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Porous titanium-hydroxyapatite composite coating obtained on titanium by cold gas spray with high bond strength for biomedical applications

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Graphical abstract



Highlights

- A titanium-hydroxyapatite coating was produced on titanium by Cold Gas Spray
- Higher adhesion and bond strengths values were obtained compared to other methods.
- Cold Gas Spray does not produce any change in hydroxyapatite and titanium phases.
- Osteoblasts adhere, proliferate and differentiate on the Ti-HA coated surfaces.

Abstract

The lack of bioactivity of titanium (Ti) is one of the main drawbacks for its application in biomedical implants since it can considerably reduce its osseointegration capacities. One strategy to overcome this limitation is the coating of Ti with hydroxyapatite (HA), which presents similar chemical composition than bone. Nonetheless, most of the strategies currently used generate a non-stable coating and may produce the formation of amorphous phases when high temperatures are used. Herein, we proposed to generate a Ti-HA composite coating on Ti surface to improve the stability of the bioactive coating. The coating was produced by cold gas spraying, which uses relatively low temperatures, and compared to a Ti coating. The coating was thoroughly characterized in terms of morphology, roughness, porosity and phase composition. In addition, the coating was mechanically characterized using a tensile loading machine. Finally,

biological response was evaluated after seeding SaOS-2 osteoblasts and measuring cell adhesion, proliferation and differentiation. The novel Ti-HA coating presented high porosity and high adhesion and bond strengths. No change in HA phases was observed after coating formation. Moreover, osteoblast-like cells adhered, proliferated and differentiated on Ti-HA coated surfaces suggesting that the novel coating might be a good candidate for biomedical applications.

Keywords: titanium; hydroxyapatite coating; bioactivity; cold gas spray; osteoblast-like cells.

1. Introduction

Although life expectancy has exponentially increased over the last decades, organ failure and traumatic injuries considerably decrease the quality of life of older people. For instance, more than one million patients worldwide undergo annually to total hip arthroplasty (THA) or total knee arthroplasty (TKA) surgery [1]. By 2030, based on the data collected on total joint replacement surgery, the demand for THAs and TKAs is estimated to grow reaching 4 million procedures per year only in USA [2]. Therefore, there is a need to develop new long lasting orthopedic prostheses, either with new designs and/or new materials, with higher reliable performance with the aim to improve their long-term success rates.

Titanium (Ti) has long been used for biomedical applications due to its excellent resistance to corrosion, biocompatibility, good mechanical properties and osteoconductivity [3]. However, titanium implants still present some limitations that

may pose concerns in clinical practice, mainly due to its bioinert nature, i.e. lack of bioactivity [4]. Several strategies are being used to improve the bioactivity of Ti including surface roughening at the micro- and nanoscale level by mechanical [5] or acid/alkali treatments [6,7], or biochemical coating techniques [8,9]. In this latter regard, hydroxyapatite (HA) coatings on Ti result in enhanced bone formation and improved fixation to adjacent bone compared to uncoated Ti [10]. This is because HA is a calcium phosphate similar to the inorganic component of bone [11,12], although it lacks several vicarious ions [13]. However, the long-term stability of the HA coated devices is still controversial mainly due to the unsatisfactory bonding strength between coating/substrate interface, which limits their clinical applications [14].

Several methods are being used for the generation of HA coatings, which can be grouped into two main strategies reviewed in [10]: physical deposition and wet-chemical techniques. Wet-chemical techniques are biomimetic coating processes based on the induction of calcium phosphate nucleation at material surfaces through immersion of biomaterial in a supersaturated calcium phosphate solution [15], including deposition processes such as dip coating [16], sol-gel [17], electrophoretic [18] and aerosol [19], among others. Although these techniques allow the immobilization of biofunctional cues, the obtained calcium phosphate layers are mainly composed of low-crystalline apatite [20] and their dimensions are difficult to control. Physical deposition processes include many different approaches such as sputtering [21], ion-beam assisted deposition [22] and pulsed laser ablation [23–26], but most of them are based on thermal spray processes [27]. HA coatings on metallic implants have been produced by several plasma spray (PS) techniques including atmosphere (APS) [28,29], suspension (SPS) [30,31], controlled atmosphere (CAPS) [32], vacuum (VPS) [33], high velocity oxygen fuel (HVOF) [34], liquid precursor (LPPS) [35], low-pressure (LPPS) [36] and

high velocity suspension flame spraying (HVSFS) [27]. Despite this versatility, only PS processes are actually approved by the FDA to produce HA biomedical coatings.

Nonetheless, due to the extremely high working temperatures, physical deposition methods usually induce the formation of amorphous phases at the implant-coating interface, which have high dissolution rates in contact with body fluids. This HA phase transitions may reduce bonding strength between HA and Ti and also may lead to undesired biological responses related to HA particulate debris that may eventually lead to implant failure, mainly through inflammatory reactions [37,38].

Another strategy that is well accepted by the FDA is the use of porous rough titanium coatings, a Ti plasma-sprayed (TPS) coating that is engineered to have a great surface roughness and large pore size, promoting bone growth into and around the implant. Porous rough TPS coatings possess all the features of a standard PS titanium coating, such as bio-inertness, bio-compatibility, and the ability to coat dissimilar substrates [9]. However, the use of PS technology for the production of such coatings implies costly vacuum installations in order to avoid titanium oxidation.

