Driving an electrolyte through a corrugated nanopore

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ABSTRACT

We characterize the dynamics of a z - z electrolyte embedded in a varying-section channel. In the linear response regime, by means of suitable approximations, we derive the Onsager matrix associated with externally enforced gradients in electrostatic potential, chemical potential, and pressure, for both dielectric and conducting channel walls. We show here that the linear transport coefficients are particularly sensitive to the geometry and the conductive properties of the channel walls when the Debye length is comparable to the channel width. In this regime, we found that one pair of off-diagonal Onsager matrix elements increases with the corrugation of the channel transport, in contrast to all other elements which are either unaffected by or decrease with increasing corrugation. Our results have a possible impact on the design of blue-energy devices as well as on the understanding of biological ion channels through membranes.

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I. INTRODUCTION

Many biological systems¹ and synthetic devices² rely on the dynamics of electrolytes confined within micropores and nanopores.^{3–6} For example, ion channels,^{7,8} membranes,^{9,10} neuron signaling,¹ plant circulation,¹¹ and lymphatic¹² and interstitial¹³ systems rely on the transport of electrolytes across tortuous micropores and nanopores. Recent technological advances have led to the realization of nanotubes and nanopores of controllable shape^{14,15} that have been exploited to separate DNA,¹⁶ proteins, or colloids.¹⁷ Likewise, resistive-pulse sensing techniques have been developed to measure properties of tracers transported across charged nanopores.¹⁸⁻²¹ Moreover, electrolyte-immersed electrodes have been characterized^{22,23} and realized for novel energy-harvesting devices.²⁴ Recently, it has been shown that novel dynamical regimes appear when the section of the confining vessel is not constant. Indeed, asymmetric pores have been used to pump²⁵ and to rectify ionic currents.^{26–30} Moreover, recirculation has been reported

for electrolytes confined between corrugated walls,^{31–34} and the variation in the section of channels can tune their permeability.^{35,36} When an electrolyte is driven inside such conduits (see Fig. 1), the local variations in the available space will couple to the local charge and ionic density distribution leading to modulations in the mesoscopic properties of the electrolyte such as the electrostatic decay length.

In this article, we show that analytical insight into such corrections can be obtained for smoothly-varying channel sections. In this scenario, we exploit the lubrication approximation and we derive closed expressions for the geometrically-induced corrections to the local electrostatic potential, charge, and ionic density distributions, and we identify the fluxes driven by weak external driving forces through applied electric fields, pressure, or salt concentration differences for both conducting and dielectric channel walls. While for constant section channels the transport coefficients are unaffected by the wall properties, for varying section channels, we show that the transport coefficients are generally larger for dielectric walls.



FIG. 1. Schematic view of the system. The green regions are the channel walls; the red stripe represents the region, of size Debye length, where the electrostatic potential decays; and the blue stripes represent the channel walls.

Moreover, upon increasing the corrugation of the channel, we show that, as expected, the transport coefficients generally decrease. However, for some specific cases, we find an increase in the transport coefficients upon increasing the corrugation of the channel. We stress that our results strongly depend on the fact that the Debye length is comparable to the channel section, i.e., on the entropic electrokinetic regime. Indeed, in the case in which the channel section is much larger than the channel section of previous results,³⁷ the phenomena that we observe are absent.

In order to properly cast our results, i.e., to show the relevance of the entropic electrokinetic regime, and in particular of the shape of the channel, we have decided to rederive some known results about the transport of electrolytes by accounting, for the first time, for both the varying section of the channel and the partial overlap of the Debye double layers. This self-consistent derivation will allow us to put our novel results in their natural context. The structure of the text is as follows. In Sec. II, we introduce our model setup and the framework of electrokinetic equations that we use to describe solute and solvent fluxes. In Sec. III, we determine the reference equilibrium scenario. In Sec. IV, we derive the linear transport coefficients of a channel driven out of equilibrium and show that the corresponding Onsager matrix is symmetric. The Onsager matrix encodes for many physical scenarios of possible experimental interest. We discuss several of those scenarios in Sec. V. In particular, in Secs. V B 2 and V C 1, we discuss the enhancement of the transport coefficients upon increasing the channel corrugation. Finally, in Sec. VII, we present our conclusions.

II. MODEL

A. Setup

Throughout this article, we analyze a microfluidic channel filled with a z - z electrolyte in a solvent of dielectric constant ϵ . The solvent is incompressible and has a viscosity η . The channel of length L along the *x*-direction is translationally invariant in the *z*-direction and has a varying pore width in the *y*-direction: the channel half-section h(x) depends solely on *x*. We write $\bar{h} \equiv (1/L) \int_0^L h(x) dx$ for the average pore section.

At x = 0 and x = L, the channel is in contact with two chemostats at electrochemical potentials $\mu^{\pm}(x = 0)$ and $\mu^{\pm}(x = L)$, respectively.

Next to differences in these chemical potentials on either side of the channel, also a pressure difference ΔP or a potential difference ΔV can be applied. We only consider isothermal systems at temperature *T*, which means that local heat generation³⁸ is neglected. Moreover, our model does not account for surface conduction in the Stern layer.³⁹

B. Electrokinetic equations

Under the above described conditions, the steady state of our system can be modeled by the classical electrokinetic equations,

$$\epsilon \nabla^2 \psi(x, y) = -zeq(x, y), \tag{1a}$$

$$\mathbf{j}^{\pm}(x,y) = \rho^{\pm}(x,y) \big[\mathbf{v}(x,y) - D\beta \nabla \mu^{\pm}(x,y) \big], \qquad (1b)$$

$$\nabla \cdot \mathbf{j}^{\pm}(x, y) = 0, \tag{1c}$$

$$\eta \nabla^2 \mathbf{v}(x, y) = -\mathbf{F}_{\text{tot}}(x, y) + \nabla P_{\text{tot}}(x), \tag{1d}$$

$$\nabla \cdot \mathbf{v}(x, y) = 0. \tag{1e}$$

First, the electrostatic potential ψ inside the channel is determined by the Poisson equation (1a) with *e* being the elementary charge and with q(x, y) being the local charge number density (m⁻³), which is nonzero whenever there is a difference between cationic and anionic number densities (ρ^+ and ρ^- , respectively),

$$q(x,y) = \rho^+(x,y) - \rho^-(x,y).$$
 (2)

Second, we model the ionic currents \mathbf{j}^{\pm} by the Nernst-Planck equation (1b), which accounts for ionic transport by advection, diffusion, and electromigration. Equation (1b) describes the dynamics of pointlike ions: this represents a fair approximation for dilute electrolytes in small electric fields. Third, when the system is driven by an external force such as a pressure drop or an electrostatic field, the electrolyte solution will flow—at a low Reynolds number—according to the Stokes equation (1d), with $\partial_x P_{\text{tot}}(x) = \partial_x P(x) + \Delta P/L$, where $\partial_x P$ is the *x*-component of the geometrically induced local pressure gradient that is determined by the boundary conditions and by fluid incompressibility and where

$$F_{\text{tot}}(x,y) = -zeq(x,y)\partial_x\psi(x,y)$$
(3)

is the *x*-component of the total electrostatic force density acting on the fluid. Finally, Eqs. (1c) and (1e) represent the steady-state continuity equation and the incompressibility equation, respectively.

Equations (1) are subject to the following boundary conditions for conducting (cond) and dielectric (diel) walls:

$$\psi \to \begin{cases} \psi(x, \pm h(x)) &= \zeta \quad \text{cond,} \\ \mathbf{n}^{\pm} \cdot \nabla \psi(x, y)|_{y=\pm h(x)} &= \pm e\sigma \quad \text{diel,} \end{cases}$$
(4a)

$$\mathbf{n}^{\pm} \cdot \mathbf{j}^{\pm}(x, \pm h(x)) = 0, \tag{4b}$$

$$\mathbf{v}(x,\pm h(x)) = 0, \tag{4c}$$

where \mathbf{n}^{\pm} is the local normal at the channel walls. Concerning the boundary conditions on the velocity, we assume no-slip. In the case of water as a solvent, this amounts to considering hydrophilic surfaces. For hydrophobic surfaces, the partial slip can induce additional effects^{40,41} that we do not consider here. Here, the boundary conditions on $\psi(x, y)$ depend on the conductive properties of the

channel walls. For dielectric channel walls, the case for pores made from polymeric materials such as polydimethylsiloxane (PDMS), we impose a constant surface charge $e\sigma$, whereas for conducting walls, such as carbon nanotubes, we impose a constant ζ potential. We denote the electrostatic potential in either case $\psi^{\zeta}(x, y)$ or $\psi^{\sigma}(x, y)$, accordingly. Note that while for flat channels either choice is related via the capacitance, in the corrugated case, a constant ζ potential leads to an *x*-dependent surface charge $e\sigma(x)$ [cf. Eq. (17)], while a constant surface charge gives rise to a varying surface potential $\psi^{\sigma}(x, h(x))$ [cf. Eq. (14)]. The same superscript notation is also used for other variables whenever we specify quantities to either boundary condition. Equations (4b) and (4c) represent the no-flux and no-slip boundary conditions at the channel walls of the solute and solvent, respectively.

C. Lubrication approximation

In the following, we restrict to pores whose section varies smoothly. This allows us to identify a separation between longitudinal and transverse length scales according to which changes in ψ and v_x along the x-direction are much smaller than those along the y-direction. This facilitates an essential simplification of Eqs. (1a) and (1d) where ∂_x^2 terms therein become negligible as compared to ∂_{ν}^2 terms. Thanks to this "lubricationlike" approximation, both Eqs. (1a) and (1d) become analytically solvable. In order to apply the lubrication approximation consistently to both the Stokes and Poisson equations, we need to identify a common small parameter. While the relevant longitudinal length scale of both the Stokes equation and the Poisson equation is the channel length L, different transverse length scales appear in these equations: the average channel section *h*, for the Stokes equation, and the screening length λ , for the Poisson equation. To proceed, we nondimensionalize the length scales via $x = x_*L$ and $h(x) = h_*(x)h$, while for the transverse direction we use either $y = y_* h$ or $y = y_* \lambda$. We then write the Stokes equation as

$$\frac{\tilde{h}^2}{L^2}\frac{\partial^2 v_x}{\partial x_*^2} + \frac{\partial^2 v_x}{\partial y_*^2} = \frac{\tilde{h}^2}{\eta} \left[-F_{\text{tot}}(x, y) + \partial_x P_{\text{tot}}(x) \right]$$
(5)

and the Poisson equation as

$$\frac{\lambda^2}{\bar{h}^2} \frac{\bar{h}^2}{L^2} \left[\frac{\partial^2 \psi}{\partial h_*^2} \left(\frac{\partial h_*}{\partial x_*} \right)^2 + \frac{\partial \psi}{\partial h_*} \frac{\partial^2 h_*}{\partial x_*^2} \right] + \frac{\partial^2 \psi}{\partial y_*^2} = -\lambda^2 \frac{zeq}{\epsilon}.$$
 (6)

A *first-order* lubrication approximation to Eq. (5) in the small parameter $\bar{h}/L \ll 1$ amounts to dropping the term of order $\mathcal{O}(\bar{h}^2/L^2)$ (first term on the left-hand side). Similarly, in Eq. (6), we neglect the term of order $\mathcal{O}(\lambda^2/L^2)$ (first term on the left-hand side), requiring the smallness of λ/L ; hence, this term is of $\mathcal{O}(\bar{h}/L)^2$ as compared to the second one, provided that $\lambda^2/\bar{h}^2 \lesssim \mathcal{O}(1)$.

We have exploited the nondimensionalized equations (5) and (6) to identify the magnitude of different terms when the pore section is smoothly varying. However, since in the following we are going to make expansions in several small parameters, we continue our analysis with the dimensionful equations. This approach has the advantage that it allows us to keep track of all these small parameters on equal footing.

