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## Bacterial Cellulose Matrices to Develop Enzymatically Active Paper

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1 **Abstract**

2 This work studies the suitability of bacterial cellulose (BC) matrices to prepare enzymatically  
3 active nanocomposites, in a framework of more environmentally friendly methodologies. After  
4 BC production and purification, two kind of matrices were obtained: BC in aqueous suspension  
5 and BC paper. A lipase was immobilised onto the BC matrices by physical adsorption, obtaining  
6 Lipase/BC nanocomposites. Neither morphology nor crystallinity, measured by scanning electron  
7 microscopy (SEM) and X-Ray diffractometry (XRD) respectively, of the BC were affected by the  
8 binding of the protein. The activity of Lipase/BC suspension and Lipase/BC paper was tested  
9 under different conditions, and the operational properties of the enzyme were evaluated. A shift  
10 towards higher temperatures, a broader pH activity range, and slight differences in the substrate  
11 preference were observed in the immobilised lipase, compared with the free enzyme. Specific  
12 activity was higher for Lipase/BC suspension (4.2 U/mg) than for Lipase/BC paper (1.7 U/mg)  
13 nanocomposites. However, Lipase/BC paper nanocomposites showed improved thermal stability,  
14 reusability, and durability. Enzyme immobilised onto BC paper retained 60% of its activity after  
15 48 h at 60 °C. It maintained 100% of the original activity after being recycled 10 times at pH 7 at  
16 60 °C and it remained active after being stored for more than a month at room temperature. The  
17 results suggested that lipase/BC nanocomposites are promising biomaterials for the development  
18 of green biotechnological devices with potential application in industrials bioprocesses of  
19 detergents and food industry and biomedicine. Lipase/BC paper nanocomposite might be a key  
20 component of bioactive paper for developing simple, handheld, and disposable devices.

21

22 **Keywords:** Bacterial cellulose; lipase immobilization; physical adsorption; nanocomposite;  
23 bacterial cellulose biopaper

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## 1 **Introduction**

2 In recent years has been increasing interest in the design of functional nanocomposites  
3 for advanced biotechnological applications. Nanocomposites consist of the combination  
4 of two types of individual materials, the matrix and the material imbedded on it, being at  
5 least one of the two of nano size dimension. Often, the matrix acts as a scaffold and  
6 supports an organic molecule with biological activity (Mohamad et al. 2015). The matrix  
7 provides the physic-chemical characteristics to the composite, while the molecule in it  
8 imparts biological properties to the matrix. Over the last two decades, the study of  
9 cellulosic nanofibres as the supporting matrix in nanocomposites has become an  
10 increasingly topical subject (Ferreira et al. 2018).

11 Cellulose is very abundant in nature and the biopolymer of choice in many applications.  
12 Traditionally plants have been the main source of cellulose. However, plant-derived  
13 cellulose is always bound to hemicelluloses and lignin and, before further used, it needs  
14 to be purified by enzymatic, chemical and/or mechanical treatments that have a high  
15 economic and environmental impact (Abdul Khalil et al. 2012). Cellulose synthesized by  
16 bacteria is referred to as bacterial cellulose (BC), an extracellular polymer produced by  
17 some microorganisms, especially from the genera *Komagateibacter* (Bielecki et al. 2005).  
18 Apart from being chemically pure (Chawla et al. 2009), BC displays a higher degree of  
19 crystallinity, a higher tensile strength, a higher water-holding capacity and a finer three-  
20 dimensional nanofibre network, being all these features of relevant importance for  
21 practical applications (Yano et al. 2005; Lee et al. 2014). Its three-dimensional open  
22 porous network structure of nanofibres, with a large surface area, is suitable to hold a  
23 large amount of inorganic and organic molecules. Moreover, cellulose contains available  
24 hydroxyl groups in its surface that provide the possibility of molecular adsorption by the  
25 formation of hydrogen bonds and electrostatic interactions (Pahlevan et al. 2018). In fact,

1 BC has been used in the preparation of several composite materials for various  
2 applications, such as in electrical devices, batteries, biosensors, electromagnetic  
3 shielding, biomedical applications or electrochromic devices (Evans et al. 2003; Kim et  
4 al. 2011; Shi et al. 2012; Ul-Islam et al. 2012a, b; Hänninen et al. 2015; Zhou et al. 2019).  
5 Immobilization of enzymes has several advantages, such as easy separation of enzyme  
6 from products in the reaction mix, reusability of the enzyme and increased stability (Wang  
7 2006; Omagari et al. 2009; Kadokawa 2012). These characteristics have promoted the  
8 widely utilization of enzyme immobilization in industry (Klemm et al. 1998; Božič et al.  
9 2012). In biotechnology, immobilized enzyme-based biosensors have huge applications  
10 in various fields as biomedicine, the detection of environmental pollutants or the  
11 monitoring of food safety and industrial bioprocesses (Monosik et al. 2012; Nigam and  
12 Shukla 2015; Rocchitta et al. 2016). The use of nanomaterials as enzyme supports have  
13 expanded its applicability (Molinero-Abad et al. 2014). However, the development of  
14 new nanomaterials that are cheap, highly pure and non-toxic is needed (Kim et al. 2015).  
15 BC is an attractive biocompatible candidate as a carrier for the immobilization of  
16 enzymes. Its porous ultrafine network allows for high accessibility onto the active site,  
17 through low diffusion resistance and easy recoverability as well as potential applicability  
18 for continuous operations (Sulaiman et al. 2015). In addition, BC is considered not only  
19 safer but more environmentally friendly than other nanomaterials (Lu et al. 2013). An  
20 effective enzyme immobilization on BC can be achieved using methods such as covalent  
21 binding or cross linking (Yao et al. 2013; Lin and Dufresne 2014). However, these  
22 methods often require chemical modifications of the matrix and/or the use of chemical  
23 linkers that complicate the procedure limiting the functionality of the composite  
24 generating residues that are harmful for the environment (Castro et al. 2014). Physical  
25 methods for the immobilization of enzymes imply the attachment of the biomolecule to

1 the matrix through physical forces such as van der Waals, electrostatic or hydrophobic  
2 interactions, and hydrogen bonding (Credou and Berthelot 2014). They do not need  
3 chemical modification of either the matrix or the enzyme, allowing minimal configuration  
4 change of the enzyme (Choi 2004). Enzyme immobilization by physical adsorption has  
5 been described for lysozyme onto BC fibres in suspension (Bayazidi et al. 2018), for  
6 lipase onto BC nanocrystals (Kim et al. 2015) and for nisin, laccase and lipase onto BC  
7 membranes (Wu et al. 2017; Yuan et al. 2018; dos Santos et al. 2018). Nevertheless,  
8 obtaining an enzymatically active BC nanocomposite that had the physical characteristics  
9 and the handiness of the paper would be of great interest.

