Recovery of Free Energy Branches in Single Molecule Experiments

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We present a method for determining the free energy of coexisting states from irreversible work measurements. Our approach is based on a fluctuation relation that is valid for dissipative transformations in partially equilibrated systems. To illustrate the validity and usefulness of the approach, we use optical tweezers to determine the free energy branches of the native and unfolded states of a two-state molecule as a function of the pulling control parameter. We determine, within $0.6k_BT$ accuracy, the transition point where the free energies of the native and the unfolded states are equal.

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Recent developments in statistical physics [1] have provided new methods to extract equilibrium free energy differences in small systems from measurements of the mechanical work in irreversible processes (see [2,3] for reviews). In this regard, fluctuation relations [3] are generic identities that establish symmetry properties for the probability of exchanging a given amount of energy between the system and its environment along irreversible processes. If a system, initially in thermodynamic equilibrium, is strongly perturbed by fast varying a control parameter λ between two values λ_0 and λ_1 , then the system is driven out of equilibrium. The work exerted upon the system, averaged over the ensemble of all possible trajectories, reads $\langle W \rangle = \langle \int_{\lambda_0}^{\lambda_1} (\partial \mathcal{H} / \partial \lambda) d\lambda \rangle$, where \mathcal{H} is the system Hamiltonian. According to the second law of thermodynamics, $\langle W \rangle$ is always greater than the free energy difference between the initial and final states, $\Delta G = G(\lambda_1) - G(\lambda_2)$ $G(\lambda_0)$. The Crooks fluctuation relation [4] extends the predictive power of the Second Law by establishing a symmetry relation for arbitrary functionals of a trajectory Γ measured along a nonequilibrium process (forward or F process) and its time reversed one (reverse or R process). In the forward process the system starts in equilibrium at λ_0 and λ is varied from λ_0 to λ_1 for a time t_f according to an arbitrary protocol $\lambda(t)$ [i.e., $\lambda_1 = \lambda(t_f)$]. In the reverse process the system starts in equilibrium at λ_1 and λ is varied from λ_1 to λ_0 following the time reversed scheme, given by $\lambda(t_f - t)$. In its most general form the Crooks fluctuation relation reads [4]

$$\langle \mathcal{F} \exp[-\beta(W - \Delta G)] \rangle_F = \langle \hat{\mathcal{F}} \rangle_R,$$
 (1)

where \mathcal{F} stands for an arbitrary functional of the forward trajectories the system can take through phase space. β is the inverse of the thermal energy $k_B T$ where k_B is the Boltzmann constant and T the temperature of the environment. In this relation, $\hat{\mathcal{F}}$ is the time reversal of \mathcal{F} , while the averages $\langle \bullet \rangle_{F(R)}$ are taken over the ensemble of all possible

forward (reverse) trajectories. The particular case $\mathcal{F} = \delta(W - W(\Gamma))$ yields a relation between work distributions along the forward and reverse processes, $P_F(W) = P_R(-W) \exp[\beta(W - \Delta G)]$. This relation has been experimentally tested and used to extract free energy differences in single molecule experiments [5–7]. A thorough discussion on its validity domain can be found in [4].

Fluctuation relation under partial equilibrium conditions.—By considering only the trajectories that go from one specific subset of configurations to another one, Maragakis et al. [8] have derived another relation useful to extract free energy differences between subsets of states. In principle, the validity of Eq. (1) is restricted to initial conditions that are Gibbsian over the whole phase space S(what we might call global thermodynamic equilibrium). It is, however, possible to extend Eq. (1) to the case where the initial state is Gibbsian but restricted over a subset of configurations (what we might call partial thermodynamic equilibrium). A relation mathematically similar to Eq. (1)can be derived, but involving nonequilibrium processes that are in partial (rather than global) equilibrium. It is useful to rephrase here the derivation in such a way to emphasize the role played by partial equilibrium. As we will see this makes it possible to experimentally determine the free energy of coexisting states for values of λ such that the system is never globally equilibrated.

Let $P_{\lambda,S'}^{eq}(\mathcal{C})$ denote the partially equilibrated (i.e., Boltzmann–Gibbs) distribution for a given value of λ . Such distribution is restricted over a subset S' of configurations \mathcal{C} contained in S (i.e., $\mathcal{C} \in S' \subseteq S$). The case S' = S corresponds to global equilibrium: $P_{\lambda}^{EQ}(\mathcal{C}) \equiv$ $P_{\lambda,S}^{eq}(\mathcal{C})$. Partially equilibrated states satisfy $P_{\lambda,S'}^{eq}(\mathcal{C}) =$ $P_{\lambda}^{EQ}(\mathcal{C})\chi_{S'}(\mathcal{C})Z_{\lambda,S}/Z_{\lambda,S'}$, where $\chi_{S'}$ is the characteristic function defined over the subset $S' [\chi_{S'}(\mathcal{C}) = 1$ if $\mathcal{C} \in S'$ and zero otherwise], and $Z_{\lambda,S'} = \sum_{\mathcal{C} \in S'} \exp(-\beta E_{\lambda}(\mathcal{C}))$, with $E_{\lambda}(\mathcal{C})$ the energy function of the system for a given λ