III. EQUILIBRIUM

At equilibrium, the electrochemical potential⁴² is constant,

$$\beta \mu_{\rm eq}^{\pm} = \ln \left[\Lambda_{\pm}^3 \rho^{\pm}(x, y) \right] \pm \beta z e \psi(x, y), \tag{7}$$

with $\beta = (k_B T)^{-1}$ being the inverse thermal energy and Λ_{\pm} being the cationic and anionic thermal de Broglie wavelengths, which we consider to be equal, $\Lambda_{\pm} = \Lambda_{-} = \Lambda$. This implies that

$$\rho^{\pm}(x,y) = \varrho^{\pm}(x) \exp\left[\mp\beta z e \psi(x,y)\right],\tag{8}$$

with

$$q^{\pm}(x) = \frac{\exp\left[\beta\mu_{\rm eq}^{\pm}\right]}{\Lambda^3}.$$
(9)

In order to get analytical insight, we assume that the electrostatic potential is weak, $\beta z e \psi(x, y) \ll 1$, i.e., we apply the Debye-Hückel approximation. For later convenience, we retain contributions up to second order in $\beta z e \psi(x, y)$; hence, the number densities of positive and negative ions read

$$\frac{\rho^{\pm}(x,y)}{\varrho^{\pm}} = 1 \mp \beta z e \psi(x,y) + \frac{1}{2} (\beta z e \psi(x,y))^2 + \mathcal{O}(\psi^3).$$
(10)

In order to simplify the notation, we choose the zero of the electrostatic potential such that we have $\mu_{eq}^{+} = \mu_{eq}^{-} \equiv \mu_{eq}$ when the z - z electrolyte is globally electroneutral in the reservoirs. Hence, we have

$$\varrho^+ = \varrho^- \equiv \varrho. \tag{11}$$

From here on, we denote the expansion of a general variable X(x, y) in the small parameter \tilde{h}/L as $X = X_0 + X_1 + \mathcal{O}(\tilde{h}/L)^2$; hence, for instance, $\psi(x, y) = \psi_0(x, y) + \mathcal{O}(\tilde{h}/L)^2$ and $\rho^{\pm}(x, y) = \rho_0^{\pm}(x, y) + \mathcal{O}(\tilde{h}/L)^2$. Accordingly, at leading order in the lubrication expansion, we retain only the first terms of the above expansions and Eq. (1a) reads

$$\partial_{y}^{2}\psi_{0}(x,y) = k_{0}^{2}\psi_{0}(x,y) + \mathcal{O}(\psi_{0}^{3}), \qquad (12)$$

where $k_0 = \sqrt{\beta(ze)^2 \gamma_0/\epsilon}$ is the inverse Debye length and $\gamma_0 = 2\varrho_0$ is the salt number density. In order to keep the notation as simple as possible, from here on, we omit $\mathcal{O}(\psi_0^n)$ and we reintroduce it only when necessary. Finally, the electrostatic potential for conducting channel walls reads

$$\psi_0^{\zeta}(x,y) = \zeta \frac{\cosh[k_0 y]}{\cosh[k_0 h(x)]},\tag{13}$$

while for dielectric walls it reads

$$\psi_0^{\sigma}(x,y) = \frac{e\sigma}{\epsilon k_0} \frac{\cosh[k_0 y]}{\sinh[k_0 h(x)]}.$$
(14)

While we enforced global electroneutrality [cf. above Eq. (11)], local electroneutrality—the balance of the total ionic charge $ze\bar{q}(x)$ in a slab located at x by a corresponding amount of opposite local surface charge $2e\sigma(x)$ —can now be discussed. Here, $\bar{q}(x)$ is the cross-sectional total unit charge,

$$\bar{q}(x) = \int_{-h(x)}^{h(x)} q(x, y) \, \mathrm{d}y.$$
(15)

For conducting walls, at lowest order in lubrication, this amounts with Eqs. (10) and (13) to

$$\bar{q}_0^{\zeta}(x) = -2\epsilon k_0 \zeta \tanh[k_0 h(x)]. \tag{16}$$

We remark that, at first order in lubrication, the surface charge at each conducting wall can be obtained by

$$e\sigma^{\zeta}(x) = \pm \epsilon \partial_{y} \psi_{0}^{\zeta}(x, y)|_{y=\pm h(x)} = \epsilon k_{0} \zeta \tanh[k_{0}h(x)].$$
(17)

For dielectric walls, we have

$$\bar{q}_0^{\sigma}(x) = -2e\sigma. \tag{18}$$

Equations (16)–(18) show that local charge neutrality is attained. The presented theory is thus not able to reproduce the recently discovered electroneutrality breaking in narrow confinement.⁴³ To account for that, the authors of Refs. 43 and 44 had to include additional interactions beyond the ones of our model.

IV. TRANSPORT

From here on, we characterize the electrolyte-filled corrugated nanochannel driven out of equilibrium by applied external forces $\Delta P/L$, $ze\Delta V/L$, and $\Delta \mu/L$. We assume these external forces to be small, which means that $\beta LL_z h \Delta P \ll 1$, $\beta ze\Delta V \ll 1$, and $\beta \Delta \mu \ll 1$, where L_z is the thickness of the channel along the *z* direction.

A. Stokes

At leading order in lubrication, the solution $v_x(x, y)$ of the Stokes equation [Eq. (1d)] subject to no-slip boundary conditions [Eq. (4c)] reads

$$v_x(x,y) = u_P(x,y) + u_{eo}(x,y),$$
 (19a)

$$u_P(x,y) = \frac{\partial_x P_{\text{tot}}(x)}{2\eta} \left[y^2 - h(x)^2 \right], \tag{19b}$$

$$u_{\rm eo}(x,y) = \mathcal{U}(x,y) - \mathcal{U}(x,h(x)), \tag{19c}$$

$$\mathcal{U}(x,y) \equiv \frac{2e}{\eta} \int dy \int dy q(x,y) \partial_x \psi(x,y), \qquad (19d)$$

where we partitioned the velocity $v_x(x, y)$ into a pressure-driven contribution u_P and an electroosmotic contribution u_{eo} that arises when ions in an electric field drag along the solvent. The local pressure gradient appearing in Eq. (19b), $\partial_x P_{tot}(x) = \partial_x P(x) + \Delta P/L$, accounts for both the pressure drop ΔP from x = L to x = 0 and for the local pressure P(x), which ensures fluid incompressibility [Eq. (1e)]. Inserting Eq. (19) into the volumetric fluid flow,

$$Q = \int_{-h(x)}^{h(x)} v_x(x, y) \, \mathrm{d}y,$$
 (20)

and performing the *y*-integral over u_P leads to an expression for $\partial_x P_{\text{tot}}(x)$,

$$\frac{2}{3\eta}\partial_x P_{\text{tot}}(x) = \frac{1}{h^3(x)} \left[\int_{-h(x)}^{h(x)} u_{\text{eo}}(x,y) \, \mathrm{d}y - Q \right].$$
(21)

Integrating the last expression over $\int_0^L dx$, imposing fluid incompressibility $\partial_x Q = 0$, and using

$$\int_0^L \partial_x P_{\text{tot}}(x) \, \mathrm{d}x = \Delta P, \qquad (22)$$

which follows from the boundary conditions on the pressure, leads to

$$Q \equiv Q_P + Q_{\rm eo},\tag{23a}$$

$$Q_P = -\frac{2}{3H_3} \frac{\dot{h}^3 \Delta P}{\eta L},$$
(23b)

$$Q_{\rm eo} = \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{\mathrm{d}x}{h^3(x)} \int_{-h(x)}^{h(x)} u_{\rm eo}(x, y) \,\mathrm{d}y, \qquad (23c)$$

where Q_P is the pressure-driven volumetric fluid flow, Q_{eo} is the electroosmotic flow, and

$$H_{3} \equiv \frac{\bar{h}^{3}}{L} \int_{0}^{L} \frac{1}{\bar{h}^{3}(x)} dx$$
 (24)

is a dimensionless geometrical measure for the corrugation of the channel. We find $H_3 \ge 1$, with the equality holding when the channel is flat $[h(x) = \bar{h}]$. Hence, for a flat channel, Eq. (23b) simplifies to the standard result $Q_P = -2\bar{h}^3\Delta P/(3\eta L)$ of a Poiseuille flow between two flat plates. Finally, in order to determine $u_{\rm eo}$ (and $Q_{\rm eo}$), we need to characterize the ionic transport.

B. Small-force expansions

Within the Debye-Hückel regime and at first order in lubrication, we expand the nonequilibrium electric potential, charge densities, and electrochemical potential, for weak external forces, *f*, about their equilibrium values,

$$\psi(x,y) = \psi_0(x,y) + \psi_{0,f}(x,y) + \mathcal{O}(f^2) + \mathcal{O}(\psi_0^3), \qquad (25a)$$

$$\rho^{\pm}(x,y) = \rho_0^{\pm}(x,y) + \rho_{0,f}^{\pm}(x,y) + \mathcal{O}(f^2) + \mathcal{O}(\psi_0^2).$$
(25b)

Here, both ψ_0 and ρ_0^{\pm} carry corrections of $\mathcal{O}(\psi_0^3)$. Hence, this expansion for small values of f about the Debye-Hückel solution is meaningful provided that contributions of order $\mathcal{O}(f)$ are larger than those of order $\mathcal{O}(\psi_0^3)$. For notation ease, in all $\mathcal{O}(f)$ terms that we write from here on, we drop mentioning the lubrication approximation, in particular $\rho_{0,f}^{\pm} \rightarrow \rho_f^{\pm}$ and $\psi_{0,f} \rightarrow \psi_f$. Inserting Eq. (25b) into Eq. (7), we find an expansion of the chemical potential, $\mu^{\pm}(x, y) = \mu_0^{\pm} + \mu_f^{\pm}(x, y) + \mathcal{O}(f^2)$, with

$$\beta \mu_f^{\pm}(x,y) = \frac{\rho_f^{\pm}(x,y)}{\rho_0^{\pm}(x,y)} \pm \beta z e \psi_f(x,y).$$
(26)

Assuming a small transverse Peclet number $(hv_y/D \ll 1)$, the steady state is achieved by systems that are in *local equilibrium* $\partial_y \mu(x, y) = 0$ in every section of the channel located at *x*. Accordingly, using Eq. (10) leads, at linear order in ψ_0 , to the density profiles

$$\rho_f^{\pm}(x,y) = \varrho_0 \Big[\beta \bar{\mu}_f^{\pm}(x) \mp \beta z e \psi_f(x,y) \Big] \big[1 \mp \beta z e \psi_0(x,y) \big].$$
(27)

With μ_f^{\pm} , we define the intrinsic (electro)chemical potential as $\bar{\mu}_f^{\pm}(x) \equiv \mu_f^{\pm}(x, y)$. It is important to remark that the contribution contained in Eq. (27) is of lower order than those disregarded in Eq. (25b) and that those contributions disregarded in

Eq. (27) are of the same (or higher) order as those disregarded in Eq. (25b).

C. Transport equations

The steady-state continuity equation (1c), together with the noflux boundary condition [Eq. (4b)], implies the *x*-independence of the following cross-sectional integrals:

$$J^{\pm} = \int_{-h(x)}^{h(x)} j_x^{\pm}(x, y) \, \mathrm{d}y, \tag{28}$$

which represent the total ionic fluxes through a slab at x. In Appendix A, we find expressions for the solute $J_c = J^+ + J^-$ and charge $J_q = J^+ - J^-$ fluxes by inserting Eqs. (1b) and (27) into Eq. (28),

$$\frac{J_c}{D} = \frac{\gamma_0 Q}{D} + \beta z e \overline{\psi_0}(x) \partial_x \xi_f(x) - 2h(x) \partial_x \gamma_f(x) + \mathcal{O}(f^2), \qquad (29a)$$

$$\frac{J_q}{D} = \frac{\mathcal{J}_q(x)}{D} + \beta z e \overline{\psi_0}(x) \partial_x \gamma_f(x) - 2h(x) \partial_x \xi_f(x) + \mathcal{O}(f^2), \quad (29b)$$

where $\gamma_f(x)$, $\xi_f(x)$, $\overline{\psi_0}(x)$, and $\mathcal{J}_q(x)^{55}$ are defined as

$$\gamma_f(x) = \varrho_0 \beta \left[\bar{\mu}_f^+(x) + \bar{\mu}_f^-(x) \right],$$
 (30a)

$$\xi_f(x) = \varrho_0 \beta \Big[\bar{\mu}_f^+(x) - \bar{\mu}_f^-(x) \Big],$$
(30b)

$$\overline{\psi_0}(x) \equiv \int_{-h(x)}^{h(x)} \psi_0(x, y) \,\mathrm{d}y, \qquad (30c)$$

$$\mathcal{J}_{q}(x) \equiv \int_{-h(x)}^{h(x)} q_{0}(x, y) v_{x}(x, y) \, \mathrm{d}y.$$
(30d)

