



Carbonate and organic matter sedimentation and isotopic signatures in Lake Chungará, Chilean Altiplano, during the last 12.3 kyr

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ABSTRACT

Sediments in lakes in the Andean volcanic setting are often made up of diatomaceous ooze together with volcanoclastics and small amounts of carbonates. Despite their scarcity, carbonates along with organic matter provide significant paleoenvironmental information about lake systems. This study focuses on the carbonates in Lake Chungará, their morphologies, distribution and origin deduced from the isotopic markers. These markers reflected changes in the water and the biomass between the onset of the Holocene and around 9.6 cal kyr BP. These changes are marked by general increases in TOC, TN, and TN- $\delta^{15}\text{N}_{\text{AIR}}$, and by fluctuating values of TOC- $\delta^{13}\text{C}_{\text{VPDB}}$ in its sediments and are probably related to major shifts in the lake surface/volume associated with rises in lake level. An increase in salinity around 10 cal kyr BP is thought to be linked to a short dry period, giving rise to the onset of carbonate production. The mid-Holocene arid period between 7.3 and 3.5 cal ka BP, with a maximum of aridity around 6.0 cal kyr BP, was deduced from $\delta^{18}\text{O}_{\text{VPDB}}$ values in the endogenic carbonates. These results match the reconstructions in Lake Titicaca based on benthic diatoms and paleoshore levels.

Offshore sediments mainly consist of a diatomaceous ooze, laminated in the lower half (Unit 1), and banded-massive with tephra layers in the upper half of the sequence (Unit 2). TOC- $\delta^{13}\text{C}_{\text{VPDB}}$ and the C/N ratio confirm that phytoplankton was the main source of organic matter in these sediments. Shallower sediments (units 3 to 5) developed in platform and littoral settings, providing evidence of subaqueous macrophytes and, to a lesser extent, land plants. Carbonate content ranges between 0.1 and 6 wt.% in offshore settings (30 to 40 m water depth) and reaches the maximum values in the lower part of Unit 2. Carbonate minerals (low magnesium calcite and minor amounts of high magnesium calcite and aragonite) are scarce and are arranged in mm-thick layers, commonly forming cm-thick levels or bioclasts. Carbonate layers are made up of euhedral-to-subhedral spindle-shaped calcite crystals and, to a lesser extent, aragonite needles, all in the μm range. Aragonite spheroids coexist in littoral sediments with other carbonate shapes and charophyte remains, where carbonate reaches locally up to 20 wt.%. CO_2 photosynthetic depletions related to seasonal phytoplankton blooms were responsible for the high frequency deposition of mm-thick carbonate layers. The average values for $\delta^{13}\text{C}_{\text{VPDB}}$ in lake water, plankton and sediments of Lake Chungará (as organic matter or as carbonate) are around 15‰ higher than commonly reported values in other lakes. This ^{13}C enrichment is attributed to carbon assimilation from a DIC affected by methanogenesis, in which HCO_3^- is the dominant species. The $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$ variations and their covariation in endogenic carbonates suggest that lake water volume and lake level increased along the Holocene.

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1. Introduction

Carbonates in lacustrine sediments provide paleoenvironmental information about biological and hydrological changes through time. Physicochemical changes in lake water cause biota modifications that are frequently associated with carbonate deposition. Moreover, lacustrine

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carbonates contain mineralogical and geochemical markers, such as the isotopic composition of carbon and oxygen, furnishing quantitative data for paleoclimatic and paleohydrological approaches (Li and Ku, 1997; Utrilla et al., 1998; Valero-Garcés et al., 1999a; Ito, 2001; Shapley et al., 2005). Carbonate $\delta^{18}\text{O}_{\text{VPDB}}$ is dependent on temperature and on water $\delta^{18}\text{O}_{\text{VSMOW}}$ (Nelson and Smith, 1966). Shells of benthic organisms formed below the thermocline reflect changes in the isotopic composition of lake water whereas shells of planktonic organisms are mainly affected by variations in water temperature, both having positive offsets between 0 and +2‰ (von Grafenstein et al., 1999). Carbonate $\delta^{13}\text{C}_{\text{VPDB}}$ is related to carbon sources, biomass and biological processes in the catchment and DIC residence time in lake water (McKenzie, 1985; Schwab, 2003). The main controls of DIC- $\delta^{13}\text{C}_{\text{VPDB}}$ in lakes are groundwater and surface water inputs, remains of organic matter from the catchment area (reaching $\delta^{13}\text{C}_{\text{VPDB}}$ –25‰ in the case of land plants), dissolution of older carbonate (0 to +2‰ in marine carbonates), equilibration with atmospheric CO_2 , and ^{12}C selective uptake by aquatic plants during photosynthesis.

Organic matter in recent lake sediments preserves its original $\delta^{13}\text{C}$, $\delta^{15}\text{N}$ and C/N values practically without change for thousands of years (Meyers, 1994). This is essential for recognizing environmental signatures and limnological changes recorded in lacustrine sediments. Organic matter provenance and mixing patterns are well recognized because lacustrine algae (and plankton) have lower C/N values than cellulose land plants. Moreover, the different photosynthetic types of plants can be recognized by their specific $\delta^{13}\text{C}$ signatures (Meyers, 1994; Sifeddine et al., 2004). $\delta^{15}\text{N}$ provides information about contributions of algae, aquatic macrophytes and land plants, and depends on whether nitrogen proceeds from NO_3^- ($\delta^{15}\text{N}$ typically in the range +7 to +10‰) or directly from atmospheric N_2 ($\delta^{15}\text{N}$ around 0‰). Meyers (2003) also used $\delta^{15}\text{N}$ as an indicator of primary productivity.

Carbonates are minor components of sediments deposited in silica-dominated lakes that are often located in active volcanic settings. However, they may provide significant additional information about lake evolution. Diatom dominated lake sediments are common in the Andean Altiplano and have been studied as paleoenvironmental archives at different time scales (Servant-Vildary and Roux, 1990; Bao et al., 1999; Sáez et al., 1999; Sylvestre, 2002; Tapia et al., 2003; Sáez et al., 2007; Hernández et al., 2008, 2010). The present paper is focused on the Late Quaternary carbonate sedimentation in Lake Chungará (northern Chilean Altiplano). The lake is characterised by diatom-rich deposits, interbedding tephra layers, and a few carbonate-rich intervals. Earlier work on the Chungará sequence described the sedimentological evolution of the lake during the last 14 kyr and distinguished several lowstand lake stages, mostly based on the percentage of benthic diatoms (Sáez et al., 2007). A mid-Holocene dry episode has also been found using high resolution XRF-core scanner data (Moreno et al., 2007) and statistical analysis of mineralogical and chemical parameters (Giralt et al., 2008). Other approaches based on $\delta^{18}\text{O}_{\text{VSMOW}}$ from diatom frustules (Hernández et al., 2008, 2010) provide a paleohydrological reconstruction for the pure diatomaceous ooze intervals, and offer insights into processes generating the high frequency laminations of sediments. Furthermore, an isotopic study of bulk carbonate and organic matter in the short-core 93S from the NW bay of the lake (Fig. 1; Valero-Garcés et al., 2003) showed wide variations in composition and hydrology in the last centuries. Research has also been undertaken on geochemical signatures in endogenic carbonates in other lakes in the area (Laguna Seca, Miscanti and Negro Francisco) in an attempt to reconstruct climatic changes since the Last Glacial Maximum (Grosjean, 1994; Grosjean et al., 1997; Schwab et al., 1999; Valero-Garcés et al., 1999b; Grosjean et al., 2001; Valero-Garcés et al., 2003; Theissen et al., 2008).

The main aim of this paper is to evaluate the paleoenvironmental information recorded in the isotopic markers in endogenic carbonates ($\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$) and in the bulk organic matter ($\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{15}\text{N}_{\text{AIR}}$) in the sedimentary record of Lake Chungará during the last

12.3 kyr. Although carbonates are discontinuous and although their amount is extremely low in the lowest part of the lacustrine sequence, our results highlight the advantage of a comprehensive study in an effort to better understand the geochemical signatures.

2. Hydrological and geological settings

Lake Chungará is located in the Chilean Altiplano (18° 15' S, 69° 09' W, 4520 masl) at the base of the Holocene Parinacota volcano (Fig. 1). The lake was generated by a partial collapse of the earlier Parinacota stratocone that dammed the Paleo-Lauca River between 15 and 17 cal kyr BP (Hora et al., 2007). The climate of the region is semi-arid (345–394 mm yr⁻¹) with an annual average temperature of 4.2 °C. Evaporation has been estimated to be about 1200 mm yr⁻¹ (Mladinic et al., 1987).

The lake has an irregular shape with a surface area of 22.5 km² and a water volume of 426 × 10⁶ m³. It has no surface outlets and the main inlet is the Chungará River (300–460 L s⁻¹) (Fig. 1). Lake Chungará is 40 m deep with the present water level being the highest in its history (Risacher et al., 1999; Herrera et al., 2006) as evidenced by the absence of emerged paleoshores and terraces. The lake can be regarded as a superficially closed hydrological system owing to its water residence time of around 15 years (Herrera et al., 2006) and owing to the absence of surface outlets (Leng et al., 2005). The cold polymictic lake is today moderately alkaline (pH between 8.99 and 9.30), well mixed (7.6 μL L⁻¹ O₂ at 34 m deep), with moderate salinity (1.3 g L⁻¹ TDS) and waters of the Na⁺–Mg²⁺–HCO₃⁻–SO₄²⁻ type. Concentrations of main solutes in mg L⁻¹ are Na⁺ 140, K⁺ 32, Ca²⁺ 50, Mg²⁺ 99, HCO₃⁻ 450, Cl⁻ 68, and SO₄²⁻ 354, and the molar Mg/Ca ratio is around 3.26 (Risacher et al., 1999; Herrera et al., 2006; Sáez et al., 2007). In the last two decades, Lake Chungará has undergone marked inter-annual changes in the water level (around 2 to 3 m) and significant oscillations in phytoplankton concentration (Dorador et al., 2003). The phytoplankton community is made up of a few species, including diatoms and Chlorophyceae that are dominant during cold and warm seasons, respectively. Dense macrophytic vegetation patches and microbial colonies occur in the littoral zone, also contributing to primary productivity (Dorador et al., 2003).

The seismic profiles indicate that the sedimentary succession is about 10 m thick and spans the last 12.3 kyr (Giralt et al., 2008). The sediments are dominated by diatomaceous ooze with interbedded tephra layers in offshore zones, and peaty diatomaceous sediments in shallow marginal zones. A detailed study using seismic and core data enabled us to define five lithostratigraphic units (Fig. 2) (Sáez et al., 2007):

Unit 1. Shallow to deep offshore deposits

Subunit 1a. Finely laminated green diatomaceous ooze (12.3–9.8 cal kyr BP).

Subunit 1b. Laminated and massive brown diatomaceous ooze with carbonate-rich intervals (9.8–8.3 cal kyr BP).

Unit 2. Deep offshore deposits (with a shallow event).

Subunit 2a. Brown massive to banded diatomaceous ooze interbedding carbonate-rich intervals and tephra layers (ca 8.3–3.5 cal kyr BP).

Subunit 2b. Dark grey-black, massive to banded diatomaceous ooze (ca 3.5 cal kyr BP – recent).

Unit 3. Shallow platform deposits (8.3 cal kyr BP – present). This unit overlies Unit 1 deposits in the central lake areas, and changes laterally to offshore facies of Unit 2 towards the west, and to littoral facies of Unit 4 towards the east.

Unit 4. Shallow to very shallow littoral peaty, rich in charophyte remains (3.2 cal kyr BP – present).

Unit 5. Alluvial-deltaic deposits.

Three main components in the lake sediments have been identified using a combination of mineralogical and XRF-core scanner

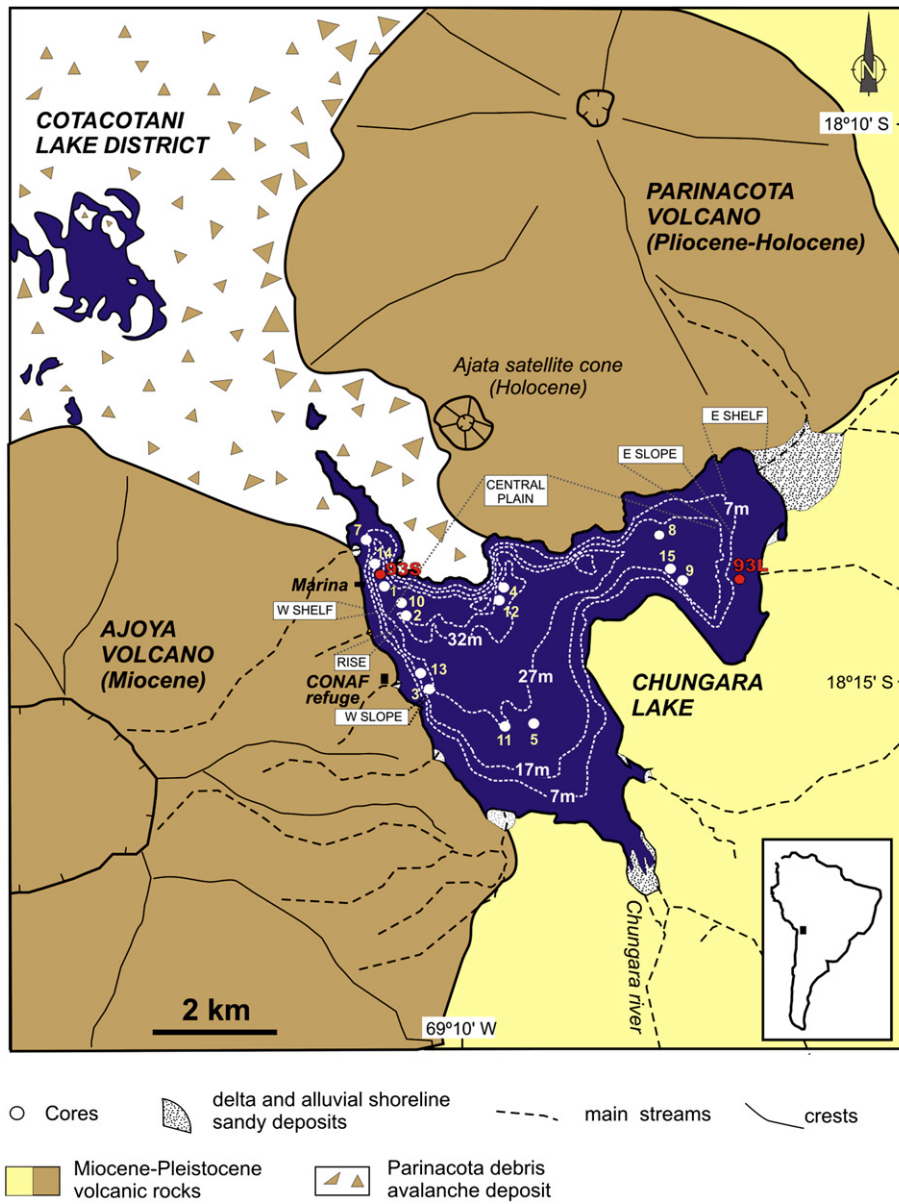


Fig. 1. Geological sketch of the Lake Chungará. Neighbouring volcanoes Ajoia and Parinacota, the Chungará River and the Cotacotani Lake district are included. Sites of all cores and the lake-bottom topography and morphology are also indicated.

data (Moreno et al., 2007; Giralt et al., 2008): a) a biogenic component mainly derived from phytoplankton, b) volcanic material (ash layers) from the nearby Parinacota Volcano, and c) endogenic carbonates. Carbonate-rich intervals and layers occur in subunits 1b, 2a, 3 and 4.