An interesting alternative is the use of Cold Gas Spraying (CGS) technique [39]. Due to the relatively low temperature used in the process, CGS allows the formation of an HA coating without observing any change in the HA phase. This can also be achieved by radio frequency (RF) magnetron sputtering [40], although thinner coatings and low bonding strengths are usually obtained. Very few attempts have been performed to produce pure HA coatings by CGS due to the difficulties for cohesion among ceramic particles [41–43]. Therefore, a ductile phase can be used to generate composite coatings. Some works deal with Ti-HA mixtures, even achieving a better bond strength (24.45MPa) than APS (10-15MPa) [44]. The HA particles usually appear exposed at the surface of the coating, resulting in enhanced mineralization ability [45], higher

corrosion current and lower corrosion resistance, and better bonding strength (25 MPa) compared to pure HA coating [46]. However, the mechanical stability of those coatings could still be a problem either for surgical operation or after implantation. In a previous study by our group, we studied the influence of different ratios of Ti and HA particle diameter and different percentages of feedstock blends on Ti-HA coating production and its bonding strength [46]. We observed that coatings were successfully obtained presenting a 60 MPa bonding strength when 30% of the blend was HA and 70% Ti, being the particle diameter ratio of Ti-HA of 2:1 approximately. In the present work we wanted to evaluate more deeply this coating and also to evaluate the biological response when osteoblastic cells were cultured on the coating.

The aim of this study was to bioactivate the Ti surface producing a high bonding strength HA coating. To this end, a 70%Ti-30%HA composite coating was obtained by CGS in order to prevent HA phase transformation. The coating was thoroughly characterized in terms of morphology, roughness, porosity, crystallinity, ion release, HA evolution and adhesion strength. The results were compared to a Ti coating also performed by CGS. Moreover, biological response was evaluated by means of adhesion, proliferation and differentiation of osteoblast-like cells to assess the biocompatibility of the coating and the feasibility for its application to biomedical implants.

2. Materials and methods

2.1. Coating production and structural characterization

Commercially pure (c.p.) titanium powder was supplied by GfE (Gesellschaft für Elektrometallurgie, Germany) with a particle size distribution of $90 \pm 22 \mu\text{m}$. The HA powder was provided by Plasma Biotal (Captal 30) (Tideswell, United Kingdom) with a particle size distribution of $50 \pm 10 \mu\text{m}$. Both pure Ti and Ti-HA blends were sprayed at the same previously defined parameters [46]. The blend ratio of the two powders for

spraying was 70%Ti-30%HA (v/v). A high pressure KINETICS® 4000 CGS thermal spraying Gun with N₂ as propelling gas was used to spray the powders, with a maximum gas pressure of 40 bars and a maximum gas temperature of 800°C.

The coating was characterized using a JEOL 5320 Scanning Electron microscope (SEM) and FESEM JEOL J-7100 operating at 15 KeV equipped with energy dispersive X-ray spectroscopy (EDX). EDX line profile and EDX mapping analyses were performed to visualize the appearance of calcium, phosphor and titanium in coating surface.

Microstructure analyses were performed following ASTM E3-01 [47], which consisted in embedding the samples in resin to grind them with different SiC papers. Then, the samples were polished using 1µm alumina abrasive suspension. The samples were chemically etched in 100 ml H₂O, 5 ml HF and 2 ml H₂O₂ solution and the microstructure was observed using a DM5000M optical microscope (Leica, Germany). The average thickness was calculated as arithmetic mean of 50 measurements using ImageJ software.

For the study of the topography, the roughness values were measured with a Map DCM 3D confocal microscope (Leica). The (micro) roughness and waviness values of the samples were extracted from the global profile with a 0.25 mm Gaussian filter.

Micro- and macro-porosity were evaluated by mercury intrusion porosimetry (MIP) using an AutoPore IV 9500 V1.07 equipment (Micrometrics, USA). Analysis was performed to determine pore entrance size distribution (PESD) within the material. Moreover, open macroporosity was evaluated for pore diameters greater than 10 µm.

Siemens D500 X-ray diffraction Bragg–Brentano type $\theta/2\theta$ apparatus, with Cu K α_1 + 2 radiation with $\alpha_1 = 1.54060$ and $\alpha_2 = 1.54443$ at 40 kV and 30 mA, was used to analyze the purity, crystallinity, phase composition and residual stress state of both types of

feedstock powders as well as of both types of coatings (Ti and Ti-HA coatings). Data were collected in 0.02° steps over the 2θ range of $10\text{--}60^\circ$, with a counting time of 2 s per step. The diffraction patterns were compared to the Joint Committee on Powder Diffraction Standards for HA (JCPDS 9-432) and Ti (JCPDS 00-044-1294).

2.2. HA evolution

Ti-HA coatings were immersed in simulated body fluid (SBF) Hank's solution to understand physiological interactions between the implant and the surrounding environment. Since Ca and P are the main ions that contribute to the osseointegration between the prosthesis and bone, an Inductively Coupled Plasma (ICP) technique was used for determining their concentration after 1, 4 and 7 days of immersion in Hank's solution in a CO_2 incubator at 37°C . In addition, release of Ti was also evaluated.

