relationships (c factors) for each lithologic type. The age

and thickness of the different units have been estimated from field data (Bover-Arnal et al., 2009a; Martín-Martín et al., 2013) and regional geology (Obis et al., 1973; Salas and Casas, 1993; Salas and Guimerà, 1996; Salas et al., 2001; Roca et al., 1994; Martín-Martín, 2004). A bulk modulus of 2 GPa has been Water assumed. thermal expansion estimated coefficients  $(\alpha)$ have been depending on fluid pressure  $(P_f)$ and temperature (T) using a linear function assuming a value of  $8.8 \cdot 10^{-5}$  K<sup>-1</sup> at  $P_{f}=0.1$  Mpa and  $T=10^{\circ}C$  (Lide, 2010) and a value of 2.4  $10^{-4}$  K<sup>-1</sup> at  $P_f=25$  Mpa and  $T=300^{\circ}$ C (Tester and Cline, 1999).

## 4. Results and discussion

The combination of the aforementioned techniques allows a multidisciplinary study of the dolomitizing fluid origin, its reactivity, the conditions under which the reaction takes place, and eventually, the comparison between fluid budgets required to dolomitize the Benicàssim rocks and the ones available from possible sources. The final outcome of the study is a conceptual model for the Benicàssim dolostones, with a quantitative evaluation of its plausibility.

# 4.1. Temperature and composition of the Benicàssim dolomitizing fluid

The isotopic composition of the original Late Aptian - Early Albian limestone is  $\delta^{13}C=$  +6.2 ‰ V-PDB and  $\delta^{18}O=+28.3$  ‰ V-SMOW, inferred from the analysis of an unaltered rudist shell. However, other host limestone samples show  $\delta^{13}C$  values from -0.4 to +2.7 ‰ and  $\delta^{18}O$  from +23.6 to +26.0 ‰, Fig. 6, Table 1). These values are clearly shifted with respect to the Aptian-Albian seawater carbonates, suggesting an isotopic modification by diagenetic fluids.

Replacive and saddle dolomite isotope values are very similar, ranging between +0.5 and +2.9 ‰ for  $\delta^{13}$ C and between +21.1 and +24.3 for  $\delta^{18}$ O. They are also similar to the values of some of the altered limestone

samples (Fig. 6; Table 1). The two types of calcite veins are also distinguished according to their stable isotope composition. Ore-stage calcite veins associated with the MVT mineralization and burial calcite cements have  $\delta^{13}$ C values of -3.1 to +0.1 ‰ and  $\delta^{18}$ O of +13.2 to +19.2, and meteoric calcite veins and cements show values ranging from -10.7 to -4.8 % for  $\delta^{13}$ C and from +22.9 to +25.4 for  $\delta^{18}$ O. Calcitized dolomite (dedolomite) isotope signature is similar to meteoric calcite, with values between -2.3 to -10.4 ‰ for  $\delta^{13}$ C and +23.2 to +25.5 for  $\delta^{18}$ O (Fig. 6, Table 1). The calcitized dolomite and meteoric calcite vein isotope values can be explained after the precipitation from a fluid that isotopically exchanged O and C with Aptian-Albian limestone at low temperatures (typically < 50°C), following the meteoric calcite line (Lohmann, 1988).

As temperature is the major parameter controlling oxygen isotope fractionation between fluids and carbonate minerals, the difference in  $\delta^{18}$ O between calcite from limestone, replacive and saddle dolomite and high-temperature calcite veins can be explained by water-rock interaction of a warm fluid with carbonates (Fig. 6; Table 2). With the modelled equations, the presented C and O isotope data are compatible with the precipitation of dolomite at 80-110°C from a fluid isotopically equilibrated with the regional limestone at different fluid to rock ratios as illustrated in Figure 6 (dashed curves). Similarly, C and O isotope composition of calcite veins and cements is constrained by model curves between 80 and 140°C (solid lines in Fig. 6). Therefore, in a  $\delta^{13}C/\delta^{18}O$  plot. data are compatible with isotope а dolomitization process caused by warm fluids interacting with the limestone host rock. The estimated temperatures are similar to those obtained from fluid inclusions in dolostones of the same age and type in different areas of the Maestrat basin (85 - 153 °C; Nadal, 2001; Grandia, 2001; Grandia et al., 2003) and also agree with isotope temperatures (100 - 120 °C) estimated by Grandia (2001) from sphaleritegalena pairs. According to our model, the dolomite isotopic data in Figure 6 can be explained from the interaction of a single fluid with the limestone host rock at a temperature range. Alternatively, the C/O isotope data could be explained through fluid mixing, as invoked for the genesis of MVT mineralization (Corbella et al. 2004). If this was the case, mixing should occur at fault zones as a consequence of the different detachment levels of the sets of active large-scale faults. However, the lack of additional evidence (fluid inclusion and crush leach data) means that we cannot corroborate this hypothesis.



