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1	Occurrence of inorganic arsenic in edible Shiitake (Lentinula edodes) products
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16	
17	Abstract
18	The present study reports arsenic speciation analysis in edible Shiitake (Lentinula
19	edodes) products. The study focused on the extraction, and accurate quantification of
20	inorganic arsenic (iAs), the most toxic form of arsenic, which was selectively separated
21	and determined using anion exchange LC-ICPMS. A wide variety of edible Shiitake
22	products (fresh mushrooms, food supplements, canned and dehydrated) were purchased

24 analysed. The extraction method showed satisfactory extraction efficiencies (>90%) and

and analysed. A cultivated Shiitake grown under controlled conditions was also

column recoveries (>85%) for all samples. Arsenic speciation revealed that iAs was the

major As compound up to 1.38 mg As kg⁻¹ dm (with a mean percentage of 84% of the total arsenic) and other organoarsenicals were found as minor species. Shiitake products had high proportions of iAs and therefore should not be ignored as potential contributors to dietary iAs exposure in populations with a high intake of Shiitake products.

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32 Keywords: Inorganic arsenic; Shiitake; *Lentinula edodes*; Mushrooms; Arsenic
33 speciation; LC-ICPMS.

34

35 1. Introduction

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The consumption of wild edible mushrooms has increased worldwide during 37 recent years. Lentinula edodes (Berk.) Pegler (also known by its Japanese name of 38 Shiitake) is one of the five most cultivated edible mushrooms in the world, being 39 40 particularly popular in China, Japan and other Asian countries (Kalač, 2013; Chang & 41 Miles, 2004). Furthermore, it is a dietary source of protein, vitamin D, B complex 42 vitamins and minerals. It is one of the best-known and best-characterized mushrooms, having been used in medicine for thousands of years. Lentinula edodes mycelium 43 extract and its purified fractions have many physiological properties including 44 45 antitumour, antiviral, antioxidant, antifungal, hypoglycaemic and immunomodulatory activity (Chang & Miles, 2004; Wasser, 2002). 46

47 Regarding the toxicological aspects of arsenic in food, inorganic arsenic (iAs,
48 (arsenite or As(III) and arsenate or As(V)) is considered to be the most dangerous form
49 due to its biological availability and physiological and toxicological effects (iAs is
50 classified as a nonthreshold, class 1 human carcinogen) (ATSDR Toxicological profile

for arsenic, 2007). Other arsenic compounds, such as arsenobetaine (AB), are non-toxic and can be consumed without concern, while arsenosugars are potentially toxic (Feldmann & Krupp, 2011). Therefore, toxicological knowledge of the different arsenic species should be considered by legislators and regulators when establishing maximum arsenic levels in food directives.

56 The ability of some mushroom species to accumulate arsenic may represent a 57 serious risk to consumer health (Dembitsky & Rezanka, 2003; Falandysz & Borovička, 2013; Kalač, 2010; Vetter, 2004). The arsenic content of mushrooms is regulated by 58 59 genetic factors and natural conditions (type of soil, bedrock, habitat, environmental 60 factors) (Falandysz & Borovička, 2013; Vetter, 2004). More than 50 different naturally 61 occurring As-containing compounds have been identified, comprising both organic and inorganic forms (EFSA Panel on Contaminants in the Food Chain (CONTAM), 2009). 62 Some of these have been found in mushrooms, including methylarsonate (MA), 63 dimethylarsinate (DMA), As(V), As(III), AB, arsenocholine (AC), trimethylarsine 64 oxide (TMAO), tetramethylarsonium cation (TETRA) and arsenosugars (Koch et al., 65 2013; Koch, Wang, Reimer, & Cullen, 2000; Larsen, Hansen, & Gössler, 1998; 66 67 Niedzielski, Mleczek, Magdziak, Siwulski, & Kozak, 2013; Šlejkovec, Byrne, Stijve, Goessler, & Irgolic, 1997; Smith, Koch, & Reimer, 2007; Soeroes, Kienzl, Ipolyi, 68 69 Dernovics, Fodor, & Kuehnelt, 2005).