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and C. Given a forward trajectory Γ , going from configuration C_0 when $\lambda = \lambda_0$ to C_1 for $\lambda = \lambda_1$, let $S_0(S_1)$ be the subset of S over which the system is partially equilibrated at $\lambda_0(\lambda_1)$. Consider now the following transformation of the functional \mathcal{F} in Eq. (1): $\mathcal{F}(\Gamma) \to \chi_{S_0}(C_0)\mathcal{F}(\Gamma)\chi_{S_1}(C_1)$. Under previous conditions the following identity can be proved [9]:

$$\frac{\langle \mathcal{F} \exp[-\beta(W - \Delta G_{\mathcal{S}_0, \lambda_0}^{\mathcal{S}_1, \lambda_1})] \rangle_F^{\mathcal{S}_0 \to \mathcal{S}_1}}{\langle \hat{\mathcal{F}} \rangle_R^{\mathcal{S}_0 \to \mathcal{S}_1}} = \frac{p_R^{\mathcal{S}_0 \to \mathcal{S}_1}}{p_F^{\mathcal{S}_0 \to \mathcal{S}_1}}, \quad (2)$$

where the average $\langle \bullet \rangle_{F(R)}^{S_0 \to S_1(S_0 \to S_1)}$ is now restricted to forward (reverse) trajectories that start in partially equilibrated state S_0 (S_1) at λ_0 (λ_1) and end in S_1 (S_0) at λ_1 (λ_0). $p_F^{S_0 \to S_1}$ ($p_R^{S_0 \to S_1}$) stands for the probability to be in S_1 (S_0) at the end of the forward (reverse) process defined above, and $\Delta G_{S_0,\lambda_0}^{S_1,\lambda_1} = G_{S_1}(\lambda_1) - G_{S_0}(\lambda_0)$ is the free energy difference between partially equilibrated states S_0 and S_1 . In the following, we will drop the subscript (F, R), leaving the direction of the arrow to distinguish forward from reverse. Moreover, we will adopt the shorthand notation $\mathcal{P}_{S_0}^{S_1} \equiv p^{S_0 \to S_1}/p^{S_0 \leftarrow S_1}$. If $\mathcal{F} = 1$ we obtain a generalization of the Jarzynski equality, $\mathcal{P}_{S_0}^{S_1} \langle \exp[-\beta(W - \Delta G_{S_0,\lambda_0}^{S_1,\lambda_1})] \rangle = 1$. Whereas for the particular case $\mathcal{F} = \delta(W - W(\Gamma))$, we get the relation

$$\mathcal{P}_{\mathcal{S}_{0}}^{\mathcal{S}_{1}} \frac{P^{\mathcal{S}_{0} \to \mathcal{S}_{1}}(W)}{P^{\mathcal{S}_{0} \leftarrow \mathcal{S}_{1}}(-W)} = \exp[\beta(W - \Delta G_{\mathcal{S}_{0},\lambda_{0}}^{\mathcal{S}_{1},\lambda_{1}})], \quad (3)$$

which has been used in [8] in the case of global equilibrium initial conditions.

Experimental test.—Here we test the validity of Eq. (3) by performing single molecule experiments using optical tweezers. Let us consider an experiment where force is applied to the ends of a DNA hairpin that unfolds or refolds in a two-state manner. The conformation of the hairpin can be characterized by two states, the unfolded state (U) and the native or folded state (N)—see Fig. 1. The thermodynamic state of the molecule can be controlled by moving the position of the optical trap relative to a pipette (Fig. 2, upper panel). The relative position of the trap along the x axis defines the control parameter in our experiments, $\lambda =$ x. Depending on the value of x the molecule switches between the two states according to a rate that is a function of the instantaneous force applied to the molecule [10]. In a ramping protocol the value of x is changed at constant pulling speed from an initial value x_0 (where the molecule is always folded) to a final value x_1 (where the molecule is always unfolded) and the force f (measured by the optical trap) versus distance x curves recorded. By computing (i) the fraction of forward trajectories (i.e., increasing x) that go from $N (\equiv S_0)$ at x_0 to $U (\equiv S_1)$ at x and (ii) the fraction of reverse trajectories (decreasing extension) that go from U at x to N at x_0 , we can determine \mathcal{P}_N^U . Then by measuring the corresponding work values for each of these trajectories, we can use Eq. (3) to estimate the free energy