**Fig. 6.**  $\delta^{13}$ C (‰ PDB) vs  $\delta^{18}$ O (‰ SMOW) plot of limestones, dolomite, calcite/dolomite cements and calcite/ankerite veins from the Benicàssim outcrops. C-O isotope model curves have been calculated in terms of fluid-rock interaction for dolomite (dashed lines) and calcite (solid lines) at different temperatures and fluid/rock ratios according to the method of Zheng and Hoefs (1993). Input data for fluid-rock exchange fractionation trajectories is shown in Table 1. The new data presented in this contribution is compared to that presented in Martín-Martín et al. (2010) and Corbella et al. (2014).

The obtained <sup>87</sup>Sr/<sup>86</sup>Sr value for host-rock limestone matches the range of late Aptian to early Albian marine carbonates (Jones and Jenkyns, 2001; Fig. 7). However, all other carbonate phases are more radiogenic, indicating the presence of a <sup>87</sup>Sr-enriched fluid during precipitation of replacive dolomite, saddle dolomite as well as burial calcite cements and veins. An <sup>87</sup>Sr enrichment of hydrothermal fluids can be explained after the

interaction with Rb-bearing minerals, such as clays, K-feldspar or K-mica bearing rocks. These minerals are present in the underlying Permian-Triassic and Paleozoic basement rocks, suggesting that fluids circulated through and interacted with deeper rocks before precipitating the hydrothermal carbonates. This hypothesis is in agreement with the interpretation of Sr-isotope data of replacement dolostones of the same age

Sample	Type of sample	$\delta^{13}C$ (PDB)	δ <sup>18</sup> O (SMOW)	
Locality:	Mortorum Hill			
1	Limestone (grainstone, bulk rock)	+1.7	+25.7	
2	Limestone (packstone, bulk rock)	-0.4	+26.0	
3	Limestone (packstone, bulk rock)	+0.7	+23.6	
4	Limestone (packstone, bulk rock)	+1.4	+25.4	
5	Limestone (wackestone, bulk rock)	+2.7	+25.8	
6	Replacive dolomite (bulk rock)	+0.5	+24.3	
7	Replacive dolomite (bulk rock)	+1.8	+24.2	
8	Replacive dolomite (bulk rock)	+1.9	+23.2	
9	Calcitized dolomite	-2.3	+23.2	
10	Calcitized dolomite	-10.4	+25.5	
11	Calcitized dolomite	-7.2	+25.2	
12	Calcitized dolomite	-5.7	+24.2	
13	Calcitized dolomite	-5.7	+24.1	
14	Calcitized dolomite	-2.4	+24.4	
15	Calcitized dolomite	-4.3	+23.5	
16	Calcitized dolomite	-2.6	+24.4	
17	Burial calcite cement	-5.1	+17.8	
18	Meteoric calcite vein	-6.3	+24.9	
19	Meteoric calcite vein	-10.4	+24.8	
20	Meteoric calcite vein	-10.7	+25.4	
21	Meteoric calcite vein	-4.8	+22.9	
Locality:	Racó del Moro			
22	Rudist shell in limestone (packstone)	+6.2	+28.3	
23	Replacive dolomite (bulk rock)	+1.7	+22.4	
24	Replacive dolomite (bulk rock)	+1.7	+22.9	
25	Replacive dolomite (bulk rock)	+2.9	+21.1	
26	Replacive dolomite (bulk rock)	+1.3	+22.9	
27	Replacive dolomite (bulk rock)	+1.4	+22.6	
28	Replacive dolomite (bulk rock)	+2.5	+22.6	
29	Replacive dolomite (bulk rock)	+2.9	+23.1	
30	Replacive dolomite (bulk rock)	+2.9	+22.8	
31	Saddle dolomite in moldic porosity	+4.5	+21.9	
32	Saddle dolomite in fracture	+3.7	+21.8	
33	Ore-stage calcite vein	+0.4	+22.3	
34	Ore-stage calcite vein	+3.2	+20.8	
35	Ore-stage calcite vein	+3.2	+22.1	
36	Burial calcite cement	-2.0	+19.0	
37	Burial calcite cement	-1.3	+19.7	
Locality:	Campello mine			
38	Saddle dolomite	+1.6	+20.4	
39	Saddle dolomite	+0.4	+19.3	
40	Ore-stage calcite vein	-0.8	+18.4	
41	Ore-stage calcite vein	+0.1	+19.2	
42	Ore-stage calcite vein	-0.4	+16.7	
43	Ore-stage calcite vein	+0.0	+16.9	
44	Ore-stage calcite vein	-3.1	+13.2	
45	Meteoric calcite-ankerite vein	-7.1	+24.6	
Locality:	Racó de Rita	<b>_</b> .		
46	Ore-stage calcite vein	+2.6	+17.3	