3. Materials and methods

Lacustrine sediments were sampled from 15 Kulleberg cores, each up to 8 m long, recovered from Lake Chungará in November 2002 and stored at 4 °C (Sáez et al., 2007). Short gravity cores were also collected near the margins of the lake (cores 93L and 93S) and drag samples were taken near the sites of the large Kulleberg cores in order to obtain information about the first centimetres of sediment and the sediment–water interface.

Selected core sequences were sampled every 5 cm, each sample being 1 cm thick, for carbonate studies, smear slides, and elemental and stable isotope geochemistry. Samples for ostracode valves were taken every 20 cm. Carbonates were characterised by petrographic, mineralogical and geochemical analyses. A preliminary estimation of

the morphology and amount of carbonates was made from the smear slides and was complemented by XRD and SEM-EDS analyses.

Samples for XRD were dried at 60 °C for 24 h and manually ground using an agate mill. The X-ray diffractions revealed that most samples were composed of a crystalline and an amorphous fraction (mostly opal), characterised by the presence of a broad peak centred between 20 and 25° 2θ. The identification and quantification of the main mineralogical species present in the crystalline fraction were carried out following a standard procedure (Chung, 1974). The amorphous fraction was quantified against artificial standards obtained by mixing pure diatomaceous ooze with increasing amounts of calcite.

A morphological description using a Jeol JSM-840 SEM-EDS and a Hitachi H-4100FE field emission SEM was made on selected samples. Owing to the high water content, samples were dried in two steps: first, most of the water was eliminated by capillarity using filter paper, and second, samples were freeze-dried and vacuum-stored prior to carbon coating. Secondary and backscattered electron images and X-ray emission spectra were used systematically to characterise the components.

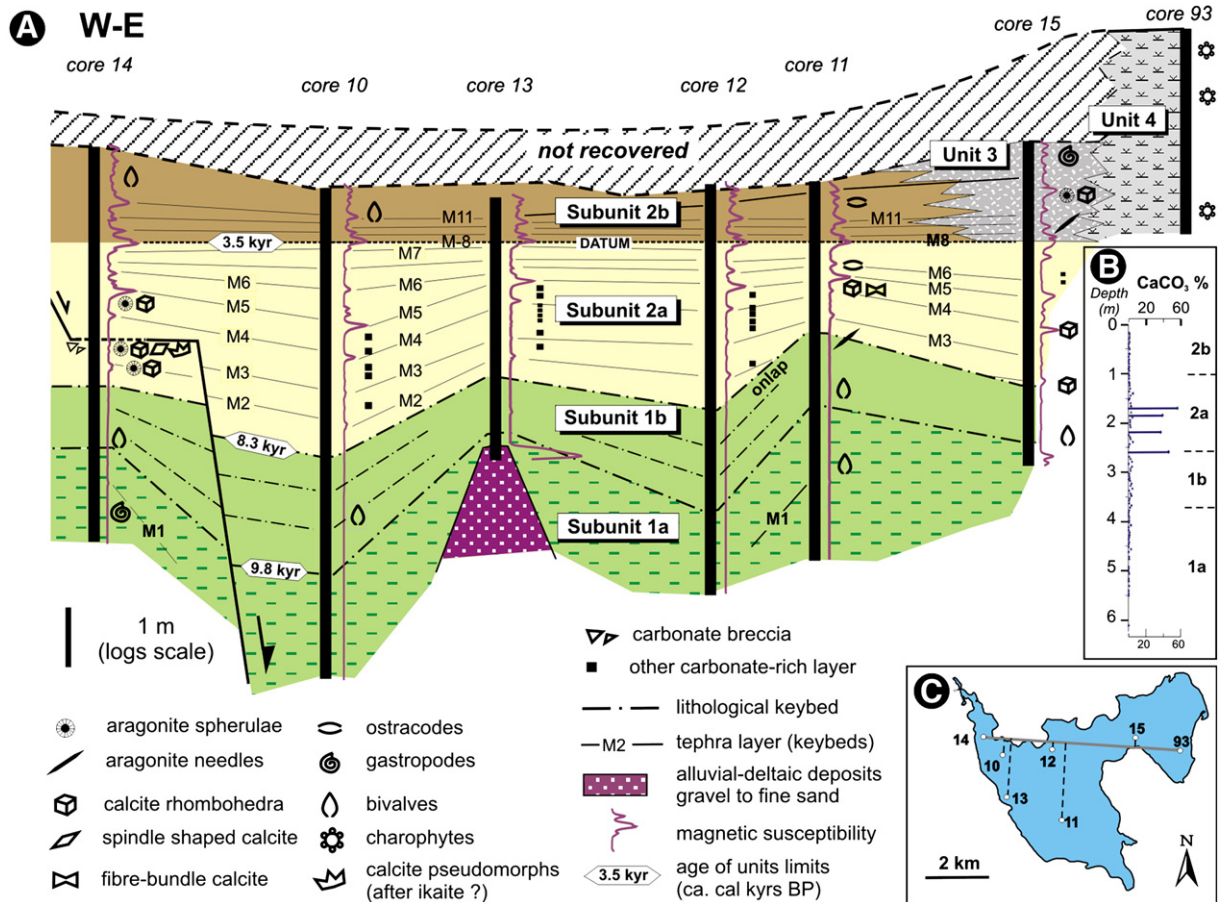


Fig. 2. A) Distribution of the carbonate fraction in the Lake Chungará sediments: the general correlation, lithological units and carbonate components are indicated (modified from Sáez et al., 2007). B) Vertical profile of carbonate content (wt.%) vs. depth in core 11 (placed in offshore setting). The maximum values (around 60 wt.%) correspond to isolated carbonate layer. C) Core sites in Lake Chungará and the W–E transect shown in A.

Isotopic compositions ($\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$) of the carbonates were determined following two procedures: a) Carbonate-rich levels were analysed using bulk samples (about 10 mg) cleaned from organic matter with a concentrated sodium hypochlorite solution. A classic offline method using phosphoric acid with a silver phosphate trap was used to eliminate sulphurous gas species that were present in small amounts. The carbonate content was quantified by XRD to determine the amount of sample to be processed. b) Bulk microsamples of sediment (with low content of endogenic carbonate) and microsamples (about 60 μg) of single carbonate components (*Chara* coatings, ostracode valves, isolated carbonate crystals) were analysed with a Kiel II online preparation device online connected to a MAT 252 IRMS. Analytical precisions (as σ) for $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$ determinations were 0.03 and 0.06‰, respectively.

Isotopic composition DIC- $\delta^{13}\text{C}_{\text{VPDB}}$ was determined from filtered (0.45 μm) water samples. Inorganic species of carbon (DIC, dissolved inorganic carbon) were precipitated adding BaCl_2 excess in a basic media. Purified CO_2 was obtained from the precipitated Ba-carbonate in a conventional line using phosphoric acid. The amount of co-precipitated Ba-sulphate was previously estimated by XRD in order to define the sample weight for analysis. Analytical precision expressed as σ was 0.03‰.

TOC (total organic carbon) and TIC (total inorganic carbon) in sediments and biomass samples were determined directly and after treatment at 400 °C in a Leco SC-144DR. TN (total nitrogen) was determined in a Variomax C/N following the Dumas' method (Ma and Gutterson, 1970).

Isotopic compositions ($\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{15}\text{N}_{\text{AIR}}$) of organic matter were measured in sediments and biomass. Sediments were treated

with diluted (1:4) HCl to eliminate minor amounts of carbonate. Plankton was separated using silica filters (0.45 μm). An elemental analyser, connected on line to a continuous flux Finnigan MAT Delta Plus IRMS, was used to determine $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{15}\text{N}_{\text{AIR}}$. Analytical precision (as σ) for $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{15}\text{N}_{\text{AIR}}$ determinations was 0.2‰, in both cases.

Isotopic compositions ($\delta^{18}\text{O}_{\text{VSMOW}}$ and $\delta\text{D}_{\text{VSMOW}}$) of waters were also analysed by CO_2 and H_2 equilibration (using Pt as catalyser) in an automatic unit connected online to a double inlet Finnigan MAT Delta S IRMS. Analytical precision expressed as σ was 0.2 and 1.5‰, respectively.

Isotopic standards used as reference were NBS-19 for carbonate $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$ and for water DIC- $\delta^{13}\text{C}_{\text{VPDB}}$; IAEA-N1, IAEA-N2, IAEA-NO3 and USGS-40 for $\delta^{15}\text{N}_{\text{AIR}}$ and IAEA-CH6, IAEA-CH7 and USGS-40 for $\delta^{13}\text{C}_{\text{VPDB}}$ in organic matter; and VSMOW, SLAP and GISP for $\delta\text{D}_{\text{VSMOW}}$ and $\delta^{18}\text{O}_{\text{VSMOW}}$ in water samples.

The chronological model of Lake Chungara was based on 17 radiocarbon AMS dates and on one $^{238}\text{U}/^{230}\text{Th}$ date (Giralt et al., 2008). The present day reservoir effect of the lake water was determined by dating the dissolved inorganic carbon (DIC) present in the surface lake water, and an apparent age of 2320 ± 20 radiocarbon years BP was obtained. This apparent reservoir effect age was corrected to offset the contamination of the thermonuclear tests carried out in the 1950s and 1960s (see Giralt et al., 2008 for further details). The calibration of the radiocarbon dates was performed using the CALIB 5.02 software and the INTCAL98 curve (Stuiver et al., 1998; Reimer et al., 2004). The software described in Heegaard et al. (2005) was employed to construct a reliable age–depth model, furnishing a corrected age for each calibrated date.

4. Results

4.1. Carbonate types

A detailed petrographic study of the carbonates in Lake Chungará allowed the identification of several occurrences (Fig. 3):

4.1.1. Euhedral to subhedral calcite crystals

Crystals may show flat faces but commonly display a fibrous inner structure with fibres (approximately 1 μm thick) arranged in the [001] direction (Fig. 3A). Only in a very few cases was high magnesium calcite observed. Calcite crystals display euhedral, spindle-shaped, and fibre-bundle (Fig. 3B) morphologies, the last two shapes being predominant. Crystal sizes vary from 6×10 to $80 \times 200 \mu\text{m}$ with the most common sizes ranging between 10×30 and $50 \times 80 \mu\text{m}$. Subunit 2a has the highest numbers of calcite crystals, but calcite already occurs at the top of subunit 1b. Subunits 1a (the oldest lacustrine sediments), 2b, 3 and 4

have very low endogenic carbonate contents. Calcite crystals are scattered among other components (diatoms and siliciclastics) or are concentrated forming white mm-thick levels of almost pure calcite. Similar morphologies in calcite (and high-Mg calcite) have been extensively described in the literature (Fernández-Díaz et al., 1996; González-Muñoz et al., 2000; Braissant et al., 2003; Jiménez-López et al., 2003). The origin of these crystals is attributed to biologically induced precipitation in the epilimnion (Robbins and Blackwelder, 1992; Thompson et al., 1997).

