

## Highlights

- We propose that liquid chromatography coupled to high-resolution mass spectrometry (LC-HRMS) can be a very useful technique for multi-toxin detection and quantification, due to the capacity to acquire in full scan with good sensitivity and better selectivity.
- A LC-HRMS method for determination of major groups of lipophilic toxins has been developed and validated.
- The use of HRMS confirmation criteria can help to avoid false positives.
- The identification and confirmation criteria validated in the present study can contribute to define new parameters to implement HRMS in complex analysis, such as it is the case for lipophilic marine toxins in mussels.

1     **Determination of lipophilic marine toxins in**  
2     **mussels by liquid chromatography coupled to**  
3     **high resolution mass spectrometry. Validation**  
4     **study.**

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22    *identification and confirmation, quantification, validation method, uncertainty.*

23

24

25     **Abstract**

26     A multitoxin method has been developed for quantification and confirmation of  
27     lipophilic marine biotoxins in mussels by liquid chromatography coupled to high  
28     resolution mass spectrometry (HRMS), using an Orbitrap-Exactive HCD mass  
29     spectrometer. Okadaic acid (OA), Yessotoxin, Azaspiracide 1, Gymnodimine, 13-  
30     desmethyl spirolide C, Pectenotoxin 2 and Brevetoxin B were analyzed as  
31     representative compounds of each lipophilic toxin group. HRMS identification  
32     and confirmation criteria were established. Fragment and isotope ions and ion  
33     ratios were studied and evaluated for confirmation purpose. In depth

characterization of full scan and fragmentation spectrum of the main toxins was carried out. Accuracy (trueness and precision), linearity, calibration curve check, limit of quantification (LOQ) and specificity were the parameters established for the method validation. The validation was performed at 0.5 times the current European Union permitted levels. The method performed very well for the parameters investigated. The trueness, expressed as recovery, range from 80 to 94 %, the precision, expressed as intralaboratory reproducibility, range from 5 % to 22 % and the LOQs range from 0.9 to 4.8 pg on column. Uncertainty of the method was also estimated for OA, using a certified reference material. A top-down approach considering two main contributions: those arising from the **veracity studies and those coming from the precision's determination, was used.** An overall expanded uncertainty of 38 % was obtained.

## 1. Introduction

Lipophilic marine biotoxins accumulate in filter-feeding shellfish and can develop into a food safety risk [1–3]. These toxins are produced by diverse microorganisms as deeply detailed in Paz et al., [3]. Lipophilic marine toxins can be classified in several groups such as okadaic acid (OA) and dinyphystoxins (DTXs), yessotoxins (YTXs), azaspiracides (AZAs), pectenotoxins (PTXs), cyclic immines and brevetoxins [3]. A toxin of each group has been selected as representative tracer. OA, yessotoxin (YTX), azaspiracide 1 (AZA1), gymnodimine (GYM), 13-desmethyl spirolide C (SPX1), pectenotoxin 2 (PTX2) and brevetoxin B (PbTx-2) are selected as are the most common as it emerges from the EFSA opinions [4–9] and certified reference materials are available [10]. The current permitted levels by the legislation in shellfish are: for the sum of OA, DTXs and PTXs 160  $\mu\text{g kg}^{-1}$  of OA equivalents, for the sum of YTXs 1000  $\mu\text{g kg}^{-1}$  YTX equivalents and for the sum of AZAs 160  $\mu\text{g kg}^{-1}$  AZA1 equivalents [11]. For the cyclic immines group (spirolides and gymnodimines) and for the brevetoxins group there are no legal limits yet. However, the European Food Safety Authority (EFSA) is issuing several opinions for each toxin group, which recommends a revision of these legal limits (lowering it, except for YTXs) [4–9].

66 Since it is known that the official reference methods, the rat bioassay and the  
67 mouse bioassay, have to be replaced for analytical and ethical questions,  
68 alternative methods are necessary. European Union proposed to replace it for  
69 liquid chromatography coupled to tandem mass spectrometry (LC-MS/MS) [12],  
70 before 31 December 2014.

71

72 To date, LC-MS/MS has been the most used technique, providing high sensitivity  
73 and selectivity [13]. However, with this technique it is mandatory to detect  
74 compounds that are pre-selected and it is sensible to matrix effects. Several  
75 methods of multitoxin analysis are described in the literature [13–16]. Some of  
76 them are intra- or inter-laboratory validated [15, 17, 18], but the analysis of  
77 lipophilic toxins by means of mass spectrometry still generates some  
78 controversies [19–22].

79

80 In the present study, liquid chromatography coupled to high-resolution mass  
81 spectrometry (LC-HRMS) is used. It has been shown that it is a very useful  
82 technique for toxins detection due to the capacity to resolve interferences from  
83 complex matrixes [23–27], such as mussels, acquiring in full scan with good  
84 sensitivity and better selectivity. Confirmation and quantification is essential to  
85 verify the results, avoiding false positives. For this reason, having as much  
86 identification and confirmation criteria as possible will be very useful.

87

88 To the best of our knowledge, none of the LC-HRMS methods for toxin analysis  
89 in literature combine both quantification and confirmation criteria. The aim of this  
90 study is to develop a method for the quantitative determination of lipophilic  
91 marine toxins in mussels based on HRMS. Identification criteria using high  
92 resolution (50,000  $m/z$  200 full width at half maximum – FWHM) and mass  
93 accuracy better than 5 ppm (in all the mass range of the study) were used.  
94 Fragment and isotope ions and ion ratios were studied and evaluated for  
95 confirmation purpose. In depth characterization of full scan and fragmentation  
96 spectrum of the main toxins was carried out. Moreover, the performance of the  
97 quantification method using HRMS was evaluated by a validation study. The  
98 following validation parameters, accuracy (trueness and precision), linearity,  
99 calibration curve check, limit of quantification (LOQ) and specificity were

100 established for all the toxins and last but not least, the uncertainty of the method  
101 was estimated for OA.

102

## 103 **2. Experimental Section**

### 104 **2.1 Chemicals and materials**

105 Methanol (SupraSolv) was acquired from Merck (Darmstadt, Germany).  
106 Acetonitrile (LC-MS, Chromasolv,  $\geq 99.9\%$ ) and ammonium hydroxide solution  
107 ( $\geq 25\%$ ) were purchased from Sigma-Aldrich (Stenheim, Germany). Water was  
108 deionized and passed through a Milli-Q water-purification system (Millipore,  
109 Billerica, MA, USA).