**Table 1.** Stable isotope data of limestone, dolomite, calcite/dolomite cements and calcite/ankerite veins. See Fig. 2 for outcrop locations.

elsewhere in the Maestrat basin discussed by Grandia (2001), and points to a similar source for the dolomitizing fluids.

The high content of Fe in replacement and saddle dolomites (Grandia, 2001; Martín-Martín et al., 2010), which was probably derived from the underlying Permian-Triassic red clays and conglomerates, may be a further indication of circulation of deep fluids along faults. Moreover, these red beds are laterally connected with the Early Cretaceous succession by the Benicàssim fault.

According to the aforementioned data, dolomitizing fluids were in thermal disequilibrium with the country rocks; therefore, fluids must have acquired their temperature at depths considerably greater than the host rock burial depth. Depths on the order of 1 km are common for the emplacement of MVT ore deposits, as is the circulation of mineralizing fluids warmer than host rocks (Grandia et al., 2003). Flow of warm fluids in the Benicàssim area during Late Cretaceous times was probably related to one of the following reasons or to both of them: (1) the existence of an igneous intrusion Chain Iberian below the or (2)the heterogeneous thinning of the Earth's crust below Iberia as a consequence of Late Permian Triassic and Late Jurassic to Early to Cretaceous rifting cycles (Salas and Casas, 1993; Salas et al., 2001; Martín-Chivelet et al., 2002). Salas et al. (2005b) proposed the existence of a hot spot that developed in relation to the opening of the oceanic North Atlantic and Bay of Biscay basins. As a consequence, an alkali basaltic volcanism would have developed in the Basque-Cantabrian and Central Pyrenean areas and a deep basic pluton would have intruded below Iberian Chain, producing the thermal anomalies during the Late Cretaceous.

Thermal anomalies were common in the southern part of the Maestrat Basin during the Late Cretaceous (Salas et al., 2005b). It has been also suggested that a small rifting episode took place at the end of the Cretaceous and Early Paleocene, during the deposition of the Garumnian facies (Meléndez et al., 1985), although this hypothesis has not been confirmed. If dolomitization occurred at this stage, just before the onset of MVT deposits (dated at about 62.5 Ma by Grandia et al., 2000), deep fluids could have been released as a consequence of fault reactivation.



**Fig. 7.**  ${}^{87}$ Sr/ ${}^{86}$ Sr isotope ratios vs.  $\delta^{18}$ O (‰ SMOW) values measured from dolomite and calcite samples at Benicàssim.

Table 2. Input data for fluid-rock	exchange models presented	in Fig.	6.
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1		0	1	0	
Mineral	$ \delta^{13}C_{fluid} (\% PDB) $	$\frac{\delta^{13}C_{rock}}{(\% PDB)}$	$\delta^{18}O_{fluid}$ (‰ SMOW)	$\delta^{18}O_{rock}$ (‰ SMOW)	$X_{\Sigma C}$
Dolomite	-10.0	+6.2	+3.0	+28.2	0.05 (H <sub>2</sub> CO <sub>3</sub> )
Calcite	-10.0	+6.2	+3.0	+28.2	0.05 (H <sub>2</sub> CO <sub>3</sub> )

## 4.2. Magnesium sources and requirements for the Benicàssim dolomitization

Based on mapping (Fig. 2) and outcrop observations we have estimated that the total volume of dolomitized rock in the ~ $60 \text{ km}^2$ Benicàssim area was on the order of ~ $4 \cdot 10^9$ m<sup>3</sup>, including rocks eroded during the Cenozoic uplift. This volume is about ~ $6 \cdot 10^8$   $m^3$  without considering the eroded material. The three mapped tabular bodies of dolostone that are visible throughout the field area, from the sedimentary high of the Dessert de Les Palmes area in the South, to the Campello fault in the North, were considered. A constant thickness and a composition of 90% dolomite and 1% porosity have been assumed. In order to account for the eroded dolostone we have **Table 3.** Summary of the different Mg sources considered in this study (see text for further details). The total amount of Mg required to dolomitize the Benicassim Aptian-Albian limestones is  $\sim 5 \cdot 10^{13}$  mol.