EDX line profile and EDX mapping analyses were performed after 7 days of incubation in Hank's solution to determine the elemental composition of coatings.

2.3. Mechanical Characterization

The adhesion strength of the coatings was measured according to ASTM C633-13 standard [48] at atmospheric temperature on the Ti and Ti-HA CGS coatings glued to uncoated sand-blasted specimens using F1000 glue. Mechanical testing was done in a tensile loading machine with self-aligning devices at $0,025\text{ mm}\cdot\text{min}^{-1}$ of displacement rate. For each coating type, the adhesion of five test pieces was measured in order to ensure a statistically representative average value.

2.4. Cell culture

Human osteoblast-like SaOS-2 cells (ATCC, USA) were cultured in McCoy's 5A medium (Sigma-Aldrich, USA) supplemented with 10% fetal bovine serum (FBS), 2 mM L-glutamine, 2 mM sodium pyruvate, penicillin/streptomycin (50 U ml^{-1} and $50\text{ }\mu\text{g ml}^{-1}$, respectively) and 20 mM HEPES buffer solution (all from Invitrogen, USA) at

37°C with 5% CO₂ in a humidified atmosphere. Medium was changed every two days and subconfluent cells were trypsinized using TrypLE solution (Invitrogen). Cells were seeded at a density of 10 x 10³ cells/sample in each of the following experiments.

2.4.1. Cytotoxicity

Cytotoxicity tests were performed following the ISO 10993-5 standard [49]. Briefly, samples were incubated for 72 h at 37°C in complete medium and supernatants were collected. Serial dilutions of these extracts in complete medium (Non-dilution, 1:1, 1:10, 1:100, 1:1000) were transferred to new microplate wells containing adhered SaOS-2 cells (10.000 cells). Afterwards, cells were incubated for 24 h, washed with phosphate-buffered saline (PBS) and lysed with 100 µl of M-PER (Mammalian Protein Extraction Reagent; Thermo Scientific, USA). Viable cells were quantified measuring the enzymatic activity of lactate dehydrogenase (LDH) using the Cytotoxicity Detection Kit^{PLUS} LDH (Roche Applied Sciences, Germany). The enzymatic activity was measured spectrophotometrically at 492 nm in a PowerWave HT microplate reader (Bio-Tek, USA) and the percentage of cell viability was calculated following the guidelines of the kit.

2.4.2. Cell morphology

Cells were seeded on samples and allowed to adhere for 6h. Then, cells were washed in 0.1 M phosphate buffer (PB) and fixed with 2.5% glutaraldehyde solution in PB for 1 h. Afterwards, samples were washed thrice in PB and dehydrated in ethanol. Complete dehydration was performed in hexamethyldisilazane (HMDS) for 15 min and samples were dry stored in a desiccator. Samples were covered with a thin carbon layer and visualized with a FESEM JEOL J-7100 (JEOL Ltd., Japan) at an operating voltage of 15 kV.

2.4.3. Cell proliferation

Cells were seeded and after 6 h, 3 days, 7 days, 14 days and 21 days samples were rinsed in PBS and lysed with 300 μ l of M-PER. The LDH enzymatic activity of cells at each specified time was measured using the Cytotoxicity Detection Kit^{PLUS}. The number of cells was quantified using a calibration curve with increasing number of cells. Tissue culture polystyrene (TCPS) was used as control.

2.4.4. Cell differentiation

The alkaline phosphatase (ALP) activity was quantified using the same cell lysates obtained in the cell proliferation assay. ALP activity was measured using the SensoLyte pNPP Alkaline Phosphatase Assay Kit (AnaSpec Inc., USA). Reactions were incubated at 37°C and absorbances were acquired at 405 nm using a PowerWave HT Microplate reader (Bio-Tek). A calibration curve was performed using purified ALP provided by the kit. Results were normalized versus their corresponding cell numbers obtained in the proliferation assay and the time of incubation.

2.5. Statistical analysis

All the experiments were performed in triplicate. Results were expressed as mean values \pm standard error of the mean. Non-parametric Kruskal-Wallis tests followed by Mann-Whitney tests with Bonferroni correction were used to determine statistical significant ($p < 0.05$) differences between the means of the different groups.

3. Results and Discussion

3.1. Microstructural characterization of the coating

Figure 1a and 1b show general optical cross section micrographs of Ti and Ti-HA coatings with inner porosity, respectively. The mean Ti-HA coating thickness was 596 ± 74 μ m. Ti-HA interfaces are rather smooth without gaps, presenting a nice bonding. Further examination of the Ti-HA coating transversal section demonstrates the presence

of a porosity gradient growing from the coating/substrate interface to the coating top surface. The typical inner inter-splat porosity showed inside the metallic Ti sprayed coatings was filled by the HA particles in the composite Ti-HA coatings, assuring the mechanical embedding of the HA particles between the deformed Ti spread particles as well as producing interconnected mesh-like porosity paths. There were no cracks, pores or any kind of defects near the coating/substrate interface, indicating the good quality of the sprayed coating.