110 Some of the Certified Reference Materials (CRMs) comprising calibration  
111 solutions and mussel tissue were acquired from the NRC Certified Reference  
112 Materials Program (Halifax, NS, Canada): AZA1 (CRM-AZA1  $1.24 \pm 0.07 \mu\text{g}$   
113  $\text{mL}^{-1}$ ), PTX2 (CRM-PTX2  $8.6 \pm 0.3 \mu\text{g mL}^{-1}$ ), SPX1 (CRM-SPX1  $7.06 \pm 0.4 \mu\text{g}$   
114  $\text{mL}^{-1}$ ), GYM (CRM-GYM  $5 \pm 0.2 \mu\text{g mL}^{-1}$ ), blank mussel tissue with OA (CRM-  
115 DSP-MUS-b  $10.1 \pm 0.8 \mu\text{g g}^{-1}$ ) and mussel tissue matrix (CRM-Zero-Mus). OA  
116 ( $10 \mu\text{g mL}^{-1}$ ) and YTX ( $3 \mu\text{g mL}^{-1}$ ) **were purchased from N'Tox** (Saint Jean  
117 d'Illac, France). PbTx-2 ( $100 \mu\text{g}$ ,  $95\%$  purity) was purchased from Latoxan  
118 (Valence, France).

119

### 120 **2.2 Analytical procedure**

121 Matrix-matched calibration curves were prepared with homogenate blank mussel  
122 extracts in the range between  $0.2$  and  $150 \text{ ng mL}^{-1}$ , which corresponded to  $2 -$   
123  $1500 \mu\text{g kg}^{-1}$  in sample, depending on each toxin. The ranges of the matrix-  
124 matched calibration curves for each toxin are shown in Table 1 Mussel extractions  
125 were made following the EU-Harmonised Standard Operating Procedure [10].  
126 To carry out the method validation for the lipophilic toxins included in the study  
127 at the level of  $0.5$  times the legislation limit, a blank mussel (CRM-Zero-Mus)  
128 was spiked for each toxin:  $80 \mu\text{g kg}^{-1}$  for OA, AZA1, PTX2 and  $500 \mu\text{g kg}^{-1}$  for  
129 YTX. For SPX1 and GYM the lowest concentration,  $80 \mu\text{g kg}^{-1}$ , was considered  
130 as no legal limit was set. Moreover, to estimate the uncertainty of the method for

131 OA the CRM-DSP-MUS was analyzed five times. The instrumental performance  
132 was evaluated with matrix-matched standards at the level of 1 and 25 ng mL<sup>-1</sup>; as  
133 if they were real sample extracts.

134

## 135 **2.3 Instrumentation**

### 136 ***2.3.1 Liquid Chromatography***

137 The LC system consisted of a Surveyor MS Plus pump and an Accela Open AS  
138 autosampler kept at 15 °C (**Thermo Fisher Scientific, San Jose, California**). A 5  
139 µL injection volume was used. A HypersilGold C18 (50 mm x 2.1 mm, 1.9 µm)  
140 (Thermo Fisher, Scientific, Bremen, Germany) was used for the separation of  
141 toxins at a flow rate of 300 µL min<sup>-1</sup>. Mobile phase A was water and B was  
142 acetonitrile/water (90:10), both containing 6.7 mM ammonium hydroxide [13].  
143 The gradient started at 20 % of B and was kept this composition for 3.5 min.  
144 Then, it was increased to 90 % of B in 16 min and kept 3 min, then returns to  
145 initial conditions of 20 % of B maintaining it for 11 min for the column  
146 equilibration. The total duration of the method was 30 min.

147

### 148 ***2.3.2 High Resolution Mass Spectrometry***

149 Mass spectrometry analyses were carried out with an Orbitrap-Exactive HCD  
150 (**Thermo Fisher Scientific, Bremen, Germany**) equipped with a heated  
151 electrospray source (H-ESI II). Nitrogen (purity > 99.999 %) was used as sheath  
152 gas, auxiliary gas and collision gas. The instrument was daily calibrated in both  
153 positive and negative modes. Three time segments were set. First segment (0 min  
154 – 3.5 min) working in negative mode without and with all ion fragmentation (AIF)  
155 (HCD 60 eV), second segment (3.5 min – 8.25 min) in positive mode without and  
156 with AIF (HCD 50 eV) and third segment (8.25 min – 30 min) in positive mode  
157 without and with AIF (HCD 20 eV). The mass range was  $m/z$  400–1250 in full  
158 scan and  $m/z$  70–1200 in AIF mode.

159

160 The resolution was 50,000 ( $m/z$  200, FWHM) at a scan rate 2 Hz, the automatic  
161 **gain control (AGC)** was set as “high dynamic range” with a maximum injection  
162 time of 100 ms.

163

164 Optimized parameters were spray voltage, capillary voltage, skimmer voltage,  
165 tube lens voltage, capillary temperature, heater temperature, sheath gas flow rate  
166 and auxiliary gas flow rate. OA was used for optimization in negative mode and  
167 SPX1 and PbTx-2 for positive mode. In both modes the final parameters were:  
168 spray voltage of 3.25 kV, capillary temperature of 375 °C, heater temperature of  
169 250 °C, sheath gas flow rate of 45 (arbitrary units) and auxiliary gas flow rate of  
170 15 (arbitrary units). In negative ESI capillary voltage of -92.5 V, tube lens voltage  
171 of -190 V and skimmer voltage of -44 V were used. In positive ESI capillary  
172 voltage of 77.5 V, tube lens voltage of 175 V and skimmer voltage of 32 V were  
173 used.

174

175 The data was processed with Xcalibur 2.1 software (Thermo Fisher Scientific,  
176 Bremen, Germany). Automatic identification/quantification can be performed.  
177 The peaks are found in the chromatogram by the exact mass of diagnostic,  
178 fragment and isotope ion, the mass accuracy ( $\pm 5$  ppm extraction window) and the  
179 retention time window. However, a manual verification is necessary to avoid false  
180 positives or false negatives and correct peak integration. To calculate the  
181 theoretical accurate mass ( $m/z$  calculated), the mass of the electron has been taken  
182 into account as 0.00055 Da [28]. Moreover, to apply the identification and  
183 confirmation criteria of the present study, our own excel files had to be built.

184

## 185 **2.4 Validation**

186 The quantification method for lipophilic toxins was proved to be fit for purpose  
187 carrying out a validation study to assure reliable results and prevent false positive  
188 and false negative results. The validation was based on diverse methodologies,  
189 such as EU Commission Decision 2002/657/EC [29] and the studies presented in  
190 Table 2, as no specific guidelines are set for analysis of marine biotoxins using  
191 HRMS. Statistical validation of the method developed was performed evaluating  
192 the parameters described in Table 2. Uncertainty of the method was also estimated  
193 for OA using a blank mussel tissue with a certified amount of OA.

194



## 195        **3. Results and Discussion**

### 196        **3.1 Optimization of conditions**

197        In addition to optimizing the experimental parameters for efficient toxin  
198        ionization, in the development of the method, there were some critical aspects that  
199        had to be addressed.