10

11 The aim of this paper was to prepare Enzyme/BC nanocomposites to evaluate the  
12 suitability of BC matrices as supports for enzyme immobilization by physical adsorption.  
13 Among the great variety of enzymes, lipases (EC 3.1.1.3, triacylglycerol hydrolases)  
14 have gained much attention as the most powerful biocatalyst for applications in areas such  
15 as food technology, detergent formulation, flavour and drug production and biofuel  
16 synthesis, among others (Angajala et al. 2016). Therefore, a lipase was chosen due to its  
17 enormous relevance in the development of bioassays and biosensors (Pohanka 2019).  
18 Protein loading, hydrolytic activity and enzymatic stability of lipases immobilized on BC  
19 matrices were evaluated at different conditions of pH and temperature and compared to  
20 free lipase. Matrices of both BC suspension (BCS) and BC paper (BCP) were compared.  
21 To the best of our knowledge, this is the first description of enzyme immobilization in  
22 BC paper, a matrix that combine the high surface-to-volume ratio of the BC nanofibres  
23 with the stiffness and the mechanical properties of paper, and that could lead to the design  
24 of devices for high performance applications.

25

## 1 **Experimental section**

### 2 **Materials**

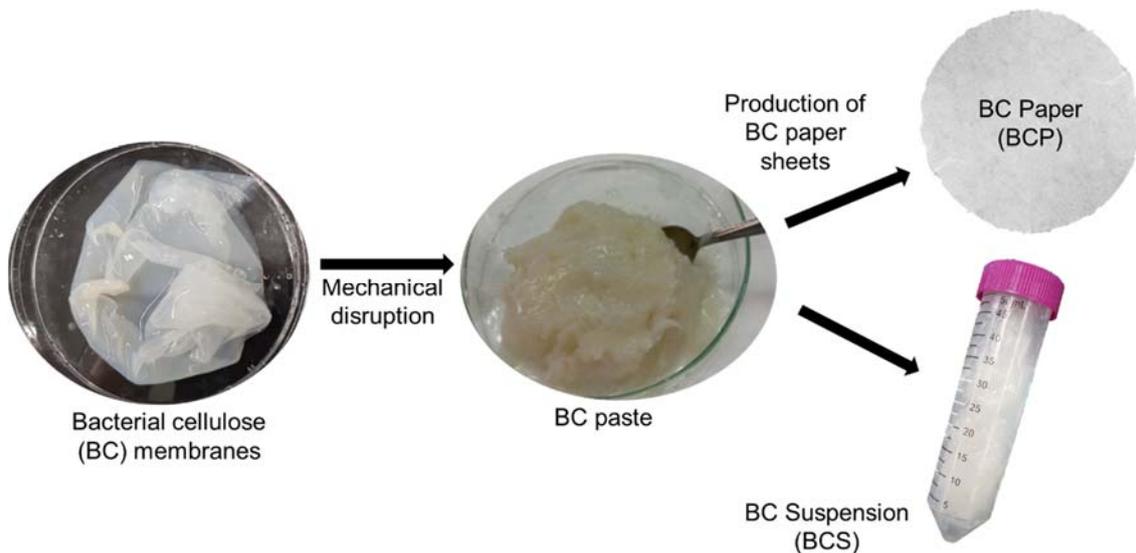
3 *Komagataeibacter intermedius* JF2, a bacterial cellulose producer, was previously  
4 isolated in the laboratory (Fernández et al. 2019). Callera™ Trans L, a commercial liquid  
5 formulation of *Thermomyces lanuginosus* lipase (Nordblad et al. 2014) was supplied by  
6 Novozymes. Lipase (37 kDa, pI 4.4) was purified (elution buffer: 20 mM TrisHCl pH 7,  
7 500 mM NaCl and 0.02 % sodium azide) from the commercial preparation by ionic  
8 exchange chromatography using HiTrap™ Q HP (GE Healthcare) columns in an  
9 AKTA™ FPLC protein purification system.

10

### 11 **Preparation of bacterial cellulose matrices**

12 To produce BC, *K. intermedius* JF2 was grown on the Hestrin and Schramm (HS)  
13 medium, containing 20 g/L glucose, 20 g/L peptone, 10 g/L yeast extract, 1.15 g/L citric  
14 acid, 6.8 g/L Na<sub>2</sub>HPO<sub>4</sub>, pH 6. The cultures were statically incubated at 25 – 28 °C for 7  
15 days. After incubation, bacterial cellulose membranes generated in the air/liquid interface  
16 of the culture media were harvested, rinsed with water and incubated in 1% NaOH at 70  
17 °C overnight. Finally, the BC membranes were thoroughly washed in deionized water  
18 until the pH reached neutrality. Membranes were mechanically disrupted with a blender  
19 and homogenized (Homogenizing System UNIDRIVE X1000) to obtain a BC paste  
20 containing a suspension of BC fibres. The amount of BC in the suspension was  
21 determinate by drying samples of known weight at 60 °C until constant weight was  
22 reached. The bacterial cellulose paste was used to produce BC paper sheets using a  
23 Rapid-Köthen laboratory former (Frank-PTI) following the ISO-5269:2004 standard  
24 method, obtaining a bacterial cellulose paper (BCP) matrix of a weight of 70 g/m<sup>2</sup>. The

- 1 BC in aqueous suspension (BCS) matrix was obtained diluting the BC paste at to 7 mg/ml.
- 2 Procedures for the generation of CB matrices are schematized in **Fig. 1**.