Possible Mg sources	Volume of of fluid (m <sup>3</sup> )	Mg concentration (mol/cm <sup>3</sup> )	Amount of Mg available for dolomitization (mol)	Viable method (yes/no?)
(a) fluids that calcitized Triassic and/or Jurassic dolomites	depends on concentration	?	1.25.10 <sup>12</sup> (assuming 30% of rock was dedolomitized)	Not alone, but could have contributed if calcitization of Triassic-Jurassic rocks was synchronous with Late Cretaceous dolomitization
(b) brines that dissolved Jurassic and/or Triassic evaporites	neglectable	-	neglectable	No. Evaporite levels are very thin, and they do not contain Mg-rich evaporites
(c) Mg-rich fluids derived from illitization of Permian-Triassic clays	neglectable	-	neglectable	No. Permian-Triassic layers only have very small amounts of Mg-bearing
clays				
(d) fluids derived from dewatering of Cretaceous marls and limestones	2.85·10 <sup>10</sup>	3.64·10 <sup>-5</sup>	$1.03 \cdot 10^{12}$	Not alone, but could have contributed if dolomite formed during compaction. Does not match dolomite temperature and formation time
(e) deep brines (unknown origin) coming from Palaeozoic basement	unknown	?	?	Not alone, but could have contributed if fluid mixing occurred
(f) seawater (pristine or concentrated)	unknown	3.64·10 <sup>-5</sup> (pristine)	depends on fluid volume	Yes. Fluid volume must be quantified (see section 5.4)

The total amount of Mg required to dolomitize the Benicàssim Aptian-Albian limestones is  $\sim 5.1 \cdot 10^{13}$  mol

projected the tabular bodies on cross-sections and measured the resulting volumes. Considering stoichiometric dolomite and a mole-by-mole replacement of calcite, at least  $\sim 5 \cdot 10^{13}$ moles of Mg would have been required to account for the partial dolomitization of the Benicàssim carbonate ramp. This amount would be about  $\sim 10^{13}$ moles of Mg if we do not take into account the dolostone rock volumes that have been eroded as a consequence of post-dolomitization largescale fault movements and uplift.

Taking into account the rocks present in the Benicàssim and surrounding areas. the magnesium required to cause the described dolomitization could have been derived from the following fluids (Table 3): (a) fluids that calcitized the underlying Triassic and/or Jurassic dolostones, (b) brines that dissolved underlying Upper Triassic and Lower Jurassic evaporites, (c) Mg-rich fluids derived from the alteration of Permian-Triassic clay minerals, (d) fluids from dewatering of Cretaceous marls and limestones due to compaction, (e) deep brines of unknown origin that had interacted with Paleozoic basement rocks and (f)seawater, either pristine, concentrated, or evolved (by interaction with other sources).

Source (a) (fluids that calcitized Triassic and/or Jurassic dolostones) is ruled out because Triassic and Jurassic dolostones from the adjacent Desert de Les Palmes area have not been extensively calcitized. Based on thin section analysis, we have estimated that these dolostones have about 30% calcite volume at present time. Moreover, the thickness of these units is very small compared to that of the Aptian-Albian dolostones (Fig. 3). The calculations show that the amount of Mg that could be obtained from this source is between one and two orders of magnitude less than required (Table 3).

Source (b) (brines that dissolved underlying Upper Triassic evaporites) is not plausible because Triassic evaporites (Keuper facies) were mostly eroded in the area before the deposition of the Aptian-Albian carbonates. The Keuper facies unit in the Desert de Les Palmes area is only a few meters thick and probably had a maximum thickness of 50 m during Aptian-Albian times (Roca et al., 1994). Moreover, the only evaporite mineral containing Mg in the Triassic-Jurassic of the Maestrat basin is polyhalite (Ortí and Salvany, 1990). The fluid volumes that could have been produced from these sources are negligible compared to the required ones to account for the Benicàssim dolomitization.

The illitization of smectite (source c) can release magnesium (Schrijver et al., 1996). However, very small amounts of Mg-bearing clays are reported from the Permian-Triassic in the area (Martín-Martín, 2004; Martín-Martín et al., 2006), so that there could be no massive Mg release.