The arsenic compounds in edible mushrooms are obviously of concern to the consumer and the regulatory authorities, but currently, no limits exist in the European Union (EU) on arsenic, either total or inorganic, in foods (European Union Regulation 1881/2006). On the other hand, China has a maximum allowable concentration of total arsenic in mushrooms of 0.5 and 1.0 mg As kg⁻¹, for fresh and dry mushrooms, respectively (MHC, 2003; MHC, 2005). Given this situation, the European Food Safety

Authority (EFSA) (EFSA Panel on Contaminants in the Food Chain (CONTAM), 2009) 76 and the FAO/WHO Joint Expert Committee on Food Additives (JECFA) (FAO/WHO, 77 Evaluation of certain contaminants in food, 2011) have evaluated dietary exposure to 78 79 As. Both reported the urgent need for further data on arsenic species, particularly iAs data, in food commodities, in order to improve the background data for future risk 80 81 assessment analysis. Furthermore, mushrooms were included among the foods that 82 contribute to iAs exposure in the general European population (EFSA Panel on Contaminants in the Food Chain (CONTAM), 2009). The report also highlighted the 83 84 need for a robust validated analytical method for the determination of iAs in a range of 85 food items. To this end, several proficiency tests (PTs) on iAs in different foodstuffs 86 have been organised (de la Calle et al., 2011; Baer et al., 2011; de la Calle et al., 2012). Satisfactory performance was generally found for the determination of iAs in rice, 87 wheat and vegetable food; and it was also emphasized that there is no reason to consider 88 89 the option of introducing possible maximum levels for iAs in rice, wheat, vegetable food and algae, in further discussions on risk management. 90

Due to the increasing focus on inorganic arsenic in food and given that 91 92 mushroom consumption had increased considerably in recent years due to their nutritional properties, two PTs, using the same test item, IMEP-116 and IMEP-39, were 93 organised by the Institute for Reference Materials and Measurements (IRMM) 94 (Cordeiro et al., 2013). Thus, the total and inorganic arsenic content in mushrooms is a 95 96 topic of current priority for the Directorate for Health and Consumers (DG SANCO) of 97 the European Commission. The iAs concentration in the Shiitake test sample was quite high, at around 0.3 mg As kg⁻¹, accounting for 50% of the total As. Therefore, arsenic 98 speciation data, particularly iAs data, for Shiitake samples are needed to estimate the 99 100 health risk associated with dietary As exposure.

Although Shiitake has medicinal properties and is one of the most consumed and 101 cultivated mushrooms, few studies of arsenic speciation appear in the literature 102 103 (Wuilloud, Kannamkumarath, & Caruso, 2004). Thus, more studies on Shiitake are required to provide information about iAs levels, which would be useful in toxicological 104 risk assessments. Therefore, the main goal of this study was to determine total arsenic 105 106 and arsenic species in several edible Shiitake products. The study focused on the 107 extraction, identification and accurate quantification of the toxic inorganic arsenic species. In addition, a preliminary study of Shiitake cultivation was performed in a 108 109 small-scale mushroom facility in order to estimate the possible health risks of home-110 cultivated Shiitake grown on a commercial substrate. Fruiting bodies and substrate 111 samples were investigated for total arsenic and arsenic species.

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113 2. Materials and methods

114 2.1. Reagents and standards

All solutions were prepared with doubly deionised water obtained from Millipore water purification systems (Elix & Rios) (18.2 M Ω cm⁻¹ resistivity and total organic carbon <30 µg L⁻¹). Nitric acid (69%, Panreac, Hiperpur) and hydrogen peroxide (31%, Merck, Selectipur) were used for the digestion and extraction procedures. Ammonium dihydrogen phosphate (Panreac, p.a.), ammonia solution (25%, Panreac, p.a.), pyridine (Scharlau, p.a.) and formic acid (98%, Panreac, p.a.) were used to prepare mobile phases.

External calibration standards for total As were prepared daily by dilution of a standard stock solution traceable to the National Institute of Standards and Technology (NIST), with a certified concentration of 1000 ± 5 mg As L⁻¹ (Inorganic Ventures 125 Standards). An arsenate standard solution of $1000 \pm 5 \text{ mg As } \text{L}^{-1}$ (Merck) was used for 126 external quality control in total arsenic and arsenic speciation measurements.