3  
4 **Fig. 1** Schematic representation of the process to obtain bacterial cellulose matrices

5

### 6 **Preparation of Lipase/bacterial cellulose nanocomposites**

7 Adsorption of lipase to BC matrices was conducted as follows: for BCP matrices, pieces  
8 of 1 cm<sup>2</sup> (7 mg ± 0.2) were immersed into the lipase binding solution (10 µg/ml, 20 mM  
9 TrisHCl pH 7), and incubated at 22 °C with slight shaking for 18 h. Then, samples were  
10 removed, washed twice by dipping in buffer solution (20 mM TrisHCl, pH 7), air-dried  
11 and stored at room temperature. For adsorption of lipase in BCS matrices, a volume  
12 containing 7 mg of dry BC was centrifuged 5 min at 4000 rpm in an Allegra™ X-22R  
13 benchtop centrifuge (Beckman Coulter) to remove excess of water and resuspended in  
14 the same volume of lipase binding solution. After incubation at 22 °C with slight shaking  
15 for 18 h, the solids were separated by centrifugation and washed twice with the buffer  
16 solution to remove the unbound enzyme. Finally, pellet was resuspended in the same  
17 buffer and stored at 4 °C before used.

18



1 crystallinity index (CI) of produced bacterial cellulose was calculated based on **equation**  
2 **(2)** (Segal et al. 1959):

$$3 \quad CI(\%) = \frac{I_c - I_{am}}{I_c} \times 100$$

4 (2)

5 where  $I_c$  is the maximum intensity of the lattice diffraction and  $I_{am}$  is the intensity of the  
6 peak at  $2\Theta = 18^\circ$ , which corresponds to the amorphous part of cellulose. The intensity of  
7 the peaks was measured as the maximum value obtained for the peak considering a  
8 baseline.

9

### 10 **Water Absorption Capacity**

11 To assess the water absorption capacity of the BC paper, samples were weighted and  
12 immersed in deionized water for 24 h. After 24 h, excess of water was removed, and the  
13 weight was measured. The WAC was expressed according **equation (3)**:

$$14 \quad WAC = \frac{W_{wet} - W_{dry}}{W_{dry}}$$

15 (3)

16 where  $W_{wet}$  is the weight of wet BC paper and  $W_{dry}$  is the initial weight of the dried BC  
17 paper.

18

### 19 **Operational properties of the immobilized lipase**

20 Lipase hydrolytic activity was analysed by measuring the release of MUF  
21 (methylumbelliferone) from MUF-derivate fatty acid (C4, C7, and C18) substrates  
22 (Sigma-Aldrich). Stock solutions of MUF-substrates were prepared at 25 mM in ethylene  
23 glycol methyl ester (EGME). The working solution contained 250  $\mu$ M of MUF-substrates

1 in 20 mM TrisHCl pH 7. MUF was measured using a Varian Cary Eclipse  
2 spectrofluorometer (Agilent Technologies) equipped with a microplate reader, as  
3 previously reported (Panizza et al. 2013). One unit of activity was defined as the amount  
4 of enzyme that released one mmol of MUF per minute under the assay conditions. The  
5 specific activity and the recovery of lipase activity were calculated using **equation (4)**  
6 and **equation (5)**.

$$\begin{aligned} & \text{Specific activity of immobilized lipase } \left( \frac{\text{U}}{\text{mg protein}} \right) \\ & = \frac{\text{Activity of immobilized enzyme } \left( \frac{\text{U/ml}}{\text{g BC}} \right)}{\text{Protein loading } \left( \frac{\text{mg protein/ml}}{\text{g BC}} \right)} \end{aligned} \quad (4)$$

10

$$\text{Recovery of lipase activity (\%)} = \frac{\text{Specific activity of immobilized lipase}}{\text{Specific activity of free lipase}} \times 100 \quad (5)$$

13 The determination of the operational characterization and properties of Lipase/BC  
14 nanocomposites was carried out with samples containing 7 mg BC (dry weigh)/ml. For  
15 Lipase/BCP, the nanocomposites were immersed into the appropriated buffer at the  
16 conditions being analysed. For Lipase/BCS nanocomposites, samples were centrifuged,  
17 and solids were suspended into the appropriated buffer, at the conditions being analysed.  
18 Assays of free lipase activity were run in parallel.

19

#### 20 *Influence of temperature and thermal stability*

21 Optimum temperature of free and absorbed lipase was determined by the analysis of the  
22 activity over a range from 30 °C to 90 °C at pH 7. Long-term thermal stability was

1 analysed based on the residual activity of lipase measured after incubation at 60 °C in 20  
2 mM TrisHCl pH 7 for a determinate period of time.

3

#### 4 *Influence of pH*

5 Optimum pH of free and absorbed lipase was determined by analysis of the activity at  
6 various pH values with the appropriate buffers 20 mM: acetate buffer (pH 4 and 5),  
7 phosphate buffer (pH 6) and TrisHCl (pH 7, 8 and 9).

8

#### 9 *Determination of kinetics constants*

10 The determination of the Michaelis-Menten constant ( $K_m$ ) and the maximum reaction  
11 rate ( $V_{max}$ ) of both free and immobilized lipase were carried out using MUF-butyrate as  
12 substrate, with initial concentrations varying from 50  $\mu$ M to 1000  $\mu$ M. The kinetic  
13 parameters were calculated by fitting hyperbolic Michaelis-Menten curves with  
14 GraphPad Prism 6 software (San Diego, California).

15

#### 16 **Statistical analysis**

17 All determinations of enzyme activity were performed after two replicas of triplicates (6  
18 determinations per sample). Experimental data were expressed as means  $\pm$  standard  
19 deviations and were analysed statistically by the paired Student's *t*-test method and  
20 analysis of variance (ANOVA) in STATGRAPHICS Centurion XVIII software  
21 (Statgraphics.Net, Madrid) among more than two groups. Scheffe's multiple range test  
22 was used to detect differences among mean values. A value of  $p \leq 0.05$  was considered  
23 statistically significant. Bartlett's test was used to test homogeneity of variance for all  
24 samples. Assumption that the residuals were normally distributed was tested with the  
25 Shapiro-Wilk test.