4.1.2. Needle-shaped aragonite crystals

Some calcium carbonate-rich white layers (mm-thick; see cores 11 and 15; Fig. 2) are scattered in the sedimentary sequence and are made up of aragonite needle cumulates. Aragonite needles are crystals elongated after [001], commonly around $2 \times 10 \mu\text{m}$ in size (Fig. 3C). Aragonite needles have been described in the literature and are related to whittings. They are also produced by calcareous algae and

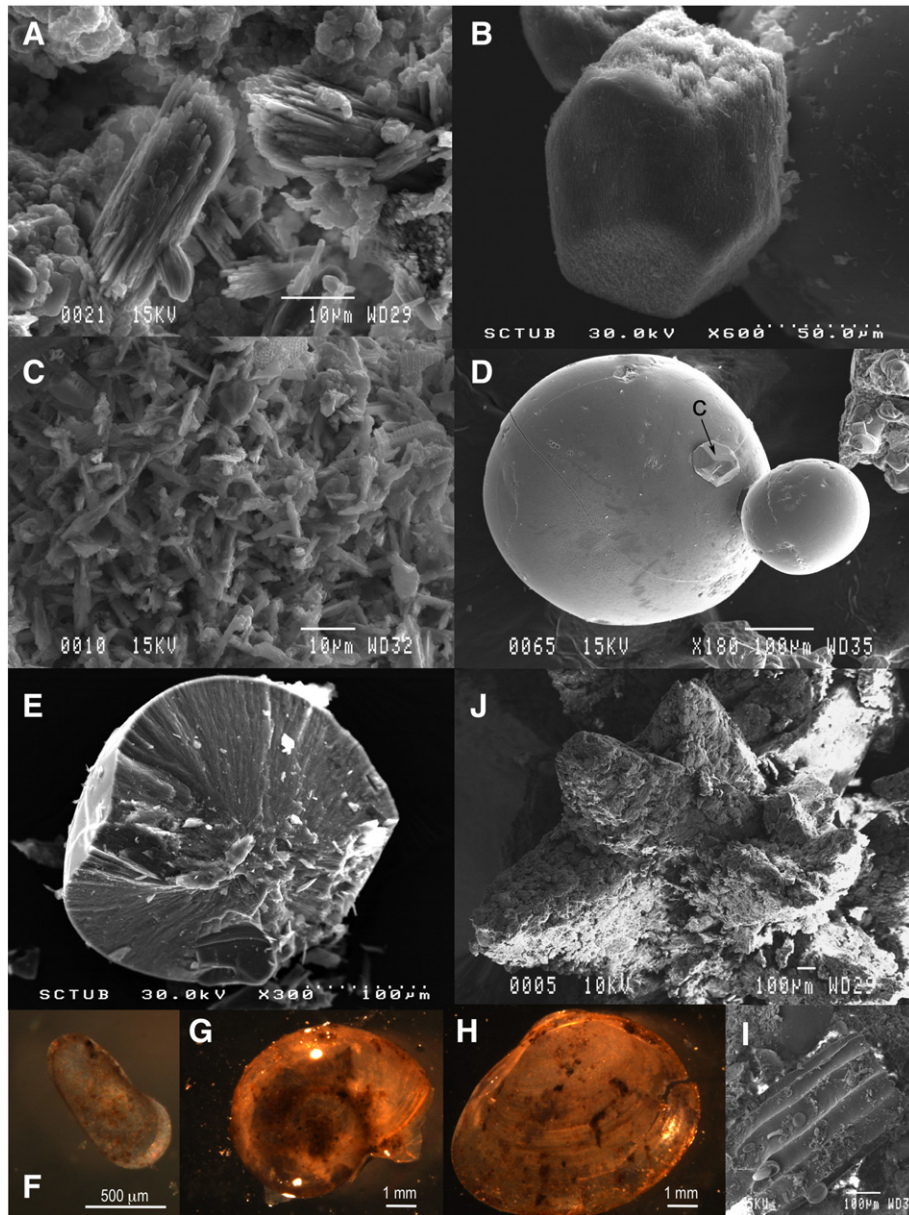


Fig. 3. Carbonate components in the Lake Chungará sediments: A) Fibre-bundle calcite crystals. B) Subhedral fibre-bundle calcite crystal. C) Needle-shaped aragonite crystals. D) Aragonite spheroids. Aragonite spheroids coexist with calcite crystals (C). E) Inner fibrous-radiated structure of an aragonite spheroid. Ostracoda valves (*Limnocythere* sp.; F), gastropoda (*Biomphalaria*; G) and bivalvia (*Pisidium*; H) shells and charophyta (*Chara*; I) stems and gyrogonites are the main bioclasts in Lake Chungará sediments. J) Calcite pseudomorphs (after ikaite?) in core 14.

microorganisms in marine and lacustrine environments (Lowenstam, 1955; Lowenstam and Epstein, 1956; Macintyre and Reid, 1992; Pedone and Folk, 1996), or precipitated experimentally (Callame and Dupuis, 1972; Zhou and Zheng, 2003).

4.1.3. Aragonite spheroids

Aragonite is also present as spheroids (Fig. 3D) commonly mixed with calcite crystals. Spheroids display variable diameters between 70 and 400 μm and an inner fibrous-radiated structure (Fig. 3E) with radial fibres of aragonite about 0.5 μm thick. Aragonite spheroids are very compact and are yellowish in colour with a pearl-like aspect. Spheroids of aragonite and other CaCO_3 minerals (calcite, vaterite, and nesquehonite) have been precipitated under experimental conditions (Bischoff and Fyfe, 1968; Suess and Fütterer, 1972; Fernández-Díaz et al., 1996; Zhou and Zheng, 2003; Wu et al., 2008; Xiang et al., 2008) and have been described in marine and lacustrine environments (Monaghan and Lytle, 1956; Oppenheimer, 1961; Giralte et al., 2001).

4.1.4. Bioclasts

Calcareous skeletal components consist of ostracode valves (*Limnocythere sp.*; Fig. 3F), gastropod (*Biomphalaria*; Fig. 3G) and bivalve (*Pisidium*; Fig. 3H) shells and corticated charophyte stems (*Chara sp.*) (Fig. 3I). Ostracode, gastropod and bivalve remains are distributed in varying amounts throughout the entire sequence. Charophyte stems and gyrogonites are restricted to littoral sites. Bivalve shells and ostracode valves are composed of low-Mg calcite, and gastropod shells of aragonite. Charophyte stems commonly consist of low- and sometimes high-Mg calcite.

4.1.5. Calcite cement

Calcite cements that trap other components and replace previous minerals are 1 to 2 mm in size (Fig. 3J), and were found in core 15. Carbonate cement shows grain sizes from 20 to 50 μm . Some of these pseudomorphs resembled those described by Grosjean (1994) in sediments from Laguna Lejía and were interpreted as calcite pseudomorphs versus ikaite.

4.2. Carbonate distribution

Carbonate displays a very heterogeneous distribution through the sequence and along an offshore-littoral transect (Fig. 2A and B). Carbonate bioclasts are scattered throughout the sequence, being more abundant in the dark green levels (rich in green algae, C facies in Sáez et al., 2007) and almost absent in the finely laminated diatomaceous ooze (A facies) of subunit 1a. Calcite crystals and aragonite needles and spheroids are commonly concentrated in white to grey thin layers. These layers are present in the entire series but frequently form discrete carbonate-rich intervals in subunit 1b (B facies) and mainly in subunit 2a.

The shallow to deep offshore deposits of Unit 1 are poor in carbonate. The lowest part (subunit 1a), which consists of a finely laminated green diatomaceous ooze, is practically devoid of carbonate crystals and ostracode remains (0.15 wt.% CaCO_3). The only carbonates are shells of gastropods and bivalves. The upper part of the unit (subunit 1b), which is made up of a laminated and massive brown diatomaceous ooze, contains some carbonate-rich cm-thick intervals (2.25 wt.% CaCO_3) composed of mm-thick carbonate layers of calcite (mostly) or aragonite. Euhedral, 50 μm long, calcite crystals and acicular aragonite crystals, 10 μm long and 2 μm wide, are scattered in the diatomaceous ooze. The upper part of subunit 1b also contains whole and fragmented shells of bivalves and gastropods, but ostracode valves are absent. The interval from 2.56 to 2.66 m, at the top of Unit 1 in core 11, includes two carbonate layers.

The deep offshore deposits of Unit 2 contain higher amounts of carbonate than unit 1. The lower part (subunit 2a) is formed by brown massive diatomaceous ooze and tephra and includes some cm-thick

carbonate-rich levels, and ostracode, bivalve and gastropod remains. Carbonate-rich levels account for up to 5% of the total thickness of subunit 2a (reaching around 6 wt.% CaCO_3) and are mostly concentrated in the lower half of the unit. Many levels are bunches of discrete whitish to pinkish mm-thick layers that are composed of calcite, minor amounts of high magnesium calcite and aragonite, and some traces of dolomite. The proportion of the carbonate mineral varies in the same carbonate-rich level throughout the basin. A change from calcite to aragonite was detected (with an accuracy around 1 cm) along several carbonate levels between cores located about 4 km from each other at water depths between 10 and 25 m (Fig. 4). Calcite-rich levels are composed of fibre-bundle crystals (Fig. 3A), spindle-shaped aggregates, rice-shaped crystals and dumbbells (10 to 200 μm long and 6 to 80 μm wide), euhedral crystals (50 to 100 μm in size) and irregular aragonite spheroids (70 to 140 μm diameter). Aragonite-rich levels show needle-shaped crystals 10 μm long and 1 to 3 μm wide. A carbonate-breccia recorded as a unique layer in core 14 (279 to 321 cm depth) with an erosive contact with the underlying sediments is composed of angular shaped, cm-thick carbonate clasts in a diatomaceous brown matrix. The upper part of the unit (subunit 2b, 3.55 cal kyr BP to recent) is formed by a massive dark grey diatomaceous ooze practically devoid of endogenic carbonate crystals (0.24 wt.% CaCO_3 for total endogenic carbonate). Ostracode valves are relatively abundant. Bivalve and gastropod shells are also present in some levels. The chlorophycean *Botryococcus braunii* records the highest concentration in the whole sequence, which is consistent with an offshore deposition around 30–40 m water depth for this unit (Sáez et al., 2007).

Shallow platform deposits of Unit 3 are composed of dark green diatomaceous ooze with abundant macrophyte remains and tephra levels. These sediments display gastropod and bivalve shells in discrete, mm- to cm-thick layers and carbonate-rich levels equivalent to the ones described in subunit 2a.

Shallow to very shallow littoral deposits of Unit 4 (less than 5 m depth) mainly consist of macrophytic peaty deposits rich in diatoms, some intercalated fine-grained tephra layers and carbonate-rich, charophyte-dominated deposits with gastropod and bivalve shells. Carbonates consist of calcite and Mg-calcite, and reach up to 25 wt.% in charophyte-rich levels. Carbonate components are mainly constituted by mineralized intercellular areas and calcite-covered charophyte stems. Ostracode and gastropod remains are also abundant.

As regards the paleogeographic distribution, carbonates predominate in littoral settings. Apart from core 93L, located on the eastern lacustrine shelf (Fig. 1), cores 14 and 15 (westwards and towards the eastern platform, respectively) contain more carbonate and its crystals are larger. Thus, the only carbonates recorded in core 11 are calcite (as fibre-bundle crystals) and aragonite needles. Aragonite spheroids are only present in littoral settings, being abundant in core 14 and, to a lesser extent, in core 15. Short-core 93S retrieved in the western sub-basin at 19 m water depth describes the most recent centuries of sedimentation in off shore settings in Lake Chungará. Sediments are composed of black, diatomaceous and organic-rich (up to 25% OM) laminated mud with a low carbonate content (6–8%). Two cm-thick levels show a high carbonate content (up to 50%) composed of euhedral calcite crystals and charophyte fragments (Valero-Garcés et al., 2003).

4.3. Isotopic markers of Lake Chungará

4.3.1. Recent water isotopic compositions ($\delta\text{D}_{\text{VSMOW}}$, $\delta^{18}\text{O}_{\text{VSMOW}}$ and $\text{DIC}-\delta^{13}\text{C}_{\text{VPDB}}$)

The water in Lake Chungará is significantly heavier than the water inputs (Chungará River, springs) and the rain and snow in the Lauca Basin (Herrera et al., 2006). All the samples of the lake surface and deep lake waters (November 2002 survey) display homogeneous values in $\delta\text{D}_{\text{VSMOW}}$ (–43‰ in average, ranging from –46 to –40‰)

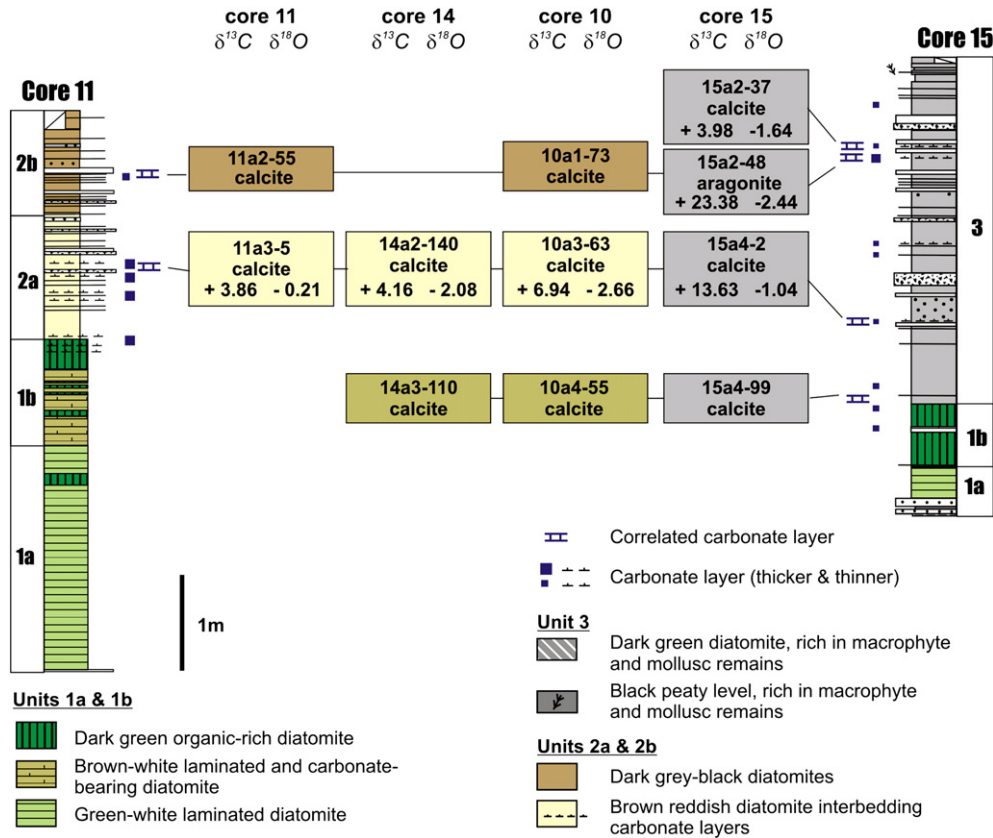


Fig. 4. Some carbonate layers in cores 10, 11, 14 and 15, placed at distances of several kilometres, are correlated. Lateral changes in carbonate mineralogy and in isotopic compositions ($\delta^{13}C_{VPDB}$ and $\delta^{18}O_{VPDB}$) are reported. Lithologic and stratigraphic units of 11 and 15 cores are included. Name, mineralogy and isotopic compositions for each sample are shown. Samples 15a2-48 and 15a4-2 (in core 15) showing very high $\delta^{13}C_{VPDB}$ values reflect methanogenic fermentation.

and $\delta^{18}O_{VSMOW}$ (-1.6% in average, -1.76 to -1.49% in range), showing δD_{VSMOW} and $\delta^{18}O_{VSMOW}$ enrichments of $+80$ and $+15\%$ with respect to average meteoric waters in the area. Values from an earlier survey (January 2001) show similar values for Chungará lake waters.