200        Acidic, neutral and basic mobile phase composed by water with acetonitrile or  
201        methanol at different proportions, were tested under two chromatographic  
202        columns, Hypersil Gold C18 (50 mm x 2.1 mm, 1.9  $\mu$ m) and Mediterranea Sea  
203        18 (100 mm x 2.1 mm, 3.0  $\mu$ m) (Teknokroma, Barcelona, Spain). Hypersil Gold  
204        was chosen because better peak shape and separation were obtained. The basic  
205        conditions were selected because the negatively charged toxins eluted early and  
206        separately from the positive ones, so it allows to set different time segments. As  
207        the mass analyzer cannot switch polarity fast enough working in both scan modes  
208        (full scan and AIF), having independent time segments for each polarity mode  
209        permits to rapidly switch full scan and AIF, providing in the same injection  
210        information from molecular ions and fragments ions.

211        Firstly, the aim was to work with only 2 time segments. Nevertheless, changes in  
212        the conditions were needed to obtain better signal for PbTx-2. After testing  
213        several gradients, PbTx-2, was delayed to the end of the chromatogram. As a  
214        consequence, 3 time segments were set: the first in ESI- for OA and YTX; the  
215        second in ESI+ for AZA1, GYM, SPX1 and PTX2; and the third for PbTx-2 in  
216        ESI+, with a lower HCD voltage. For each time segment only one HCD voltage  
217        was possible, in order to obtain enough data points per peak, so the HCD voltage  
218        was optimized in each case to had at least one intense fragment for every toxin.

219

### 220        **3.2 Mass spectral characterization**

221        Mass spectral characterization is indicated in Figure 1 showing also the fragment  
222        ions obtained by AIF experiment. At the optimum working conditions specified in  
223        the previous section, each compound was identified and several fragment ions  
224        were obtained as described below. Some of these fragment ions have been already  
225        described in literature [25-27] but in the present manuscript all the fragment ions  
226        have been evaluated with confirmation purpose.

227

228 The diagnostic ion was selected by taking the most selective and intense peak  
229 either the deprotonated / protonated molecule or an adduct. Non-desirable adducts  
230 consequence of basic mobile phase conditions were avoided by meticulous ion  
231 source parameters optimization. Major diagnostic ion and few signal distribution  
232 was achieved for almost all the toxins.

233

234 OA can be analyzed either in positive or negative ESI mode. Better sensitivity  
235 was obtained when the deprotonated molecule at  $m/z$  803.4587  $[M-H]^-$  was  
236 extracted from the full scan experiment. At 50 HCD voltage the fragment ions  
237 generated were at  $m/z$  785.4482  $[C_{44}H_{65}O_{12}]^-$ , at  $m/z$  255.1238  $[C_{13}H_{19}O_5]^-$  and at  
238  $m/z$  113.0608  $[C_6H_9O_2]^-$ . The  $m/z$  255.1238 was chosen for being the most intense.

239

240 The Orbitrap-MS mass spectrum of YTX showed an intense diagnostic ion at  $m/z$   
241 1163.4587  $[M-2H+Na]^-$ . In full scan mode there were some other characteristic  
242 ions of YTX at lower intensities, such as at  $m/z$  1141.4717  $[M-H]^-$ , at  $m/z$   
243 570.2322  $[M-2H]^{2-}$  and also in full scan there were fragment ions at  $m/z$   
244 1061.5149  $[M-SO_3]^-$  and at  $m/z$  467.1669  $[C_{42}H_{62}O_{19}S_2]^{2-}$ . Although it is not  
245 desirable to have ion source fragmentation, during the optimization of ion source  
246 conditions, it has been observed that with higher voltages and temperatures the  
247 sensitivity increased. HCD fragment ions were at  $m/z$  855.3842  $[C_{42}H_{63}O_{16}S]^-$  and  
248 at  $m/z$  96.9601  $[HSO_4]^-$ . The  $m/z$  96.9601 was the most intense, but it was  
249 considered not suitable, due to the fact that this  $m/z$  region presented many  
250 interferences from solvent and mussel matrix (data not shown). The fragment ion  
251 at  $m/z$  467.1669 was chosen, although it was a source fragment, because it  
252 presented good stability in all the concentration range.

253

254 AZA1 produced the protonated molecule at  $m/z$  842.5049  $[M+H]^+$ . The fragment  
255 ions generated in the HCD cell were a water loss of the protonated molecule at  
256  $m/z$  824.4943  $[C_{47}H_{70}NO_{11}]^+$ , two water losses of the protonated molecule at  $m/z$   
257 806.4838  $[C_{47}H_{68}NO_{10}]^+$  and at  $m/z$  672.4106  $[C_{38}H_{58}NO_9]^+$ . The  $m/z$  824.4943  
258 was used as fragment ion due to its high intensity.

259

260 The GYM mass spectrum revealed that the water loss of the protonated molecule  
261 at  $m/z$  490.3316  $[\text{C}_{32}\text{H}_{44}\text{NO}_3]^+$  was more intense than the protonated molecule at  
262  $m/z$  508.3421  $[\text{M}+\text{H}]^+$ . That occurred as a consequence of the high voltage and  
263 temperature of the method that were necessary for the other toxins. HCD fragment  
264 ions were at  $m/z$  392.2948  $[\text{C}_{27}\text{H}_{38}\text{NO}]^+$ , at  $m/z$  162.1277  $[\text{C}_{11}\text{H}_{16}\text{N}]^+$ , at  $m/z$   
265 136.1121  $[\text{C}_9\text{H}_{13}\text{N}]^+$  and at  $m/z$  121.0886  $[\text{C}_8\text{H}_{11}\text{N}]^+$ . The fragment ion at  $m/z$   
266 121.0886 was chosen because it was the most intense.

267

268 SPX1 produced the protonated molecule at  $m/z$  692.4521  $[\text{M}+\text{H}]^+$  and a water loss  
269 at  $m/z$  674.4415  $[\text{C}_{42}\text{H}_{60}\text{NO}_6]^+$ . The HCD mass spectrum showed several fragment  
270 ions: a water loss at  $m/z$  674.4415  $[\text{C}_{42}\text{H}_{60}\text{NO}_6]^+$ , at  $m/z$  444.3108  $[\text{C}_{27}\text{H}_{42}\text{NO}_4]^+$   
271 and at  $m/z$  164.1430  $[\text{C}_{11}\text{H}_{18}\text{N}]^+$ . The fragment ion at  $m/z$  164.1430 was chosen  
272 because it was the most intense and characteristic.