FIG. 1 (color online). A two-state molecule in a pulling experiment. (a) Schematic picture of the free energy landscape in a two-state folder as a function of the reaction coordinate. The blue and red colors represent the subsets of configurations that define the N and U states, respectively. (b) Depending on the value of the control parameter x, the shape of the free energy landscape is tilted toward one state (either N or U). For each value of x, the Boltzmann–Gibbs equilibrium value of the force restricted to each state defines the native (blue) and unfolded (red) force-distance branches. (c) During a ramping protocol, x is changed at a constant pulling speed from x_0 to x_1 (from x_1 to x_0 in the reverse case) and the molecule can visit both states as indicated by the colored circles. In order to measure $G_U(x)$ [all free energies are computed with respect to $G_N(x_0)$] at a given value of x (gray area), only forward and reverse trajectories that are in N at x_0 and in U at x have to be considered (which is true only for the reverse trajectory in the example shown in the picture).

of the unfolded branch $G_U(x)$ as a function of x. By repeating the same operation with N instead of U, the free energy of the folded branch $G_N(x)$ can be measured as well. Note that we adopt the convention of measuring all free energies with respect to the free energy $G_N(x_0)$ of the native state at x_0 . We are also able to compute the free energy difference between the two branches, $\Delta G_N^U(x) =$ $G_U(x) - G_N(x)$.

We have pulled a 20 bps DNA hairpin using a miniaturized dual-beam laser optical tweezers apparatus [11]. Molecules have been pulled at two low pulling speeds (40 and 50 nm/s) and two fast pulling speeds (300 and 400 nm/s), corresponding to average loading rates ranging between 2.6 and 26 pN/s, from $x_0 = 0$ to $x_1 = 110.26$ nm (for convenience we take the initial value of the relative distance trap-pipette equal to 0). A few representative force-distance curves are shown in Fig. 2 (inset of lower panel). We have then selected a value $x = \bar{x} = 61.75$ nm,



FIG. 2 (color online). (Upper panel) Experimental setup. (Lower panel) Unfolding and refolding work distributions $P^{N \to U}(W)$, $P^{N \to U}(-W)$ at $x = \bar{x} = 61.75$ nm measured at two pulling speeds: 300 nm/s (red, unfolding and green, refolding) and 40 nm/s (blue, unfolding and orange, refolding). In the inset we show two force-distance cycles corresponding to low (40 nm/s) and fast (300 nm/s) pulling speeds. The gray area indicates the mechanical work $(=\int_{\bar{x}}^{x_0} f dx)$ exerted along the green refolding trajectory. Statistics (number of molecules, total number of unfolding/refolding cycles): 40 nm/s (2, 223), 50 nm/s (2, 183), 300 nm/s (3, 337), 400 nm/s (1, 551). More details about the data analysis procedure can be found in the appendices [9].

close to the expected coexistence value of x where both N and U states have the same free energy (see below). Such value of x is chosen in order to have good statistics for the evaluation of the unfolding and refolding work distributions. The system is out of equilibrium at the four pulling speeds. To extract the free energy of the unfolded branch $G_U(\bar{x})$, we have measured the work values $W^{N \to U} =$ $\int_{x_0}^{\bar{x}} f dx, W^{N \leftarrow U} = \int_{\bar{x}}^{x_0} f dx \text{ along the unfolding and refold-}$ ing trajectories, respectively, and then determined the distributions $P^{N \to U}(W)$ and $P^{N \leftarrow U}(-W)$. In the main panel of Fig. 2 we show the work distributions obtained for a slow and fast pulling process. Note that the support of the unfolding work distributions is bounded by the maximum amount of work that can be exerted on a molecule between x_0 and \bar{x} . This bound corresponds to the work of those unfolding trajectories that have never unfolded before reaching \bar{x} .

As a direct test of the validity of Eq. (3), in the upper panel of Fig. 3 we plot the quantity $\log(P^{N \to U}(W)/P^{N \to U}(-W)) + \log \mathcal{P}_N^U(\bar{x})$ against W (in k_BT units). As expected, all data fall into straight lines of slope close to 1. The intersections of such lines with the W axis provide an estimate of $G_U(\bar{x})$. Note that both fast and slow pulling speeds intercept the horizontal axis around the same value within $0.75k_BT$ of error. With this method, we estimate $G_U(\bar{x}) = 185.9(5)k_BT$. Note that this free energy estimate, as all the others in this Letter, refers to the whole system, comprising hairpin, handles, and trap.