Isotopic composition of inorganic carbon species dissolved in lake water ($DIC-\delta^{13}C_{VPDB}$) is also significantly heavier than the water inputs (Table 1). Fifteen samples of lake water, 1 sample from the Chungará River and 5 samples from the springs around the lake were also analysed. The $DIC-\delta^{13}C_{VPDB}$ average values are $+4.7\%$ (range between $+3.21$ and $+5.97$) for Chungará lake water, -1.1% for Chungará River and -7.14% (-10.14 to -2.69%) for the springs.

4.3.2. Recent and old organic matter C/N atomic ratio and isotopic compositions ($\delta^{13}C_{VPDB}$ and $\delta^{15}N_{AIR}$)

Land plant samples in the catchment and aquatic plants, plankton and sediments in the lake were analysed for C/N, $\delta^{13}C_{VPDB}$ and $\delta^{15}N_{AIR}$ (Table 2; Fig. 5) to obtain information about biomass provenances in the lake. The C/N ratio of recent biomass in Lake Chungará yields values between 40 and 80 for land plants, between 4 and 25 for aquatic plants, and around 8.7 for plankton. Carbon isotopic composition ranges from values around -25% for land plants, -6% for aquatic plants (between -10 and -4%), and around -13% for plankton. Nitrogen isotopic composition ranges between -1 and $+5\%$ for land plants, around 0% for cyanobacterial mats, and between 0 and $+12\%$ for aquatic plants.

Recent sediments (2 cm below the water-sediment interface) from short cores, and core 15 show C/N values between 12 and 14, $\delta^{13}C_{VPDB}$ values around -15% and $\delta^{15}N_{AIR}$ between $+5$ and $+7\%$. Recent offshore sediments from short-core 93S (Valero-Garcés et al., 2003) show $\delta^{13}C_{VPDB}$ values between -14 and -20% . Older sediments from core 15 show C/N ratios between 12 and 17, $\delta^{13}C_{VPDB}$ around -16% ,

and $\delta^{15}N_{AIR}$ from $+4$ to $+7\%$ close to the short-core values. Core 11 sediments display lower values of $\delta^{13}C_{VPDB}$ (between -16 and -22%), $\delta^{15}N_{AIR}$ between $+10$ and $+1\%$, and C/N values between 8 and 17.

4.3.3. Isotopic compositions ($\delta^{13}C_{VPDB}$ and $\delta^{18}O_{VPDB}$) of carbonates

Carbonates were selected from four cores (11, 14, 15 and 93L; Fig. 1). Endogenic carbonates from cores 11, 14 (in offshore areas of the lake) and 15 (on the eastern platform) consist of calcite crystals and aragonite spheroids and, in smaller amounts, needle-like aragonite. All these materials were analysed either as bulk samples (mixed with diatomaceous ooze) when separation of the carbonate fraction was not possible, or as individual components (calcite crystals, spheroids, and aragonite needles). Moreover, bivalve shells and ostracode valves were analysed in offshore core 11. Carbonate samples from core 93L, in the eastern shallow platform, which is dominated by *Chara* gyrogonites and stem incrustations, were also analysed.

Bulk samples containing endogenic carbonate crystals from cores 10, 11, 14, 15 and 93L, show varying isotopic compositions (Fig. 6A). Core 11 located in the central area of the lake (28 m deep) displays values of $\delta^{13}C_{VPDB}$ between $+3$ and $+5\%$, and $\delta^{18}O_{VPDB}$ ranging between -0.5 and 0% . Core 14, also in a deeper area (27 m, but close to the NW shore) shows higher $\delta^{13}C_{VPDB}$ values ($+3.5$ to $+6\%$) and lower $\delta^{18}O_{VPDB}$ values (-2 to 0%). Samples from core 93L show even more clearly the shift to higher and lower values in both $\delta^{13}C_{VPDB}$ ($+7$ to $+11\%$) and $\delta^{18}O_{VPDB}$ (and -4 to 0%) at around 5 m depth. The available bulk samples from core 15 (Fig. 6A), which are located on the eastern platform at an intermediate depth of 24 m, show the highest values of $\delta^{13}C_{VPDB}$ (i.e.: $+14$, $+23\%$) found in the lake. The bulk samples from recent off shore sediments show greater variation

Table 1

Isotopic composition (δD_{VSMOW} , $\delta^{18}O_{VSMOW}$ and $DIC-\delta^{13}C_{VPDB}$) of surface waters and springs at Chungará–Cotacotani (November 2002 survey; Herrera et al., 2006).

Sample	Site	δD	$\delta^{18}O$	$\delta^{13}C$ DIC
Choquelin	Rain 6-10-02	-132.3	-19.40	
RIO-1	Chungará river	-113.1	-16.09	-1.15
MAN-1	Spring Ajata	-105.9	-15.28	-6.86
MAN-2	Spring CONAF	-121.9	-17.04	-8.97
MAN-3	Spring	-116.6	-16.49	-7.04
MAN-4	Spring basalt Parinacota	-112.1	-16.23	-2.69
MAN-5	Spring N of Ajata	-121.7	-17.12	-10.14
CHU-1	Chungara lake, isolated very shallow shore	-91.9	-11.81	-0.43
CHU-2	Chungara lake, harbour	-44.4	-2.05	+4.18
CHUN-1A	Chungara lake	-42.0	-1.52	+4.96
CHUN-1B	Chungara lake	-42.6	-1.62	+4.26
CHUN-2A	Chungara lake	-39.9	-1.56	+3.21
CHUN-2B	Chungara lake	-43.4	-1.54	+5.17
CHUN-2C	Chungara lake	-43.4	-1.578	+4.56
CHUN-010	Chungara lake	-44.0	-1.54	
CHUN-011	Chungara lake	-44.7	-1.58	+4.62
CHUN-013	Chungara lake			+3.86
CHUN-014A	Chungara lake	-43.2	-1.53	+5.52
CHUN-014B	Chungara lake	-44.4	-1.58	+5.34
CHUN-015A	Chungara lake	-44.2	-1.49	+5.43
CHUN-015B	Chungara lake	-45.8	-1.61	+3.55
CHUN-016A	Chungara lake	-44.3	-1.49	+5.97
CHUN-016B	Chungara lake	-41.7	-1.76	+3.68
CHUN-017	Chungara lake	-43.0	-1.74	+5.52
COT-1	Cotacotani, Laguna verde	-12.3	+3.02	-7.43
COT-2	Cotacotani, Laguna verde	-100.2	-12.82	-5.45
COT-3	Cotacotani lake	-65.8	-5.20	+4.67
COT-4	Cotacotani lake	-79.4	-7.93	-3.94
COT-5	Cotacotani lake	-68.4	-7.23	+2.13
COT-6	Cotacotani, isolated pool rich in m.o.	-52.0	-2.19	-6.65
COT-8	Cotacotani, isolated pool near road	-9.6	+9.46	-2.88

(+1 to -13‰ for $\delta^{18}O_{VPDB}$ and +8 to -25‰ for $\delta^{13}C_{VPDB}$; Valero-Garcés et al., 2003).

Apart from the bulk samples, microsamples of individual components (ostracode, gastropod and bivalve remains, calcite crystals, aragonite needles and spheroids; Table 3, Fig. 6A) were analysed in the same cores. Excluding charophyte stems from core 93L, $\delta^{13}C_{VPDB}$ and $\delta^{18}O_{VPDB}$ range from -2 to +6 and from -1 to +4‰, respectively. As a rule, bivalve shells show widespread values and are always rather depleted in ^{13}C (about -5‰) and enriched in ^{18}O (about +3‰) with respect to crystals (calcite or aragonite), spheroids (aragonite) or ostracode remains. Ostracode valves analysed from core 11 display a pattern similar to that of the bulk samples ($\delta^{13}C_{VPDB}$ between +1 and +5‰, and $\delta^{18}O_{VPDB}$ between -0.5 and +2‰). Aragonite and calcite crystals, and aragonite spheroids from cores 11 and 14 also show values similar to those of the bulk samples (Fig. 6A). Likewise, isolated *Chara* stems show the same isotopic dispersion as bulk samples (strongly dominated by *Chara* remains) from core 93L.

5. Discussion

5.1. Origin of organic matter

Organic matter in recent sediments in Lake Chungará (upper 2 cm of short-core samples) displays $TOC-\delta^{13}C_{VPDB}$ values close to plankton (-15 and -13‰, respectively; Table 2, Fig. 5). These $\delta^{13}C_{VPDB}$ values are higher than those commonly attributed to plankton (Fig. 5B; Meyers, 1994, 2003) and are related to high values of $TOC-\delta^{13}C_{VPDB}$ and $\delta^{13}C_{VPDB}$ of carbonates. They are also consistent with the +4‰ $DIC-\delta^{13}C_{VPDB}$ values measured in the recent lake water (about 20‰ heavier) (Table 1). The high $\delta^{13}C_{VPDB}$ values along the whole carbon cycle in lakes (DIC, biomass, TOC in sediments and endogenic

Table 2

TOC, TN, and isotopic compositions ($\delta^{13}C_{VPDB}$ and $\delta^{15}N_{AIR}$) of recent biomass and the organic fraction in bulk sediments from cores 11, 15 and short-core samples) from Lake Chungará. C and N concentrations in plankton biomass (using filters >0.045 μm) are also indicated. C/N is expressed as atomic ratio.

Sample	TOC (%)	$\delta^{13}C$	TN (%)	$\delta^{15}N$	C/N	
<i>Recent vegetal samples into the lake and the catchment</i>						
1. Lor4 aquatic plant	39.76	-7.45	3.70	+4.28	12.5	
2. Cyanobacterial mats	43.72	-10.04	11.43	-0.25	4.5	
3. Lor 8 Seca. Aquatic plant	32.90	-6.10	3.39	+9.26	11.3	
4. Chungara lake. Aquatic plant	31.36	-11.33	2.10	+8.94	17.4	
5. Chungara lake. Aquatic plant	42.94	-5.72	5.07	+11.74	9.9	
6. Chungara lake. Terrestrial pt.	33.35	-26.24	0.92	+1.37	42.2	
11. Chungara lake. Aquatic plant	40.34	-3.95	1.99	+10.93	23.6	
7. Terrestrial plant. 'Paja brava'	43.31	-24.85	0.65	+4.04	77.5	
8. Terrestrial plant. 'Llaretá'	25.36	-25.06	0.60	+1.07	49.0	
9. Terrestrial plant.	47.52	-23.39	1.09	+1.80	50.7	
10. Terrestrial plant.	53.14	-24.60	1.08	-0.65	57.5	
<i>Plankton samples</i>						
Filter 1	22.9	-13.3	3.11	+3.43	8.6	
Filter 2	21.3	-14.2	2.85	+5.32	8.7	
Filter 3	43.6	-11.6	5.46	+3.53	9.3	
<i>C and N concentration in water (from plankton)</i>						
Sample	mg C L ⁻¹	Mg N L ⁻¹				
Filter 1	2.04	276.6				
Filter 2	2.83	379.7				
Filter 3	10.26	1284.2				
<i>Short-core samples</i>						
Sample	TOC (%)	$\delta^{13}C$	TN (%)	$\delta^{15}N$	C/N	
1A 1M 0-1 cm	12.70	-14.22	1.20	+6.02	12.3	
8A 1M 0-1 cm	16.77	-15.06	1.47	+5.35	13.3	
9A 1M 0-1 cm	22.47	-14.48	1.90	+5.03	13.8	
10A 1M 0-1 cm	10.17	-16.1	0.95	+6.23	12.4	
11A 1M 0-1 cm	8.88	-15.62	0.78	+5.51	13.2	
12A 1M 0-1 cm	10.82	-15.74	0.96	+5.76	13.1	
13A 1M 0-1 cm	10.49	-15.9	0.97	+6.18	12.6	
14A 1M 0-1 cm	12.56	-14.1	1.19	+5.29	12.3	
<i>Core 15</i>						
Sample depth (m)	TOC (%)	$\delta^{13}C$	TN (%)	$\delta^{15}N$	C/N	
0.08	0.44	-15.0	0.04	+5.24	12.3	
0.25	1.95	-15.5	0.16	+4.64	14.2	
0.43	18.04	-16.4	1.42	+4.12	14.9	
0.63	6.73	-15.6	0.48	+4.96	16.2	
1.01	12.21	-16.1	0.94	+6.72	15.2	
1.39	0.64	-15.6	0.05	+6.45	13.8	
<i>Core 11</i>						
Sample	Depth (m)	TOC (%)	$\delta^{13}C$	TN (%)	$\delta^{15}N$	C/N
Section 2. 1	0.16	6.20	-16.75	0.49	+6.55	14.8
Section 2. 10	0.25	9.70	-16.30	0.87	+6.64	13.0
JJP1/A 1	0.30	7.55		0.63		13.9
Section 2. 20	0.35	8.20	-17.25	0.62	+7.08	15.4
JJP1/A 2	0.40	10.90		0.95		13.4
Section 2. 30	0.45	1.40	-17.62	0.13	+7.31	12.5
Section 2. 41	0.56	4.90	-17.76	0.47	+7.26	12.2
Section 2. 50	0.65	1.50	-20.37	0.16	+6.27	10.7
Section 2. 60	0.68	5.40	-18.09	0.52	+7.76	12.1
JJP1/A 3	0.70	8.52		0.72		13.7
Section 2. 70	0.76	5.20		0.46		13.1
Section 2. 80	0.85	8.00	-18.12	0.63	+7.83	14.8
Section 2.90	0.95	4.60	-18.38	0.38	+7.40	14.4
Section 2.100	1.05	3.60	-18.86	0.33	+8.32	12.6
Section 2.110	1.15	5.90	-17.57	0.43	+9.14	16.2
Section 2.121	1.25	6.20	-18.57	0.52	+8.15	13.9
JJP1/A 4	1.29	8.59		0.71		14.1
JJP1/A 5	1.36	5.37	-19.25	0.48	+9.12	13.1
Section 2.130	1.42	6.00		0.52		13.4
Section 2.139	1.54	1.60	-19.32	0.16	+8.35	11.4
JJP1/A 6	1.58	7.35		0.69		12.5
Section 3.1	1.67	7.90	-18.00	0.66	+6.07	14.1