273

274 PTX2 formed some adducts at full scan mode, with ammonium at  $m/z$  876.5104  
275  $[\text{M}+\text{NH}_4]^+$ , with sodium at  $m/z$  881.4658  $[\text{M}+\text{Na}]^+$  and with potassium at  $m/z$   
276 897.4397  $[\text{M}+\text{K}]^+$ . In full scan there was also a double water loss of the  
277 protonated molecule at  $m/z$  823.4647  $[\text{C}_{47}\text{H}_{67}\text{O}_{12}]^+$ . In the HCD mass spectrum  
278 there were several water losses from the protonated molecule (for instance, at  $m/z$   
279 823.4647 and at  $m/z$  805.4512) and some fragment ions generated were at  $m/z$   
280 213.1121  $[\text{C}_{11}\text{H}_{17}\text{O}_4]^+$  and at  $m/z$  195.1016  $[\text{C}_{11}\text{H}_{15}\text{O}_3]^+$  (data not shown). The  
281 fragment ion at  $m/z$  823.4647 was chosen, because although it was a source  
282 fragment, it presented good stability in all the concentration range.

283

284 PbTx-2 produced the protonated molecule at  $m/z$  895.4838  $[\text{M}+\text{H}]^+$  and the  
285 sodium and potassium adducts at  $m/z$  917.4658  $[\text{M}+\text{Na}]^+$  and at  $m/z$  933.4397  
286  $[\text{M}+\text{K}]^+$ . HCD fragment ions were several water losses from the protonated  
287 molecule at  $m/z$  877.4733  $[\text{C}_{50}\text{H}_{69}\text{O}_{13}]^+$  and at  $m/z$  859.4627  $[\text{C}_{50}\text{H}_{67}\text{O}_{12}]^+$ . The  
288 fragment ion at  $m/z$  877.4733 was chosen for being the most intense.

289

290

### 291 **3.3 Determination of lipophilic marine toxins**

292 Lipophilic toxins were separated by reverse phase chromatography coupled to an  
293 Orbitrap–Exactive HCD mass spectrometer. As shown in Figure 2 seven toxins  
294 were separated in 10 min. After an in depth characterization of full scan and  
295 fragmentation spectra, a diagnostic ion, a fragment ion and an isotope ion were  
296 chosen for each toxin and were included in Table 3. The choice of the diagnostic  
297 ions changed depending on background interferences [30]. The fragment ion  
298 chosen was the one giving higher signal and the most stable ion fragment ratio.  
299 The M+1 isotope ion was chosen in all the cases for the calculation of the isotope  
300 ion ratio.

301

302 In the present study, the chromatographic separation of PbTx–2 was achieved in  
303 alkaline conditions. However, it should be highlighted that PbTx–2 was poorly  
304 ionized by the conditions of the method, although meticulous optimization of  
305 parameters was carried out. For these reasons, the identification and confirmation  
306 was not possible at the concentrations of interest. Further validation was not  
307 performed.

308

#### 309 ***3.3.1 Identification and confirmation criteria***

310 The identification and confirmation criteria adopted in the present study are  
311 detailed in Table 2. The use of confirmation by a second ion is very helpful to  
312 prevent false positives [23].

313

##### 314 **3.3.1.1 Mass accuracy and precision**

315 In the present study, mass accuracy and precision expressed as parts per million  
316 (ppm) were used. Precision has received limited coverage in the literature, but it is  
317 important, and both accuracy and precision should be considered, when dealing  
318 with accurate mass measurements [31]. In Table 3 are listed mass accuracy and  
319 precision of each toxin obtained from matrix–matched calibration curves.  
320 Observed  $m/z$ , mass accuracy and mass precision are averages from all the points  
321 of the calibration curve, except for PbTx–2. Mass accuracy and precision were  
322 calculated using root–mean–square to avoid positive and negative values

cancelling each other. Mass accuracy in most of cases was ranging between 1 and 2.5 ppm (see Table 3) for diagnostic, fragment ions and isotope ions. So a maximum of 5 ppm of error was permitted to the software for peak identification. Precision was ranging from 0.23 to 2.12 ppm, which indicates a good stability in the mass measurement. For PTX2 a high variability in the mass measurement was observed between high and low points of the calibration curve.

#### 3.3.1.2 High resolution

The experimental resolution was better than 20,000 (FWHM) in all the mass range of interest, in order to achieve the criteria listed in Table 2. As it is shown in Figure 3a) high resolution is necessary to resolve the interferences coming from the matrix.

#### 3.3.1.3 Retention time

Retention time in standards and samples must agree, so a restrictive tolerance has been set. Only peaks with values lower than three times the standard deviation (SD) from the mean of the matrix-matched calibration curve retention times had been considered.

#### 3.3.1.4 Ion ratio

As it has been previously said, the incorporation of additional parameters and criteria for confirmation of positive results is recommended. In the present study, the ion ratio is defined as the ratio between the diagnostic ion and the confirmation ion. The confirmation ion can be a fragment ion or an isotope ion, so two different ion ratios were evaluated. The tolerance of the ion ratios must not exceed those from Decision 2002/657/EC [29]: if the ion ratio is under 2, a  $\pm 20$  % of maximum ratio tolerance is accepted, if it is between 2 and 5,  $\pm 25$  % is accepted, if it is between 5 and 10, a  $\pm 30$  % is accepted and if it is more than 10, a  $\pm 50$  % is accepted.

#### Fragment ion ratio

The fragment ion ratio, defined as the ratio between the area of the diagnostic ion and the area of the fragment ion, has been used to confirm peak identity in the samples. The average ion ratio for each toxin has been established. This was done

357 after studying the ion ratios of the diagnostic ions with all the fragments obtained,  
358 evaluating its stability in all the concentration range. HCD fragments were  
359 preferably used, but for some toxins it was mandatory to use source fragments due  
360 to the hard ionization conditions. The definitive ion ratios are listed in Table 3.  
361 For OA the fragment ion ratio was 6.39 with a relative standard deviation (RSD)  
362 of 7 %. In the case of YTX a value of 0.72 was obtained with a RSD of 17 %. For  
363 AZA1 the ion ratio was 4.47 with a RSD of 17 %. GYM obtained a value of 14.44  
364 with a RSD of 13 %. For SPX1 the fragment ion ratio was 5.13 with an RSD of 15  
365 %. PTX2 obtained a value of 5.28 with an RSD of 40 %.

366

367 To confirm a finding as an actual positive the ion ratio of the sample should be in  
368 agreement with the ion ratio of the matrix-matched calibration curve.