Stock standard solutions (1000 mg As L^{-1}) for arsenic speciation were prepared 127 128 as follows: As(III), from As₂O₃ (NIST, USA, Oxidimetric Primary Standard 83d, 99.99%) dissolved in 4 g L⁻¹ NaOH (Merck, Suprapure); As(V), from Na₂HAsO₄·7H₂O 129 130 (Carlo Erba) dissolved in water; MA, prepared from (CH₃)AsO(ONa)₂·6H₂O (Carlo Erba) dissolved in water; and DMA, prepared from (CH₃)₂AsNaO₂·3H₂O (Fluka) 131 dissolved in water. AC from (CH₃)₃As⁺(CH₂)CH₂OHBr⁻ was supplied by the "Service 132 Central d'Analyse" (CNRS Vernaison, France); and a certified reference material of AB 133 from (CH₃)₃As⁺CH₂COO⁻ was supplied by National Metrology Institute of Japan 134 (NMIJ, Japan) as NMIJ CRM 7901-a, standard solution. TMAO was prepared from 135 (CH₃)₃AsO (Argus Chemicals srl) dissolved in water. Arsenate, arsenite, DMA, MA, 136 AC, TMAO and AB were standardised against As₂O₃ for our internal quality control. 137 All stock solutions were kept at 4 °C, and further diluted solutions for the speciation 138 analysis were prepared daily. 139

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141 *2.2 Samples and certified reference materials*

Different types of Shiitake-based food commodities that are representative of all types of edible Shiitake products consumed in Spain, were purchased from markets, local supermarkets and retail stores in Barcelona, Spain, during 2012. A selection of edible Shiitake products was analysed: five fresh, four dehydrated, three canned and two food supplement samples. The three canned Shiitake are commercialised in glass vessels. According to the manufacturer, Shiitake food supplements contain both mycelium and primordia (young fruit body) cultivated into a biomass that is grown on a sterilised (autoclaved) substrate. Various brands were purchased and all samples were brought to the laboratory on the day of purchase and kept for no more than a day in the refrigerator until sample preparation, which was performed before the recommended time of consumption.

In addition, Shiitake was home-cultivated in a small-scale facility, from which mushrooms were collected as samples for further analysis, to expand the information collected in the study.

Two certified reference materials (CRMs) and a reference material (RM) were analysed during the study. NIST SRM 1570a spinach leaves was obtained from the NIST (Gaithersburg, MD, USA). WEPAL IPE-120 reference material *Agaricus bisporus* mushroom was produced by the Wageningen Evaluating Programs For Analytical Laboratories (WEPAL, Wageningen, The Netherlands). ERM-BC211 rice was obtained from the IRMM of the European Commission's Joint Research Centre (Geel, Belgium).

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164 2.3 Apparatus and instrumentation

165 Mushroom samples were dried in an oven with natural convection (Digitronic, 166 JPSelecta, Spain). The dried mushrooms were minced using a commercial mincer (Multiquick 5 Hand Processor, Spain) Braun). A microwave digestion system (Ethos 167 Touch Control, Milestone), was used for the digestion and extraction procedures. An 168 Agilent 7500ce inductively coupled plasma mass spectrometer (ICPMS) (Agilent 169 Technologies, Germany) was used to determine total arsenic content. An Agilent 1200 170 Series LC system (Agilent Technologies, Germany) was used as the chromatographic 171 system for arsenic speciation via coupling LC-ICPMS. The separations were performed 172

on an anion-exchange column (Hamilton Company, USA) and cation-exchange column
(Agilent Technologies, Germany) (Table 1). The outlet of the LC column was
connected via polyetheretherketone capillary tubing to the nebuliser (Burgener Research
Inc, Mississauga, Canada) of the ICPMS system (Table 1).

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178 2.4. Cultivation of Shiitake

179 Cultivation of Shiitake was performed in a small-scale mushroom facility belonging to the University of Barcelona. Fruiting bodies of Shiitake were produced on 180 a commercial pasteurised substrate inoculated with mycelium intended to be grown at 181 homemade cultivation. The cultivation procedure followed the instructions supplied by 182 183 the manufacturer. The mushrooms were grown under controlled conditions following the manufacturer's guidelines, and a large number of fruiting bodies were produced. 184 The original substrate was submerged in tap water in a controlled chamber for 24 hours. 185 186 Then, the substrate was placed in a cool damp place at a temperature of 17-20 °C, under natural indoor light cycles. After a week the fungi began to fruit and all mushrooms 187 188 were harvested. After the first harvest, the substrate was air dried for 20 days. After this 189 time, the substrate was submerged in tap water for another 24 hours and the whole 190 process was repeated. This enabled a second Shiitake mushroom harvest.