We now proceed as follows: from Eq. (29), we will derive expressions for $\gamma_f(x)$ and $\xi_f(x)$ in terms of the fluxes J_c , J_q , and Q [cf. Eqs. (33) and (34)]. Since $\gamma_f(x)$ and $\xi_f(x)$ are defined in terms of the intrinsic chemical potentials $\bar{\mu}_f^{\pm}(x)$, which must adhere to externally enforced boundary values $\bar{\mu}_f^{\pm}(0)$ and $\bar{\mu}_f^{\pm}(L)$, these expressions can, in turn, be inverted to yield the fluxes in terms of driving forces. To do all that, we start by rewriting Eq. (29b),

$$\partial_x \xi_f(x) = \frac{\beta z e \overline{\psi_0}(x)}{2h(x)} \partial_x \gamma_f(x) + \frac{\mathcal{J}_{q,f}(x) - J_{q,f}}{2Dh(x)}, \tag{31}$$

where $J_q = J_{q,f} + \mathcal{O}(f^2)$ and $\mathcal{J}_q = \mathcal{J}_{q,f} + \mathcal{O}(f^2)$. With a slight abuse of notation, we drop the subscript f in $J_{q,f}$ and $\mathcal{J}_{q,f}$ from here on, and we will do the same for $J_{c,f}$ and Q_f , which are defined analogously to $J_{q,f}$ and $\mathcal{J}_{q,f}$. As usual,⁴⁶ instead of J_c , from here on we will consider the "excess" solute flow not caused by advection, $J'_c = J_c - \gamma_0 Q$. Inserting Eq. (31) into Eq. (29a), we find

$$\partial_x \gamma_f(x) = -\frac{1}{2Dh(x)} \left[J'_c + J_q \frac{\beta z e \overline{\psi_0}(x)}{2h(x)} \right], \tag{32}$$

which upon integrating yields

$$\gamma_f(x) = \gamma_f(0) - \frac{J'_c}{2D} \int_0^x \frac{dx'}{h(x')} - \frac{J_q}{4D} \int_0^x \frac{\beta z e \overline{\psi_0}(x')}{h^2(x')} dx'.$$
 (33)

Similarly, substituting Eq. (32) into Eq. (31) and integrating, at leading order in ψ_0 , yields

$$\xi_{f}(x) = \xi_{f}(0) - \frac{J_{q}}{2D} \int_{0}^{x} \frac{dx'}{h(x')} - \frac{J_{c}'}{4D} \int_{0}^{x} \frac{\beta z e \overline{\psi_{0}}(x')}{h^{2}(x')} dx' + \frac{1}{2D} \int_{0}^{x} \frac{\mathcal{J}_{q}(x')}{h(x')} dx'.$$
(34)

Evaluating the above two equations at x = L gives

$$\frac{\Delta \gamma}{L} = -\frac{H_1}{2D\bar{h}} \left[J_c' + J_q \Phi \Upsilon_1 \right], \tag{35a}$$

$$\frac{\Delta\xi}{L} = -\frac{H_1}{2D\bar{h}} \left[J_q + J_c' \Phi \Upsilon_1 \right] + \frac{\Phi \Upsilon_3 \gamma_0}{Dk_0^2 \eta} \frac{\Delta P}{L},$$
(35b)

where we defined $\Delta \gamma = \gamma_f(L) - \gamma_f(0)$ and $\Delta \xi = \xi_f(L) - \xi_f(0)$, and where we used the following new functions:

$$H_1 \equiv \frac{\bar{h}}{L} \int_0^L \frac{1}{h(x)} \,\mathrm{d}x,\tag{36a}$$

$$\Phi \equiv \beta z e \times \begin{cases} \zeta & \text{cond,} \\ \frac{e\sigma}{\epsilon k_0} & \text{diel,} \end{cases}$$
(36b)

$$\Upsilon_1 \equiv \frac{\bar{h}}{H_1 L} \int_0^L dx \, \frac{\beta z e \overline{\psi}_0(x)}{2 h^2(x) \Phi},\tag{36c}$$

$$\Upsilon_{3} \equiv \frac{\bar{h}^{3}}{H_{3}L} \int_{0}^{L} dx \, \frac{\beta z e[2h(x)\psi_{0}(x,h(x)) - \overline{\psi}_{0}(x)]}{2 \, h^{4}(x)\Phi}.$$
 (36d)

First, similar to H_3 , H_1 is a measure for the corrugation of the channel. Second, $\Phi/(\beta z e)$ equals the surface potential $\psi^{\zeta}(x, h(x))$ for conducting walls, while for dielectric surfaces it differs from $\psi^{\sigma}(x, h(x))$ by a factor $\coth[k_0h(x)]$ [cf. Eq. (14)]. Third, the Y functions are dimensionless and depend solely on the parameter $k_0\bar{h}$ and the channel shape h(x). We report their functional dependence on these parameters for both boundary conditions in Eq. (42). Finally, the Y₃ term in Eq. (35b) stems from the $\mathcal{J}_q(x)$ term in Eq. (34) [see Appendix B].

D. Onsager matrix

Reshuffling Eq. (35) gives

$$J_c' = -J_q \Phi \Upsilon_1 - \frac{2D\bar{h}}{H_1} \frac{\Delta \gamma}{L},$$
(37a)

$$J_q = -J'_c \Phi \Upsilon_1, -\frac{2D\bar{h}}{H_1} \frac{\Delta \xi}{L} + \frac{2\Phi \Upsilon_3}{H_1} \frac{\gamma_0 \bar{h}}{k_0^2 \eta} \frac{\Delta P}{L}.$$
 (37b)

Using the formalism developed in this section, in Appendix C, we determine the missing piece of Q [Eq. (23)],

$$Q = Q_P - J_q \frac{\Phi \Upsilon_3}{D\beta\eta k_0^2},\tag{38}$$

for both electric boundary conditions [cf. Eqs. (C16a) and (C16b)], provided that the channel satisfies h(0) = h(L). Inserting Eq. (37a) into Eq. (37b), Eq. (37b) into Eq. (37a), and Eq. (37b) into Eq. (38), at leading order in Φ , leads to

$$J_q = -\frac{2D\bar{h}\beta\gamma_0}{H_1}\frac{\Delta\xi}{\beta\gamma_0L} + 2\Phi\Upsilon_1\frac{D\bar{h}}{H_1}\frac{\Delta\gamma}{L} + \frac{2\Phi\Upsilon_3}{H_1}\frac{\gamma_0\bar{h}}{k_0^2\eta}\frac{\Delta P}{L},\qquad(39a)$$

$$J_c' = 2\Phi \Upsilon_1 \frac{D\bar{h}\beta\gamma_0}{H_1} \frac{\Delta\xi}{\beta\gamma_0 L} - \frac{2D\bar{h}}{H_1} \frac{\Delta\gamma}{L},$$
(39b)

$$Q = \frac{2\Phi\Upsilon_3}{H_1} \frac{\gamma_0\bar{h}}{\eta k_0^2} \frac{\Delta\xi}{\beta y_0 L} - \frac{2\bar{h}^3}{3H_3\eta} \frac{\Delta P}{L}.$$
(39c)

In Eq. (39), we identify three effective force densities, namely, $\Delta \xi / (\beta \gamma_0 L)$, $\Delta \gamma / (\beta \gamma_0 L)$, and $\Delta P / L$. We use Eq. (26) to rewrite $\Delta \xi = \beta Q_0 [\Delta \bar{\mu}_f^+ - \Delta \bar{\mu}_f^-] = \beta \gamma_0 z e \Delta V$ and $\Delta \gamma = \beta Q_0 [\Delta \bar{\mu}_f^+ + \Delta \bar{\mu}_f^-] = \beta \gamma_0 \Delta \bar{\mu}$ in terms of the more familiar ionic chemical potential $\Delta \bar{\mu}^{56}$ and external potential drop ΔV .⁵⁷ We can then relate the three fluxes and three forces in Eq. (39) via a 3 × 3 conductivity matrix \mathcal{L} , the Onsager matrix of the out-of-equilibrium corrugated nanochannel,

$$\begin{pmatrix} J_q \\ J'_c \\ Q \end{pmatrix} = \begin{pmatrix} \mathcal{L}_{11} & \mathcal{L}_{12} & \mathcal{L}_{13} \\ \mathcal{L}_{21} & \mathcal{L}_{22} & 0 \\ \mathcal{L}_{31} & 0 & \mathcal{L}_{33} \end{pmatrix} \begin{pmatrix} ze\Delta V \\ \Delta \bar{\mu} \\ \Delta P \end{pmatrix} \frac{1}{L},$$
(40)

where the coefficients read

$$\mathcal{L}_{11} = \mathcal{L}_{22} = -2\gamma_0 \frac{\ddot{h}}{H_1} \frac{\mu_{\rm ion}}{ze},$$
 (41a)

$$\mathcal{L}_{12} = \mathcal{L}_{21} = -2\gamma_0 \frac{\bar{h}}{H_1} \frac{\mu_{\rm ion}}{ze} \Phi \Upsilon_1, \tag{41b}$$

$$\mathcal{L}_{13} = \mathcal{L}_{31} = -2\gamma_0 \frac{\bar{h}}{H_1} \frac{1}{\eta k_0^2} \Phi \Upsilon_3, \qquad (41c)$$

$$\mathcal{L}_{33} = -\frac{2}{3H_3} \frac{\bar{h}^3}{\eta},$$
 (41d)

where $\mu_{ion} = D\beta ze$ is the ionic mobility. Clearly, the matrix \mathcal{L} in Eq. (40) is symmetric; hence, Onsager's reciprocal relations are fulfilled. While Eq. (40) is a well known result,^{47–49} the derivation of Eq. (41) from a microscopic theory is, up to our knowledge, unprecedented. We discuss properties of the Onsager matrix and its coefficients in Sec. V.

V. RESULTS

A. General properties of the Onsager matrix

We list a few general properties of the Onsager matrix:

1. Equation (40) relates three fluxes (J_q, J'_c, Q) to three thermodynamic forces $(ze\Delta V/L, \Delta \mu/L, \Delta P/L)$ via four independent nonzero transport coefficients $(\mathcal{L}_{11}, \mathcal{L}_{12}, \mathcal{L}_{13}, \mathcal{L}_{33})$. Note that the off-diagonal matrix elements vanish $(\mathcal{L}_{12} = \mathcal{L}_{13} = 0)$ when the channel walls are uncharged ($\Phi = 0$). In that case, the charge flow J_q , the solute flow J'_c , and the flow of the solution Q respond solely to the electrostatic potential drop, chemical potential differences, and pressure differences, respectively. Conversely, for $\Phi \neq 0$, the off-diagonal terms of the Onsager matrix do not vanish ($\mathcal{L}_{i\neq j} \neq 0$) and Eq. (40) encodes a rich nonequilibrium behavior.

- 2. In bulk electrolytes, a salt gradient does not drive fluid flow. In the presence of a solid substrate, the interactions between the ions and the surface drive a phoretic flow $v \sim \nabla \mu / dr \ r(\exp[\beta U(r)] - 1)$ with U being the interaction potential between the ions and the walls. Within the Debye-Hückel approximation, the electrostatic potential is small. Hence, reversing the sign of the interaction leads to a reversal of the sign of the phoretic flow. This means that in the presence of a gradient $\nabla \mu$, the first nonzero contribution to the fluid flow is of $\mathcal{O}(f \psi^2)$, in agreement with Ref. 47.
- 3. Equation (41a) states that $\mathcal{L}_{11} = \mathcal{L}_{22}$. This implies that provided our approximations are applicable, the knowledge of the diagonal coefficient, \mathcal{L}_{11} , associated with the electric current induced solely by an electrostatic potential drop, determines the diagonal coefficient \mathcal{L}_{22} associated with the ionic current under the action of an ionic chemical potential imbalance $\Delta \bar{\mu}$.⁵⁸
- 4. The diagonal terms are controlled solely by H_1 and H_3 . Since these functions do not depend on the boundary conditions (constant σ or constant ζ) on the electrostatic potential nor do the diagonal terms.
- 5. In contrast to the diagonal elements, the off-diagonal terms are sensitive to the electrostatic boundary conditions. This is evident after inserting Eqs. (13) and (14) into Eqs. (36c) and (36d),

$$\Upsilon_{1}^{\zeta} = \frac{\bar{h}}{H_{1}L} \int_{0}^{L} \mathrm{d}x \, \frac{\mathscr{G}[k_{0}h(x)]}{h(x)},\tag{42a}$$

$$Y_1^{\sigma} = \frac{h}{H_1 L} \int_0^L dx \, \frac{1}{k_0 h(x)^2},$$
(42b)

$$\Upsilon_{3}^{\zeta} = \frac{\bar{h}^{3}}{H_{3}L} \int_{0}^{L} dx \, \frac{1 - \mathscr{G}[k_{0}h(x)]}{h^{3}(x)}, \qquad (42c)$$

$$\Upsilon_{3}^{\sigma} = \frac{\bar{h}^{3}}{H_{3}L} \int_{0}^{L} dx \, \frac{\mathscr{L}[k_{0}h(x)]}{h^{3}(x)},$$
(42d)

where

$$\mathscr{G}(x) = \frac{\tanh(x)}{x}, \qquad \mathscr{L}(x) = \coth(x) - \frac{1}{x}, \qquad (43)$$

with $\mathscr{L}(x)$ known as the Langevin function.

6. According to our formulation, the entropy production Σ reads

$$T\Sigma = -J_q \times ze\Delta V - J'_c \times \Delta\bar{\mu} - Q \times \Delta P \tag{44}$$

in terms of the thermodynamics forces $ze\Delta V$, $\Delta \bar{\mu}$, and ΔP . This expression is in agreement with Refs. 46 and 47.