Noteworthy HA particles were detected over the top surface and through the entire thickness of the composite coating, as well as over the entire inner surface of the pores. Further examination clearly shows the preferential location of the HA particles inside the grain boundaries formed between the adjacent Ti sprayed particles (Figure S1a and S1b). Although this particularity may appear detrimental for both mechanical resistance and adhesion of coatings under mechanical loading, could become a crucial factor to stimulate bone tissue ingrowth inside the porous coating.

A considerable amount of craters was found over the surface of the Ti-HA coatings mainly caused by the impacting HA particles and their subsequent detachment compared to Ti coated surfaces (Figure S1c and S1d), as previously observed [46].

EDX line and EDX mapping demonstrated that the particles observed by SEM are composed of calcium and phosphate (Figure S2).

3.2. Topographic surface characterization

The top surface of the Ti-HA coating is characterized by the presence of deep wells, where HA is preferentially/usually found at the bottom. The sintered feedstock particles have fractured and the crystallites have scattered, a typical behavior in CGS coatings that has been explained in a previous work by our group [41]. At much higher energetic

conditions, typical for spraying Ti, the presence of the wells was more disperse and non-uniform, while at lower temperature conditions, to which HA had been previously sprayed, there was no proper deposition at all. Therefore, an intermediate set-up of spraying conditions was selected as optimal for the deposition of this blend.

The synergic combination of high surface waviness with open macroporosity over the top surface of the coating is considerable beneficial from the mechanical stabilization point of view. The presence of an open surface macroporosity as well as inner porosity should also be beneficial in terms of mechanical interlocking and bone ingrowth, leading to new bone tissue ingrowth with optimal vascularization after implantation of devices.

From Figure 2, it can be observed that the coating surfaces were characterized by a micro-roughness and submicro-roughness profile. The numerical results presented in Table 1 indicated that the waviness and roughness contribution was much higher in the composite Ti-HA coating than in the Ti coating. The average values of the Ti-HA composite presented by Ra, related to the 2D profile of the microroughness topography, were two times higher and the Wa was more than three times higher compared to the values of the Ti coatings. Such differences, also observed in Rz and Wz values, were the result of the HA influence.

3.3. Physical characterization

Figure 3 shows the pore size distribution curves obtained by MIP for both Ti and Ti-HA coatings sprayed by CGS methodology. It is worth noticing that the void associated to the surface macroporous crater-like geometry was mainly measured in the first step of Hg intrusion, while intergranular porosity was measured under greater pressures. The total porosity and the pore average size were smaller for the Ti coating than for Ti-HA

coating. This result confirmed that higher packing density of the splat structure was obtained compared to the single-phase metallic of the Ti coating.

During the CGS progress, the total porosity and the pore size progressively increased from the substrate/coating interface to the top surface. This behavior may be attributed to two main reasons: the pore coarsening effect and the consequent masking effect [50], as well as the progressive reduction of base material density with the consequent loss of compaction effect in the latest films sprayed. This process resulted in the aforementioned high roughness and open crater-like porosity on the surface.

One of the main concerns of coated prostheses is the poor adherence of the HA plasma coating which may lead to failure causing severe problems for the patient. For instance, TS ceramic coatings based on HA or HA/TiO₂ have dealt with this issue, limiting its commercial use [51–53]. In the present study the addition of Ti to HA slightly enhanced the adhesion of the coating to the substrate without reflecting loss in the bonding strength (Figure 4). However, the values of both Ti and Ti-HA coatings were very similar (60 ± 3 MPa and 65 ± 4 MPa, respectively). Tensile adhesion tests of Ti-HA coatings showed promising mechanical properties for clinical applications being higher compared to the Plasma Spraying coatings with FDA certification.

Adhesion values obtained by Ti-HA composite coatings deposited by CGS showed higher values than those obtained by ceramic HA coatings deposited by other thermal spray processes, such as APS (from 5 GPa to 23 MPa) [54,55], AAPS ($26,7 \pm 1,7$ MPa) [56], HVOF (24 ± 8 and 31 ± 2 MPa) [57] or HVSFS (25 MPa) [27]. Furthermore, cold sprayed Ti-HA composite coatings showed higher adhesion values than Ti6Al4V and Ti metallic coatings deposited by APS (with values up to 38 MPa) [55,58,59].

In addition, the adhesion values obtained in the present study were also significantly higher than other Ti-HA coatings deposited by thermal spraying processes, such as APS

(from 23 MPa to 40 MPa) [27,58,60,61], HVOF (from 20 MPa to 40 MPa) [62], AAPS (38.2 MPa) [63], as well as slightly higher than the values obtained by VPS (42 MPa) [27].

The stability of the Ti-HA coating was analyzed after 4 weeks of immersion in SBF solution. The tensile bond strength slightly decreased (55 ± 5 MPa; data not shown) without statistically significant differences compared to non-immersed samples ($p > 0.05$). As a proof of concept, we wanted to evaluate the stability of the Ti-HA coating in SBF to simulate in vitro the effect of body fluids. However, the in vivo scenario is much more complex to reproduce in vitro, since many other factors play a significant role in the coating stability. Considering this limitation, the stability of the coating in the present study maintained similar values after 4 weeks whereas other authors observed a slight degradation in the bond strength of Ti-HA composite coatings [64]. In contrast, pure HA coatings usually experience a very high decrease in bond strength after soaking in SBF [65], indicating that Ti-HA composite coatings are more stable and less prone to biological demineralization.