191 The substrate was randomly sampled in triplicate three times during the cultivation 192 study. Care was taken to collect substrate in which mycelium was not visible to the 193 naked eye. The original substrate was first sampled before submersion and cultivation. 194 A second sample was taken after the first cultivation (medium substrate) and a third 195 sample after the second cultivation (final substrate). The tap water and waste water, and 196 the water remaining after substrate submersion, were also sampled during the cultivation study. Total arsenic and arsenic species were analysed by ICPMS and LCICPMS, respectively, in the three substrate samples and the tap and waste water
samples.

200

201 2.5 Sample pretreatment

202 Fresh Shiitake mushrooms were cleaned by hand of substrate and foreign matter. 203 The end of the stalk (in contact with the substrate) was removed using a stainless steel knife. Damaged or soiled parts were cut off with a knife and smaller particles were 204 205 removed using a fine brush. Only the edible parts of the mushrooms were used for the 206 analysis. Mushrooms were cut into small pieces that were then air-dried on filter paper 207 and further dried in an oven at 40°C for 24-48 hours. The dried mushrooms were minced using a commercial mincer made of stainless steel until complete 208 homogenization. Care was taken to avoid contamination. Between samples, the mincer 209 210 was washed once with soap and water, rinsed once with HNO_3 (about 10%), rinsed several times with deionised water, and then rinsed three times with doubly deionised 211 water, before drying with cleaning wipes. 212

213 Shiitake food supplements, which are commercially available as tablets, were 214 pulverized with an agate mortar, homogenised and stored over silica gel in a desiccator 215 until analysis.

Canned Shiitake samples were drained and then dried in an oven at 40°C for 24–
48 hours and finally minced using a commercial mincer until complete homogenisation.
Powdered samples were stored over silica gel in a desiccator until analysis.

Dehydrated Shiitake samples were cut into small pieces and then minced using a commercial mincer until complete homogenisation and stored over silica gel in a desiccator until analysis. Cultivated Shiitake were pretreated in the same way as the purchased fresh mushrooms. Substrate samples were pulverized, homogenised, and stored over silica gel in a desiccator until analysis of arsenic and arsenic species. Tap water and waste water were filtered through PET filters (Chromafil® PET, Macherey–Nagel, pore size 0.45 μm) and stored at 4°C before analysis of total arsenic and arsenic species.

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228 2.6. Moisture determination

Aliquots of 0.5 g samples were dried, in triplicate, at 102 ± 3 °C to constant weight in an oven. All the results in the study are expressed as dry mass.

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232 2.7 Total arsenic determination

The total arsenic content of the mushroom samples, CRMs, RM and substrate 233 234 samples was determined by ICPMS measurement after microwave digestion (Llorente-Mirandes, Calderón, López-Sánchez, Centrich, & Rubio, 2012). Helium gas was used in 235 the collision cell to remove interferences in the ICPMS measurements. A solution of 236 ⁹Be, ¹⁰³Rh and ²⁰⁵Tl was used as an internal standard. Each sample was digested and 237 analysed in triplicate. The digestion blanks were also measured. Arsenic content in the 238 samples was quantified by means of an external calibration curve for the standards. For 239 quality control purposes, the standards of the calibration curve were run before and after 240 each sample series. The detection (LOD) and quantification limits (LOQ) were 241 estimated and were 0.006 and 0.021 mg As kg⁻¹, respectively. 242