In order to proceed with our analysis of the Onsager matrix and the functions H_1 , H_3 , Υ_1 , and Υ_3 appearing therein, we need to restrict to a particular channel shape. Accordingly, we choose

$$h(x) = \bar{h} + \Delta h \cos\left(2\pi \frac{x}{L}\right). \tag{45}$$

A more general shape of the channel may include a "phase" in the argument of the cosine. However, Eqs. (41) and (36) show that the transport coefficients depend solely on the integral of the channel shape and are thus phase independent.⁵⁹ While the dimensionless combination $\Delta h/\dot{h}$ already gives a sense of the channel corrugation, in the following we prefer to use the related "entropic barrier" defined as

$$\Delta S \equiv \ln \left[\frac{\bar{h} + \Delta h}{\bar{h} - \Delta h} \right] = 2 \tanh^{-1} (\Delta h / \bar{h}), \tag{46}$$

which takes values between $\Delta S = 0$ for flat channels and $\Delta S > 0$ for corrugated channels. Inserting Eq. (45) into H_1 [Eq. (36a)], H_3 [Eq. (24)], and Υ_1^{σ} [Eq. (42b)] gives

$$H_1 = \cosh \frac{\Delta S}{2},\tag{47a}$$

$$H_3 = \left[\frac{3}{2}\cosh^2\left(\frac{\Delta S}{2}\right) - \frac{1}{2}\right]\cosh^3\frac{\Delta S}{2},\tag{47b}$$

$$\Upsilon_1^{\sigma} = \frac{\cosh^2(\Delta S/2)}{k_0 \bar{h}}.$$
(47c)

We plot H_1 and H_3 in Fig. 2.

1. Diagonal terms

As noted earlier, the diagonal matrix elements $\mathcal{L}_{11} \sim 1/H_1$ and $\mathcal{L}_{33} \sim 1/H_3$ depend on the shape of the channel but not on the boundary conditions on the electrostatic potential at the channel walls, this is in agreement with Refs. 50–52. Because H_1 and H_3 increase with ΔS , \mathcal{L}_{11} and \mathcal{L}_{33} decrease therewith. This implies that the pressure-driven volumetric fluid flow Q and the chemical potential-driven excess solute flow J'_c diminish with increasing ΔS .

2. Off-diagonal terms

The off-diagonal terms in Eq. (40) are controlled by the functions $\Upsilon^{\sigma,\zeta}$ and by H_1 . In the following, we will focus on the dependence of \mathcal{L}_{12} and \mathcal{L}_{13} on $\Upsilon_1^{\sigma,\zeta}$ and $\Upsilon_3^{\sigma,\zeta}$, respectively. Figure 3(a) shows for conducting channel walls that Υ_1^{ζ} (and thus \mathcal{L}_{12}^{ζ}) is almost independent of ΔS , whereas Υ_1^{σ} (and thus \mathcal{L}_{12}^{τ}) increases more drastically as $\Upsilon_1^{\sigma} \propto \exp[\Delta S]$ at large ΔS [cf. Eq. (47c)]. Hence, counterintuitively, increasing the corrugation of the channel enhances

FIG. 2. Plots of the functions H_1 [Eq. (36a)] (black line) and H_3 [Eq. (24)] (gray dashed line) as a function of ΔS for the channel shape given in Eq. (45).



some off-diagonal transport coefficients. Equation Eq. (40) shows that this enhancement occurs for the electric current driven by a chemical potential drop $\Delta \mu$ (with $\Delta P = \Delta V = 0$) and for the excess solute flow J'_c driven by an external electrostatic field $\Delta V/L$ (with $\Delta P = \Delta \mu = 0$).

Figure 3(b) shows that the sensitivity of Υ_1^{ζ} , and hence \mathcal{L}_{12} , on the boundary conditions disappears when the Debye length is much smaller than the channel section, $k_0 \bar{h} \gg 1$, whereas it becomes significant when $k_0 \bar{h} \leq 1$. In particular, Fig. 3(b) shows that in the latter regime, \mathcal{L}_{12}^{ζ} keeps the linear dependence on $k_0 \bar{h}$ whereas $\mathcal{L}_{12}^{\sigma}$ reaches a plateau.

The dependence of \mathcal{L}_{13} , on the top of its H_1 -dependence, on both ΔS and $k_0 \bar{h}$ is encoded in $\Upsilon_3^{\sigma,\zeta}$. Similar to \mathcal{L}_{12} , also \mathcal{L}_{13} shows an explicit dependence on the boundary conditions. In particular, Fig. 3(a) shows that both $\mathcal{L}_{13}^{\sigma,\zeta}$ decrease with increasing ΔS , and for large values of ΔS , we have $\mathcal{L}_{13}^{\sigma,\zeta} \exp[-2\Delta S]$ and $\mathcal{L}_{13}^{\zeta} \propto \exp[-\Delta S]$. This means that the electric current induced by a pressure drop ΔP and the electroosmotic flow induced by ΔV decay exponentially with ΔS . The dependence of $\mathcal{L}_{13}^{\sigma,\zeta}$ on $k_0\bar{h}$ is shown in Fig. 3(b). Interestingly, for large values of $k_0\bar{h}$, both $\mathcal{L}_{13}^{\sigma}$ and \mathcal{L}_{13}^{ζ} reach a plateau, whereas for smaller values of $k_0\bar{h}$.

To understand the influence of the channel walls (i.e., conducting or dielectric) on the Υ functions, it is insightful to look at their relative differences through the combination

$$\Delta \Upsilon_{1,3} = 2 \frac{\Upsilon_{1,3}^{\sigma} - \Upsilon_{1,3}^{\zeta}}{\Upsilon_{1,3}^{\sigma} + \Upsilon_{1,3}^{\zeta}}.$$
(48)







Figure 4 shows that the functions $\Upsilon_{1,3}$ have a surprisingly similar sensitivity to the boundary conditions: whenever Υ_1 changes by switching from conducting to dielectric walls, so does Υ_3 and, remarkably, by almost the same amount. This means that there is no regime in which some of the Onsager coefficients are more sensitive than others upon changing the electrostatic properties of the walls. Finally, we notice from Fig. 4 that $\Delta \Upsilon_{1,3} > 0$, which means [cf. Eq. (48)] that the Onsager coefficients for dielectric walls are always larger than their counterparts for conducting walls. This can be understood by looking at the functional form of $\Upsilon^{\sigma,\zeta}_{1,3}$; see Eq. (36) together with Eq. (30c). Concerning $\Upsilon_1^{\sigma,\zeta}$, we notice that the integrand is proportional to $\overline{\psi_0(x)}/\Phi$. Interestingly, $\overline{\psi_0(x)}/\Phi$ is proportional to the charge-to-surface potential ratio. Hence, the fact that $\Delta\Upsilon_1>0$ is associated with the fact that, per unit surface potential, dielectric walls can enclose a larger charge. Concerning $\Upsilon_1^{\sigma,\zeta}$, we notice that the integrand is proportional, $2h(x)\psi(x,h(x)) - \overline{\phi_0}(x)$, i.e., to the difference between the potential averaged over the channel section, $\overline{\psi_0}(x)$, and the (maximum) value the average can attain, namely, $2h(x)\psi(x, h(x))$. In this regard, the fact that $\Delta \Upsilon_3 > 0$ is a reminder of the fact that, per unit surface potential, dielectric walls lead to a large variation of the potential between the channel walls as compared to conduction walls.

B. Single external force

So far, we have discussed the general properties of the Onsager matrix and their relation to some relevant cases. In the following, we discuss in detail several transport phenomena. In order to emphasize the role of the geometry and the onset of the entropic



1. Electrostatic driven flows

In the case that $\Delta P = \Delta \bar{\mu} = 0$, a potential difference $\Delta V \neq 0$ drives an ionic current, salt flow, and electroosmotic fluid flow. In particular, the electroosmotic flow (per unit length in the z-direction) reads

$$Q_{\rm eo} = \mathcal{L}_{13} \frac{ze\Delta V}{L} = \frac{2\Phi\epsilon\bar{h}}{\eta} \frac{\Delta V}{L} \times \frac{\Upsilon_3}{H_1}.$$
 (49)

This amounts to

$$Q_{\rm eo}^{\zeta} = \frac{2\zeta\epsilon\bar{h}}{\eta}\frac{\Delta V}{L} \times \frac{Y_3^{\zeta}}{H_1},\tag{50a}$$

$$Q_{\rm eo}^{\sigma} = \frac{2e\sigma\bar{h}^2}{\eta} \frac{\Delta V}{L} \times \frac{\Upsilon_3^{\sigma}}{k_0\bar{h}H_1}.$$
 (50b)

We note that Eq. (50b) coincides with Eq. (40) of Ref. 52 and that the combination $\zeta \epsilon / \eta$ is known as the "electroosmotic mobility."⁵¹ For the channel as specified in Eq. (45), we show Eq. (50) as a function of ΔS in Fig. 5(a) and as a function of $k_0 h$ in Fig. 5(b). From this figure, we see that both Q_{eo}^{ζ} and Q_{eo}^{σ} vanish for highly corrugated channels (when $\Delta S \gg 0$). Meanwhile, both Q_{eo}^{ζ} and Q_{eo}^{σ} diminish upon



FIG. 5. Electroosmotic flow Q_{eo} [Eq. (50)] and streaming current I_{str} [Eq. (53)] for conducting (red lines) and dielectric (blue lines) channel walls. Panel (a) shows Δ S-dependence at $k_0h = 1$ (solid) and $k_0h = 10$ (dashed). Panel (b) shows ΔV -dependence at $\Delta S = 1$ (solid) and $\Delta S = 1$ (dashed).

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FIG. 4. Plots of the functions $\Delta \Upsilon_1$ (downward triangles) and $\Delta \Upsilon_3$ (upward trian-

gles), as a function of ΔS [panel (a)] and $k_0 h$ [panel (b)] for the channel shape

given in Eq. (45). In (a), we use $k_0h = 0.1, 1, 10$: larger points stand for larger

values of k_0h . In (b), we use $\Delta S = 0, 1, 3$: larger points stand for larger values of

decreasing kh. Interestingly, the transition between the two plateaus occurs for $k_0 h \in [1:10]$, the entropic electrokinetic regime.^{33–36} For a straight channel, the integrals in Υ [Eq. (42)] become trivial and Eq. (50) simplifies to

$$Q_{\rm eo}^{\zeta} = \frac{2\zeta\epsilon h}{\eta} \frac{\Delta V}{L} \times \left[1 - \mathscr{G}(k_0 \bar{h})\right], \tag{51a}$$

$$Q_{\rm eo}^{\sigma} = \frac{2e\sigma\bar{h}^2}{\eta} \frac{\Delta V}{L} \times \frac{\mathscr{L}(k_0\bar{h})}{k_0\bar{h}}.$$
 (51b)

We note that Eq. (51a) is in agreement with Eq. (50) of Ref. 53. Moreover, for a flat channel, $\psi^{\zeta}(x, y) = \psi^{\sigma}(x, y)$ provided that the surface potentials are the same: $\psi^{\zeta}(x, h) = \psi^{\sigma}(x, h) \Rightarrow \zeta = e\sigma \coth[k_0 h]/(\epsilon k_0).$ Inserting this into Eq. (51a), we confirm Eq. (51b).

2. Pressure driven flows

In the case that $\Delta V = \Delta \bar{\mu} = 0$, a pressure difference $\Delta P \neq 0$ drives a *streaming current* (per unit length in the *z*-direction),

$$I_{\rm str} = \mathcal{L}_{13} z e \frac{\Delta P}{L} = \frac{2\Phi\epsilon h}{\eta} \frac{\Delta P}{L} \times \frac{\Upsilon_3}{H_1},$$
 (52)

which amounts to

$$I_{\rm str}^{\zeta} = \frac{2\zeta\epsilon\bar{h}}{\eta}\frac{\Delta P}{L} \times \frac{\Upsilon_3^{\zeta}}{H_1},\tag{53a}$$

$$I_{\rm str}^{\sigma} = \frac{2e\sigma\bar{h}^2}{\eta} \frac{\Delta P}{L} \times \frac{\Upsilon_3^{\sigma}}{k_0\bar{h}H_1}.$$
 (53b)

Clearly, I_{str} is governed by the same matrix element \mathcal{L}_{13} as the electroosmotic flow (its reciprocal phenomenon) discussed above. As a consequence, Q^{σ} and \hat{I}_{str}^{σ} share the same term $\Upsilon_{3}^{\sigma}/(k_{0}\bar{h}H_{1})$ and, hence, display the same ΔS and $k_0 h$ dependence (see Fig. 5).

From Eq. (52), it is easy to determine the streaming current between two parallel plates and to check that this agrees with Eq. (37) of Ref. 53.