Dewatering of marls and limestones by compaction (source d) would probably liberate seawater. either pristine or evolved/concentrated, trapped within the pore system at the time of sedimentation. This process has been claimed for the dolomitization of Upper Jurassic rocks in the Swabian Alb, south-western Germany (Reinhold, 1998). In a similar way, we have estimated the fluid volume that could be expelled from Early Cretaceous rocks due to compaction (Table 3), after calculating the original pore space volume before and after burial. For this purpose, we have used the values proposed by Sclater and Christie (1980) original carbonate porosity for and the porosity-depth equation of Schmoker and Halley (1982) for porosity evolution after burial. Assuming that the porosity volume loss corresponds to the expelled fluid volume, the amount of Mg that could be obtained from this source is at least 50 times less than the volume required to form the estimated total volume of dolomite. Moreover, if such fluids had dolomitization potential, they would have reacted with the host limestones during their time, long residence losing their dolomitization capacity.



**Fig. 8.** (a) Sketch illustrating the concept of pore volume. A fluid with certain composition ([x]) and at a certain temperature (T) is inserted in the pore space. After reaction with n fluid pore volumes the rock is completely dolomitized; (b) quantification of fluid/rock ratios that are required to dolomitize a volume of rock depending on composition and temperature. The dolomitization temperature range in the area was ~80 - 150°C.

Table 4. Chemical composition of the solutions used for geochemical modelling. Concentration units are in mol/kg
water. References: (1) Stumm and Morgan (1981); (3) Shanks and Bischoff (1997) and (4) Kharaka and Thordsen
(1992). All these fluids were equilibrated with calcite prior to calculations.

	(1) Present-day	(2) 5 times-	(3) 7 times-	(4) Brine A	(5) Brine B
	seawater	concentrated seawater	concentrated seawater		
Ca	0.0103	0.0516	0.0723	0.1285	1.1627
Mg	0.0531	0.2654	0.3719	0.0314	0.1255
Na	0.4680	2.3390	3.2770	4.0279	2.6838
Cl	0.5460	2.7290	3.8240	4.4010	5.5849
% wt eq NaC	Cl 3.5	17.5	24.5	25	32

Source (e) (deep brines of unknown origin that had interacted with Paleozoic basement rocks) is impossible to quantify with the current geologic knowledge of the area. On the other hand, seawater-derived fluids (source f) must have been available at any time, given that the depo-setting was mostly marine during the Cretaceous. Hence, the geochemical and geological data from Benicàssim and adjacent areas suggest that the possible sources of Mg pristine/evaporated/evolved are seawater (source f), basement brines (source e), which can also be seawater-derived, or a mixture of both. Dewatering of carbonates by compaction (source d) could have been an extra source of Mg or for local dolomitization (Table 3). The possible contribution of fluids derived from such sources is evaluated in detail in the following section.

## 4.3. Reactivity of potential dolomitizing fluids

The dolomitizing capacity of five types of fluids that could have sourced sufficient Mg (as explained above) has been compared at a temperature range between 50°C and 150°C, which is a range larger than suggested by the geochemical data. These fluids are (Table 4): 1) present-day seawater, 2) a fluid that is 5 times-concentrated seawater, 3) a fluid that is 7 times-concentrated seawater, 4) brine A (Shanks and Bischoff, 1997) and 5) brine B (Kharaka and Thordsen, 1992). Late Cretaceous seawater had a lower Mg/Ca ratio than modern seawater (e.g. Wallmann, 2001). However, we prefer to perform our estimations with present-day seawater. because its chemical behaviour is well known. If Late Cretaceous seawater was used instead, the volume of fluid required to completely dolomitize a certain rock volume would be higher. Brines A and B were selected from the literature as representative of fluids that interacted with carbonates and siliciclastic at similar burial depths rocks to the Benicàssim rocks. These brines contain relatively high Mg concentrations, so they can be considered to have a high dolomitization capacity.

The calculations indicate that the minimum fluid/rock ratio needed to dolomitize the Benicàssim limestone would vary between ~46 and ~61, depending on the selected method. These would correspond to 7 timesconcentrated seawater at 50 to 100 °C (Fig. 8, Table 4). The worst-case scenario would require a fluid/rock ratio of ~900, which corresponds to brine A at 50 °C (Fig. 8, Table 4). It is worth noting that the dolomitizing capacity of these solutions (Fig. 8b) is directly proportional neither to the Mg concentration nor to the salinity of the fluid, even though the concentrated present-day seawater, with twice more Mg than brine B and eight times more Mg than brine A, has a higher dolomitizing capacity.