Table 2 (continued)

Core 11						
Sample	Depth (m)	TOC (%)	$\delta^{13}\text{C}$	TN (%)	$\delta^{15}\text{N}$	C/N
Section 3.10	1.76	7.70	-21.40	0.53	+8,00	16,9
Section 3.20	1.86	6.70	-18.60	0.53	+7,32	14,7
Section 3.30	1.96	4.80	-19.67	0.34	+8,08	16,3
Section 3.40	2.06	7.00	-18.93	0.69	+9,15	11,8
JJP1/A 7	2.09	5.21		0.46		13,3
JJP1/A 8	2.10	4.86		0.42		13,4
JJP1/A 9	2.11	6.93		0.60		13,4
Section 3.50	2.16	6.30	-21.30	0.48	+10,35	15,2
JJP1/A10	2.20	7.80		0.72		12,6
Section 3.60	2.26	6.50	-18.49	0.60	+8,92	12,6
JJP1/A11	2.31	6.29		0.59		12,5
Section 3.70	2.36	6.90	-17.59	0.70	+9,75	11,5
JJP1/A12	2.41	5.82		0.54		12,6
JJP1/B 1	2.46	5.78	-18.40	0.51	+9,84	13,1
JJP1/B 2	2.51	5.34	-18.90	0.51	+8,46	12,3
Section 3.80	2.66	9.20	-19.10	0.73	+9,97	14,8
Section 3.90	2.76	5.00	-18.70	0.61	+9,92	9,5
Section 3.100	2.81	4.70		0.52		10,6
Section 3.110	2.86	6.10	-18.45	0.63	+9,18	11,3
Section 3.120	2.96	8.50	-18.30	0.67	+8,73	14,7
Section 3.130	3.06	6.90	-18.06	0.64	+10,44	12,6
Section 3.140	3.11	8.00		0.64		14,6
Section 3.149	3.15	7.30	-17.60	0.79	+6,63	10,8
Section 4.1	3.17	6.70	-18.80	0.90	+8,73	8,7
Section 4.10	3.26	8.60		0.69		14,5
Section 4.20	3.36	4.90		0.58		9,8
Section 4.30	3.41	7.80		0.65		14,0
Section 4.40	3.61	6.00		0.62		11,2
Section 4.50	3.66	6.80		0.62		12,7
Section 4.60	3.75	5.70		0.59		11,2
Section 4.70	3.86	6.00		0.57		12,4
Section 4.80	3.96	5.50		0.57		11,2
Section 4.90	4.06	5.10	-19.23	0.53	+8,52	11,3
Section 4.100	4.16	5.30		0.50		12,4
Section 4.110	4.26	5.00	-21.53	0.47	+8,02	12,5
Section 4.120	4.36	6.50		0.64		11,8
Section 4.130	4.46	8.30		0.73		13,2
Section 4.140	4.56	6.40	-21.08	0.59	+8,21	12,7
Section 5.1	4.67	5.60		0.51		12,7
Section 5.10	4.76	5.10	-22.13	0.43	+6,70	14,0
Section 5.20	4.86	4.40	-21.51	0.39	+5,90	13,1
Section 5.30	4.96	5.10		0.45		13,3
Section 5.40	5.06	4.10	-21.67	0.47	+5,33	10,3
Section 5.50	5.16	5.00	-20.77	0.48	+4,15	12,2
Section 5.60	5.26	4.70		0.42		13,1
Section 5.70	5.36	4.60	-20.67	0.46	+4,15	11,8
Section 5.80	5.46	4.50		0.44		11,9
Section 6.1	5.51	2.50		0.36		8,1
Section 6.10	5.66	3.90		0.42		10,7
Section 6.20	5.71	3.90	-20.40	0.42	+4,77	10,7
Section 6.30	5.81	4.40		0.45		11,4
Section 6.40	5.91	4.90	-20.53	0.53	+2,91	10,9
Section 6.50	5.98	4.10		0.46		10,4
Section 6.60	6.11	4.70	-21.60	0.45	+1,38	12,3
Section 6.70	6.16	5.20		0.47		12,9

carbonates) have been ascribed to internal or external causes. Internal causes are high algal productivity and selective ^{12}C uptake linked to persistent eutrophic conditions (Hollander and McKenzie, 1991; Schelske and Hodell, 1995; Gu and Schelske, 1996; Brenner et al., 1999; Meyers, 2003), elimination of light carbon by methanogenesis (Talbot, 1990; Talbot and Kelts, 1990) or high lake water pH and alkalinity (Bade et al., 2004) in lakes with high reservoir effect (Wu et al., 2010). External sources of heavy carbon have been attributed to older carbonate recycling or thermal (and magmatic) CO_2 inputs (Valero-Garcés et al., 1999a; Wu et al., 2010).

$\delta^{15}\text{N}_{\text{AIR}}$ of plankton and aquatic plants in Lake Chungará (+4 to +12‰) are in the range of reported values (Meyers, 2003), reflecting the assimilation of nitrate as NO_3^- , about +8‰ enriched with respect

to atmospheric N_2 . The C/N ratio of plankton and aquatic plants is similar (around 9), being close to that of the recent sediments in the lake bottom (around 11) and far from local land plants (50 on average) (Table 2). These data provide evidence that the main source of organic matter in recent sediments is phytoplankton with minor contributions of aquatic macrophytes (Fritz et al., 2006).

Older sediments show variable isotopic compositions depending on the depositional environment. Higher $\delta^{13}\text{C}$ values on the shallow platform (core 15) than in deeper offshore sediments (core 11) may be ascribed to a greater contribution of aquatic plants (macrophytes) and cyanobacterial colonies from littoral environments. The sediments of core 15 show similar $\delta^{13}\text{C}_{\text{VPDB}}$ values and slightly higher C/N values than the recent sediments and are about 3‰ lighter than plankton, reflecting a macrophytic contribution. Organic matter in offshore sediments (core 11) is mainly derived from phytoplankton. Its low $\delta^{13}\text{C}_{\text{VPDB}}$ values cannot be attributed to mixing with land plants (enriched in ^{12}C), as is demonstrated by their C/N ratios (Table 2). It seems more reasonable to suggest that these $\delta^{13}\text{C}_{\text{VPDB}}$ values (-17‰ in the more recent to -22‰ in the oldest sediments) together with C/N ratios between 8 and 17 (12.7 in average) reflect the algal contribution to the total organic matter in the lake sediments (Meyers, 1997; Meyers and Teranes, 2001).

$\text{TOC}-\delta^{13}\text{C}_{\text{VPDB}}$ (and to a lesser extent $\delta^{13}\text{C}$ in carbonates) of Lake Chungará displays increasing values throughout the sedimentary sequence (Fig. 7). A similar trend from lower to higher $\delta^{13}\text{C}_{\text{VPDB}}$ has been reported in other lake systems and ascribed to increases in aquatic productivity (Hollander and McKenzie, 1991; Meyers, 2003). Isotopic fractionation produced in the early stages of microbial degradation due to a partial oxidation of organic matter (Lehmann et al., 2002) should be ruled out in the absence of significant changes in the C/N ratio with depth (Table 2). Therefore, other heavy carbon inputs are necessary to account for all the $\delta^{13}\text{C}_{\text{VPDB}}$ values in Lake Chungará.

The highest $\text{DIC}-\delta^{13}\text{C}_{\text{VPDB}}$ values in natural lacustrine environments reported in the literature are around +26‰ and correspond to pore-water affected by methanogenesis in Lake Bosumtwi (Ghana; Talbot and Kelts, 1986) and Lake Apopka (Florida, USA; Gu et al., 2004) (Fig. 8). ^{13}C enrichment is smaller in the water column (Lake Apopka $\text{DIC}-\delta^{13}\text{C}_{\text{VPDB}}$ from +5 to +13‰) and has been attributed to advection of heavy carbon through the water-sediment interface and to the DIC speciation at high pH values (>8.3) where HCO_3^- is the dominant ionic form (Deuser and Degens, 1967; Hassan et al., 1997; Keeley and Sandquist, 2006). The high $\delta^{13}\text{C}_{\text{VPDB}}$ values in Lake Chungará (in DIC, endogenic carbonate and TOC; Tables 1, 2 and 3) are similar to those reported in Lake Apopka. $\text{DIC}-\delta^{13}\text{C}_{\text{VPDB}}$ pore-water values, calculated from endogenic carbonate sediments of Lake Chungará, are in the highest range (around +23‰). Moreover, the DIC system is currently dominated by HCO_3^- (Mühlhauser et al., 1995; Risacher et al., 1999; Herrera et al., 2006). Recent $\text{DIC}-\delta^{13}\text{C}_{\text{VPDB}}$ values in the water column (+3 to +6‰) and $\delta^{13}\text{C}_{\text{VPDB}}$ values in endogenic carbonate crystals (+3 to +6‰ in cores 11 and 14, and +12‰ in core 93L, this last core located in a littoral setting) are similar to those reported in Lake Apopka (Gu et al., 2004; Fig. 8).

$\delta^{15}\text{N}_{\text{AIR}}$ values show a contrasting behaviour with respect to $\delta^{13}\text{C}_{\text{VPDB}}$. The values are more positive in offshore (core 11) than in littoral sediments (core 15; Fig. 5A) of Lake Chungará. Lower marginal $\delta^{15}\text{N}_{\text{AIR}}$ values probably reflect cyanobacterial activity and land inputs. The gradual increase in $\delta^{15}\text{N}_{\text{AIR}}$ (from +1 to +10‰) and TN between 12.3 and 7.2 cal kyr BP (Fig. 7) may be linked to an increase in algal productivity, matching approximately the behaviour of carbon (Meyers and Teranes, 2001).

From 12.3 (the beginning of the sequence) to 10.2 cal kyr BP, in subunit 1a, the increase in TOC, TN, $\text{TOC}-\delta^{13}\text{C}_{\text{VPDB}}$ and $\text{TN}-\delta^{15}\text{N}_{\text{AIR}}$ suggests a rise in productivity with a maximum around 11 cal yr BP (Fig. 7). This increase in productivity could be related to external inputs associated with runoff during a lake level rise (Sáez et al., 2007).

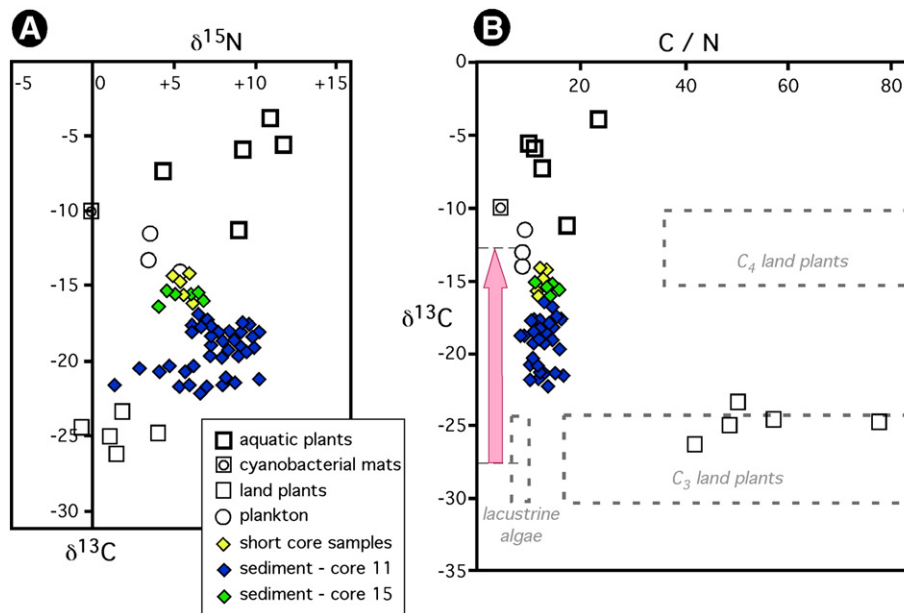


Fig. 5. A) Isotopic compositions ($\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{15}\text{N}_{\text{AIR}}$) of the organic fraction of bulk samples in the Lake Chungará sediments (cores 11, 15 and short cores), in recent plankton, as well as in aquatic and land vegetation (in the catchment) are plotted. B) Organic carbon isotopic composition versus C/N for recent biomass and sediments are plotted. Common values for lacustrine phytoplankton and C_3 and C_4 land plants are indicated (after Meyers, 1994). Recent plankton from Lake Chungará shows ^{13}C enrichment around +15‰.

Furthermore, this change could have increased the surface of the photic zone of the lake, reflecting higher contributions of cyanobacterial colonies and other algae (*Botryococcus*) as biological producers. This matches the values reported for living macrophytes of Lake Chungará (Table 2, Fig. 5) and other altiplanic lakes (Lake Titicaca, in the Peru-Bolivia border; Fritz et al., 2006). The low TN- $\delta^{15}\text{N}_{\text{AIR}}$ values, which are close to 0‰ in the early and shallower stages of the recorded sedimentary sequence, suggest a biomass dominated by cyanobacteria (Meyers, 2003).