3. Chemical potential steps Δµ

Finally, we consider the case in which flows are driven solely by a chemical potential drop $\Delta \bar{\mu}$. Accordingly, the electric current reads

$$J_q = -2\gamma_0 \bar{h} \frac{\mu_{\rm ion}}{ze} \Phi \Delta \bar{\mu} \times \frac{\Upsilon_1}{H_1},\tag{54}$$

which amounts to

$$J_{q}^{\zeta} = -2\gamma_{0}\bar{h}\beta\mu_{\rm ion}\zeta\Delta\bar{\mu}\times\frac{\Upsilon_{1}^{\zeta}}{H_{1}},$$
(55a)

$$J_{q}^{\sigma} = -2\gamma_{0}\bar{h}^{2}\beta\mu_{\rm ion}\frac{e\sigma}{\epsilon}\Delta\bar{\mu}\times\frac{\Upsilon_{1}^{\sigma}}{k_{0}\bar{h}H_{1}}.$$
(55b)

For dielectric channel walls, we find with Eq. (47c) that $J_q^{\sigma}(\Delta S)/J_q^{\sigma}(\Delta S = 0) = \cosh[\Delta S/2]$; hence, this ratio grows monotonically with increasing corrugation of the channel ΔS . Conversely, for conducting channel walls, Fig. 6(a) displays that J_a^{σ} has a maximum for a finite value of ΔS . The ionic charge currents as discussed in this section are the relevant physical phenomenon underlying





FIG. 6. Electric current J_q [Eq. (55)] driven by a chemical potential difference, as a function of ΔS [panel (a)] and $k_0 h$ [panel (b)] for conducting (red lines) and dielectric (blue lines) channel walls. Panel (a) shows ΔS -dependence at $k_0 h = 1$ (solid) and $k_0h = 10$ (dashed). Panel (b) shows ΔV -dependence at $\Delta S = 1$ (solid) and $\Delta S = 1$ (dashed).

reverse electrodialysis, whereby electrical energy is generated from a salt concentration difference.⁴

C. Membrane

In the following, we characterize several cases in which the channel is in series with a membrane that selectively impedes the passage of (any combination of) solvent and ions. In this scenario, we can control Q and the fluxes of positive, $2J^+ = J_c + J_q$, and negative, $2J_{-} = J_{c} - J_{q}$, ions. Due to its physical interest, in this section, we focus on the full solute flow $J_c = J'_c + \gamma_0 Q$ rather than on J'_c . The general solution to Eq. (40), under the above constraints, reads

$$\Delta \bar{\mu} = \frac{J_c - (\mathcal{L}_{12} + \gamma_0 \mathcal{L}_{13}) z e \Delta V - \gamma_0 \mathcal{L}_{33} \Delta P}{\mathcal{L}_{22}},$$
(56a)

$$ze\Delta V = \frac{\mathcal{L}_{22}J_q - \mathcal{L}_{12}J_c + (\mathcal{L}_{12}\gamma_0\mathcal{L}_{33} - \mathcal{L}_{22}\mathcal{L}_{13})\Delta P}{\mathcal{L}_{22}^2},$$
 (56b)

$$Q = \frac{\mathcal{L}_{13}}{\mathcal{L}_{22}} J_q + \mathcal{L}_{33} \Delta P.$$
(56c)

1. Electrodes

First, we consider a membrane that allows for a net electric current, $J_q \neq 0$, but not for mass fluxes, $Q = J_c = 0$. This looks like having some electrodes that close the electric circuit at zero solvent and ionic flow. When only ΔV is nonzero, Eqs. (56), at linear order in Φ, give

 10^{1}

 10^{0}

$$\Delta \bar{\mu}_V = \frac{\mathcal{L}_{12}}{\mathcal{L}_{22}} z e \Delta V, \qquad (57a)$$

$$\Delta P_V = \frac{\mathcal{L}_{13}}{\mathcal{L}_{33}} z e \Delta V, \tag{57b}$$

$$J_q = \mathcal{L}_{22} z e \Delta V, \tag{57c}$$

or, likewise,

$$\Delta \bar{\mu}_V = z e \Delta V \Phi \times \Upsilon_1, \tag{58a}$$

$$\Delta P_V = \frac{1}{3} \frac{\epsilon}{\beta z e} \frac{1}{\bar{h}^2} \Delta V \Phi \times \frac{H_3}{H_1} \Upsilon_3, \tag{58b}$$

$$J_q = -2\frac{\epsilon}{\beta ze}\bar{h}\frac{\mu_{\rm ion}}{ze}\Delta V \times \frac{1}{H_1}.$$
(58c)

Figure 7(a) shows the dependence of $\Delta \bar{\mu}_V$ on ΔS . In particular, for both conducting and dielectric channel walls, $\Delta \bar{\mu}_V$ increases with the corrugation of the channel ΔS . In contrast, the dependence of $\Delta \bar{\mu}_V$ on k_0h is more sensitive to the conductive properties of the channel walls. While for dielectric walls $\Delta \bar{\mu}_V$ is independent on $k_0\bar{h}$, for conducting walls it reaches a plateau for both $k_0\bar{h} \ll 1$ and $k_0\bar{h} \gg 1$ and it grows for $k_0\bar{h} \in [1 : 10]$ [see Fig. 7(b)]. Interestingly, the very same behavior is observed for the electrostatic potential drop, ΔV_{μ} , induced by an applied chemical potential, $\Delta \bar{\mu}$, when the electric current is set to zero, $J_q = 0$. In this case, ΔV_{μ} grows with ΔS for both kinds of channel walls [see Fig. 7(a)]. Since both $\Delta \bar{\mu}_V$ and ΔV_{μ}



FIG. 7. Chemical potential drop, $\Delta\mu$ [Eq. (58a)], as a function of ΔS [panel (a)] and $k_0\bar{h}$ [panel (b)] for conducting (red lines) and dielectric (blue lines) channel walls. Panel (a) shows ΔS -dependence at $k_0\bar{h} = 1$ (solid) and $k_0\bar{h} = 10$ (dashed). Panel (b) shows ΔV -dependence at $\Delta S = 1$ (solid) and $\Delta S = 1$ (dashed). The same quantitative behavior holds for ΔV induced by an applied chemical potential, $\Delta\mu$, when the electric current is set to zero, Jq = 0 (see text).

are proportional to Υ_1 , they increase upon enlarging ΔS as already shown in Fig. 6. Finally, we remark that ΔP_v [Eq. (58b)] is also known as the *electroosmotic back/counter pressure*.⁵¹

2. Open electric circuit

Second, we consider a membrane that allows for mass flows, $Q \neq 0$ and $J_c \neq 0$, but not for electric current, $J_q = 0$. In this case, at linear order in Φ , Eqs. (56) read

$$\Delta \bar{\mu} = \frac{J_c - (\mathcal{L}_{12} + \gamma_0 \mathcal{L}_{13}) z e \Delta V - \gamma_0 \mathcal{L}_{33} \Delta P}{\mathcal{L}_{22}},$$
 (59a)

$$ze\Delta V = \frac{-\mathcal{L}_{12}J_c + (\mathcal{L}_{12}\gamma_0\mathcal{L}_{33} - \mathcal{L}_{22}\mathcal{L}_{13})\Delta P}{\mathcal{L}_{22}^2},$$
 (59b)

$$Q = \mathcal{L}_{33} \Delta P. \tag{59c}$$

In particular, for $\Delta V = 0$, this amounts to

(

$$\Delta \bar{\mu} = -\frac{ze\bar{h}^2}{\mu_{\rm ion}\eta} \Delta P \times \frac{1}{k_0^2 \bar{h}^2} \frac{\Upsilon_3}{\Upsilon_1},\tag{60a}$$

$$J_c = -2\frac{\epsilon \bar{h}}{\beta(ze)^2 \eta} \Delta P \times \left[\frac{1}{3}\frac{k_0^2 \bar{h}^2}{H_3} - \frac{1}{H_1}\frac{\Upsilon_3}{\Upsilon_1}\right],\tag{60b}$$

$$Q = -\frac{2}{3} \frac{\bar{h}^3}{\eta} \Delta P \times \frac{1}{H_3}.$$
 (60c)

As shown in Fig. 8(a), the solute current J_c decreases monotonically for both dielectric and conducting channel walls upon increasing the channel corrugation ΔS . More surprising is the dependence of



FIG. 8. Solute current, J_c [Eq. (60b)], as a function of ΔS [panel (a)] and $k_0 h$ [panel (b)] for conducting (red lines) and dielectric (blue lines) channel walls. Panel (a) shows ΔS -dependence at $k_0 h = 1$ (solid) and $k_0 h = 10$ (dashed). Panel (b) shows ΔV -dependence at $\Delta S = 1$ (solid) and $\Delta S = 1$ (dashed).

 J_c on $k_0 h$. Indeed, Fig. 8(b) shows that J_c reaches a plateau for both $k_0 h \ll 1$ and $k_0 h \gg 1$, and for $k_0 h \simeq 1$, it displays a nonmonotonous dependence on $k_0 h$.

3. Solvent permeable membrane

Third, we consider a membrane that selectively permits the flow of solute but not of ions; hence, $J_c = J_q = 0$ and $J'_c = \gamma_0 Q$. Accordingly, we obtain

$$\Delta \bar{\mu} = -\frac{\gamma_0 \mathcal{L}_{33}}{\mathcal{L}_{22}} \Delta P, \tag{61a}$$

$$ze\Delta V = \frac{\mathcal{L}_{12}\gamma_0\mathcal{L}_{33} - \mathcal{L}_{22}\mathcal{L}_{13}}{\mathcal{L}_{22}^2}\Delta P,$$
 (61b)

$$Q = \mathcal{L}_{33} \Delta P, \tag{61c}$$

from Eq. (56). This amounts to

$$\Delta \bar{\mu} = -\frac{1}{3} \frac{\bar{h}^2}{\eta} \frac{ze}{\mu_{\rm ion}} \Delta P \times \frac{H_1}{H_3}, \qquad (62a)$$

$$ze\Delta V = \frac{\bar{h}^2}{\eta} \frac{ze}{\mu_{\rm ion}} \Phi \Delta P \times \left[\frac{1}{3} \frac{H_1}{H_3} \Upsilon_1 - \frac{1}{k_0^2 \bar{h}^2} \Upsilon_3 \right], \qquad (62b)$$

$$Q = -\frac{2}{3}\frac{\bar{h}^2}{\eta}\Delta P \times \frac{1}{H_3}.$$
 (62c)

Interestingly, in order to sustain a nonvanishing fluid flow, $Q \neq 0$, the system will excite all three external forces, i.e., we have $\Delta V \neq 0$, $\Delta P \neq 0$, and $\Delta \bar{\mu} \neq 0$. In particular, when ΔP is the only externally applied force, then from Eq. (62b) we can read off the induced *streaming potential*. Figure 9(a) shows that ΔV decays monotonically with ΔS for both conducting and dielectric channel walls, whereas Fig. 9(b) shows that ΔV reaches a plateau for both $k_0 \bar{h} \ll 1$ and $k_0 \bar{h} \gg 1$ and that the sensitivity of ΔV on $k_0 \bar{h}$ is maximum for $k_0 \bar{h} \simeq 1$, i.e., in the entropic electrokinetic regime. Finally, by inverting Eq. (62a), Eqs. (62) show that net fluid flow can be obtained by applying a chemical potential gradient $\Delta \bar{\mu}$. However, the net fluid flow is not a *direct consequence* of $\Delta \bar{\mu}$ (we recall that $\mathcal{L}_{23} = \mathcal{L}_{32} = 0$). Rather, $\Delta \bar{\mu}$ induces an osmotic pressure drop that eventually drives the flow.