A more accurate test of the validity of Eq. (3) and a better estimation [12] of the free energy $G_U(\bar{x})$ can be obtained through the Bennett acceptance ratio method [13]. In Bennett's method we define the following functions:

$$z_{\text{UNF}}(u) = \log \left\langle \frac{1}{1 + \frac{n_{\text{UNF}}}{n_{\text{REF}}}} \exp[\beta(W - u)] \right\rangle_{\text{UNF}},$$
$$z_{\text{REF}}(u) = \log \left\langle \frac{\exp(-W)}{1 + \frac{n_{\text{UNF}}}{n_{\text{REF}}}} \exp[-\beta(W + u)] \right\rangle_{\text{REF}}.$$

From Eq. (3) it has been proved [12,13] that the solution of the equation z(u) = u, where

$$z(u) \equiv z_{\text{REF}}(u) - z_{\text{UNF}}(u) - \log \mathcal{P}_N^U(\bar{x}), \qquad (4)$$

is the optimal (minimal variance) estimate of $G_U(\bar{x})$.

In Fig. 3 (lower panel) we plot the function z(u) for different pulling speeds. It is quite clear that the functions z(u) are approximately constant along the u axis and cross



FIG. 3 (color online). Experimental verification of the fluctuation relation Eq. (3). (Upper panel) Plot of $\log \mathcal{P}_N^U(\bar{x}) + \log(\mathcal{P}^{N \to U}(W)/\mathcal{P}^{N \leftarrow U}(-W))$ as a function of W (in $k_B T$ units) at different pulling speeds. A least squares fitting method of the experimental data to straight lines gives the following slopes: 0.91(4) (400 nm/s), 1.07(8) (300 nm/s), 1.02(6) (50 nm/s), 1.07(3) (40 nm/s). (Lower panel) Bennett's acceptance ratio method. We plot the function z(u) Eq. (4) for different pulling speeds. The thick black line corresponds to the function z = u. Error bars correspond to the standard deviation of the Bennett estimate, as determined in Ref. [12]. Data statistics are reported in the caption of Fig. 2.



FIG. 4 (color online). Free energy of the folded and unfolded branches. The purple line is $G_N(x)$, the dark green line is $G_U(x)$, both computed with respect to $G_N(x_0)$. The black solid line and the gray area mark the expected value $x_c = 62.0(7)$ nm and its standard error, while the dashed black line is the crossing point $x_c \approx 61.28$ nm. Both left and right insets show a magnified image of the coexistence region. In the right one, the dashed lines represent the free energies that we would obtain if we had overlooked the factor \mathcal{P}_N^U . The top panel shows the functions $\Delta G_N^U(x) - \Delta G_U^U(x)$ and $\Delta G_N^N(x) - \Delta G_U^N(x)$, which, by definition, should be independent of x and equal to $G_U(x_1)$. Again, the dashed lines represent the result if we do not include the factor \mathcal{P}_N^U .

the line z = u around the same value $G_U(\bar{x}) = 186.0(3)k_BT$. A distinctive aspect of Eq. (3) is the presence of the factor \mathcal{P}_N^U . If such correction was not taken into account then the fluctuation relation would not be satisfied anymore. We have verified that if \mathcal{P}_N^U is not included in the analysis, then the Bennett acceptance ratio method gives free energy estimates that depend on the pulling speed [9].

Free energy branches.--After having verified that the fluctuation relation Eq. (3) holds and that it can be used to extract the free energy of the unfolded $(G_U(\bar{x}))$ and folded $(G_N(\bar{x}), \text{ data not shown})$ branches, we have repeated the same procedure in a wide range of x values. The range of values of x is such that at least 8 trajectories go through N(or U) (ensuring that we get a reasonable statistical significance). The results of the reconstruction of the free energy branches are shown in Fig. 4. The two free energy branches cross each other at a coexistence value x_c at which the two states (N and U) are equally probable. This is defined by $\Delta G_N^U(x_c) = 0$. We get $x_c \approx 61.28$ nm, which is in good agreement with another estimate, $x_c = 62.0(7)$ nm, obtained interpreting experimental data according to a simple phenomenological model [9]. In the insets of Fig. 4 we zoom the crossing region. It is interesting to recall again the importance of the aforementioned correction term (\mathcal{P}_N^U) to Eq. (3). If such term is not included in the analysis then the two reconstructed branches never cross (bottom right inset). This result is incompatible with the existence of the unfolding/refolding transition in the hairpin, showing that the factor \mathcal{P}_N^U is key to measure free energy branches.

Equation (2) is valid in the very general situation of partially equilibrated initial states which, however, are arbitrarily far from global equilibrium. This makes the particular case Eq. (3) a very useful identity to recover the free energy of states that cannot be observed in conditions of thermodynamic global equilibrium. We have shown how it is possible to apply Eq. (3) to recover free energy differences of thermodynamic branches of folded and unfolded states in a two-state DNA hairpin. These methods can be further extended to the recovery of free energies of non-native states such as misfolded or intermediates states.

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