243

244 2.8 Arsenic speciation analysis

245	The extraction of arsenic species was based on our previous studies (Llorente-
246	Mirandes et al., 2012; Llorente-Mirandes, Calderón, Centrich, Rubio, & López-
247	Sánchez, 2014) and was applied here to mushroom samples, CRMs and the RM and
248	substrate samples. Briefly, 0.25 g aliquots of the samples were weighed in PTFE vessels
249	and then extracted by adding 10 mL of 0.2% (w/v) HNO_3 and 1% (w/v) $\rm H_2O_2$ solution
250	in a microwave system. This extraction method completely oxidises As(III) into As(V),
251	without conversion of the other organoarsenicals into iAs. After extraction, arsenic
252	speciation was carried out in extracts by LC-ICPMS (Llorente-Mirandes et al., 2010 and
253	2011) using the conditions shown in Table 1. The total arsenic in the extracts was
254	determined by ICPMS (as described above). Arsenic species were quantified by external
255	calibration curves. Extraction blanks were also analysed in each batch of samples. Each
256	sample was extracted and analysed in triplicate. LOD and LOQ were estimated for each
257	As species. The LODs for As(III), DMA, MA, As(V), AB, TMAO and AC were
258	0.0010, 0.0014, 0.0017, 0.0024, 0.0010, 0.0028 and 0.0018 mg As kg ⁻¹ , respectively.
259	The LOQs for As(III), DMA, MA, As(V), AB, TMAO and AC were 0.0033, 0.0047,
260	0.0056, 0.0080, 0.0033, 0.0093 and 0.0060 mg As kg ⁻¹ , respectively.

261

262 **3. Results and Discussion**

263

264 3.1 Quality assessment in the determination of total arsenic and arsenic species

265 *3.1.1 Total arsenic*

To evaluate the accuracy of total arsenic measurements a RM and two CRMs were analysed with every batch of samples. The present results in these CRMs showed good agreement with the certified values, as shown in Table 2. The percentage accuracy was 102% and 99% for NIST SRM 1570a and ERM-BC211, respectively. 270

271 *3.1.2 Extraction efficiency*

272 Extraction efficiencies (calculated as the ratio of total As in the extract to total As in the sample) were calculated. Several extraction solvents have been used for the 273 speciation of arsenic in mushrooms. Extraction efficiencies appear to be highly variable, 274 275 depending on the mushroom species and extraction solution, ranging from 7% to 129% (Koch et al., 2000; Larsen et al., 1998; Šlejkovec et al., 1997; Slekovec, et al., 1999; 276 Smith et al., 2007; Wuilloud et al., 2004). The present values ranged from 94 to 103% 277 278 and extracted on average 98% of total arsenic (Table 3). These results indicated full 279 extraction of the arsenic species that may exist in Shiitake mushrooms. The extraction 280 efficiency of ERM-BC211, NIST SRM 1570a and WEPAL-IPE-120 was 98%, 93% and 99%, respectively (Table 2). 281

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283 3.1.3 Column recovery

Column recovery (calculated as the ratio of the sum of the species eluted from 284 the chromatographic columns to the total arsenic in the extract injected into the column) 285 286 was calculated to guarantee the correctness of the chromatographic separation. This parameter, assessed in replicates with good reproducibility, allowed us to evaluate the 287 quantification of the As species in mushroom samples. Values close to 100% usually 288 indicate that all arsenic extracted was recovered from the analytical column. The present 289 290 values obtained for column recoveries ranged between 87 and 104% and showed 291 average column recoveries of 97% (Table 3). Satisfactory values were also obtained for the RMs: 102%, 92% and 94% for ERM-BC211, NIST SRM 1570a and WEPAL-IPE-292 120, respectively (Table 2). 293

294

295 *3.1.4 Spiking experiments of inorganic arsenic*

To assure the accurate identification and quantification of inorganic As species, 296 297 three Shiitake samples were spiked by adding As(III) and As(V) standards to solid samples and then homogenised. The mixtures were left to stand for 30 min before 298 extraction. Arsenate was the only inorganic species found in the spiked samples, 299 300 showing the quantitative oxidation of As(III) to As (V) without conversion of the other 301 organoarsenicals into iAs. The concentration of iAs was quantified as As(V) and determined via anion LC-ICPMS. The recovery of iAs from fresh, cultivated and food 302 303 supplement samples was: 93 ± 6 , 97 ± 5 and 94 ± 5 , respectively (mean $\% \pm$ SD, n = 3). 304 The results show that all of the iAs was recovered successfully (average recoveries of 305 95% for iAs in Shiitake samples). Furthermore, the ERM-BC211 rice material, which is certified in inorganic arsenic, was also spiked by adding As(III) and As(V) standards. 306 The concentration of iAs was quantified as As(V) and the recovery of iAs was 307 308 satisfactory: $102 \pm 4\%$, n=3.