369

370 Isotope ion ratio

371 The isotope ion ratio, defined as the ratio between the monoisotope ion  
372 (diagnostic ion) and the isotope ion (M+1, corresponding to the natural isotope  
373  $^{13}\text{C}$ ), has been calculated as a confirmation criteria. It was possible to perform it  
374 because in every case the diagnostic ion has relatively high  $m/z$  (490.3316 to  
375 1163.4537), M+1 was always around 50% of it and the isotope ion ratio was  
376 stable in all the concentration range. Moreover, with the software used it is  
377 possible to determine the theoretical isotope ion ratio of each compound and it has  
378 been observed that all isotope ion ratios were very similar to the theoretical ratio.  
379 The isotope ion ratios are listed in Table 3. For OA it was 2.17 with a RSD of 15  
380 %. In the case of YTX a value of 1.62 was obtained with a RSD of 7 %. For  
381 AZA1 the isotope ratio was 2.08 with a RSD of 10 %. GYM obtained a value of  
382 2.96 with a RSD of 9 %. For SPX1 the isotope ion ratio was not possible to  
383 calculate in all the levels of the calibration curve, due to interferences in the M+1  
384 and in M+2. PTX2 obtained a value of 1.84 with a RSD of 18 %.

385

386 After analyzing all the data, an optimum confirmation ion for each toxin was  
387 selected [23]. For OA, although fragment ion ratio had acceptable values, at low  
388 concentration levels of the matrix-matched calibration curve was not possible to  
389 use it, so the isotope ion ratio was used as confirmation criteria for this toxin. In  
390 the case of YTX, AZA1 and GYM both ratios can be used as they had the same

391 sensibility and acceptable values of the ion ratios in the studied concentration  
392 range of the calibration curves. For SPX1 the fragment ion ratio should be used as  
393 there were interferences due to the complex matrix in the M+1 and in M+2  
394 isotope ions. The interference was detected by analyzing the isotope ion ratio of a  
395 calibration curve without matrix. For PTX2 the isotope ion ratio was used because  
396 better sensibility can be achieved, instead of using the fragment ion ratio, as this  
397 had an unacceptable RSD.

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## 400 **3.4 Validation study**

401 The suitability of the quantification method for lipophilic toxins was evaluated by  
402 a validation study. Firstly, a freeze-dried blank mussel spiked at 0.5 times the  
403 legislation limit was used as no reference material containing all the toxins at low  
404 level ( $80 \mu\text{g kg}^{-1}$ , except for YTX  $500 \mu\text{g kg}^{-1}$ ) was available. Validation was  
405 performed at this level as we are near de MRL as it's recommended by the  
406 Commission Decision 2002/657/EC [29]. Afterwards, a mussel tissue reference  
407 material containing a certified amount of OA was analyzed to estimate the  
408 uncertainty of the method.

409

### 410 **3.4.1 Validation Parameters**

#### 411 3.4.1.1 Accuracy

412 The accuracy of a method can be defined taking in consideration its trueness  
413 (closeness of agreement between the average of a number of tests results and an  
414 accepted reference value) and its precision (closeness of agreement between test  
415 results) [32].

416

#### 417 Trueness

418 In the present study, trueness is expressed as the recovery of fortified mussel  
419 samples ( $n=3$ ), spiked at concentration levels of 0.5 times the legislation limit.  
420 Table 4 shows that recoveries were in the range of 80 – 94 %. These values are  
421 acceptable according to Commission Decision 2002/657/EC [29], which states that

the accuracy (as recovery) of a method with analyte levels above  $10 \mu\text{g kg}^{-1}$  must be ranging between 80 – 110 %.

#### Precision

The precision, expressed as intralaboratory reproducibility of the method, was determined in terms of relative standard deviation ( $\text{RSD}_R$ ) from  $n=3$  recovery experiments at 0.5 times the legislation limit.

The precision of the method was ranging between 5 % and 22 % as listed in Table 4. This precision is totally acceptable according to the Horwitz equation [29]. It should be highlighted that this equation gives unacceptable high values for concentrations below  $100 \mu\text{g kg}^{-1}$ . As set in the Commission Decision 2002/657/EC [29], the highest variation acceptable is 23 % at  $100 \mu\text{g kg}^{-1}$ , and this method presents a maximum variation of 22 % for OA at lowest concentration ( $80 \mu\text{g kg}^{-1}$ ), so the values obtained were acceptable.

#### 3.4.1.2 Linearity

Matrix-matched calibration curves were run every day. A minimum of 5 points for each calibration curve were required. Linearity was considered acceptable when the regression coefficient was  $\geq 0.98$  [10] with residuals lower than 30 % [33]. For all the matrix-matched calibration curves injected the correlation was acceptable, obtaining values between 0.9806 and 0.9993 (Table 1). Due to the high linear range chosen for the curves (the range of concentration studied ranged from 0.2 to  $150 \text{ ng mL}^{-1}$ ) to fulfill with residuals values lower than 30 %, a weighted curve in concentration was adopted ( $1/x$ ) and it was not forced to go through the origin.

#### 3.4.1.3 Calibration Curve Check (intra-batch response drift)

The response drift of the method was checked by comparing a level of the matrix-matched calibration curve at the beginning of the analysis with the same level analyzed after the samples. The difference could not exceed a 30%. Fresh calibration curves were needed. It is especially important in this case as no internal standards are available and evaporation of matrix-matched calibration curves may occur.



#### 3.4.1.4 Limit of Quantification (LOQ)

LOQ was determined by the lowest point of the calibration curve which was possible to confirm fulfilling the criteria established in section 3.3.1 (see Identification and Confirmation Criteria). Table 1 shows the instrumental LOQ, expressed as picograms on column, obtained for each toxin. In Figure 3 the spectrum and the extracted ion chromatogram from SPX1 ( $m/z$  692.4521) at 0.18 pg  $\mu\text{l}^{-1}$  are shown. These low values from LOQ are quite interesting in a middle future because the EFSA has proposed new legislation limits and, for the majority of them, these are much lower than the actual ones.

#### 3.4.1.5 Blank Quality Control (QC) / Specificity

Extracted blank mussel ( $n=15$ ) was analyzed as a real sample to study signals obtained from the matrix and to evaluate if interferences that lead to false positive results were obtained. The good specificity of the technique (working in high resolution 50,000 –  $m/z$  200, FWHM and with extracted ion window of 5 ppm) makes possible to have no signal at all in the blank mussel for any of the toxins, except for SPX1. Nevertheless it must be noted that the blank mussel is not certified for the absence of SPX1, so the obtained signals could be attributed to its presence in the sample. However, in all the positive results for SPX1, the concentration was below the LOQ or it can not be confirmed.

### 3.4.2 Uncertainty estimation for OA

Uncertainty is a quantitative indicator of the confidence in the analytical data and describes the range around a reported or experimental result within which the true value can be expected to lie within a defined probability (confidence level) [30].

The uncertainty of the whole method at the interest level was estimated following a top-down approach [34] considering two main contributions: those arising from the veracity studies and those coming from the precision's determination.

487 Those values have been derived from the analysis of a reference material (with a  
488 certified value of  $10.1 \pm 0.8 \mu\text{g g}^{-1}$ ). To achieve the interest level, a dilution of  
489 1/50 was done following the EU-Harmonised Standard Operating Procedure [10].