The optimum temperature for dolomitization fewer pore volumes (i.e. required) ranges between 75 and 125 °C (Fig. 8b-d). The sensitivity of fluid reactivity to temperature depends on fluid composition. In fact, fluids that are capable of dolomitizing at a certain temperature can calcitize when they are cooler (Escorcia et al., 2013). This implies that one single fluid in a basin could be responsible for both dolomitization when warm and dedolomitization when cooled to surface temperature.

## 4.4. Fluid volume constraints

Following the method of Staude et al. (2009) the estimation of fluid release for dolomitization was performed in three different end-member scenarios (Fig. 9):

- interstitial fluids trapped within Late Jurassic to Early Cretaceous rocks, overpressured below the uppermost marly unit at the base of the Benassal Fm (Fig. 4). For simplicity, these fluids were considered under lithostatic pressure and were decompressed as a consequence of seal breaking (Fig. 9b).
- (2) interstitial fluids trapped within all rocks down to 10 km depth, overpressured below the uppermost

marly unit at the base of the Benassal Fm. These fluids were at lithostatic pressure, and were decompressed as a consequence of seal breaking (Fig. 9c).

(3) interstitial fluids trapped within all rocks down to 10 km depth, strongly heated due to an hypothetic anomalous geothermal gradient  $(50^{\circ}C/km)$ . Gradients higher than that have been reported from several parts of the Iberian Chain during the Late Cretaceous (Salas et al., 2005b). Fluid pressure was considered to be hydrostatic, with connected porosity (Fig. 9d).

In the first two cases, the seal breaking would release fluid rapidly, in the form of a fluid burst as a consequence of fast fluid decompression due to the bulk modulus differences between fluids and host rocks. The fluids buried and compressed at the end of Early Cretaceous rifting cycle could be released during the Late Cretaceous post-rift cycle due to seal breaking as a consequence of isostatic rebound or exhumation. Fluid release could also occur due to fault reactivation during a small rift cycle during the Cretaceous - Paleocene transition (contemporaneously with the deposition of the Garumnian facies). In case (3) a high geothermal gradient, caused by either a thin Earth's crust or an igneous intrusion (Salas et al. 2005b), would warm up fluids and host rocks. In such a case, fluid release would be slower and extend over a longer period due to the fluid thermal expansion during heating. High-temperatures are reported in the whole Maestrat basin during the Late Cretaceous post-rift cycle (Tritlla, 1994; Tritlla and Cardellach, 1997; Tritlla and Solé, 1999; Tritlla et al., 2001; Grandia et al., 2000; 2003; Martín-Martín, 2004; Martín-Martín et al., 2013), so case 3 is a very plausible scenario.

The total fluid volume that could have been expelled due to a fast decompression would have been about  $6 \cdot 10^8 \text{ m}^3$  and  $4 \cdot 10^9 \text{ m}^3$  under scenarios (1) and (2), respectively (Table 5), in

one instantaneous pulse. This volume is 3 to 4 orders of magnitude smaller than the minimum required to account for the Benicàssim dolostones, which would have been equivalent to  $5 \cdot 10^{11}$  m<sup>3</sup> of fluid for the most optimistic scenario (Table 5). About 6 times less fluid would be needed if the eroded dolostone volumes are not considered. However, these differences would be considerably higher if the reaction conditions were not optimal, as our required fluid volume calculations assume a mole-by-mole replacement and 100% efficiency in the dolomitization reaction (i.e. all the fluid totally reacts). The difference between required and expelled fluid would have been even higher if the pressure below the Benassal Fm was lower than lithostatic and/or if other less reactive fluids were considered. These results suggest that these scenarios can only provide a limited mass of Mg for the dolomitization reaction when taking into account rock volumes in the Benicàssim half graben. Therefore, even if a larger area is considered, the orders of magnitude difference between required and rules available fluid volumes out overpressured fluids as the only dolomitizing source.

As pores cannot expand as much as the fluid when heated, some excess fluid would be transported upwards through the connected pore network or by hydrofracturing under scenario (3). The calculations indicate that the volume of fluid expelled would have been on the order of  $6 \cdot 10^9$  m<sup>3</sup>. This value is still 2 to 3 orders of magnitude smaller than minimum required one (Table 5). Consequently, fluid release by heating can also not be considered as the only source for the Benicàssim dolomitization fluids. The uncertainty associated with the calculations presented in this section is negligible when compared to the required versus available fluid budgets, which are different by orders of magnitude. An alternative dolomitizing fluid source and flow mechanism involving a larger area should be invoked. In a sedimentary basin, thermal convection is a very likely alternative capable of driving large volumes of fluids to the reaction site through faults and permeable layers during long time periods (e.g. Person et al., 1996).