26

## 1 **Results and Discussion**

### 2 **Adsorption of lipase in BC matrices and activity of the BC nanocomposites**

3 pH plays a key role in the immobilization process by physical adsorption of proteins onto  
4 cellulose (Lin et al. 2015). The optimal pH for lipase immobilization into BC paper was  
5 tested in acetate buffer at pH 3 and 5, and TrisHCl buffer pH 7. Buffers concentration of  
6 the lipase binding solution was 20 mM, as it is described that lower ionic strength can  
7 enhance the formation of the protein/polysaccharide composites (Chai et al. 2014). The  
8 amount of protein adsorbed into BC paper at the different pHs is shown in **Table 1**.  
9 Results indicated that the highest efficiency of lipase adsorption was at pH 3, suggesting  
10 electrostatic bounding between the positive charged protein and the overall negative  
11 charge cellulose due to hydroxyl groups and molecular dipole. However, when enzymatic  
12 activity of the obtained Lipase/BC paper nanocomposites was measured under standard  
13 conditions, at pH 7, the enzyme immobilized at pH 3 showed less specific activity than  
14 the immobilized at pH 5 and at pH 7 (Table 1). These results suggested that the buffer at  
15 pH 3 used for the immobilization process, while favouring its binding, inactivated the  
16 enzyme. Therefore, and to maintain the same working conditions as in the determinations  
17 of enzymatic activity, further adsorption experiments were conducted with 20 mM Tris  
18 at pH 7, conditions that allowed both good adsorption and specific activity.

19

20 **Table 1.** Effect of the pH on the adsorption of the lipase onto BC paper and enzymatic  
21 activity of the adsorbed lipase.

pH	Lipase adsorbed ( $\mu\text{g}/\text{cm}^2$ )	Specific activity (U/mg protein)
pH 3	$5.29 \pm 0.4$	$0.89 \pm 0.03$
pH 5	$2.90 \pm 0.36$	$1.99 \pm 0.02$
pH 7	$2.65 \pm 0.11$	$1.69 \pm 0.11$

22

1 Lipase was physically adsorbed onto BC cellulose fibres, both in aqueous suspension  
2 (BCS) and in paper (BCP) obtaining Lipase/BCS and Lipase/BCP nanocomposites,  
3 respectively (**Table 2**). BCS showed higher capacity to adsorb protein than BCP, which  
4 could be attributed to the difference in the density of their nanofibrils' network that  
5 influences the accessibility of the protein to the matrix of cellulose. Moreover, during the  
6 process of paper production to obtain BCP matrices, the fibres of cellulose undergo  
7 dehydration through evaporation of water. The loss of the molecules of water produce  
8 irreversible formation of new hydrogen bounds between the hydroxyl groups of adjacent  
9 glucan chains that would hinder the diffusion of the protein (Seves et al. 2001). To test if  
10 this structural modification would affect the adsorption properties of the BC matrix, its  
11 water absorption capacity (WAC) was measured (equation 3). BCP showed a WAC of  
12  $263 \pm 28\%$ , around 37 times its dry weight. As expected, WAC of BC paper was lower  
13 than that reported for never dried native BC membranes (Meftahi et al. 2010). However,  
14 Fernandez *et al.* (Fernández et al. 2019) reported values of WAC of only 10 – 20 % for  
15 dry films of BC membranes. The results obtained indicated that the dry BC paper matrices  
16 maintained enough WAC to carry out adsorption assays by immersing the paper matrix  
17 in the aqueous solution of the enzyme for its immobilization. Nevertheless, in BCP,  
18 probably most of the protein binding is taking place only in the most superficial layers of  
19 fibres of the matrix.

20 The obtained nanocomposites were enzymatically active, although specific activity of the  
21 immobilized enzyme decreased with respect to that of the free enzyme (Table 2). This is  
22 a common phenomenon described previously for a variety of enzymes and immobilizer  
23 supports (Lian et al. 2012). However, differences in the specific activity between the two  
24 types of Lipase/nanocomposites were observed. Enzyme bounded to BCS maintained  
25 about 68 % of activity respect to the free enzyme, while enzyme bounded to BCP

1 maintained only 28%, approximately (Table 2). The decrease of lipase activity after  
 2 immobilization may be due to the changes in structural conformation of lipase and lower  
 3 accessibility of substrate to its active sites (Kim et al. 2015). After the enzyme is  
 4 entrapped and immobilized in the porous network of BC, more mass transfer resistance  
 5 forms compared to the free enzyme, impairing the binding efficiency between the enzyme  
 6 and the substrate (Chen et al. 2015). This effect is more accused for the lipase adsorbed  
 7 onto BCP, which is a less porous matrix than BCS owing to its higher fibre density after  
 8 water evaporation. Consequently, the lipase has less diffusional mechanisms, influencing  
 9 the activity of the enzyme (Estevinho et al. 2014).

10

11 **Table 2.** Characteristics of lipase immobilized onto BCP and BCS matrices.

<b>Matrix</b>	<b>Adsorbed protein (<math>\mu\text{g/g BC}</math>)</b>	<b>Specific activity (U/mg protein)</b>	<b>Recovered activity (%)</b>
Free enzyme	-	$6.13 \pm 0.4$	-
Lipase/BCP nanocomposite	$416.37 \pm 85$	$1.69 \pm 0.11$	27.6
Lipase/BCS nanocomposite	$737.35 \pm 106$	$4.15 \pm 0.14$	67.7