3.4. Morphological changes upon immersion in Hank's solution

Figure S3 shows the surface morphology of the Ti-HA coating after 1, 4 and 7 days of immersion in the Hank's solution and the release of calcium and phosphate ions. HA could be mainly observed at the valleys of the surface topography over the surface. This means that, by immersion, the dissolution and precipitation processes take place mainly at those sites, which may stimulate the bone ingrowth. Interestingly, some biomimetic apatite particles could be observed in higher magnification images.

After 4 days of immersion, the biomimetic apatite layer grew considerably on the Ti-HA coatings (Figure S3b). The growth of such layer with a very fine structure results

from the complex dissolution-precipitation behavior [66,67]. After 7 days, the structure was preserved but the thickness of the apatite layer increased. Manipulation of the samples during dehydration and/or sputter coating may be the responsible for drying shrinkage and surface crack formation (Figure S3c).

The calcium phosphate layer on the Ti-HA surface increased after 7 days of immersion in Hank's solution, as can be observed in Figure S4. Interestingly, accumulation of calcium and phosphate was observed on areas with no HA particles, as can be observed in EDX line profile (Figure S4b).

Figure S3d shows the evolution of Ca and P ions release during the time of immersion in Hank's solution. An increase of both calcium and phosphate concentrations in the solution after one day of immersion of Ti-HA coated substrates was observed, which indicates dissolution of ions from the coating. Afterwards, calcium and phosphate concentration slightly decreased and kept approximately constant during the immersion, which would indicate a balance between precipitation/dissolution. In contrast, both calcium and phosphate ions maintained constant during the time of immersion and with the same levels than Hank's solution concentrations.

Interestingly, no release of Ti was observed during Hank's immersion in both Ti and Ti-HA coatings. Metallosis, the potentially harmful local and systemic effects of ions and/or particles released from implants, may be produced by dissolution, fretting and wear effects [68]. In the case of CGS, cohesion of titanium particles is usually very high, thereby reducing the risk of metallosis by particle detachment.

3.5. Phase identification of the coatings

XRD results (Figure 5) show that, for both types of CGS coatings, the main phases of the coatings were practically the same than the feedstock powders, i.e. pristine HA and hcp Ti. However, some small peaks appeared that were attributed to oxide titanium phases, such as rutile and anatase (TiO_2) as well as TiO, although the low intensity of both peaks indicate limited amount of these phases. The formation of TiO_2 phase is generally caused by spontaneous oxidation of titanium and titanium alloys in contact with air, while the appearance of TiO could be attributed to the TiO_2 reduction by Ti in an oxygen deficient atmosphere [62,69].

In addition, the peak intensity of the two main phases was reduced and their peak width was broadened after spraying process of the Ti-HA powders. The slightly peak broadening could be explained by the refining of initial feedstock powders due to the fact that particles might be crushed during spraying [70]. Big broad bands were not observed due to the absence of amorphous phases produced by the thermal degradation of HA and Ti during CGS spraying. Noteworthy, TiO and TiO_2 phases were detected in feedstock powders, but no growth of the titanium oxide peak intensity was detected on as sprayed Ti-HA CGS coatings. In this same regard, CaO phase has not been identified neither on the feedstock powder nor over sprayed Ti-HA coatings.

3.6. *Biological characterization*

The percentage of cell viability after 24 h of incubation with different dilutions of medium is shown in Figure 6A. The percentage of cell viability in all the tested conditions was above 90% demonstrating no cytotoxic effects.

The number of cells attached to Ti and Ti coatings were very similar, being approximately half of the seeded cells (Figure 6B). Similar results were obtained in previous works where Ti was coated thorough thermal spraying techniques [71,72].

Surprisingly, the number of cells seeded on Ti-HA was reduced to 25% compared to TCPS. This may be attributed to the elevated roughness obtained after spraying both Ti and HA, since HA coatings obtained through other strategies have demonstrated to induce cellular adhesion [73,74]. Although this roughness could be detrimental for cellular adhesion it is expected that would be beneficial for osteointegration and mechanical fixation of the implant [75–77].

Proliferation assay showed a continuously increase of cell number over time on control Ti samples presenting similar levels than TCPS. In contrast, cells seeded on Ti and Ti-HA coated surfaces slowly increased during the cell culture periods studied (Figure 6C). This delayed increase may indicate that cells are differentiating into osteoblasts, since this process induces cell growth arrest [78]. Simultaneously to such process, a sequential and selective expression of genes is activated resulting in the differentiation into osteoblast lineage [79]. In the present work, both CGS Ti and Ti-HA coatings exhibited significantly higher ALP activity values compared to non-coated Ti and TCPS (Figure 6D). These higher values imply that probably surface topography is playing a significant role in the differentiation of SaOS-2 cells. In fact, roughened surfaces play a very important role in cell response. This could be because roughened surfaces adsorb more proteins, which could mediate integrin-mediated [80] signaling affecting cell differentiation and phenotype [81]. Noteworthy, when surfaces were sprayed with Ti-HA composite highest ALP activity values at each time point were obtained. It may be important to highlight that the presence of HA modified the surface topography of the composite coating compared to Ti coating. In addition, presence of HA crystals on the Ti surface may play a synergistic effect on cell differentiation [82], since these crystals may increase the HA nucleation potential and mineralization initiation [83].