## 4.5. Conceptual model for the Benicàssim dolomitization

Based on the presented field and geochemical data and the mass-balance calculations, we argue that the most plausible model for the formation of the Benicàssim high-temperature replacive dolostone includes a system in which a warm brine, derived from concentrated seawater that interacted with Permian-Triassic rocks and/or Paleozoic basement, circulated upwards along largescale faults and high permeability beds (Fig. 9e). The most likely mechanism driving this fluid advection in Benicàssim during late Cretaceous to early Tertiary times is thermal convection. According to Morrow (1988) thermal convection can support long-lived flow systems that are capable of recycling subsurface solutions many times through the rock mass. This mechanism could have been driven by a thermal anomaly during the postrift cycle, as explained above.

**Table 5.** Mass-balance calculations comparing available vs. required fluid budgets to account for the Benicàssim replacement dolostone. This table shows data for the scenarios described in section 4.6.

fluid type	min. required fluid/	total volume	total volume	difference (ratio					
	rock ratio at 100°C	required (m <sup>3</sup> )	available (m <sup>3</sup> )	required/estimated)					
Scenario 1 – c	Scenario 1 – overpressured pore fluids trapped within Late Jurassic - Early Cretaceous rocks:								
seawater	302	$2.57 \cdot 10^{12}$	$5.7 \cdot 10^8$	x 4,519					
5 x conc. seaw	vater 60	$5.12 \cdot 10^{11}$		x 900					
7 x conc. seaw	vater 43	$3.62 \cdot 10^{11}$		x 637					
brine A	556	$4.73 \cdot 10^{12}$		x 8,310					
brine B	378	$3.12 \cdot 10^{12}$		x 5,651					
Scenario 2 – c	overpressured pore flu	ids trapped down	to a depth of 10 km:						
seawater	302	$2.57 \cdot 10^{12}$	$3.6 \cdot 10^9$	x 704					
5 x conc. seaw	vater 60	$5.12 \cdot 10^{11}$		x 140					
7 x conc. seaw	vater 43	$3.62 \cdot 10^{11}$		x 99					
brine A	556	$4.73 \cdot 10^{12}$		x 1,294					
brine B	378	$3.12 \cdot 10^{12}$		x 880					
Scenario 3 – pore fluids trapped down to a depth of 10 km and heated with a geothermal gradient of 50°C/km:									
seawater	302	$2.57 \cdot 10^{12}$	$5.7 \cdot 10^{9}$	x 455					
5 x conc. seaw	vater 60	$5.12 \cdot 10^{11}$		x 91					
7 x conc. seaw	vater 43	$3.62 \cdot 10^{11}$		x 64					
brine A	556	$4.73 \cdot 10^{12}$		x 836					
brine B	378	$3.12 \cdot 10^{12}$		x 569					

In order to calculate the minimum size of the advection system that would account for the Benicàssim dolomitization, we consider a geothermal gradient of 30 °C/km. In the worst-case scenario of no additional heat source, a minimum depth of 5 km for the convection cells is needed for the fluid to reach an equilibrium temperature of 150-200°C, which would allow for some cooling when flowing upwards. A total volume of ~2 · 10<sup>11</sup> m<sup>3</sup> could be stored within pores of the first 5 km of rocks in the Benicàssim half graben (the same

porosity values as for the previous fluid budget calculations were assumed). If all this fluid volume circulated at some stage through the high permeability layers, and assuming that all fluid reacted, then an advective system approximately 3 times bigger than the Benicàssim half graben would have been required to form the dolostone geobodies. This calculation was carried out considering the most reactive dolomitizing fluid and 100% reaction efficiency with the host rock. The rock volume involved in the advective system would probably have been larger, because not





**Figure 9.** (a) Sketch illustrating the Benicàssim half graben configuration during the Late Cretaceous. Arrows indicate fluid flow of deeper warm fluids that flowed upwards along large-scale faults and laterally invaded high-permeability beds, giving rise to a stratabound dolomitization pattern. (b-d) Dolomitization scenarios according to 1-D fluid release calculations: (b) scenario (1), (c) scenario (2) and (d) scenario (3). (e) SSW-NNE cross-section along the Maestrat Basin illustrating a thermal convection conceptual model for the southern part of the basin during the Late Cretaceous. The location of this cross section is indicated by the line A-A' in Fig. 1.

all the fluid reacts until reaching equilibrium in a convection cell. The presence of dolostones of the same type and age in other parts of the Maestrat basin (Nadal, 2001; Grandia, 2001) suggests that the flow system would have involved at least the whole southern part of the Maestrat basin.