12

### 13 **Physical characterization of Lipase/BC nanocomposites**

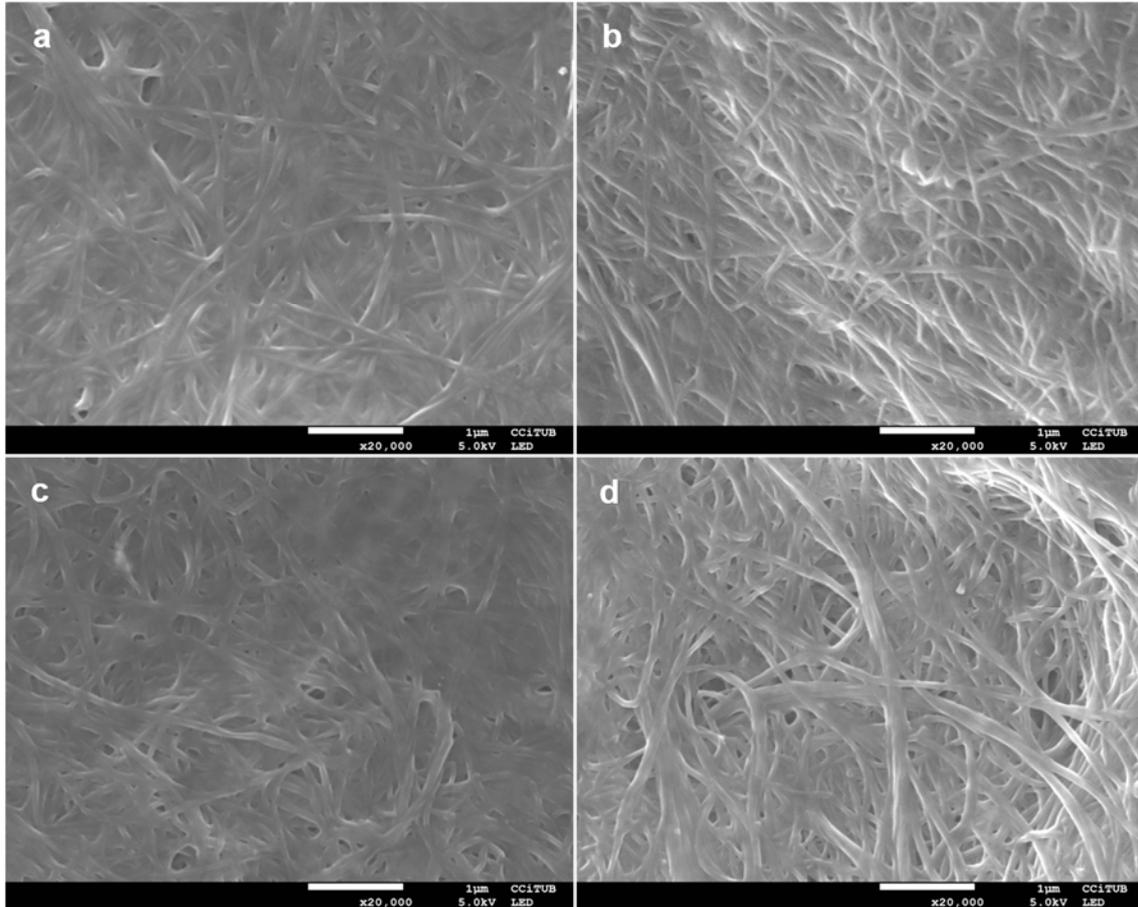
14 Lipase/BC nanocomposites were characterized in terms of morphology and chemical  
 15 structure and crystallinity by SEM and XRD, respectively.

16

#### 17 *Morphology observation by SEM*

18 SEM images of BCP and BCS matrices are shown in **Fig. 2**. In both of them, it could be  
 19 observed a connected structure consisting of ultrafine cellulose fibrils with a diameter of  
 20 about 50 - 70 nm, which results in large surface area. This high surface area and porous  
 21 features of BC would provide microchannels to entrap enzyme and would improve the  
 22 contact area exposed to protein molecules (Chen et al. 2015). BCP fibres disposition was

1 more flawless than in BCS fibres, where the fibres displayed a higher density. No changes  
2 were observed in the morphology or in the arrangement of the nanofibers after  
3 immobilization of the lipase.



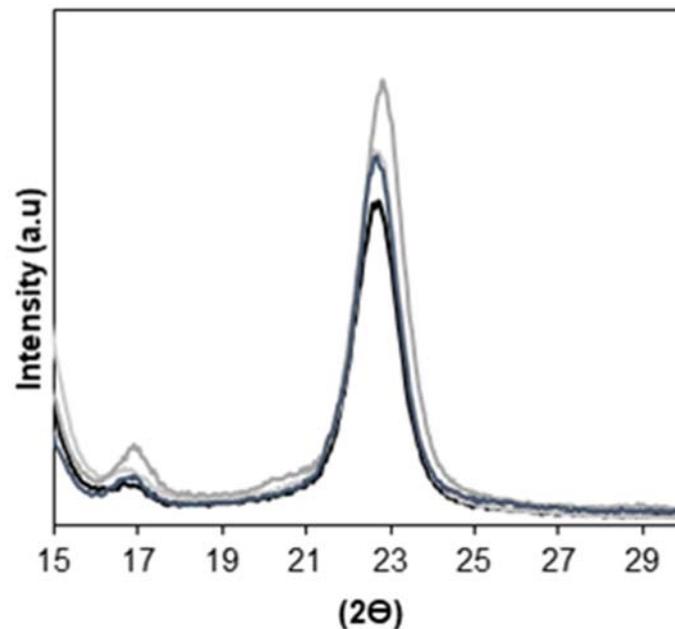
4  
5 **Fig. 2** Scanning electron microscopic (SEM) images of BCP **(a)**, BCS **(b)**, Lipase/BCP  
6 nanocomposite **(c)** and Lipase/BCS nanocomposite **(d)**

7

### 8 *Crystallinity*

9 XRD patterns of Lipase/BC nanocomposites were measured. **Figure 3** shows diffraction  
10 peaks at  $2\theta$  angles around  $18.4^\circ$  and  $22.7^\circ$ ; the presence of which were ascribed to the  
11 typical profile of cellulose I (natural cellulose) in crystalline form (Chen et al. 2015) for  
12 BCP and BCS matrices and their Lipase/nanocomposites. Even though immobilization of  
13 lipase caused a slight broadening of all peaks, intensities were not dramatically changed.  
14 The estimated degree of crystallinity index (equation 2) of the pure BC was 94% for BCP

1 and 93% for BCS. With the introduction of lipase, the crystallinity index did not change  
2 (94% for both nanocomposites). These results indicated that no changes in the crystalline  
3 structure within the cellulose fibres did occur during the incorporation of lipase by  
4 physical adsorption, suggesting that characteristics as mechanical strength and interfacial  
5 properties of the cellulose fiber were not modified (Huang et al. 2014).