The interaction of cells with the different surfaces was also visualized by FESEM, showing cells completely spread on Ti control surfaces with very few protrusions (Figure S5a and S5b). In contrast, cells cultured on both Ti (Figure S5c and S5d) and Ti-HA (Figure S5e and S5f) coated surfaces presented an elongated shape with huge quantity of filopodia. Interestingly, in Ti-HA coated surfaces these cytoplasmic projections interacted directly with HA particles probably sensing them as chemotropic cues. This behavior has been previously observed on nanocrystalline HA coatings and has been related to preference of cells for nanostructured surfaces [84,85]. In fact, the presence of these projections were also observed on HA and silicon-substituted HA scaffolds, where cells formed numerous filopodia and lamellipodia [86,87].

4. Conclusions

A Ti-HA coating presenting high porosity has been successfully obtained on Ti surface by CGS. The main advantage of this method is the low temperature achieved, which decreased the probability of titanium and/or calcium oxidation and undesirable formation of secondary metastable and/or amorphous HA phases. In this regard, use of this methodology opens the possibility to coat Ti surfaces with desired calcium phosphate formulations without any phase change during the process. In addition, adhesion and bond strength values obtained in the present work suggest long stability of the coatings obtained by CGS. Finally, the novel Ti-HA coating could be a promising candidate for bone applications due to the observed cellular response, especially in terms of osteoblastic cell differentiation.

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Appendix A. Supplementary data

The following is Supplementary data to this article:

The supporting information is available free of charge. Morphology of Ti and Ti-HA coatings by SEM, Morphology and ion release after immersion in Hank's solution for several days, Cell morphology on the uncoated and Ti- and T-HA-coated surfaces.

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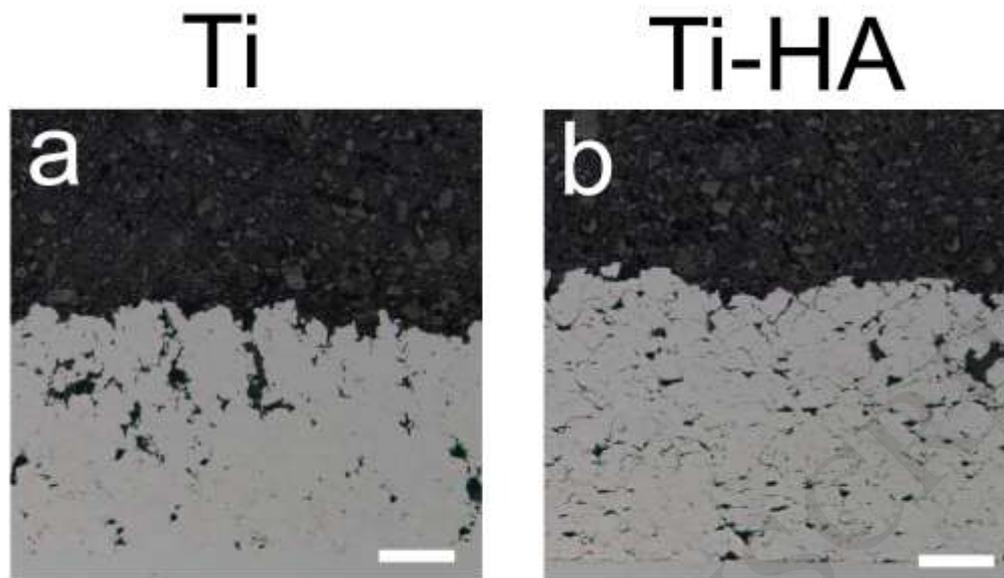


Figure 1: Cross-sectional optical micrographs showing the morphology of the Ti (a) and Ti-HA coatings (b). Scale bar denotes 200 μm .

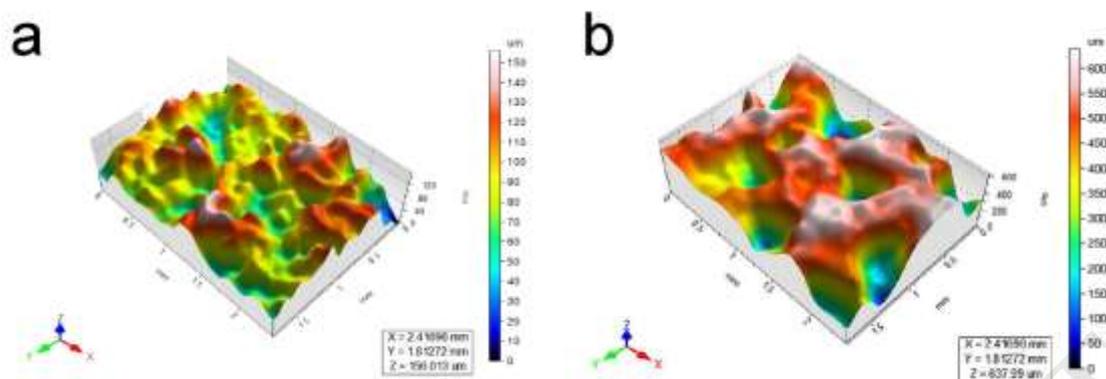


Figure 2. Representative confocal 3D-reconstructed micrographs of Ti coating (a) and Ti-HA composite coating (b).