5.2. Origin of carbonates

The carbon isotopic composition of carbonates from the Lake Chungará sediments increases from offshore towards the littoral sediments (Fig. 6A). Charophyte-rich sediments in the eastern margin (core 93L, about +9‰ in average) and calcite-rich laminae in recent sediments (+9 to +5‰; Valero-Garcés et al., 2003) record the heaviest values. Isotopic analyses ($\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$) were performed along the same carbonate keybed, identified in cores 11 and 14 (both located in offshore settings), displaying similar values around +4‰ for $\delta^{13}\text{C}_{\text{VPDB}}$ and from 0 to -2‰ for $\delta^{18}\text{O}_{\text{VPDB}}$ (Fig. 4). Slightly higher $\delta^{13}\text{C}_{\text{VPDB}}$ in core 14 than in core 11 can also be ascribed to the local influence of higher productivity in a more littoral setting. Evaporation in littoral areas and the local effect of photosynthesis associated with macrophytes would account for the increase in $\delta^{13}\text{C}_{\text{VPDB}}$. Nevertheless, values exceeding +7‰ suggest other sources of heavy carbon. In Lake Chungará, where there is evidence of anoxia and gas accumulation in the bottom sediments, synsedimentary methanogenesis has to be considered as a possible origin for heavy isotopic carbonate compositions. Evidence of trapped gas (methane?) has been interpreted in the seismic profiles of the eastern platform, where core 15 is located (Fig. 4 in Sáez et al., 2007). Two carbonate samples from core 15, on the eastern platform of the lake, show the highest values of $\delta^{13}\text{C}_{\text{VPDB}}$ (+13 and +23‰) (Table 3, Fig. 6A). Both samples are the only ones that show an incipient early cementation that replaces earlier minerals (probably CaCO_3 precursors such as ikaite) (Fig. 3). Calcite spherulites with similar methanogenic signatures have been described in Lake Bosumtwi (Talbot and Kelts, 1986). Methanogenic fermentation in the sediments may have caused methane releases

to the water column, thus increasing the pore-water DIC, rich in ^{13}C , resulting in Ca-carbonate local replacements. These higher $\delta^{13}\text{C}_{\text{VPDB}}$ values in early diagenetic cements are evidence of bacterial methanogenesis (Turner and Fritz, 1983). The general trend towards higher $\delta^{13}\text{C}_{\text{VPDB}}$ values in carbonates along the whole sequence (Fig. 7) would be a consequence of changes in biological (increase in organic productivity) and chemical (increase in pH and alkalinity) parameters causing higher DIC- $\delta^{13}\text{C}_{\text{VPDB}}$. The influence of geothermal and volcanic CO_2 and degassing during groundwater discharge has been proposed in the shallow lakes of the Andean Altiplano as a ^{13}C enrichment mechanism (Schwalb et al., 1999; Valero-Garcés et al., 1999a, 2000; Gibert et al., 2008). Nevertheless, these processes cannot be applied here. Lake Chungará is neither a saline nor a shallow lake, and it provides no evidence of thermal inputs. Moreover, it is a young lake without a long history of interactions with the surrounding volcanic environment.

The oxygen isotopic composition of carbonates depends on the isotopic composition and on the temperature of the host water. In high-altitude Andean lakes, the isotopic water composition is the main factor that is dependent on the input waters (rainfall, groundwaters and runoff) and on evaporative enrichment (Aravena et al., 1999). The Lake Chungará waters are heavier than inputs (springs, rivers, and rainfall) due to the strong evaporation in the Altiplano arid environment (Table 1; Fig. 7 in Herrera et al., 2006). The $\delta^{18}\text{O}_{\text{VPDB}}$ values of carbonates, ranging roughly from +4 to -4‰ (Table 3, Fig. 6A) show a clear relationship with the depositional environment: samples from littoral zones of the lake are isotopically lighter than those from offshore areas. Although the variability of the bulk sediment samples is very large, the calcite-rich levels from the short-core 93S show a range similar (0 to -4‰) to that of the separated charophyte stems (-1 to -5‰). Lower isotopic compositions of carbonates suggest inputs of lighter waters and the effect of warmer temperatures in coastal settings. The most likely source is freshwater input and seepage of groundwater in the NE basin.

5.3. Controls on carbonate precipitation

Changes in the carbonate content and mineralogy in the sediments of Lake Chungará record variations in the water chemistry and also in

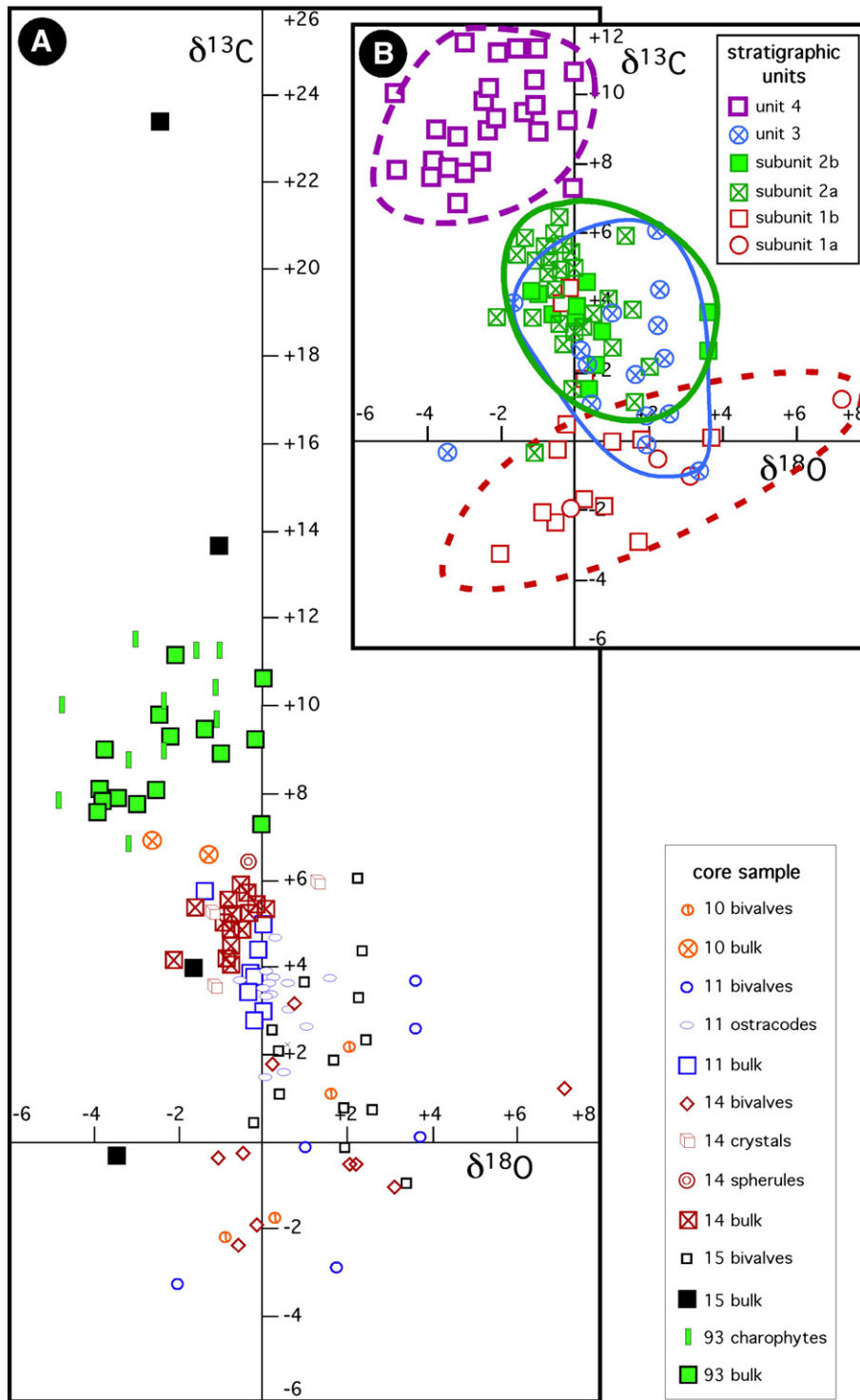


Fig. 6. Isotopic compositions ($\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$) in the carbonate fraction in sediments of Lake Chungará. A) Samples are separated by cores and components. Gastropods are not plotted. B) Samples are separated by stratigraphic units.

the type and productivity of biomass in space and time. Thus, a low calcium concentration in waters during the early stages of lake evolution would be responsible for the negligible amount of calcium carbonate precipitation (i.e. in Unit 1, 12.3 to 8.3 cal kyr BP, Figs. 2 and 9). A possible explanation for relatively Ca-poor waters during the early lake stage could be the short leaching time of volcanic rocks in the catchment after the collapse that had dammed the River Lauca, creating the lake. Thereafter, the availability of calcium in waters

could have increased with time, giving rise to conditions that favoured a higher (although never dominant) calcium carbonate precipitation, mainly in subunit 2a, 8.3 to 3.55 cal kyr BP. The rise in calcium in lake waters is also consistent with the marked increase in tephra intercalations starting from the beginning of Unit 2 (Sáez et al., 2007). In fact, the onset of a significant precipitation of carbonates can be related to extreme photosynthetic activity associated with massive diatom blooms during a short dry period from 10.0 to 9.6 cal yr BP

Table 3

Isotopic compositions ($\delta^{13}\text{C}_{\text{VPDB}}$, $\delta^{18}\text{O}_{\text{VPDB}}$) of the carbonate fraction in the Lake Chungará sediments. Results are grouped by cores (10, 11, 14, 15 and 93). The kind of analysed samples (bulk samples, microsamples, isolated components) is indicated in the 'compo' column. Gastropods are aragonitic, bivalves and ostracods are calcitic, carbonate crystals are calcite (only one aragonite level has been analysed) and spheroids are aragonitic.

Sample	Depth—cm	^{14}C age ^a	$\delta^{13}\text{C}_{\text{PDB}}$	$\delta^{18}\text{O}_{\text{PDB}}$	Compo ^b
<i>Core 10</i>					
10a1–24	24		1.52	0.39	Bivalve — LMC
10a1–44	44		2.21	0.57	Bivalve — LMC
10a4–13	370		1.09	1.64	Bivalve — LMC
10a4–96	453		2.19	2.06	Bivalve (?)
10a4–112	469		–1.96	0.77	Bivalve — LMC
10a4–114	471		–1.76	0.28	Bivalve — LMC
10a4–118	475		–2.14	–0.84	Bivalve — LMC
<i>Core 11</i>					
11a2–1	16		3.71	–0.54	Ostracode — LMC
11a2–3	18		3.70	3.65	Bivalve (?)
11a2–10	25		4.29	–1.14	Ostracode — LMC
11a2–20	35		4.23	–0.82	Ostracode — LMC
11a2–30	45		3.90	0.05	Ostracode — LMC
11a2–40	55	2075	2.59	3.64	Bivalve (?)
11a2–41	56		3.62	0.15	Ostracode — LMC
11a2–50	65		3.52	0.04	Ostracode — LMC
11a2–80	95	3470	4.64	0.35	Ostracode — LMC
11a2–90	105		3.61	0.60	Ostracode — LMC
11a2–100	115		4.07	0.93	Ostracode — LMC
11a2–110	125		3.75	0.26	Ostracode — LMC
11a2–121	136		3.35	0.17	Ostracode — LMC
11a2–130	145		3.74	0.11	Ostracode — LMC
11a2–139	154		3.72	1.62	Ostracode — LMC
11a3–5	171	6390	3.86	–0.21	Bulk ^c
11a3–10	176		3.42	0.05	Ostracode — LMC
11a3–10	176		4.98	0.00	Bulk
11a3–20	186		2.62	1.04	Ostracode — LMC
11a3–20	186		2.76	–0.21	Bulk
11a3–30	196		1.54	0.53	Ostracode — LMC
11a3–40	206		1.48	0.07	Ostracode — LMC
11a3–50	216		3.03	0.69	Ostracode — LMC
11a3–50	216		3.42	–0.34	Bulk
11a3–55	221		2.97	0.00	Bulk
11a3–90	256	8150	5.76	–1.35	Bulk
11a3–93	259		–0.06	1.03	Bivalve — LMC
11a3–95	261		4.41	–0.09	Bulk
11a3–100	266		3.88	–0.30	Bulk — A
11a3–108	274		–3.27	–1.99	Bivalve — LMC
11a3–112	278		–2.94	1.78	Bivalve — LMC
11a3–115	281	8800	0.08	3.77	Bivalve — LMC
11a3–132	298		3.01	–0.66	Bulk microsample ^d
11a3–133	299		3.01	–0.83	Bulk microsample
11a3–139,5	305.5		1.89	–1.31	Bulk microsample
11a3–140	306		–0.07	1.94	Bivalve — LMC
11a4–4	320		–0.07	–2.52	Bulk microsample
11a4–6	322		1.82	–2.14	Bulk microsample
11a4–10	326	9610	3.18	–1.24	Bulk microsample
11a4–11,5	327.5		2.32	–1.41	Bulk microsample
11a4–14	330		3.47	–0.98	Bulk microsample
11a4–17,5	333.5		0.35	–1.66	Bulk microsample
11a4–40	356		0.61	–2.86	Bulk microsample
11a4–42	358		5.07	–2.24	Bulk microsample
11a4–44	360		3.61	–2.43	Bulk microsample
11a4–46	362		–1.74	–2.52	Bulk microsample
11a4–51,5	367.5	10,370	6.79	–0.88	Bulk microsample
<i>Core 15</i>					
15a1–37	37		3.65	1.02	Gastropod — A
15a1–43	43		6.05	2.24	Bivalve — LMC
15a1–45	45		4.37	2.36	Gastropod — A
15a1–63	63		2.35	2.44	Bivalve — LMC
15a2–37	134		3.98	–1.64	Bulk
15a2–48	145		23.38	–2.44	Bulk
15a2–70	152		2.57	0.21	Bivalve — LMC
15a2–72	154		2.11	0.38	Bivalve — LMC
15a3–76	244		0.73	1.97	Gastropod — A
15a3–130	316		2.06	0.37	Bivalve — LMC
15a3–134	320		–0.14	1.96	Bivalve — LMC
15a3–139	325		0.75	2.60	Bivalve — LMC
15a4–3	321		13.63	–1.04	Bulk

Table 3 (continued)