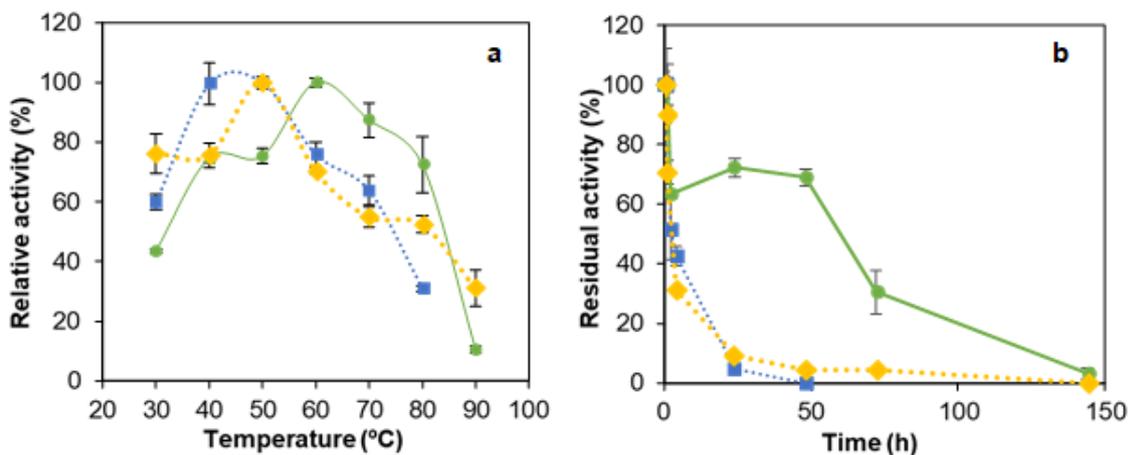
6  
7 **Fig. 3** XRD patterns of BCP (black line), BCS (grey line), Lipase/BCP nanocomposite (light grey  
8 line) and Lipase/BCS nanocomposite (dark grey line)

## 10 **Operational properties of Lipase BC/nanocomposites**

### 11 *Effect of the temperature and thermal stability*

12 The effect of temperature on the activity of free and immobilized lipase was studied in  
13 the temperature range of 30 – 90 °C (**Fig. 4a**). Free enzyme had its optimum temperature  
14 activity between 40 – 50 °C, while Lipase/BCS nanocomposite retained its maxim activity  
15 at 50 °C. Remarkably, Lipase/BCP nanocomposite shifted its optimal temperature to 60  
16 – 70 °C and, in addition, it broadened the range of temperature where the enzyme can be  
17 active. This shift of the optimum temperature suggested an increase in the thermal

1 stability of the immobilized lipase, and it could be related to the change of structural  
2 stabilisation of the immobilized enzyme (Chen et al. 2015). In fact, the thermal stability  
3 of lipase at 60 °C was highly enhanced by the adsorption of lipase onto BCP (**Fig. 4b**).  
4 After 2 h of incubation, the residual activity of free lipase and Lipase/BCS nanocomposite  
5 was about 50 % whereas the lipase immobilized onto BCP retained more than 60 % of  
6 enzymatic activity after 48 h. These results highlight that the BCP matrix provided a  
7 framework of great stability for the activity of the lipase at elevated temperatures. This  
8 enhanced stability could be attributed to the restricted conformational mobility of the  
9 entrapped lipase molecules after immobilization (Frazão et al. 2014) onto cellulose  
10 matrix, delaying the rate of inactivation (Yuan et al. 2018), and it has been reported by  
11 other authors for other BC supports as BC membranes (Yuan et al. 2018) and BC  
12 nanocrystals (Kim et al. 2015).



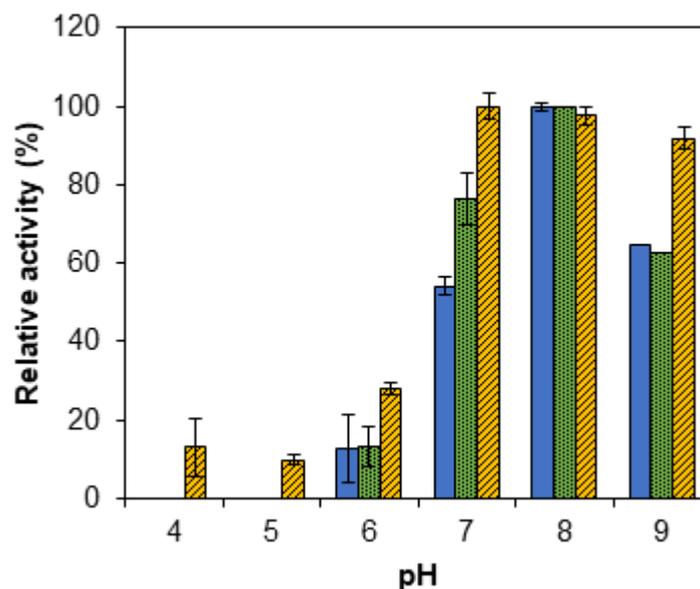
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14 **Fig. 4** Lipase activity at different temperatures (**a**). Activity was expressed in relative values, with  
15 the highest activity denoting 100%. (**b**) Thermal stability at 60 °C under different times of  
16 incubation, where residual activity was expressed as percentage of the initial activity at time zero.  
17 Squares line = free lipase, dots line = Lipase/BCP nanocomposite, rhombus line = Lipase/BCS  
18 nanocomposite

19

1 *Effect of pH*

2 The effect of pH on the activity of free and immobilized lipase was tested under various  
3 pH (**Fig. 5**). The general profiles of the pH dependency were very similar; pH 8 was  
4 optimal for free enzyme and Lipase/BCP nanocomposites. Nevertheless, the higher  
5 activity of Lipase/BCS nanocomposites was preserved at a wider range of pH, showing  
6 the highest activity between pH 7 and 9. Very low activity was detected at pH lower than  
7 6.