Understanding the main processes that control the distribution and properties of dolomitized geobodies is essential for the hydrocarbon industry. The Benicàssim outcrop analogue is an excellent example of faultcontrolled stratabound extensive dolomitization. The presented case study summarizes an attempt to quantify some of these processes in order to find general rules and improve the ability of prediction making. We have shown that, according to the calculations and conceptual model presented in this contribution, this type of extensive fault-controlled, stratabound hydrothermal dolomitization would only occur if huge fluid volumes are mobilized and transported, in a system driven by thermal convection. This process can take place when the geothermal gradient causes a high enough fluid density contrast between shallower and deeper fluids. Other examples hydrothermal of dolomitization caused by thermal convection are discussed in Morrow et al. (1988), Wilson et al. (1990), Spencer-Cervato and Mullis (1992), Coniglio et al. (1994), Wendte et al. (1998), Gasparrini et al. (2006) or Nader et al. (2012). The Benicàssim case study illustrates the strong interplay between fault zones, seals, basin configuration and fluid flow has a significant impact on the distribution of diagenetic/hydrothermal alterations and hence on reservoir quality.

## 5. Conclusions

This contribution presents an integrated approach to the principal constraints on the distribution of fault-controlled stratabound hydrothermal dolomitization of the Aptian-Albian limestone of the Benicàssim outcrop analogue. Isotopic data is used to infer fluid characteristics and methods to quantify reactivity of dolomitizing fluids as well as mass and volume balances are introduced. As a result, a dolomitization model for this case study is proposed. The analysis performed in this study led to the following conclusions:

- $\delta^{18}$ O and  $\delta^{13}$ C stable isotope data and 1. isotope fractionation trajectories of fluid-rock interaction indicate that dolomite formed at a temperature between 80 to 110°C from a fluid that interacted with the regional limestone.  $\delta^{18}$ O becomes lighter with progressive replacement. Similarly, hydrothermal calcite veins and burial calcite cements were formed between 80 and 140°C, with characteristic light  $\delta^{18}$ O values. Meteoric calcite cements and veins as well as calcitized dolomite are also identified and characterized by heavier  $\delta^{18}$ O and lighter  $\delta^{13}$ C.
- $^{87}\mathrm{Sr}/^{86}\mathrm{Sr}$  data combined with pre-2. existing crush-leach analyses and radiogenic isotopes suggest that the Benicàssim dolomitization resulted from the circulation of brines that interacted with underlying Triassic and/or basement rocks. Thus, hightemperature fluids circulated along high permeability layers, in thermal disequilibrium with the surrounding rocks, after flowing through basement rocks and large-scale faults.
- 3. Several potential fluid sources with Benicàssim magnesium for the dolomitization are analyzed. Geologic constraints and geochemical data indicate that seawater (or evolved/concentrated seawater), basement brines (which could also be evolved seawater) or a mixture of both were the most likely dolomitizing fluids.
- 4. The evaluation of the reactivity of such fluids by means of geochemical

simulations results in an optimum range of dolomitization temperatures 75 to 125°C. Concentrated seawater is the fluid with the most dolomitizing capacity, requiring a minimum fluid/rock ratio of ~50 in order to completely dolomitize a volume of rock. The reactivity of the modelled fluids is not directly proportional to Mg concentration or salinity.

- In order to account for the volume of 5. dolostone geobodies at Benicàssim, fluid volumes on the order of  $10^{11}$  - $10^{12}$  m<sup>3</sup> are necessary to provide the  $5 \cdot 10^{13}$  moles of Mg required. Release of overpressured fluids below seals during the Late Cretaceous or Early Paleocene could only provide fluid volumes 3-4 orders of magnitude Similarly, smaller than required. release of fluids by heating could only provide fluid volumes 2-3 order of magnitude smaller. Neither of those sources is volumetrically sufficient to dolomitize the Benicàssim limestones. Therefore, a long-lasting thermal convection system during the Late Cretaceous post-rift stage is the most plausible fluid flow mechanism for this dolomitization, since a fluid volume of the same order of magnitude as the required fluid can be stored within the rocks of the Benicàssim half graben.
- 6. Dolomitization similar to that of Benicàssim could have developed in basins with similar histories of rifting and post-rifting stages. Such geologic history leads to the development of a heterogeneous crust which facilitates long-term deep fluid convection.

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