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310 *3.1.5 Arsenic species in the reference materials*

Arsenic speciation was performed on CRMs and the RM and the results are summarised in Table 2. To date, no CRMs are available for arsenic species in mushrooms. Therefore, the ERM-BC211 rice was used throughout the study to assess the accuracy and reliability of the As speciation results. The material was analysed and the results were in agreement with the certified values. The percentage accuracy was 98% and 105% for iAs and DMA in ERM-BC211, respectively (Table 2).

The As speciation results in WEPAL-IPE-120 showed that AB was the major As species (40% of the total As). The inorganic arsenic content was 0.033 ± 0.001 mg As kg⁻¹ (corresponding to 20% of the total As), while DMA accounted for 28% of the total

As. No arsenic speciation studies on this RM mushroom have been found in the 320 literature. However, studies on Agaricus sp. found that AB predominated in this 321 mushroom genus (Koch et al., 2013; Šlejkovec et al., 1997; Smith et al., 2007; Soeroes 322 et al., 2005), which is in agreement with the present results. Although WEPAL-IPE-120 323 (Agaricus bisporus) is not certified for arsenic species, the sum of the As species (0.156 324 \pm 0.010 mg As kg⁻¹) compared well with the indicative total As value of 0.137 \pm 0.067 325 mg As kg⁻¹. An unknown compound was found by the cationic column with a retention 326 time of 380 s and could be attributed to TETRA due to the matching of the retention 327 328 times when using the same chromatographic conditions (Kirby et al., 2004). However, it 329 was not possible to check this attribution due to the lack of appropriate standards.

Regarding As species in the NIST SRM 1570a, inorganic arsenic was the major compound at 0.059 ± 0.005 mg As kg⁻¹, which was in agreement with the reference value assigned by expert laboratories in the proficiency test IMEP-112: 0.054 ± 0.012 mg As kg⁻¹ (de la Calle et al., 2012).

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335 *3.1.6 External quality control*

336 This method was tested with participation as an expert laboratory in two recent proficiency tests organised by the European Union Reference Laboratory for Heavy 337 Metals in Feed and Food (EURL-HM) and the International Measurement Evaluation 338 Program (IMEP) from the IRMM, IMEP-116 and IMEP-39, Determination of total Cd, 339 340 Pb, As, Hg and inorganic As in mushrooms. Satisfactory results were obtained 341 compared with the assigned value for iAs, which demonstrates the validity and reliability of the present method (Cordeiro et al., 2013). Therefore, this method could be 342 recommended for the quantification of inorganic arsenic in edible mushrooms. 343

344

The total arsenic content in the purchased edible Shiitake products is shown in 346 Table 3 and ranged from 0.11 to 1.44 mg As kg⁻¹ dry mass (dm). The mean arsenic 347 concentration of 14 samples was 0.51 mg As kg⁻¹ dm. Total arsenic was highest in fresh 348 samples (n=5): 0.90 ± 0.57 mg As kg⁻¹ dm (mean \pm SD) with wide variability between 349 the samples. The total arsenic content for dehydrated (n=4) and canned (n=3) samples 350 was 0.26 ± 0.08 and 0.33 ± 0.29 mg As kg⁻¹ dm, respectively. Two food supplements of 351 different brands were analysed and the total arsenic content was 0.45 ± 0.01 and $0.12 \pm$ 352 0.01 mg As kg⁻¹ dm. Four of the fresh Shiitake samples exceeded the limit of 0.5 mg As 353 kg⁻¹ established by China for fresh mushrooms (MHC, 2003; MHC, 2005). However, 354 none of the dehydrated Shiitake exceeded the limit of 1.0 mg As kg⁻¹ established by 355 China for dry mushrooms (MHC, 2003; MHC, 2005). 356