4. Ion exchange membrane

Finally, we consider the channel to be in series with a membrane that selectively allows for the flow of solvent and one ionic species, impeding the other species. Recalling that $J_c = (J^+ + J^-)/2$ and $J_q = (J^+ - J^-)/2$, when only positive ions are flowing $(J^- = 0)$, we have $J_c = J_q$, whereas when only negative ions are flowing $(J^+ = 0)$, we have $J_c = -J_q$. Hence, we have

$$J_{c} = \frac{\mathcal{L}_{22}^{2}}{\pm \mathcal{L}_{22} - \mathcal{L}_{12}} ze\Delta V - \frac{\mathcal{L}_{12}\gamma_{0}\mathcal{L}_{33} - \mathcal{L}_{22}\mathcal{L}_{13}}{\pm \mathcal{L}_{22} - \mathcal{L}_{12}}\Delta P,$$
 (63a)

$$\Delta \bar{\mu} = -\frac{\mathcal{L}_{12} + \gamma_0 \mathcal{L}_{13}}{\mathcal{L}_{22}} z e \Delta V - \frac{\gamma_0 \mathcal{L}_{33}}{\mathcal{L}_{22}} \Delta P + J_c, \tag{63b}$$

$$Q = \mathcal{L}_{13}\Delta V + \mathcal{L}_{33}\Delta P, \tag{63c}$$





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FIG. 9. Electrostatic potential drop ΔV [Eq. (62b)] across a solvent permeable membrane, as a function of ΔS [panel (a)] and $k_0 h$ [panel (b)] for conducting (red lines) and dielectric (blue lines) channel walls. Panel (a) shows ΔS -dependence at $k_0 h = 1$ (solid) and $k_0 h = 10$ (dashed). Panel (b) shows ΔV -dependence at $\Delta S = 1$ (solid) and $\Delta S = 1$ (dashed).

where the sign in front of \mathcal{L}_{22} is positive when positive ions are flowing and negative otherwise. We remark that within linear response, $\mathcal{L}_{22} \geq \mathcal{L}_{12}$.⁴⁶ In contrast to the solvent permeable membrane, for the ion exchange membrane we can put one of the thermodynamic forces to zero. In particular, for electrostatic driven flows, $\Delta V \neq 0$ with $\Delta P = 0$, Eqs. (63) read⁶⁰

$$J_c = -2\frac{\mu_{\rm ion}\epsilon}{\beta(ze)^2\bar{h}}\Delta V \times \frac{k_0^2\bar{h}^2}{H_1}\frac{1}{\pm 1 - \Phi\Upsilon_1},$$
(64a)

$$\Delta \tilde{\mu} = ze\Delta V \Phi \times \left[\Upsilon_1 + \frac{\epsilon}{\beta z e \mu_{\rm ion} \eta} \Upsilon_3 \right] + J_c, \tag{64b}$$

$$Q = -2\frac{\epsilon}{\beta ze}\Delta V\Phi\bar{h}\frac{1}{\eta} \times \frac{\Upsilon_3}{H_1}.$$
 (64c)

Equation (64a) is remarkable for two reasons. First, as commented earlier, if it diverges when $\pm 1 - \Phi Y_1 \rightarrow 0$ it would require higher order corrections. Second, in Eq. (64a), the dimensionless potential, Φ , plays a major role, i.e., it modulates the relative magnitude of the two terms in the denominator, and it is not simply a multiplicative constant as is in all previous cases. In order to proceed with a numerical inspection of Eq. (64a), we *fix* the value of Φ such that $\pm 1 - \Phi Y_1$ never vanishes. This allows us to fulfill the constraint $\mathcal{L}_{22} \ge \mathcal{L}_{12}$. Fixing the value of Φ is crucial for the dielectric case since, in order to keep the magnitude of the potential fixed, the surface charge density decreases with k_0 (Fig. 10).



FIG. 10. Solute flux J_c [Eq. (64a)] through a semipermeable membrane as a function of ΔS [panel (a)] and $k_0 \bar{h}$ [panel (b)] for conducting (red lines) and dielectric (blue lines) channel walls. Panel (a) shows ΔS -dependence at $k_0 \bar{h} = 1$ (solid) and $k_0 \bar{h} = 10$ (dashed) (all lines are overlapping each other). Panel (b) shows ΔV -dependence at $\Delta S = 1$ (solid) and $\Delta S = 1$ (dashed).

VI. MICROSCOPIC PERSPECTIVE

So far, we have discussed the macroscopic transport properties of the channel. However, our framework also allows us to discuss some microscopic details, such as the local electrostatic potential, $\psi_f(x, y)$, induced by the external forces modulated by the geometry of the channel. At first order in lubrication, the leading order forceinduced correction of the Poisson equation [Eq. (1a)] reads

$$\partial_y^2 \psi_f(x,y) = k_0^2 \left[\psi_f(x,y) + \psi_0(x,y) \frac{\gamma_f(x)}{\gamma_0} \right] - \frac{ze\xi_f(x)}{\epsilon}, \qquad (65)$$

where the right hand side follows from the perturbed ionic charge density linear in the external force, $q_f = \rho_f^+ - \rho_f^-$. Equation (65) is solved in Appendix D for both dielectric and conducting boundary conditions. In the case of conducting walls, $\psi_f(x, y)$ reads

$$\psi_{f}^{\zeta}(x,y) = \frac{\psi_{0}^{\zeta}(x,y)}{2} \frac{\gamma_{f}(x)}{\gamma_{0}} (k_{0}y \tanh[k_{0}y] - k_{0}h(x) \tanh[k_{0}h(x)]) + \frac{1}{\beta z e} \frac{\xi_{f}(x)}{\gamma_{0}} \left(1 - \frac{\cosh[k_{0}y]}{\cosh[k_{0}h(x)]}\right),$$
(66)

while for dielectric walls we find

$$\psi_f^{\sigma}(x,y) = \frac{\psi_0^{\sigma}(x,y)}{2} \frac{\gamma_f(x)}{\gamma_0} (k_0 y \tanh[k_0 y] - k_0 h(x) \coth[k_0 h(x)] - 1)$$

$$+\frac{1}{\beta ze}\frac{\xi_f(x)}{\gamma_0},\tag{67}$$

with $\psi_0^{\zeta}(x, y)$ and $\psi_0^{\sigma}(x, y)$ given by Eqs. (13) and (14), respectively. We note that the contributions in Eqs. (66) and (67) are within the approximations in Eq. (25a).

A. Local charge electroneutrality

At the end of Sec. III, we showed that local charge neutrality is fulfilled at equilibrium in our system of interest. For the out-ofequilibrium case discussed in this section, one can show that local charge neutrality holds at the lowest order in the applied forces as well: Using Eqs. (13) and (14), we show in Appendix E that the total ionic charge $q_{\text{tot,f}}$ in a slab at x precisely balances the local surface charge for both boundary conditions $[2e\sigma_f^{\zeta}(x)]$ and $2e\sigma_f = 0$, respectively]. Global charge neutrality is then obviously satisfied at this order of approximation as well.

B. Local Debye length

Using Eqs. (26) and (30a), we can define a local Debye length

$$k(x) = \sqrt{\frac{\beta(ze)^2}{\epsilon}(\gamma_0 + \gamma_f(x))} \simeq k_0 + k_1(x), \qquad (68)$$

where

$$k_1(x) = \frac{k_0}{2} \frac{\gamma_f(x)}{\gamma_0}.$$
 (69)

Interestingly, assuming a local Debye length $k(x) = k_0 + k_1(x)$ and expanding Eqs. (13) and (14) for small values of $k_1(x)$, the terms proportional to γ_f in Eqs. (66) and (67) are retrieved. Hence, our results show that the leading corrections to the local electrostatic potential proportional to the local salt concentration $\gamma_f(x)$ can be interpreted as being caused by a local Debye length proportional to $\gamma_f(x)$.

VII. CONCLUSIONS

We have characterized the dynamics of a z - z electrolyte embedded in a varying-section channel. We focused our analysis on channels whose section is varying smoothly enough so that we can apply a lubrication approximation to the (linearized) Poisson-Boltzmann equation—governing the electrostatic potential—as well as for the Stokes equation-governing the fluid flow. At equilibrium, we found that the Debye length stays constant up to first order in the lubrication expansion. For driven systems, we have focused on the linear response of the electrolyte to the external driving. Such a limit is relevant for weak external forces for which higher-order contributions are negligibly small. Applying these approximations enabled us to derive analytical expressions for the corrections induced by the external driving to the local properties of the electrolyte. In such a regime, we have identified the set of thermodynamic forces and fluxes for which the Onsager matrix is symmetric. We remark that our results are valid in the regime of small Dukhin number (see Appendix F).

Exploiting these results, we have investigated several cases of experimental interest. In general, we found that increasing the channel corrugation ΔS leads to smaller transport coefficients. However, our model shows that there are a few counterexamples for which the opposite holds. Indeed, the electric current induced by an ionic chemical potential imbalance $\Delta \bar{\mu}$ grows with ΔS (Fig. 6). A

similar effect can be obtained when multiple thermodynamic forces are applied. For example, both the ionic chemical potential drop, $\Delta \bar{\mu}_V$, induced by an applied voltage ΔV when $Q = J_c = 0$ and the electric potential drop ΔV_{μ} induced by a chemical potential drop $\Delta \bar{\mu}$ for $J_q = 0$ grow with the channel corrugation ΔS .

Finally, we have investigated the role of the conductive properties of the channel walls on the transport coefficients. In contrast to the case of planar channel walls for which it is possible to map the solution for dielectric channel walls into that of conducting channel walls by properly rescaling the potential at the wall, when $\Delta S \neq 0$ this mapping does not hold anymore and different behavior appears. Interestingly, our results show that the difference between the transport coefficients calculated for dielectric or conducting channel walls can be significant (see Fig. 4). In particular, dielectric walls typically lead to larger transport coefficients than conducting walls. This difference is relevant only in the entropic electrokinetic regime, where the Debye length is comparable to the channel bottleneck, $k_0 h_{\min} \simeq 1$, but not to the channel widest section, $k_0 h_{\max} \simeq 1$. This clearly requires $\Delta S = \ln[h_{\text{max}}/h_{\text{min}}] \neq 0$. Indeed, as shown in Fig. 4, the difference between the transport coefficients vanishes for $k_0 h \rightarrow 0$, i.e., when the Debye length is too small and also for $\Delta S \rightarrow 0$, i.e., for straight channels. In particular, the difference between the transport coefficients for conducting and dielectric walls becomes not only quantitative but also qualitative in the case of the electric current driven by an ionic chemical potential drop, $\Delta \bar{\mu}$, and with $Q = J_c = 0$. For this case, the current grows monotonically with ΔS for dielectric channel walls, whereas it shows a maximum for conducting channel walls.

Our results show that electrolytes embedded in varying-section channels can display a rich dynamics. We believe that these results can open new routes for the realization of synthetic devices aiming at energy harvesting or water desalination and can be insightful for the understanding of biological processes such as ionic transport across pores and membranes.

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APPENDIX A: DERIVATION OF EQ. (29)

Inserting the chemical potential μ^{\pm} [Eq. (26)] into the Nernst-Planck equation (1b), at linear order in the external force f and up to quadratic in the equilibrium electrostatic potential ψ_0 (but disregarding $\mathcal{O}(f\psi_0^2)$), we find

$$j_x^{\pm}(x,y) = v_x(x,y)\rho_0^{\pm}(x,y) - D\beta\rho_0^{\pm}(x,y)\partial_x\mu_f^{\pm}(x,y) + \mathcal{O}(f^2).$$
(A1)

Using the local equilibrium approximation, $\partial_y \mu_f^{\pm} = 0$, we find

$$J^{+} = \int_{-h(x)}^{h(x)} \rho^{+}(x, y) v_{x}(x, y) \, \mathrm{d}y - D\beta \partial_{x} \bar{\mu}_{f}^{+}(x) \int_{-h(x)}^{h(x)} \rho^{+}(x, y) \, \mathrm{d}y, \qquad (A2a)$$

$$J^{-} = \int_{-h(x)}^{h(x)} \rho^{-}(x, y) v_{x}(x, y) \, \mathrm{d}y - D\beta \partial_{x} \bar{\mu}_{f}^{-}(x) \int_{-h(x)}^{h(x)} \rho^{-}(x, y) \, \mathrm{d}y.$$
(A2b)

Recalling that $\rho_0^{\pm}(x, y) \simeq \varrho_0(1 \mp \psi_0(x, y))$, $\gamma_0 = 2\varrho_0$, and defining $J^{\pm}(x) = \int_{-h(x)}^{h(x)} j_x^{\pm}(x, y) \, dy$ [Eq. (28)], $J_c = J^+ + J^-$, and $J_q = J^+ - J^-$, we obtain

$$\frac{J_c}{D} = \beta z e \overline{\psi_0}(x) \partial_x \xi_f(x) - 2h(x) \partial_x \gamma_f(x) + \frac{\gamma_0 Q}{D},$$
(A3a)

$$\frac{J_q}{D} = \beta z e \overline{\psi_0}(x) \partial_x \gamma_f(x) - 2h(x) \partial_x \xi_f(x) + \frac{\mathcal{J}_q(x)}{D}, \qquad (A3b)$$

where we identified Q [cf. Eq. (20)] in the last term of Eq. (A3a) and where we defined⁶¹

$$\overline{\psi_0}(x) = \int_{-h(x)}^{h(x)} \psi_0(x, y) \,\mathrm{d}y,\tag{A4}$$

$$\mathcal{J}_{q}(x) = \int_{-h(x)}^{h(x)} q_{0}(x, y) v_{x}(x, y) \,\mathrm{d}y.$$
(A5)

APPENDIX B: DERIVATION OF EQ. (35b)