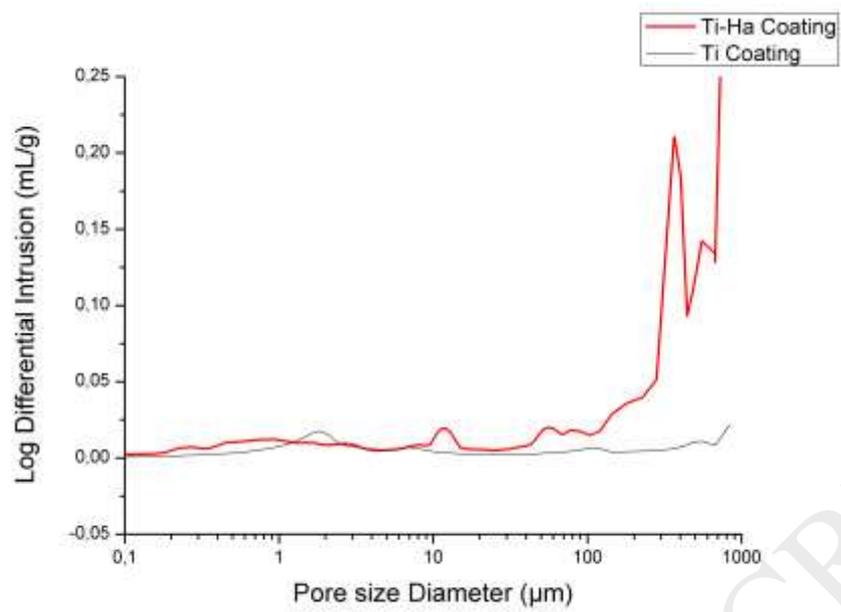


Figure 3. Pore diameter size distribution vs Log Differential intrusion of Ti and Ti-HA CGS coatings.

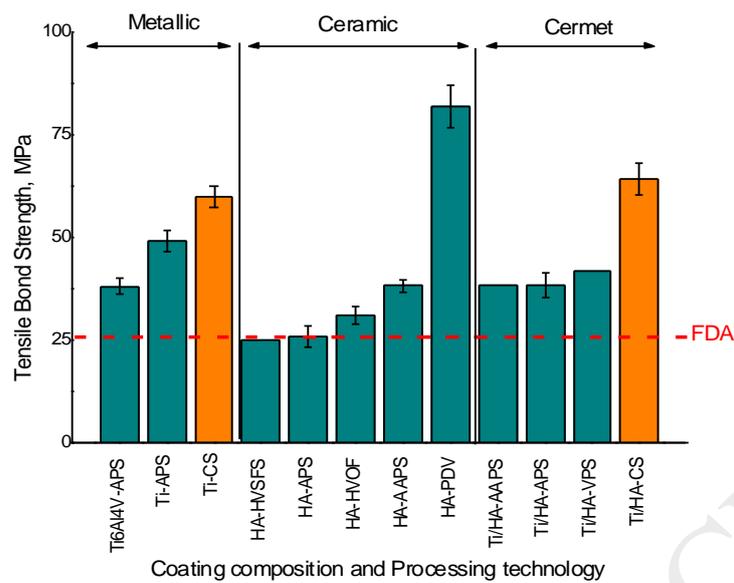


Figure 4. Tensile Bond strength values of coatings (metallic, ceramic and cermet) produced by different technologies including the CGS coatings produced in this study (highlighted in orange). Values in green were obtained from the literature [27,58,60,62,88–90].