490

491 **Before the final uncertainty's estimation, the compatibility index between the**  
492 results from our laboratory and the CRM was checked. The two values were  
493 compared following the methodology proposed by the Institute for Reference  
494 Materials and Measurements [35]. This procedure takes into account the  
495 difference between the certified value and the measurement result, as well as their  
496 respective uncertainties. No significant difference between the measurement result  
497 and the certified value was detected.

498

499 A value for expanded uncertainty ( $k = 2$ ) of 38 % was obtained, which is in  
500 agreement with the expected value arising from the specialized literature and  
501 meets the criteria of SANCO 12495/2011 [30].

502

### 503 **3.5 Instrumental Quality Parameters**

504 In addition to the validation of the overall method (extraction and instrumental  
505 analysis), instrumental quality parameters were assessed. As shown in Table 5  
506 instrumental trueness, repeatability ( $i\text{RSD}_r$ ) and reproducibility were evaluated at  
507 two levels (1 and 25  $\text{ng mL}^{-1}$  matrix-matched standards). Instrumental trueness ( $n$   
508 = 6) were ranging from 91 to 116 % in 1  $\text{ng mL}^{-1}$  and from 94 to 111 % in 25  $\text{ng}$   
509  $\text{mL}^{-1}$ . Repeatability ( $i\text{RSD}_r$ ) ( $n = 5$ ) values were ranging from 4 to 10 % in 1  $\text{ng}$   
510  $\text{mL}^{-1}$  and from 1 to 7 % in 25  $\text{ng mL}^{-1}$ . Reproducibility ( $i\text{RSD}_R$ ) was tested in 6  
511 different days obtaining values from 7 to 16 % in 1  $\text{ng mL}^{-1}$  and from 9 to 14% in  
512 25  $\text{ng mL}^{-1}$ .

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## 518 **Conclusions**

519 A sensitive LC–HRMS method for quantification of major groups of marine  
520 lipophilic toxins has been developed and validated. The method performed very  
521 well for the parameters investigated. Ion ratios as confirmation criteria were  
522 deeply studied. It was observed, that both fragment ion ratio and isotope ion ratio  
523 can be used to confirm a positive result, but for each compound one or the other  
524 can be more suitable. The use of the HRMS criteria can help to prevent false  
525 results. Interferences coming from the matrix can be identified because data is  
526 acquired in full scan mode so matrix effects are minimized. It has been shown that  
527 HRMS provides incomparable confirmatory performances with excellent  
528 quantitative capabilities. Further studies are necessary to include more toxins of  
529 each group studied and more toxin groups. Moreover, this study can contribute to  
530 define new parameters based on HRMS, for complex matrix analysis, as it is the  
531 case for lipophilic marine toxins in mussels.

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613 Figure captions

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615 Figure 1. Mass spectral characterization of all the toxins. For okadaic acid the  
616 fragmentation spectrum at 60 HCD voltage is shown. For azaspiracide 1, gymnodimine,  
617 13-desmethyl spirolide C the fragmentation spectra at 50 HCD voltage are shown. For  
618 yessotoxin and pectenotoxin 2 the full scan spectra are shown. For brevetoxin B the  
619 fragmentation spectrum at 20 HCD voltage is shown. All the spectra correspond to  
620 matrix-matched standard at 50 ng mL<sup>-1</sup>, except for brevetoxin B that correspond to 1 µg  
621 mL<sup>-1</sup>.

622 Figure 2. Extracted ion chromatogram of the lipophilic marine toxins, showing a)  
623 diagnostic ions and b) fragment ions, with an extraction window of 5 ppm.

624 Figure 3. a) Spectrum and b) Extracted ion chromatogram from SPX1 (*m/z* 692.4521) at  
625 0.18 pg µL<sup>-1</sup>.

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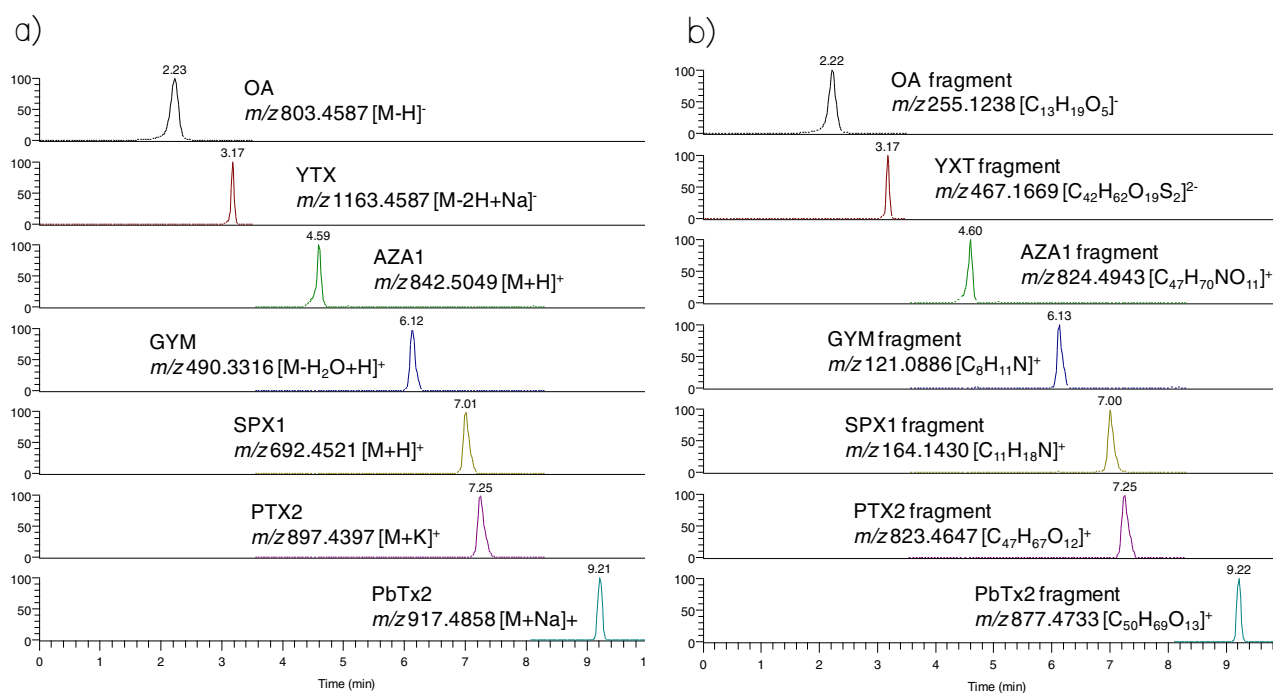


Figure 2.