Sample	Depth—cm	^{14}C age ^a	$\delta^{13}\text{C}_{\text{PDB}}$	$\delta^{18}\text{O}_{\text{PDB}}$	Compo ^b
15a4–68	386		1.89	1.70	Bivalve (fragment)
15a4–72	390		1.06	0.41	Bivalve — LMC
15a4–80	398		3.31	2.29	
15a4–95	413		–0.94	3.39	
15a4–99	417		–0.34	–3.45	Bulk
15a4–137	455		0.44	–0.20	Bivalve — LMC
<i>Core 14</i>					
14a1–15	15		3.17	0.75	Bivalve — LMC
14a2–140	415		3.54	–2.14	Bulk
14a3–6	281		5.90	1.35	Carbonate cryst. LMC
14a3–8	283		5.88	–0.50	Bulk
14a3–9	284		5.70	–0.41	Bulk
14a3–12	287		5.37	–1.54	Bulk
14a3–8	293		5.19	–1.00	Carbonate cryst. LMC
14a3–18	293		6.44	–0.41	Spherulæ — A
14a3–20	295		4.18	–0.76	Bulk
14a3–22	297		3.50	–1.17	Carbonate cryst. LMC
14a3–24	299		5.51	–0.80	Bulk
14a3–26	301		5.25	–0.72	Bulk
14a3–28	303		5.39	–0.12	Bulk
14a3–30	305		5.32	–0.03	Bulk
14a3–32	307		5.25	–0.30	Bulk
14a3–34	309		5.21	–0.75	Bulk
14a3–36	311		5.13	–0.88	Bulk
14a3–40	315		5.09	–0.72	Bulk
14a3–42	317		4.84	–0.45	Bulk
14a3–44	319		4.70	–0.64	Bulk
14a3–46	321		4.52	–0.74	Bulk
14a3–82	357		4.07	–0.71	Bulk
14a3–105	380		–0.36	–1.05	Bivalve — LMC
14a3–127	402		–0.54	2.19	Bivalve — LMC
14a3–129	404		–0.27	–0.47	Bivalve — LMC
14a3–131	406		–2.38	–0.51	Bivalve — LMC
14a3–145	420		1.78	0.26	Gastropod — A
14a4–25	450		–1.94	–0.08	Bivalve — LMC
14a4–32	457		–0.50	2.27	Bivalve — LMC
14a4–32	457		1.22	7.26	Bivalve — LMC
14a4–35	460		–1.05	3.15	Bivalve — LMC
<i>Core 93L</i>					
1993–16	16		10.13	–2.33	Charophytes ^e
1993–24	24		8.97	–3.73	Bulk
1993–30	30		9.77	–2.43	Bulk
1993–35	35		11.48	–3.00	Charophytes
1993–49	49		11.16	–2.06	Bulk
1993–50	50		8.90	–0.99	Bulk
1993–58	58		10.61	0.03	Bulk
1993–82	82		9.44	–1.37	Bulk
1993–120	120		9.28	–2.17	Bulk
1993–123	123		11.27	–1.56	Charophytes
1993–136	136		7.84	–3.80	Bulk
1993–139	139		8.76	–3.16	Charophytes
1993–150	150		8.07	–3.84	Bulk
1993–155	155		7.87	–3.41	Bulk
1993–165	165		8.04	–2.53	Bulk
1993–171	171		7.72	–2.98	Bulk
1993–176	176		7.59	–3.89	Bulk
1993–180	180		6.82	–3.15	Charophytes
1993–183	183		7.82	–4.80	Charophytes
1993–192	192		10.02	–4.87	Charophytes
1993–285	285		7.26	–0.02	Bulk
1993–287	287		9.67	–1.06	Charophytes
1993–298	298		10.40	–1.10	Charophytes
1993–301	301		11.27	–1.01	Charophytes
1993–310	310		9.22	–0.17	Bulk
1993–312	312		8.96	–2.33	Charophytes

^a Calendar years BP.

^b Sample type (bulk, bioclast, crystal), main mineralogy (A, LMC, HMC). A: aragonite. LMC: low magnesium calcite. HMC: high magnesium calcite.

^c Bulk samples mostly contain carbonate crystals with diatomitic ooze. Nevertheless, some mixing with bioclasts is possible. In core 93L bulk microsamples contain charophyte remains and diatomitic ooze.

^d Bulk microsamples contain carbonate crystals mixed with diatomitic ooze.

^e Charophyte samples are commonly stems and mixtures of LMC and HMC.

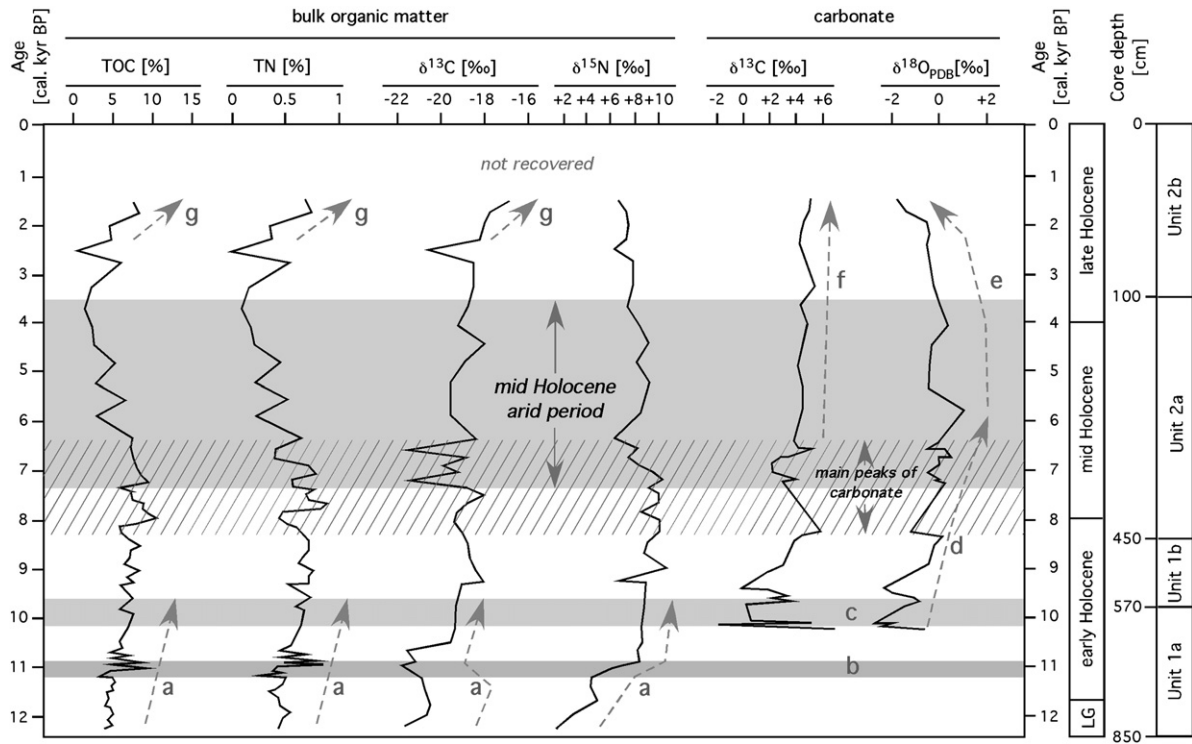


Fig. 7. Left to right: Vertical distribution vs. time (cal years BP) of TOC, TN, $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{15}\text{N}_{\text{AIR}}$ in the organic fraction, and $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$ in the carbonate fraction (ostracode valves and endogenic carbonate crystals) of core 11 from Lake Chungará. It was not possible to obtain data of isotopic composition of carbonate in samples older than 10.3 kyr. The following trends and events are indicated: a) Increase in algal productivity. b) Peak productivity attributed to inputs of nutrients (Sáez et al., 2007). c) Marked changes in surface/volume of Lake Chungará. High increases in productivity and in % of benthic diatoms. Start of significant endogenic carbonate precipitation. d) Increase in $\delta^{18}\text{O}_{\text{VPDB}}$ in carbonates (endogenic carbonate crystals and ostracode remains) until the aridity maximum (around 6 kyr). e) Decrease in $\delta^{18}\text{O}_{\text{VPDB}}$ in carbonates, reflecting the progressive inputs of less evaporated meteoric waters. f) Stability in DIC, probably reflecting few changes in the type of biomass and organic matter provenance. g) Increase in productivity.

(Bao et al., 2010). The absence of calcium was probably the limiting factor in the early stages of lake evolution (Fig. 7) when phytoplankton productivity was dominant.

Changes in type and distribution of carbonates were attributed to the type and amount of biomass and to hydrologic gradients generated by surface streams in the lake. These gradients and surface streams were controlled by dominant winds, and by surface and underground water inputs to the lake. The deposition of mm-thick carbonate layers covering the whole or a considerable part of the lake must be interpreted as an endogenic bioinduced (due to bacterial and algal activity) and hydrologically controlled precipitation. Experimental studies have shown that differences in calcite (and high-Mg calcite) grain shape and size are generated by growth in a bacterial exopolysaccharide matrix with varying contents of amino acids (Braissant et al., 2003): (a) in the presence of varying concentrations of proteins (lysozyme; Jiménez-López et al., 2003), (b) in gel bacterial cultures (González-Muñoz et al., 2000) or (c) in a gelatinous matrix with varying amounts of Mg^{2+} (Fernández-Díaz et al., 1996). The changes in morphology and mineralogy observed in the carbonates of Lake Chungará may be the result of variations in the Mg/Ca in the water body and the influence of organic matrices. Aragonite precipitation has been attributed to the loss of CO_2 in waters with Mg/Ca ratios exceeding 3 (Bischoff and Fyfe, 1968; Müller et al., 1972; Barkan et al., 2001). This is consistent with the almost monomineral nature (aragonite or calcite) of single offshore carbonate levels. The most recent offshore sediments of Lake Chungará lack aragonite levels, indicating that the Mg/Ca molar ratio in the lake waters was commonly under this value. A single change from calcite (in the deeper part of the lake) to aragonite needle-shaped crystals (on the eastern platform of the lake) was observed along an individual cm-thick carbonate-rich level at a distance of about 4 km. This can be explained by local simultaneous or quasi-simultaneous changes in the water inputs. Both minerals are usually mixed as calcite crystals and aragonite

spheroids in marginal areas of the lake (Fig. 3). This is due to the mixing of benthic aragonite spheroids with bloom-related calcite crystals. Similar spheroids in aragonite showing inner radial structure were grown in biofilms and gels (Suess and Fütterer, 1972; Fernández-Díaz et al., 1996) and have also been described for calcite and vaterite (Giralt et al., 2001), suggesting that organic mucilages such as bacterial or cyanobacterial polysaccharides could control the resultant mineralogy and shape (Braissant et al., 2003). Benthic photosynthetic cyanobacteria that are present in the coastal areas of Lake Chungará would provide the exclusive matrix in which aragonite spheroids grew (Défarge et al., 1996).

Algal blooms causing CO_2 photosynthetic depletions and Ca^{2+} inputs coming from marginal aquifers are the limiting factors in the precipitation of the mm-thick Ca-carbonate layers. In fact, Ca-carbonate precipitation in offshore areas reflects the combined effect of several factors: at short time scales the effect of algal blooms, at intermediate time scales the effect of evaporation, and at long time scales, modifications in water chemistry due to changes in watershed rock composition, direct volcanoclastic inputs to the lake, and weathering (Sáez et al., 2007). The enhancement of endogenic carbonate precipitation during wet periods has been related to increases in the aquifer recharge rate, and to Ca inputs into previously Ca-depleted waters (Julià et al., 1998; Shapley et al., 2005; Romero et al., 2006). Moreover, these inputs increased the productivity of macrophytes (*Chara*, *Myriophyllum*) and benthic algae in the marginal areas and platforms of the lake, resulting in local CO_2 depletions where the photic zone reached the lake bottom. This together with the Ca^{2+} marginal gradient accounts for the higher amount and the larger size of the Ca-carbonate components in the lake littoral sediments.

The isotopic evolution along time, given the data from cores 11, 14, 15 and 93L (separated by sedimentary units; Fig. 6B) shows changes in range and covariance that can be significant. The variability and

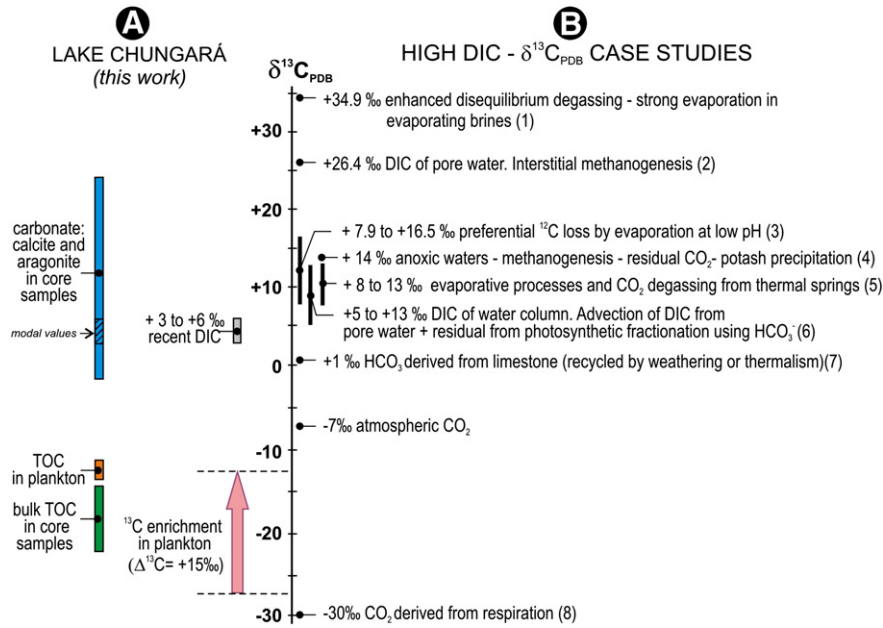


Fig. 8. On the left, $\delta^{13}\text{C}_{\text{VPDB}}$ for carbonates, TOC in recent plankton and core sediments, and recent DIC for Lake Chungará are shown. The vertical arrow represents the estimated ^{13}C enrichment for plankton. On the right, a compilation of high DIC - $\delta^{13}\text{C}_{\text{VPDB}}$ values reported in the literature and the related processes are shown: (1) Laboratory conditions (Stiller et al., 1985). (2) Pore-water DIC in Lake Bosumtwi (Talbot and Kelts, 1986) and Lake Apopka (Gu et al., 2004). (3) and (4) Dead Sea brines (Stiller et al., 1985; Gu et al., 2004). (5) El Peinado and San Francisco lakes (Valero-Garcés et al., 1999a). (6) DIC in the water column, Lake Apopka (Gu et al., 2004). (7) and (8) From the compilation by Keeley and Sandquist (2006).