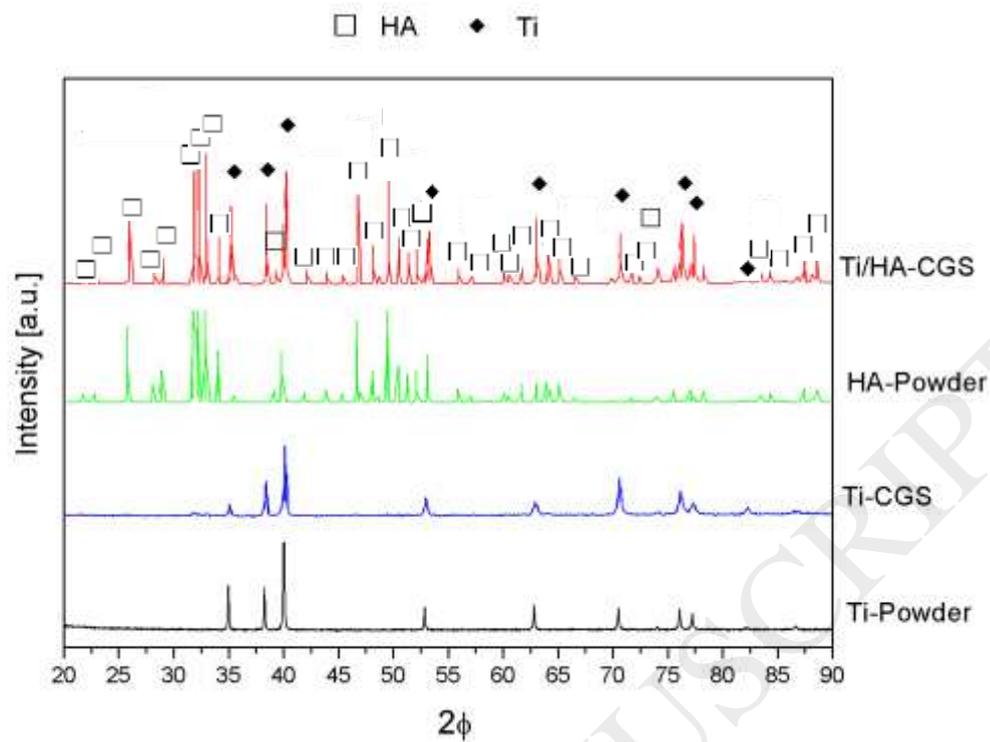


Figure 5. X-ray diffraction pattern of Ti powder, HA powder, Ti coating (Ti-CGS) and Ti-HA coating (Ti/HA-CGS). Diffraction patterns were compared to JCPDS 9-432 (HA) and CPDS 00-044-1294 (Ti).

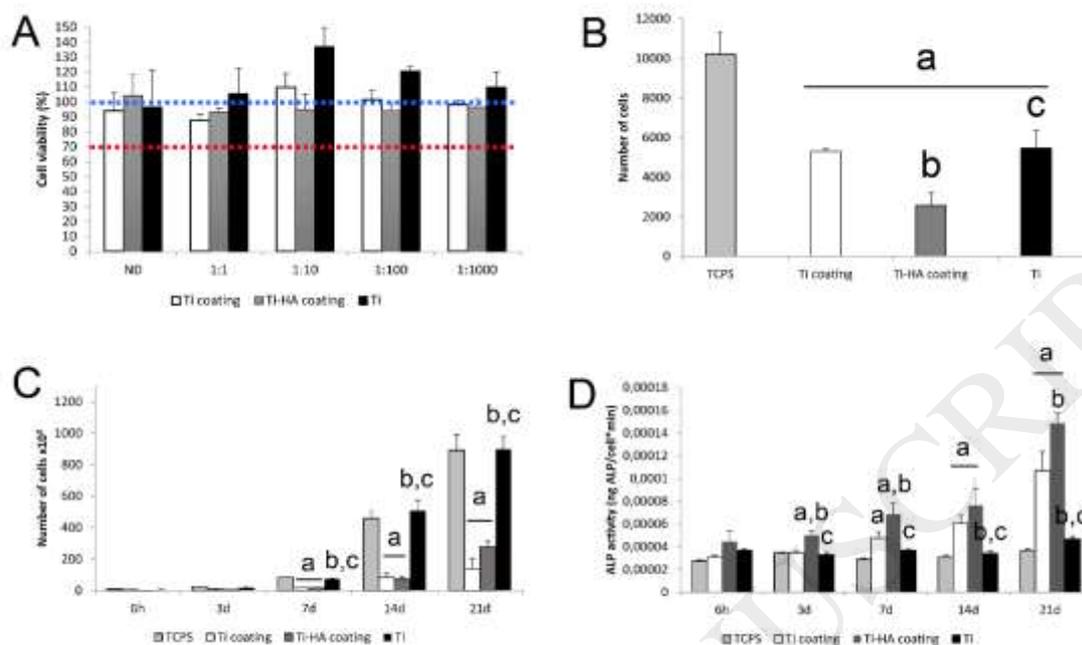


Figure 6. SaOS-2 cell behavior on the different analyzed surfaces. (A) Percentage of SaOS-2 cell viability after 24 h of exposure with different extract dilutions. Red line shows the limit of toxicity and blue line represents the optimal value. ND means non-diluted concentration. (B) Number of SaOS-2 cells after 6 h of cell adhesion on the different substrates. (C) SaOS-2 cell proliferation on the different substrates. (D) ALP activity of SaOS-2 cells cultured on the different substrates. In each figure, “a” means statistically significant differences ($p < 0.05$) compared to TCPs, “b” means statistically significant differences compared to Ti coating and “c” means statistically significant differences compared to Ti-HA coating.

Table 1. Roughness parameters (μm) obtained by confocal microscopy of the as-sprayed CGS coatings.

	Parameter	Ti	Ti-HA composite
Waviness 3D	Sa	17 ± 1	27 ± 2
	Sz	406 ± 87	768 ± 115
Waviness 2D	Wa	19 ± 4	65 ± 3
	Wz	31 ± 6	126 ± 10
Microroughness 2D	Ra	10 ± 1	25 ± 2
	Rz	73 ± 7	201 ± 9