8

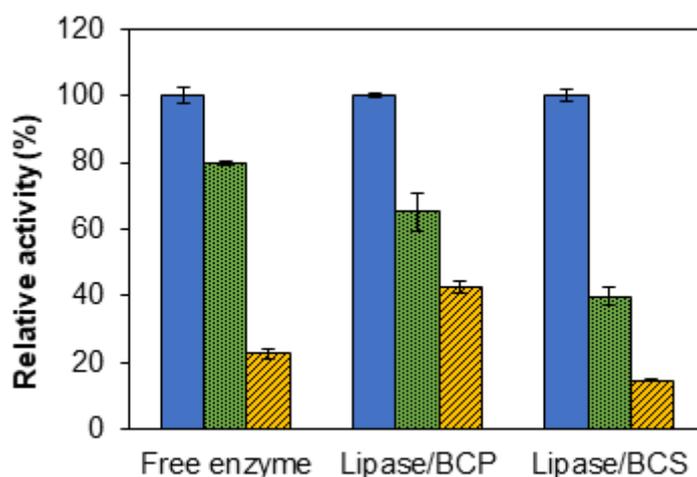
9 **Fig. 5** Effect of pH on free lipase and the Lipase/BC nanocomposites. Activity was expressed in  
10 relative values, with the highest activity denoting 100%. Solid bars = free lipase, dot bars =  
11 Lipase/BCP, line bars = Lipase/BCS

12

13 *Specificity of substrate length*

14 Specificity of substrate length of Lipase/BCS and lipase/BCP nanocomposites was tested  
15 on MUF-derivative fatty acid of different chain-length and compared with that of the free  
16 lipase. The free enzyme and the enzyme immobilized onto BCS and BCP nanocomposites  
17 showed activity on butyrate (C4) heptanoate (C7) and oleate (C18) (**Fig. 6**). Butyrate was

1 the optimum substrate, with significant differences regarding the other two assayed  
2 substrates. However, even if all of them demonstrated the same profile of relative activity,  
3 the most striking result to emerge from the data was that with oleate. Interestingly,  
4 Lipase/BCP nanocomposite showed higher activity with oleate than Lipase/BCS  
5 nanocomposite and free lipase, suggesting that the lowest water content matrix could  
6 better accommodate more hydrophobic substrates.



7  
8 **Fig. 6** Specificity of substrate for the free lipase and the Lipase/BC nanocomposites: butyrate  
9 (solid bars), heptanoate (dot bars) and oleate (line bars). Activity was expressed in relative values,  
10 with the highest activity denoting 100%

### 12 *Kinetic constants*

13 Enzyme activity was measured at different substrate concentrations (50 – 1000  $\mu$ M) with  
14 free and immobilized lipases. The kinetic data was fitted to the Michaelis-Menten  
15 equation and parameters were calculated. The kinetic parameters are summarized in  
16 **Table 3**. Both  $K_m$  and  $V_{max}$  were affected by immobilization process.  $K_m$  has higher  
17 values in immobilized enzyme than the in the free one: in Lipase/BCP nanocomposite it  
18 was almost the double, whereas in Lipase/BCS nanocomposite an approximately 4-fold  
19 increase was observed, indicating a weaker attachment of substrate to enzyme.

1 Diffusional limitations due to the immobilization of the enzyme would cause a lower  
2 affinity for the substrate.

3

4 **Table 3.** Kinetic constants of free and immobilized lipase.

	<b>K<sub>m</sub> (μM)</b>	<b>V<sub>max</sub> (U/mg protein)</b>
Free lipase	169.9 ± 25.75	3.03 ± 0.18
Lipase/BCP nanocomposite	276.9 ± 53.02	3.31 ± 0.32
Lipase/BCS nanocomposite	659.0 ± 264.3	6.44 ± 1.7

5

6 During the process of immobilization by physical adsorption, the orientation of the  
7 immobilized lipase on the matrix was not a controlled process. Therefore, an improper  
8 fixation could hinder the active site for binding of substrates to the immobilized enzyme  
9 (Yang et al. 2010). The increasing of kinetic parameters correlates favourably with  
10 previous studies on enzyme immobilization (Bayazidi et al. 2018).

11

### 12 **Leaching of the lipase from the nanocomposites**

13 The stability of immobilized lipase onto BC matrices was determined. Lipase/BC  
14 nanocomposites were incubated in buffer solution (20 mM TrisHCl pH 7) at room  
15 temperature and enzymatic activity was measured in the solution at several times. Lipase  
16 activity was not detected at times 0, 24 and 48 h. After 72 h, only about 4 % of the lipase  
17 activity was released from the Lipase/BCP nanocomposites, whereas for Lipase/BCS  
18 nanocomposites no leaching of activity was detected (**Table 4**). At this time, the activity  
19 that remained in the nanocomposites was measured. The results indicated that  
20 Lipase/BCP nanocomposites maintained 100 % of the activity, in accordance with the  
21 results obtained from the leaching of the activity. However, Lipase/BCS nanocomposites  
22 retained only 33 % of the activity, suggesting that the enzyme lost activity during the 72  
23 h incubation at room temperature (Table 4).

1 It has been described that the interactions by physical adsorption between enzymes and  
 2 plant cellulose supports would be not strong enough to ensure permanent immobilization  
 3 and to prevent, consequently, the leaking of the biomolecules (Credou and Berthelot  
 4 2014). Nevertheless, in this study, the lipase adsorbed onto BCP matrices seemed to be  
 5 strongly entrapped. Probably, the high density and specific surface area provided by BC  
 6 nanofibers resulted in more available hydroxyl groups where the lipase can be adsorbed  
 7 (Skočaj 2019). Moreover, the porous three-dimensional structure of nanofibers would  
 8 help to retain the enzyme.