covariance of Unit 1 (6 and 9 δ units for carbon and oxygen, and $r=0.9$ in subunit 1a) show that Lake Chungará was shallower and smaller, which is consistent with the sedimentological data. Units 2 and 3 show smaller variability (near 6 δ units for carbon and oxygen) and a weak and negative covariance (r around -0.4). This negative

covariance can be ascribed to the recycling of older carbon due to methanogenesis. Unit 4 (which corresponds to the 93L core samples) shows the narrower range of isotopic values along the lake evolution (around 5 δ units) and a weak covariance ($r = 0.36$). The progressive decrease in the range may be due to the fact that Lake Chungará

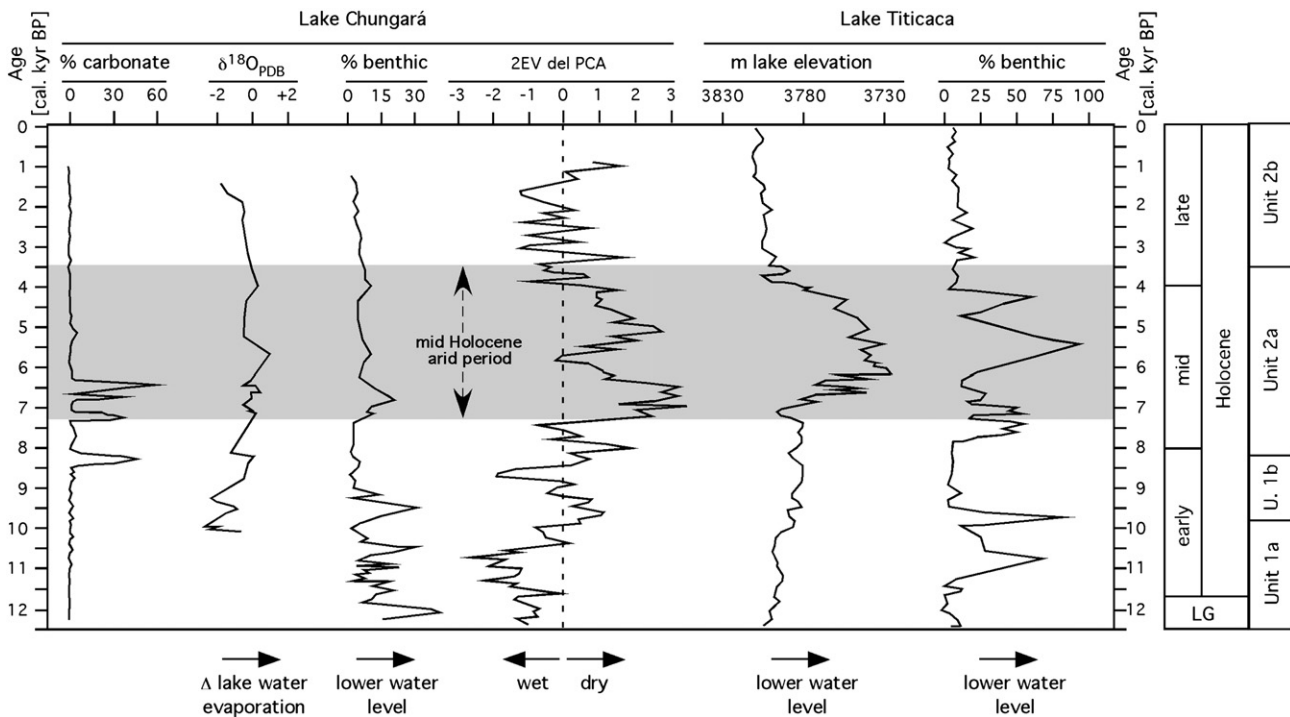


Fig. 9. Variability of proxies related to aridity along the last 12.5 cal kyr BP in the Chungará and Titicaca lacustrine sediments. From left to right: carbonate content (wt.%) (Sáez et al., 2007), $\delta^{18}\text{O}_{\text{VPDB}}$ in the carbonate fraction (this work), benthic diatoms (%) (modified from Sáez et al., 2007) and water availability (statistical approach by Giralt et al., 2008) in core 11 samples from Lake Chungará. The two graphs on the right correspond to lake water elevation (masl) (Abbot et al., 2003) and % of benthic diatoms in sediments (Baker et al., 2001b) in Lake Titicaca.

reached its highest level during the later evolutionary stages. The weak and positive covariance probably reflects the water inputs and evaporation in the shallow and coastal setting where core 93L is placed.

5.4. Mid-late Holocene paleoenvironmental changes from the carbonate record

Core profile 11 combines low magnesium calcite crystals ('inorganic' carbonate) and ostracode low magnesium calcite valves (biogenic carbonate). In the absence of experimental data on the vital effects for *Limnocythere* sp and in order to be able to combine the two carbonates, three samples in which both carbonates coexist were used to calculate the vital effects (Table 3). ^{13}C and ^{18}O relative enrichments of *Limnocythere* carbonate with respect to the coexisting inorganic carbonate was determined in -0.70 and $+0.90\text{‰}$, respectively. Calcite endogenic crystals and ostracode valves (corrected for the vital effect) display similar isotopic values in the Lake Chungará offshore area (i.e. cores 11 and 14), reflecting the $\text{DIC}-\delta^{13}\text{C}_{\text{VPDB}}$ values (now around $+4$ to $+6\text{‰}$; Table 1) in the lake waters. By contrast, bivalve remains vary considerably more because of greater vital effects and the inheritance of aquatic plants that are common in their diets (Keith et al., 1964). This prompted us to use the isotopic data from endogenic carbonate crystals and ostracode valves (corrected of vital effects) from core 11 to obtain the $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$ vertical profiles of the carbonate fraction (Fig. 7). These profiles were used to characterise the environmental fluctuations of Lake Chungará over the last 10.2 kyr.

The early stages of carbonate precipitation through the upper part of subunit 1a and in the lower part of subunit 1b (between 10.2 and 9.0 cal kyr BP) vary considerably in the $\delta^{13}\text{C}_{\text{VPDB}}$ values, between $+7$ and -2‰ , and in $\delta^{18}\text{O}_{\text{VPDB}}$, from -3‰ to 0‰ (Figs. 7 and 9). This isotopic variability coincided with significant changes in the lake surface/volume ratio (Sáez et al., 2007). First, a flooding of extensive shallow areas of the eastern platform occurred (Hernández et al., 2008). The low $\delta^{18}\text{O}_{\text{VPDB}}$ values at these times (around -3‰) indicate that lake waters had scarcely evaporated. After the flooding, a major peak of benthic diatoms exceeded 40% between 10.0 and 9.6 cal kyr BP, indicating a shallow water environment. During this dry period, lake waters progressively evaporated, and an increase in photosynthetic activity related to the development of extraordinary diatom blooms (Bao et al., 2010) could have triggered the endogenic precipitation of carbonate crystals. Moreover, variations in seasonality giving rise to calcium inputs into the lake linked to short humid seasons during arid periods (such as the recent 'Bolivian or altiplanic winter' in the Andes due to changes in the location of the Bolivian High; Vuille, 1999) could have caused intermittent carbonate precipitation (Shapley et al., 2005).

Along the upper part of subunit 1b and in the lower part of subunit 2a (between 9.0 and 7.5 cal kyr BP), $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$ vary less with higher values between $+6$ and $+2\text{‰}$ and between -1 and $+1\text{‰}$, respectively. Endogenic carbonate precipitation (mainly as low magnesium calcite crystals) increases significantly in this period, with a strong peak around 8.3 kyr (Fig. 9), which matches the increases in both $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$, indicating the evolution towards more evaporated lake waters.

Finally, in subunit 2a and in the lower part of subunit 2b (between 7.0 and 1.5 cal kyr BP) $\delta^{13}\text{C}_{\text{VPDB}}$ values are stable and vary around $+4\text{‰}$, and $\delta^{18}\text{O}_{\text{VPDB}}$ between $+2$ and -1‰ , respectively. The trend in subunit 2a towards higher $\delta^{18}\text{O}_{\text{VPDB}}$ values (Figs. 7 and 9) reflects the increase in evaporation (and salinity) and the reduction of the lake water volume and depth, coeval with the highest carbonate precipitation, between 8.3 and 6.4 cal kyr BP. The fact that higher $\delta^{18}\text{O}_{\text{VPDB}}$ values occurred during the middle and upper part of subunit 2a and the end of the main carbonate event seems to confirm this evolution: Between the onset of carbonate precipitation at 10.2 kyr and the arrival of the mid-Holocene arid period at 7.3 kyr, carbonates show relatively low $\delta^{18}\text{O}_{\text{VPDB}}$ values. The $\delta^{18}\text{O}_{\text{VPDB}}$, which increased in

carbonates during the development of the arid event, shows that lake waters became more evaporated. In the uppermost carbonate sediments (subunit 2b) the trend reverses, around 6 cal kyr BP, to lower $\delta^{18}\text{O}_{\text{VPDB}}$ values with a drop around 2.2 cal kyr BP, which is interpreted as the final decrease in aridity (Figs. 7 and 9).

The increase of about 3‰ between the base of subunit 1b and the centre of subunit 2a followed by a decline until reaching subunit 2b should be interpreted as reflecting an arid period between 7.3 and 3.5 cal kyr BP, with an aridity maximum of around 6.0 cal kyr BP (Fig. 9). An arid period in the mid-Holocene has been suggested on the basis of high resolution XRF and mineralogical data (Moreno et al., 2007) and on the basis of water availability quantified using a statistical approach (Giralte et al., 2008). Furthermore, increases in benthic diatoms at 6.9 (20%), 5.7 (10%) and 4.1 cal kyr BP (Fig. 9) reflect shallower periods along the evolution of the Chungará paleolake (Sáez et al., 2007). This arid period is coeval with the aridity crisis observed in other paleoclimatic records of the Altiplano (Paco Cocha and Titicaca lakes, Abbot et al., 2003; Tapia et al., 2003; and Salar de Uyuni, Baker et al., 2001a) (Fig. 9).

6. Conclusions

Both the sedimentary record and the isotopic markers indicate that phytoplankton is the main source of organic matter in the offshore sediments of Lake Chungará. Lower Unit 1 (12.3 to 8.3 cal kyr BP) is a diatomaceous ooze and the upper Unit 2 (8.3 cal kyr BP to present) is a diatomaceous sediment interbedded with volcanic tufa. Phytoplankton, macrophytes and, to a lesser extent, land plants are the main organic contributions to the littoral sediments.

TOC- $\delta^{13}\text{C}_{\text{VPDB}}$ values reflect the organic provenance (phytoplankton vs. macrophytes and land plants) and changes in productivity. The increase in TN- $\delta^{15}\text{N}_{\text{AIR}}$ in the early stages of lake evolution reflects the increase in aquatic productivity. Changes in TOC along the sedimentary record provide evidence of the increase in biological productivity and in the macrophytic contribution. Low values of $\delta^{15}\text{N}_{\text{AIR}}$ around 0‰ in the lowermost lacustrine sediments are probably due to the dominance of cyanobacterial biomass.

Carbonate content in the Lake Chungará sediments is commonly below 1 wt.% but is higher in the upper part of subunit 1b and the lower part of subunit 2a. In addition to its presence as bioclasts (small molluscs and ostracodes), carbonates consist of endogenic calcite (or aragonite) crystals arranged as cm-thick levels. Seasonal CO_2 photosynthetic depletions of lake water DIC may have been responsible for the mm-thick carbonate layer deposition. The long-term carbonate deposition seems to be controlled by changes in seasonality.

Euhedral to subhedral calcite and aragonite crystals and spheroids, micrometric in size, are the most common shapes. Many crystal shapes are similar to those obtained by induced precipitation in gels or exopolysaccharide and proteinic matrix, suggesting similar processes in the sediments of Lake Chungará that account for the coexistence of calcite crystals and aragonite spherules. The presence of calcite vs. aragonite has been attributed to the Mg/Ca ratio in the water and to the precipitation in an organic matrix. Lateral changes in mineralogy (calcite crystals vs. aragonite needles) detected at the same level in locations at a distance of some km were ascribed to contemporary (or quasi-contemporary) chemical changes in the water inputs into the lake.

$\delta^{13}\text{C}_{\text{VPDB}}$ in endogenic carbonate crystals and ostracode valves increased while $\delta^{18}\text{O}_{\text{VPDB}}$ decreased from basin to shore, reflecting the relative contribution of the organic producers (phytoplankton and macrophytes), the DIC inputs, and the residence time. The $\delta^{18}\text{O}_{\text{VPDB}}$ depends on the isotopic composition of water inputs into the lake and its water balance. The variations and covariation of $\delta^{13}\text{C}_{\text{VPDB}}$ and $\delta^{18}\text{O}_{\text{VPDB}}$ indicate that the lake was shallower in the early stages of its evolution and that the water level and volume increased with time. Bacterial methanogenesis in the bottom sediment gave rise to carbonate cements with very high $\delta^{13}\text{C}_{\text{VPDB}}$ values. Heavy carbon

advection to the water column is with DIC speciation (HCO_3^- dominant at pH around 9) responsible for the high $\delta^{13}\text{C}_{\text{VPDB}}$ values (around 15‰ higher than commonly reported values) in the whole carbon reservoirs of Lake Chungará.

Lake Chungará underwent major changes between the onset of the Holocene and around 9.6 cal kyr BP, which is evidenced by an overall increase in TOC, TN, and $\text{TN}-\delta^{15}\text{N}_{\text{AIR}}$ as well as by fluctuating values of $\text{TOC}-\delta^{13}\text{C}_{\text{VPDB}}$. These changes were probably linked to major shifts in the lake surface/volume ratio, associated with the rise and fall in lake level. A pulse of aridity at around 10 kyr BP in addition to the increased calcium content in the lake water favoured the onset of carbonate precipitation. $\delta^{18}\text{O}_{\text{VPDB}}$ values in ostracode valves and endogenic carbonate crystals, combined with other proxies (benthic/planktonic diatoms ratio, water availability inferred from the statistical treatment of magnetic susceptibility and geochemical data) demonstrate an arid period in the mid-Holocene, between 7.3 and 3.5 kyr B.P. and an aridity maximum around 6.0 kyr BP.

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