Inserting $v_x(x, y)$ [Eq. (19)] into $\mathcal{J}_q(x)$, to $\mathcal{O}(\psi_0)$, only the pressure driven velocity u_P remains, giving

$$\mathcal{J}_{q}(x) = \gamma_{0}\beta ze \frac{\Delta P}{2\eta L} \frac{\tilde{h}^{3}}{H_{3}h^{3}(x)} \int_{-h(x)}^{h(x)} [h^{2}(x) - y^{2}] \psi_{0}(x, y) \, \mathrm{d}y.$$
(B1)

Here, we used that $\partial_x P_{\text{tot}f}(x) = \Delta P \bar{h}^3 / [H_3 L h^3(x)] + \mathcal{O}(\psi_0)$, which follows from inserting Eq. (23) into Eq. (21). We can rewrite the integral in Eq. (B1) by inserting the Poisson equation (12). Two partial integrations then give

$$\int_{-h(x)}^{h(x)} k_0^2 [h^2(x) - y^2] \psi_0(x, y) \, dy = h^2(x) \int_{-h(x)}^{h(x)} \partial_y^2 \psi_0(x, y) \, dy - \int_{-h(x)}^{h(x)} y^2 \partial_y^2 \psi_0(x, y) \, dy$$
$$= h(x)^2 \int_{-h(x)}^{h(x)} \partial_y^2 \overline{\psi_0(x, y)} \, dy - y^2 \partial_y \psi_0(x, y) \Big|_{-h(x)}^{h(x)} + 2 \int_{-h(x)}^{h(x)} y \partial_y \psi_0(x, y) \, dy$$
$$= 2 \int_{-h(x)}^{h(x)} [\psi_0(x, h(x)) - \psi_0(x, y)] \, dy.$$
(B2)

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Evaluating Eq. (34) at x = L, a term containing $\int_0^L dx \mathcal{J}(x')/h(x')$ appears. With the above two equations and the definition of Υ_3 [cf. Eq. (36d)], we find

$$\int_{0}^{L} \frac{\mathcal{J}_{q}(x)}{h(x)} \, \mathrm{d}x = \gamma_{0} \frac{\Delta P}{k_{0}^{2} \eta L} \frac{\tilde{h}^{3}}{H_{3}} \beta z e \int_{0}^{L} \mathrm{d}x \int_{-h(x)}^{h(x)} \frac{\psi_{0}(x, h(x)) - \psi_{0}(x, y)}{h^{4}(x)} \, \mathrm{d}y = 2\gamma_{0} \frac{\Delta P}{k_{0}^{2} \eta} \Phi \Upsilon_{3}, \tag{B3}$$

which proves the appearance of the Υ_3 term in Eq. (35b). Note that the derivation in Eq. (B2) is valid for both boundary electric conditions, so is Eq. (B3).

APPENDIX C: DERIVATION OF EQ. (38)

We recall the expression of the electroosmotic flow

$$Q_{\rm eo} = \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{\mathrm{d}x}{h^3(x)} \int_{-h(x)}^{h(x)} u_{\rm eo}(x, y) \,\mathrm{d}y, \tag{C1}$$

where the electroosmotic fluid velocity [Eq. (19c)] reads

$$u_{\rm eo}(x,y) = \mathcal{U}(x,y) - \mathcal{U}(x,h(x)), \tag{C2}$$

with

$$\mathcal{U}(x,y) = \frac{ze}{\eta} \int dy \int dy q(x,y) \partial_x \psi(x,y), \qquad (C3)$$

the integrand of which, to first order in lubrication, ψ_0 , and *f*, reads

$$q(x,y)\partial_x\psi(x,y) = q_f(x,y)\partial_x\psi_0(x,y) + q_0(x,y)\partial_x\psi_f(x,y).$$
 (C4)

By inserting Eqs. (66) and (67) into Eq. (C4), we find $\mathcal{O}(\psi_0 f)$ expressions for the two terms in Eq. (C4) for both boundary conditions, which we can write compactly as

$$q_f(x,y)\partial_x\psi_0(x,y) = \xi_f(x)[1-\theta(x,y)]\partial_x\psi_0(x,y), \tag{C5}$$

$$q_0(x,y)\partial_x\psi_f(x,y) = -\psi_0(x,y)\partial_x[\xi_f(x)\theta(x,y)],$$
(C6)

with

$$\theta(x,y) = \begin{cases} 1 - \frac{\cosh[k_0 y]}{\cosh[k_0 h(x)]} & \text{cond,} \\ 1 & \text{diel.} \end{cases}$$
(C7)

After reshuffling, \mathcal{U} reads

$$\mathcal{U}_{f}(x,y) = -\frac{ze}{\eta} \int dy \int dy \Big\{ \partial_{x} \Big[\psi_{0}(x,y)\xi_{f}(x)(\theta(x,y)-1) \Big] \\ + \psi_{0}(x,y)\partial_{x}\xi_{f}(x) \Big\}.$$
(C8)

We can now determine $u_{eo}(x, y)$ for the two different boundary conditions explicitly (note that the first term of the above integrand is zero for dielectric walls and the last term can be explicitly integrated twice),

$$u_{eo}^{\zeta}(x,y) = \frac{1}{\beta z e} \frac{\epsilon \zeta}{\eta} \left\{ \frac{\partial_x \xi_f(x)}{\gamma_0} \left(1 - \frac{\cosh[k_0 y]}{\cosh[k_0 h(x)]} \right) - \partial_x \left[\frac{\xi_f(x)}{\gamma_0} \frac{2k_0^2 [y^2 - h^2(x)] + \cosh[2k_0 y] - \cosh[2k_0 h(x)]}{8\cosh^2[k_0 h(x)]} \right] \right\},$$
(C9)

$$u_{\rm eo}^{\sigma}(x,y) = \frac{e\sigma}{\eta k_0} \frac{1}{\beta z e} \frac{\partial_x \xi_f(x)}{\gamma_0} \frac{\cosh[k_0 h(x)] - \cosh[k_0 y]}{\sinh[k_0 h(x)]}.$$
 (C10)

Using $\mathscr{G}(x) = \tanh(x)/x$ and $\mathscr{L}(x) = \coth(x) - 1/x$ as defined in Eq. (43), we obtain

$$\int_{-h(x)}^{h(x)} u_{eo}^{\zeta}(x,y) \, \mathrm{d}y = \frac{2h(x)}{\beta ze} \frac{\epsilon \zeta}{\eta} \frac{\partial_x \xi_f(x)}{\gamma_0} (1 - \mathscr{G}[k_0 h(x)]) \\ - \frac{1}{\beta ze} \frac{\epsilon \zeta}{\eta k_0} \frac{\partial_x [\xi_f(x) \Gamma(x)]}{\gamma_0}, \quad (C11a)$$

$$\int_{-h(x)}^{h(x)} u_{eo}^{\sigma}(x, y) \, \mathrm{d}y = 2h(x) \frac{e\sigma}{\eta k_0} \frac{\partial_x \xi_f(x)}{\beta z e \gamma_0} \mathscr{L}[k_0 h(x)], \qquad (C11b)$$

with

$$\Gamma(x) = \frac{1}{4} \tanh[k_0 h(x)] - \frac{k_0 h(x)}{2 \cosh[k_0 h(x)]} + \frac{3k_0 h(x) - 4k_0^3 h^3(x)}{12 \cosh^2[k_0 h(x)]}.$$
(C12)

Using Eqs. (C2) and (C11), we find the following expressions for Q_{eo} [cf. Eq. (23c)]:

$$Q_{eo}^{\zeta} = 2 \frac{\epsilon \zeta}{\eta} \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{1}{\beta z e} \frac{\partial_x \xi_f(x)}{\gamma_0} \frac{1 - \mathscr{G}[k_0 h(x)]}{h^2(x)} dx - \frac{\epsilon \zeta}{\eta k_0} \frac{1}{\beta z e} \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{\partial_x [\xi_f(x) \Gamma(x)]}{\gamma_0} \frac{dx}{h^3(x)}, \qquad (C13a)$$

$$Q_{eo}^{\sigma} = 2 \frac{e\sigma}{\eta k_0} \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{1}{\beta z e} \frac{\partial_x \xi_f(x)}{\gamma_0} \frac{\mathscr{L}[k_0 h(x)]}{h^2(x)} \, \mathrm{d}x.$$
(C13b)

We obtain an expression for $\partial_x \xi_f(x)$ by substituting Eq. (32) in Eq. (31). At order $\mathcal{O}((\psi_0)^0)$, we obtain

$$\partial_x \xi_f(x) = -\frac{J_q}{2Dh(x)} + \mathcal{O}(\psi_0). \tag{C14}$$

Inserting this expression into Eq. (C13), at linear order in ψ_0 , we find

$$Q_{eo}^{\zeta} = -\frac{J_q}{D} \frac{\epsilon \zeta}{\eta} \frac{1}{\beta z e \gamma_0} \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{1 - \mathscr{G}[k_0 h(x)]}{h^3(x)} dx - \frac{\epsilon}{\eta} \frac{\zeta}{k_0} \frac{1}{\beta z e} \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{\partial_x [\xi_f(x) \Gamma(x)]}{\gamma_0} \frac{dx}{h^3(x)},$$
(C15a)

$$Q_{eo}^{\sigma} = -2\frac{e\sigma}{\eta k_0} \frac{1}{\beta z e y_0} \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{J_q}{2Dh(x)} \frac{\mathscr{L}[k_0 h(x)]}{h^2(x)} \, \mathrm{d}x.$$
(C15b)

Comparing the above equations to Eq. (42), we see that we can write Q_{eo} as

$$Q_{\text{eo}}^{\zeta} = -J_q \frac{\Phi \Upsilon_3^{\zeta}}{D\beta\eta k_0^2} - \frac{\epsilon}{\eta} \frac{\zeta}{k_0} \frac{1}{\beta z e} \frac{\bar{h}^3}{H_3 L} \int_0^L \frac{\partial_x \left[\xi_f(x) \Gamma(x)\right]}{\gamma_0 h^3(x)} \, \mathrm{d}x, \quad (C16a)$$

$$Q_{\rm eo}^{\sigma} = -J_q \frac{\Phi \Upsilon_3^{\sigma}}{D\beta \eta k_0^2},\tag{C16b}$$

which proves Eq. (38) up to the second term on the right hand side of Eq. (C16a). Since at $\mathcal{O}(\zeta^0)$ we have $\xi_f(x) = \xi_f[h(x)]$, we can write

$$\int_0^L \frac{\partial_x F[h(x)]}{h^3(x)} \, \mathrm{d}x = \int_0^L \frac{\delta F[h(x)]}{\delta h(x)} \frac{\partial_x h(x)}{h^3(x)} \, \mathrm{d}x, \tag{C17}$$

with $F[h(x)] = \xi_f(x)\Gamma(x)$, i.e., F[h(x)] depends on x solely through h(x). Without loss of generality, we can define a function G[h(x)] such that

$$\frac{1}{h^{3}(x)} \frac{\delta F[h(x)]}{\delta h(x)} = \frac{\delta G[h(x)]}{\delta h(x)}$$

$$\Rightarrow \int_{0}^{L} \frac{\partial_{x} F[h(x)]}{h^{3}(x)} dx = \int_{0}^{L} \frac{\delta G[h(x)]}{\delta h(x)} \partial_{x} h(x) dx$$

$$= G[h(L)] - G[h(0)], \quad (C18)$$

i.e., for periodic channels, h(L) = h(0), we have G(L) = G(0) and the last term in Eq. (C16a) vanishes.

APPENDIX D: DERIVATION OF EQS. (66) AND (67)

The solution to Eq. (65) reads

$$\psi_f(x, y) = -A_0(x) \frac{\gamma_f(x)}{4\gamma_0} [\cosh(k_0 y) - 2k_0 y \sinh(k_0 y)] + A_f(x) \cosh(k_0 y) + \frac{1}{\beta z e} \frac{\xi_f(x)}{\gamma_0},$$
(D1)

where A_0 is given for conducting and dielectric walls by $A_0^{\zeta} = \zeta/\cosh[k_0h(x)]$ and $A_0^{\zeta} = e\sigma/(\epsilon k_0 \sinh[k_0h(x)])$, respectively. The term $A_f(x)$ is obtained by imposing the suitable boundary conditions at the channel walls.