9

10 **Table 4.** Activity and leaching of lipase from BC nanocomposites

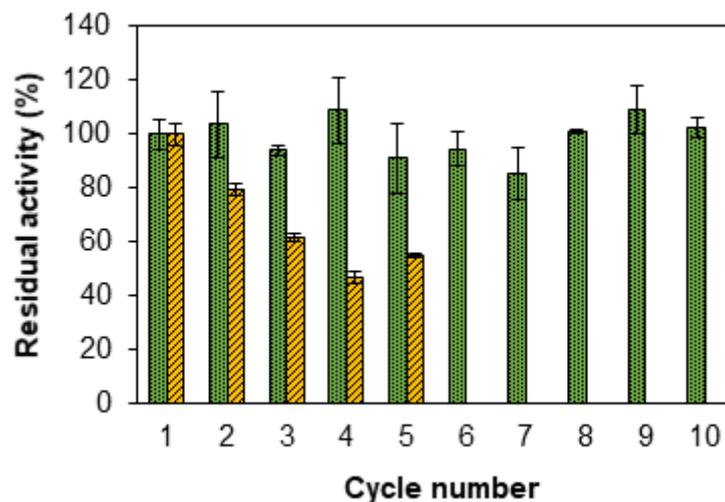
	t = 0 h	t = 72 h		
	mU/ml	mU/ml	% leaching	% remaining activity
Lipase/BCP	5.6 ± 0.03	5.74 ± 0.90	3.9	100
Lipase/BCS	20.8 ± 0.4	6.93 ± 0.62	0	33

11

12 **Reusability of Lipase/BC nanocomposites**

13 To determine the reusability of the Lipase/BC nanocomposites activity was measured.  
 14 Then, Lipase/BCP nanocomposites were rinsed twice by immersion in 20 mM TrisHCl  
 15 pH 7 and allowed to air-dry before the following activity assay. Lipase/BCS  
 16 nanocomposites were rinsed by centrifugation and resuspension of the pellet in 20 mM  
 17 TrisHCl pH 7. A third centrifugation allowed the resuspension of the pellet in the reaction  
 18 buffer for subsequent lipase activity assay. These operational cycles were repeated 10  
 19 consecutive times. Results are shown in **Fig. 7**. The activity of the BCS nanocomposites  
 20 gradually decreased with the subsequent cycles, retaining 55 % of the original activity  
 21 after five recycling times, although significant differences were already detected in the

1 second round of recycling. As for the Lipase/BCP nanocomposites, the activity did no  
 2 decrease along the reusing cycles, without any significant difference. In comparison to  
 3 other supports, as green coconut fibre, where a laccase was immobilized by physical adsorption,  
 4 the composite lost 30 % of its initial activity in the second cycle (Cristóvão et al. 2011). Therefore,  
 5 BCP would stand out as a matrix that allows a notable operational stability. Moreover, the  
 6 efficiency of reusability of lipase on BC paper was higher of that described for a crosslinking-  
 7 immobilized laccase on BC membrane, which showed 69 % of its original activity after seven  
 8 recycling times (Chen et al. 2015). Good reusability of enzyme can lead to significant reduction  
 9 of operational cost which is of utmost relevance for the industry (Silva et al. 2006) and for  
 10 practical applications as biosensors (Nigam and Shukla 2015).



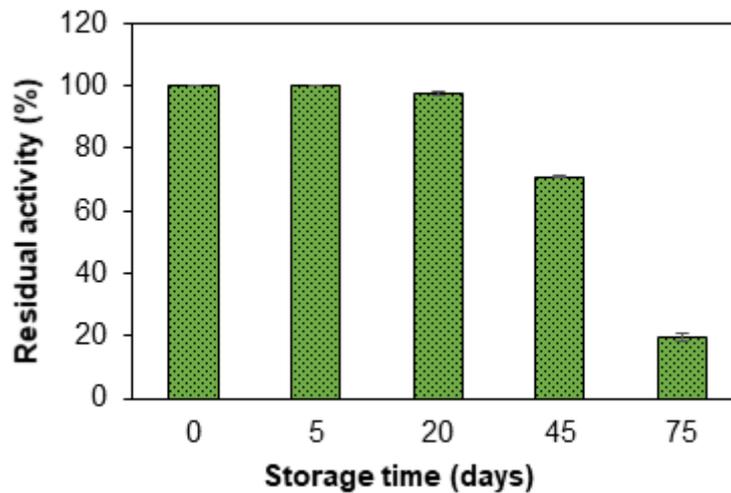
11  
 12 **Fig. 7** Reusability of the Lipase/BCP nanocomposites (dot bars) and the Lipase/BCS  
 13 nanocomposites (line bars). The reusability was expressed as the percent of remaining activity  
 14 where activity from the first run was taken as 100%

15

### 16 **Storage stability of the Lipase/BCP nanocomposites**

17 The effect of storing time on the functionality of Lipase/BCP nanocomposite was studied during  
 18 75 days period. Results showed that nanocomposites with immobilized enzyme could be stored  
 19 at room temperature during at least 20 days without any significant loss of activity, retaining 71  
 20 % after 45 days (**Fig. 8**). These results indicated that neither the enzyme activity nor the cellulose-

1 attachment of the enzyme were compromised under these conditions during several weeks. The  
2 biocompatibility and the three-dimensional network of nanofibers of the BC in  
3 conjunction with the water-free environment of BCP allowed the preservation of the  
4 activity of the enzyme without the need for special storage. Those are essential properties  
5 to be found in biopaper based devices (Crini 2005).



6  
7 **Fig. 8** Stability of Lipase/BCP nanocomposite during a 75-days storage at room temperature.  
8 Residual activity was expressed as percentage of the initial activity at time zero.

## 10 Conclusions

11 In the present work, the immobilization of lipase by physical adsorption, a cost-effective  
12 and environmentally friendly method, generated functional bacterial cellulose-based  
13 nanocomposites. BCS matrices showed higher protein adsorption capacity than BCP  
14 matrices. Likewise, Lipase/BCS presented higher specific activity than Lipase/BCP  
15 nanocomposites. However, enzyme immobilized onto BCP was able to operate at higher  
16 temperatures and showed greater thermal stability. Moreover, Lipase/BCP  
17 nanocomposites maintained their enzymatic activity after several weeks of storage at  
18 room temperature and for, at least, 10 reusing cycles. This study could be the first step  
19 in establishing a process to obtain bioactive BC paper, considering "BC paper" as the

1 material obtained from bacterial cellulose paste in the form of thin sheets that combine  
2 the characteristics of BC nanofibers with the stiffness and physical properties of paper. It  
3 is foreseeable that nanocomposites of BC paper with other enzymes could be obtained.  
4 Enzyme/BCP nanocomposites are of particular interest because they could be used as part  
5 of biosensors devices with applications in many fields including clinical diagnosis,  
6 environmental monitoring, and food quality control. Due to their operational properties,  
7 they could be suitable for point-of-use testing devices.

8

## 9 **Compliance with Ethical Standards**

10

### 11 **Declaration of interest**

12 Authors declare no conflict of interest

13

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