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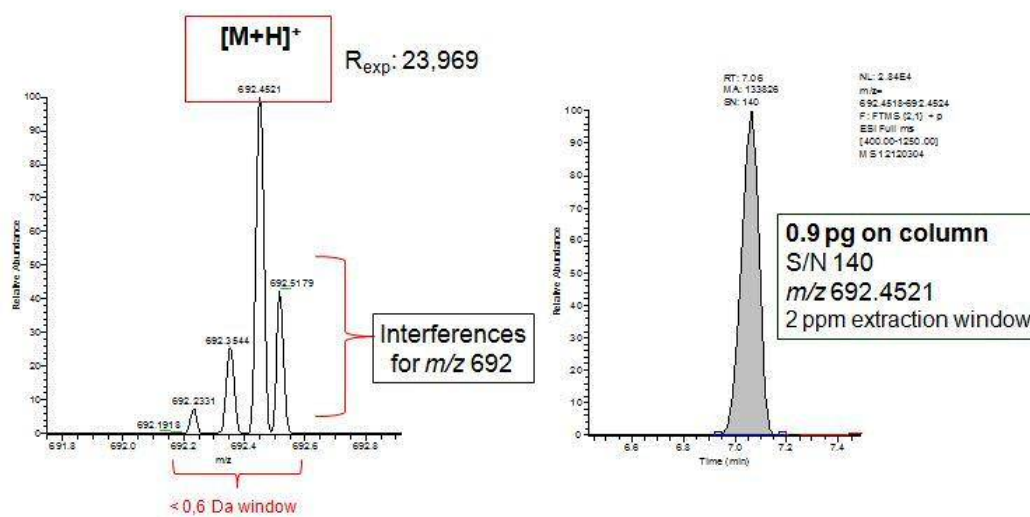


Figure 3.

687 Table 1. Identification and confirmation criteria. Validation parameters. Matrix-matched  
688 calibration curves ranges. Values from n = 6 matrix-matched calibration curves. Linearity  
689 expressed as regression coefficients and residuals. In italics data not used, as described in  
690 section 3.3.1.4.

Toxin	RT (min) $\pm$ 3·SD	Fragment ion ratio $\pm$ tolerance	Isotope ion ratio $\pm$ tolerance	Matrix-matched calibration curves range	R <sup>2</sup>	Residuals (minimum–maximum)	LOQ (pg on column)
Okadaic acid	2.27 $\pm$ 0.12	<i>6.39 <math>\pm</math> 30%</i>	2.17 $\pm$ 25%	0.5–150 ng mL <sup>-1</sup>	0.9911 – 0.9988	0%–29%	2.4
Yessotoxin	3.17 $\pm$ 0.07	0.72 $\pm$ 20%	1.62 $\pm$ 25%	1–150 ng mL <sup>-1</sup>	0.9825 – 0.9968	0%–28%	4.8
Azaspiracide-1	4.58 $\pm$ 0.08	4.47 $\pm$ 25%	2.08 $\pm$ 25%	0.5–50 ng mL <sup>-1</sup>	0.9806 – 0.9955	2%–26%	2.4
Gymnodimine	6.13 $\pm$ 0.09	14.44 $\pm$ 50%	2.96 $\pm$ 25%	0.5–150 ng mL <sup>-1</sup>	0.9832 – 0.9993	0%–23%	2.4
13-desmethyl spirolide C	7.02 $\pm$ 0.09	5.13 $\pm$ 30%	–	0.2–150 ng mL <sup>-1</sup>	0.9904 – 0.9992	0%–25%	0.9
Pectenotoxin-2	7.25 $\pm$ 0.08	<i>5.28 <math>\pm</math> 30%</i>	1.84 $\pm$ 25%	0.5–50 ng mL <sup>-1</sup>	0.9926 – 0.9961	1%–30%	3.1

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Table 2. Identification and confirmation criteria, validation parameters. Comparison between studies.

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		Pitarch, 2007 [33]	EU-RL-MB SOP [10]	Gerssen, 2010[17]	SANCO / 12495 / 2011 [30]	Mol, 2012 [23]	Present study
Analytes applied to		Priority organic micropollutants	Lipophilic toxins	Lipophilic toxins	Pesticide residues	Pesticides	Lipophilic toxins
Matrix		Water	Molluscs	Shellfish	Food and feed	Vegetables and fruits	Mussel
Analytical technique		GC-MS/MS	LC-MS/MS	LC-MS/MS	HRMS	LC-HRMS/MS	LC-HRMS/MS
Purpose		Quantification	Quantification	Quantification	Quantification	Screening	Quantification
Identification and Confirmation Criteria							
Mass accuracy		—	—	—	< 5 ppm	< 5 ppm	< 5 ppm
High Resolution (at full width at half maximum – FWHM)		—	—	—	≥ 20,000 at the mass range of interest	≥ <b>20,000</b> at the mass range of interest	≥ <b>20,000</b> at the mass range of interest
Retention time (RT) drift		Agreement in RT between samples and standards	Not exceed 3 %	5 %	2.5 %	1 %	Mean ± 3 · SD (not relative to time)
Diagnostic ions		1 or 2 precursor ions	1 precursor ion	1 precursor ion	≥ <b>2 diagnostic</b> ions	≥ <b>2</b> diagnostic ions	1 diagnostic ion
Fragment ions		At least two MS/MS transitions	At least two MS/MS transitions	Two product ions were selected for each toxin	At least 1 fragment ion	At least 1 fragment ion	1 fragment ion
Isotope ions		—	—	—	–	M+1, M+2	M+1
Ion ratio		Ratio between quantitative and confirmative transitions	Must be recorded	As described in Decision 2002/657/EC	Comparison of experimental ratio of samples and standards	Fragment ion ratio: ratio between diagnostic and fragment ion	Fragment ion ratio: ratio between diagnostic and fragment ion
						Isotope ion ratio: relation between diagnostic ion and M+1 or M+2	Isotope ion ratio: relation between diagnostic ion and M+1
Fragment-isotope ion ratio tolerance		Comparison of experimental ratio of samples and standards. As described in Decision 2002/657/EC	—	As described in Decision 2002/657/EC	As described in Decision 2002/657/EC	Independent of relative intensity between ions: ± 50%	As described in Decision 2002/657/EC
Validation parameters							
Accuracy	Trueness	Recovery 70–120 %	—	As described in Decision 2002/657/EC. Recovery 80 – 110 %	Recovery 70 – 120 %	—	As described in Decision 2002/657/EC. Recovery 80 – 110 %
	Precision	RSD <sub>r</sub> < 20 %	—	Intraday repeatability and reproducibility. HorRat < 1.0	RSD <sub>r</sub> and RSD <sub>R</sub> < 20 %	—	As described in Decision 2002/657/EC. RSD <sub>R</sub> < 23 %