For conducting channel walls, one has $\psi_f(x, \pm h(x)) = 0$, and hence

$$A_{f}^{\zeta}(x) = \frac{1}{\cosh[k_{0}h(x)]} \times \left[\frac{\zeta}{4}\frac{\gamma_{f}(x)}{\gamma_{0}}(1 - 2k_{0}h(x)\tanh[k_{0}h(x)]) - \frac{1}{\beta ze}\frac{\xi_{f}(x)}{\gamma_{0}}\right].$$
(D2)

Inserting $A_f^{\zeta}(x)$ into Eq. (D1), we find Eq. (66) of the main text,

$$\psi_{f}^{\zeta}(x,y) = \frac{\zeta}{2} \frac{\gamma_{f}(x)}{\gamma_{0}} \left(k_{0}y \frac{\sinh[k_{0}y]}{\cosh[k_{0}h(x)]} - k_{0}h(x) \frac{\cosh[k_{0}y]\sinh[k_{0}h(x)]}{\cosh^{2}[k_{0}h(x)]} \right)$$
$$+ \frac{1}{\beta z e} \frac{\xi_{f}(x)}{\gamma_{0}} \left(1 - \frac{\cosh[k_{0}y]}{\cosh[k_{0}h(x)]} \right)$$
(D3)
$$\psi_{0}^{\zeta}(x,y) \gamma_{f}(x) \quad (h \text{ wterb}[h,y] - h h(y) \tanh[h h(y)])$$

$$= \frac{1}{2} \frac{1}{\gamma_0} \frac{\xi_f(x)}{\gamma_0} \left(1 - \frac{\cosh[k_0 y]}{\cosh[k_0 h(x)]} \right).$$
(D4)

In the case of dielectric walls, we fix $A_f^{\sigma}(x)$ by demanding $\partial_y \psi_f(x, \pm h(x)) = 0$. First, we find

$$\begin{split} \psi_{f}^{\sigma}(x,y) &= A_{f}^{\sigma}(x) \cosh[k_{0}y] \\ &- \frac{e\sigma}{4\epsilon k_{0}} \frac{\gamma_{f}(x)}{\gamma_{0}} \frac{\cosh[k_{0}y] - 2k_{0}y \sinh[k_{0}y]}{\sinh[k_{0}h(x)]} \\ &+ \frac{1}{\beta z e} \frac{\xi_{f}(x)}{\gamma_{0}}, \end{split}$$
(D5)

$$\partial_{y}\psi_{f}^{\sigma}(x,h(x)) = A_{f}^{\sigma}(x)k_{0}\sinh[k_{0}h(x)]$$

. .

$$+ \frac{e\sigma}{4\epsilon} \frac{\gamma_f(x)}{\gamma_0} (1 + 2k_0 h(x) \coth[k_0 h(x)]).$$

We now fix $A_f^{\sigma}(x)$ to

$$\Lambda_f^{\sigma}(x) = -\frac{e\sigma}{4\epsilon k_0} \frac{\gamma_f(x)}{\gamma_0} \frac{1 + 2k_0 h(x) \coth[k_0 h(x)]}{\sinh[k_0 h(x)]}.$$
 (D6)

Using these results, we find Eq. (67),

$$\begin{split} \psi_{f}^{\sigma}(x,y) &= -\frac{e\sigma}{4\epsilon k_{0}} \frac{\gamma_{f}(x)}{\gamma_{0}} \bigg[\frac{\cosh[k_{0}y] - 2k_{0}y\sinh[k_{0}y]}{\sinh[k_{0}h(x)]} + \frac{\cosh[k_{0}y]}{\sinh[k_{0}h(x)]} + \frac{2k_{0}h(x)}{\tanh[k_{0}h(x)]} \frac{\cosh[k_{0}y]}{\sinh[k_{0}h(x)]} \bigg] + \frac{1}{\beta ze} \frac{\xi_{f}(x)}{\gamma_{0}} \\ &= \frac{e\sigma}{2\epsilon k_{0}} \frac{\cosh[k_{0}y]}{\sinh[k_{0}h(x)]} \frac{\gamma_{f}(x)}{\gamma_{0}} (k_{0}y\tanh[k_{0}y] - 1 - k_{0}h(x)\coth[k_{0}h(x)]) + \frac{1}{\beta ze} \frac{\xi_{f}(x)}{\gamma_{0}}, \\ \psi_{f}^{\sigma}(x,y) &= \frac{\psi_{0}^{\sigma}(x,y)}{2} \frac{\gamma_{f}(x)}{\gamma_{0}} (k_{0}y\tanh[k_{0}y] - 1 - k_{0}h(x)\coth[k_{0}h(x)]) + \frac{1}{\beta ze} \frac{\xi_{f}(x)}{\gamma_{0}}. \end{split}$$
(D7)

APPENDIX E: DERIVATION OF LOCAL CHARGE NEUTRALITY AT $\mathcal{O}(\psi f)$

The total ionic charge in a slab at x is obtained by integrating the rhs of Eq. (65) along the transverse direction,

$$ze\bar{q} = ze\bar{q}_{\bar{f}}(x) = 2zeh(x)\xi_f(x) - \beta(ze)^2 \left[\overline{\psi_0}(x)\gamma_f(x) + \overline{\psi_f}(x)\gamma_0(x)\right].$$
(E1)

Here, $\overline{\psi_0^{\zeta}} = 2\zeta \tanh[k_0 h(x)]/k_0$ and $\overline{\psi_0^{\sigma}} = 2e\sigma/(\epsilon k_0^2) = 2\sigma/(\beta z^2 e \gamma_0)$. In the conducting case,

$$\frac{\overline{\psi_{f}^{\zeta}}(x,y)}{2\gamma_{0}} = -\zeta \frac{\gamma_{f}(x)}{2\gamma_{0}} \left(2h(x) \tanh^{2}[k_{0}h(x)] - 2h(x) + \frac{2 \tanh[k_{0}h(x)]}{k_{0}} \right) + \frac{1}{\beta ze} \frac{\xi_{f}(x)}{\gamma_{0}} \left(2h(x) - \frac{2}{k_{0}} \tanh[k_{0}h(x)] \right) \\
= -\zeta h(x) \frac{\gamma_{f}(x)}{\gamma_{0}} \left(\tanh^{2}[k_{0}h(x)] - 1 + \frac{\tanh[k_{0}h(x)]}{k_{0}h(x)} \right) + \frac{2h(x)}{\beta ze} \frac{\xi_{f}(x)}{\gamma_{0}} \left(1 - \frac{\tanh[k_{0}h(x)]}{k_{0}h(x)} \right).$$
(E2)

This gives

$$\overline{q_f^{\zeta}}(x,y) = \underline{2h(x)}\xi_f(\overline{x}) - \beta z e \gamma_f(x) \frac{2\zeta \tanh[k_0 h(x)]}{k_0} + \beta z e \zeta h(x) \gamma_f(x) \left(\tanh^2[k_0 h(x)] - 1 + \frac{\tanh[k_0 h(x)]}{k_0 h(x)} \right) - 2h(x)\xi_f(x) \left(p' - \frac{\tanh[k_0 h(x)]}{k_0 h(x)} \right),$$
(E3)

$$zeq_{f}^{\overline{\zeta}}(x,y) = \beta(ze)^{2}\zeta h(x)\gamma_{f}(x) \left(\tanh^{2}[k_{0}h(x)] - 1 - \frac{\tanh[k_{0}h(x)]}{k_{0}h(x)} \right) + 2h(x)ze\xi_{f}(x)\frac{\tanh[k_{0}h(x)]}{k_{0}h(x)}.$$
(E4)

Meanwhile, the surface charge per surface is given by $e\sigma = -\epsilon \partial_y \psi_f(x, y = h(x))$. We find for the two surfaces

$$2e\sigma(x) = \epsilon \zeta \frac{\gamma_f(x)}{\gamma_0} \Big[k_0^2 h(x) \tanh^2 [k_0 h(x)] - k_0 \tanh[k_0 h(x)] - k_0^2 h(x) \Big] + 2ze\xi_f(x) \frac{\tanh[k_0 h(x)]}{k_0}$$

$$2e\sigma^{\zeta}(x) = \beta(ze)^2 \zeta h(x) \gamma_f(x) \Big[\tanh^2 [k_0 h(x)] - \frac{\tanh[k_0 h(x)]}{k_0 h(x)} - 1 \Big] + \frac{2ze\xi_f(x)}{k_0} \tanh[k_0 h(x)].$$
(E5)

Clearly, the local surface charge $2e\sigma^{\zeta}(x)$ balances the ionic charge $zeq_f^{\zeta}(x, y)$ at each *x*.

In the dielectric case, there are no perturbations to the surface charge: $-\epsilon \partial_y \psi_f(x, y) = 0$, which means that local charge neutrality is satisfied only if the total perturbed ionic density vanishes $\overline{q_f^{\sigma}}(x, y) = 0$. We find

$$\psi_{f}^{\sigma}(x,y) = -\frac{e\sigma}{2\epsilon k_{0}} \frac{\gamma_{f}(x)}{\gamma_{0}} \left(\frac{\cosh[k_{0}y]}{\sinh[k_{0}h(x)]} + k_{0}h(x) \frac{\cosh[k_{0}y]\cosh[k_{0}h(x)]}{\sinh^{2}[k_{0}h(x)]} - k_{0}y \frac{\sinh[k_{0}y]}{\sinh[k_{0}h(x)]} \right) + \frac{1}{\beta z e} \frac{\xi_{f}(x)}{\gamma_{0}}$$

$$\Rightarrow \overline{\psi_{f}^{\sigma}}(x,y) = -\frac{e\sigma}{2\epsilon k_{0}} \frac{\gamma_{f}(x)}{\gamma_{0}} \left(\frac{2}{k_{0}} + \frac{2h(x)}{\tanh[k_{0}h(x)]} + \frac{2}{k_{0}} - \frac{2h(x)}{\tanh[k_{0}h(x)]} \right) + \frac{2h(x)}{\beta z e} \frac{\xi_{f}(x)}{\gamma_{0}}, \quad (E6)$$

$$\beta z e \gamma_{0} \overline{\psi_{f}^{\sigma}}(x,y) = -\frac{2e\sigma}{z e} \frac{\gamma_{f}(x)}{\gamma_{0}} + 2h(x)\xi_{f}(x).$$

With Eq. (E1), we then indeed find that

$$\overline{q_{f}^{\sigma}}(x,y) = 2h(x)\xi_{f}(x) - \gamma_{f}(x)\frac{2\sigma}{z\gamma_{0}} + \frac{2\sigma}{z}\frac{\gamma_{f}(x)}{\gamma_{0}} - 2h(x)\xi_{f}(x) = 0.$$
(E7)

APPENDIX F: DUKHIN NUMBER

Here, we derive the values of the Dukhin number that fit within our approximations. Specifically, we have 54

$$Du = \frac{\lambda^2}{l_{GC}\bar{h}},\tag{F1}$$

$$l_{GC} = 2 \frac{\epsilon k_B T}{z^2 e \sigma},\tag{F2}$$

where the Guy-Chapman length l_{GC} is the length at which the surface electrostatic potential energy equals the thermal energy and λ is the Debye length. We define ψ_{wall} as the dimensionless magnitude of the surface potential

$$\psi_{wall} = \frac{\sigma 1/k_0}{\epsilon k_B T};\tag{F3}$$

hence, we obtain

$$l_{GC} = \frac{2/k_0}{z} \frac{1}{\psi_{wall}} \tag{F4}$$

and

$$Du = \frac{1}{2}\psi_{wall}\frac{1/k_0}{\bar{h}}.$$
 (F5)

In the Debye-Hückel regime, we can assume $\psi_{wall} \lesssim 1$, and hence, we get

$$Du \simeq \frac{1/k_0}{\bar{h}}.$$
 (F6)

Hence, the values of Dukhin number that fit within our lubrication approximation are $Du \lesssim 1$.

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⁵⁵ze $\mathcal{J}_q(x)$ coincides with the usual definition of the streaming current I_{str} [see, for instance, Eq. (1) of Ref. 45] if the channel is flat but differs from I_{str} for a corrugated channel.

⁵⁶From here on, we will omit subscripts *f* when we denote chemical potential differences because $\Delta \tilde{\mu}$ is enforced upon the system, while the local perturbed chemical potential $\mu_f(x)$ is a reaction to that thermodynamic force.

⁵⁷With Eqs. (26) and (27), it is easy to show that $\Delta \gamma = \mu^+(x, y) + \mu^-(x, y)$ and $\Delta \xi = \mu^+(x, y) - \mu^-(x, y)$, i.e., that $\Delta \gamma$ and $\Delta \xi$ are the sum and difference of the full chemical potentials at order $\mathcal{O}(f)$.

⁵⁸Since a chemical potential drop alone cannot induce a solvent flow, $J'_c = J_c$.

⁵⁹This differs from what has been reported for a tracer (see Ref. 35). In the latter case, a phase dependence arose because the concentration of tracers is not affecting the local electric field.

⁶⁰We remark that by using Stokes-Einstein, $D = k_B T/(6\pi\eta R)$ where $R \simeq 0.1$ nm is the linear size of an ion, and the prefactor in Eq. (64b) reads $\epsilon/(\beta ze\mu_{ion}\eta) \simeq 10^{-1}$. ⁶¹We recall that $zeq_0(x,y) = -\epsilon k^2 \psi_0(x,y) = -2_0 \beta \psi_0(x,y)$.