The present arsenic results are in the usual range found in mushrooms from 357 unpolluted areas (0.5 to 5 mg As kg⁻¹, Kalač, 2010). However, arsenic content appears 358 to be highly variable, with significant differences according to the soil arsenic 359 concentration as well as the ability of mushroom species to accumulate arsenic 360 (Falandysz & Borovička, 2013; Kalač, 2010). To date, there are few studies of arsenic 361 content in Shiitake in the literature. Several Shiitake purchased in Brazil contained As 362 in concentrations ranging between 0.083 and 0.210 mg As kg⁻¹ dm (Maihara, Moura, 363 Catharino, Castro, & Figueira, 2008). Another study reported an arsenic content of 1.3 364 mg As kg⁻¹ dm in a Shiitake sample (Wuilloud et al., 2004). According to Haldimann 365 366 and co-authors (Haldimann, Bajo, Haller, Venner, & Zimmerli, 1995), the As content in five Shiitake mushrooms varied from 0.04 to 0.07 mg As kg⁻¹ dm. The available results 367 on arsenic in Shiitake-based food are limited and conflicting. Given the number of 368 samples analysed in the present study and the small amount of data available in the 369

370 literature, the present values of arsenic content cannot be generalized to indicate the371 concentrations commonly present in Shiitake mushrooms.

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373 *3.3 Arsenic species in purchased Shiitake*

The arsenic speciation results for the purchased edible Shiitake products are 374 shown in Table 3. Inorganic arsenic was the predominant As compound in all Shiitake 375 products and ranged from 0.086 to 1.38 mg As kg⁻¹ dm, with a mean value of 0.43 mg 376 As kg⁻¹ dm. Inorganic arsenic accounted for 53 to 99% of the total arsenic with a mean 377 percentage of 84% of the total arsenic, whereas DMA, MA, AB, and TMAO accounted 378 379 for a few percent of the total arsenic. DMA accounted for 2.7 to 28%, MA accounted 380 for 1.6 to 7.6% and AB accounted for 0.4 to 5.5% of the total arsenic. TMAO was only quantified in one sample, accounting for 3.1% of total arsenic, and AC was below the 381 LOQ in all samples. An unknown compound separated by the anionic column was 382 383 found in one sample of fresh Shiitake, with a retention time of 255 s. This unknown anionic arsenic species could be a phosphate arsenosugar. This hypothesis is supported 384 by the fact that the retention time of phosphate arsenosugar, present in *Fucus serratus* 385 extract, matches the retention time of the present unknown peak, when using the same 386 chromatographic conditions (Madsen, Goessler, Pedersen, & Francesconi, 2000). 387 However, due to the lack of appropriate standards, this identification was not checked. 388

The finding that almost all the arsenic in the present edible Shiitake products was present as inorganic As is shown in Figure 1. An example of this behaviour is illustrated in an anion exchange chromatogram of fresh Shiitake extract in which iAs was identified as the main arsenic species; DMA was also clearly detected and traces of MA and cationic species were also present.

To date and to our knowledge, few studies on arsenic speciation in Shiitake are 394 present in the literature. A study on inorganic arsenic content in Hong Kong foods 395 found an iAs value ranging from 0.036 to 0.053 mg As kg⁻¹ dm in dehydrated Shiitake 396 samples (Wong, Chung, Chan, Ho, & Xiao, 2013). Our results on iAs in dehydrated 397 samples (n=4) are consistent with this study, with a mean value of 0.21 mg As kg⁻¹ dm 398 corresponding to 79% of the total arsenic. Wuilloud and colleagues (Wuilloud et al., 399 400 2004) analysed Shiitake samples by size-exclusion liquid chromatography (SEC) coupled to UV and ICPMS for detection (SEC-UV-ICPMS). In their study arsenic was 401 402 found to be associated mainly with a molecular weight (MW) fraction of 4.4-4.9 kDa 403 for all extraction solvents. The authors concluded that the arsenic species are mainly in 404 a form that is not associated with proteins or other high MW compounds, which is 405 consistent with the present results.