Sensitivity / LOD	the lowest confirmation level tested	with the lowest intensity	strongest transition			
LOQ	Lowest level that can be validated with recovery (70 – 120 %) and precision (RSD < 20 %)	—	<b>S/N ≥ 6 for the weakest transition</b>	Lowest level that can be validated with recovery (70 – 120 %) and precision (RSD < 20 %) <b>and</b> ≤ MRL	—	Lowest point of the calibration curve that fulfill all the identification and confirmation criteria
Blank Quality Control (QC) / Specificity	—	Methanol blank to be injected. No signal for lipophilic toxins ( < LOD or < 10 % of the lowest calibration point)	21 different blank samples to determine interfering peaks	Blank reagent < 30 % LOQ	—	Mussel blank samples (n = 15) to determine interfering peaks. No signal for lipophilic toxins ( < LOQ or do not fulfill the confirmation criteria)

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708 Table 3. Chemical formula, diagnostic ion, fragment ion, isotope ion, *m/z* calculated and observed for each ion. Mass  
709 error (RMS error) and in brackets the standard deviation (SD) of all the concentration levels of the calibration curve (n  
710 isotope ratio) and relative standard deviation (RSD) of the ion ratio calculated using all the concentration levels of the  
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Toxin	Chemical formula	Diagnostic ion	<i>m/z</i> calculated	<i>m/z</i> observed	Mass accuracy and precision	Fragment ion	<i>m/z</i> calculated	<i>m/z</i> observed	Mass accuracy and precision	Fragment ion ratio (% RSD)	Isotope ion
Okadaic acid	C <sub>44</sub> H <sub>68</sub> O <sub>13</sub>	[M-H] <sup>-</sup>	803.4587	803.4597	1.43 (0.53)	[C <sub>13</sub> H <sub>19</sub> O <sub>5</sub> ] <sup>-</sup>	255.1238	255.1232	2.53 (0.39)	6.39 (7)	[C <sub>43</sub> <sup>13</sup> CH <sub>67</sub> O <sub>13</sub> ] <sup>-</sup>
Yessotoxin	C <sub>55</sub> H <sub>82</sub> O <sub>21</sub> S <sub>2</sub>	[M-2H+Na] <sup>-</sup>	1163.4537	1163.4558	2.04 (0.75)	[C <sub>42</sub> H <sub>62</sub> O <sub>19</sub> S <sub>2</sub> ] <sup>2-</sup>	467.1669	467.1676	1.59 (0.25)	0.72 (17)	[C <sub>54</sub> <sup>13</sup> C H <sub>80</sub> O <sub>21</sub> S <sub>2</sub> ] <sup>-</sup>
Azaspiracide-I	C <sub>47</sub> H <sub>71</sub> NO <sub>12</sub>	[M+H] <sup>+</sup>	842.5049	842.5048	0.83(0.92)	[C <sub>47</sub> H <sub>70</sub> NO <sub>11</sub> ] <sup>+</sup>	824.4943	824.4933	1.28 (0.37)	4.47 (17)	[C <sub>46</sub> <sup>13</sup> C H <sub>72</sub> NO <sub>12</sub> ] <sup>+</sup>
Gymnodimine	C <sub>32</sub> H <sub>45</sub> NO <sub>4</sub>	[M+H+H <sub>2</sub> O] <sup>+</sup>	490.3316	490.3308	1.82 (0.86)	[C <sub>8</sub> H <sub>11</sub> N] <sup>+</sup>	121.0886	121.0885	1.41 (0.52)	14.44 (13)	[C <sub>31</sub> <sup>13</sup> C H <sub>44</sub> NO <sub>4</sub> ] <sup>+</sup>
13-desmethyl spiroside C	C <sub>42</sub> H <sub>61</sub> NO <sub>7</sub>	[M+H] <sup>+</sup>	692.4521	692.4514	1.13 (0.81)	[C <sub>11</sub> H <sub>18</sub> N] <sup>+</sup>	164.1434	164.1431	1.70 (0.23)	5.13 (15)	[C <sub>41</sub> <sup>13</sup> C H <sub>62</sub> NO <sub>7</sub> ] <sup>+</sup>
Pectenotoxin-2	C <sub>47</sub> H <sub>70</sub> O <sub>14</sub>	[M+K] <sup>+</sup>	897.4397	897.4396	1.87 (2.09)	[C <sub>47</sub> H <sub>67</sub> O <sub>12</sub> ] <sup>+</sup>	823.4647	823.4645	2.23 (2.05)	5.28 (40)	[C <sub>46</sub> <sup>13</sup> C H <sub>70</sub> O <sub>14</sub> ] <sup>+</sup>
Brevetoxin B*	C <sub>50</sub> H <sub>70</sub> O <sub>14</sub>	[M+Na] <sup>+</sup>	917.4658	917.4692	—	[C <sub>50</sub> H <sub>69</sub> O <sub>13</sub> ] <sup>+</sup>	877.4733	877.4720	—	—	[C <sub>49</sub> <sup>13</sup> C H <sub>70</sub> O <sub>14</sub> ] <sup>+</sup>

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713 \*Brevetoxin B data from one single acquisition.  
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Table 4. Validation parameters for spiked mussel: trueness expressed as recovery experiments and precision as intralaboratory reproducibility (n = 3) at 0.5 times the legislation limit.

Toxin	0.5 times the legislation limit	Trueness (%)	RSD <sub>R</sub> (%)
Okadaic acid	80 µg·kg <sup>-1</sup>	94	22
Yessotoxin	500 µg·kg <sup>-1</sup>	88	12
Azaspiracide-I	80 µg·kg <sup>-1</sup>	80	8
Gymnodimine*	80 µg·kg <sup>-1</sup>	90	5
13-desmethyl spirolide C*	80 µg·kg <sup>-1</sup>	85	8
Pectenotoxin-2	80 µg·kg <sup>-1</sup>	89	13

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\*For SPX1 and GYM as no legislation limit was set, the lowest concentration was taken.

727 Table 5. Instrumental quality parameters. Instrumental accuracy (trueness) (n=6),  
728 instrumental repeatability (iRSD<sub>r</sub>) (n=5) and intralaboratory reproducibility (iRSD<sub>R</sub>)  
729 (n=6).  
730

Toxin	Accuracy: trueness	iRSD <sub>r</sub>	iRSD <sub>R</sub>	Accuracy: trueness	iRSD <sub>r</sub>	iRSD <sub>R</sub>
	1 ng mL <sup>-1</sup>			25 ng mL <sup>-1</sup>		
Okadaic acid	102%	9%	16%	102%	1%	13%
Yessotoxin	116%	6%	7%	111%	3%	12%
Azaspiracide-I	93%	7%	9%	94%	3%	9%
Gymnodimine	100%	4%	13%	102%	1%	16%
13-desmethyl spirolide C	94%	5%	14%	108%	2%	11%
Pectenotoxin-2	91%	10%	10%	101%	7%	14%

731  
732