Different proportions of arsenic species have been reported in the literature 406 depending on the mushroom species (Dembitsky & Rezanka, 2003; Falandysz & 407 Borovička, 2013; Kalač, 2010). Gonzálvez et al. (Gonzálvez, Llorens, Cervera, 408 Armenta, & de la Guardia, 2009) reported that iAs species were the major compounds 409 in several of the studied mushrooms and that the iAs concentration ranged from 0.14 to 410 0.89 mg As kg⁻¹, similar to the present results. However, a high iAs content was found 411 412 in Lycoperdon sp. mushroom samples on a gold mine site contaminated with arsenic (Koch et al., 2000). Slekovec and co-authors (Slekovec, Goessler, & Irgolic, 1999) 413 414 reported that iAs was the predominant As compound, with the sum of arsenite and arsenate up to 35.5 mg As kg⁻¹ dm in *Thelephora terrestris*. A recently study also found 415 high levels of iAs of up to 27.1 and 40.5 mg As kg⁻¹ dm for As(III) and As(V), 416 respectively, for Xerocomus badius from different sample collection places (Niedzielski 417 418 et al., 2013). Arsenic species content could depend on the environment; the site of sample collection is an important factor that influences both the concentration and form
of As present in mushroom fruiting bodies. However, it is not entirely clear whether
mushrooms accumulate inorganic arsenic from the soil, or produce it through
biotransformations.

The occurrence of inorganic arsenic in food is a complex subject, because foods 423 that are usually high in arsenic, such as seafood and fish (Fontcuberta et al., 2011) or 424 425 algae (Llorente-Mirandes et al., 2010 and 2011), often have a low iAs content, whereas iAs can be the major arsenic species in other foods with a lower total arsenic content, 426 427 such as rice (Llorente-Mirandes et al., 2012) and cereal based-food (Llorente-Mirandes 428 et al., 2014). Despite the increased focus in the European Commission (EC) on iAs in 429 food commodities, no maximum levels have been set for iAs to date. However, there are ongoing discussions in the EC and CODEX Alimentarius on the potential future 430 regulation of inorganic arsenic in rice and rice-based products. A maximum level of 0.2 431 mg As kg⁻¹ has been proposed, but this has not been implemented in the legislation 432 (CODEX, 2012). On the other hand, Australia and New Zealand have established 433 different limits for iAs: 1 mg As kg⁻¹ for seaweed and molluscs and 2 mg As kg⁻¹ for 434 435 crustaceans and fish (ANFZA 2011). China has maximum limits for inorganic arsenic for different foodstuffs such as rice (MHC, 2005). According to our present results, 436 edible Shiitake products contained in all cases high percentages of toxic inorganic 437 arsenic (accounting for 84% of the total As). These iAs concentrations were higher than 438 439 those usually found in cereal-based products (Llorente-Mirandes et al., 2014), fish, 440 vegetable foods and meat (Fontcuberta et al., 2011) and similar to those of other widely consumed foods such as rice and rice products (Llorente-Mirandes et al., 2012), and in 441 some cases were even higher (up to $1.38 \text{ mg As } \text{kg}^{-1} \text{ dm}$, Table 3). Although it is true 442 that the quantity and frequency of Shiitake intake are relatively low compared to that of 443

rice or cereal-based food in the European population, it should not be ignored as a 444 potential contributor to dietary iAs exposure. Nevertheless, more data on As speciation 445 in edible Shiitake products are needed in order to accurately estimate the dietary 446 exposure to inorganic As in such populations. There is also lack of data on 447 bioaccessibility of iAs species in edible Shiitake products. The consideration of 448 bioaccessibility and arsenic speciation data into the exposure assessment can further 449 450 refine and improve the risk assessment process. In a recent study, high rates of As bioaccessibility from several mushrooms are reported (Koch et al., 2013). 451

452

453 *3.4 Cultivated Shiitake*

454 As well as dehydrated, fresh, canned and food supplements, another way to consume Shiitake is through its cultivation in commercial substrate inoculated with 455 mycelium intended to be grown at home. Therefore, to investigate the distribution of 456 arsenic compounds and the potential health risks involved in the consumption of 457 cultivated Shiitake, a preliminary cultivation study was performed. In addition, the 458 relationship between iAs contents in the substrate, tap water and in the mushroom 459 fruiting bodies was investigated to check whether iAs originates from the substrate or is 460 produced through biotransformation. 461

As mentioned earlier, Shiitake was cultivated according to the instructions supplied by the manufacturer. Tap and waste water solutions and substrate samples were analysed before and after each harvest. The first and second harvest produced a considerable number of mushrooms of different sizes. Differences in the total yield were found between harvests: 319 g and 222 g (wet mass) for the first and second harvest, respectively. The total arsenic concentrations and arsenic species in the substrate, water and mushroom samples over the two harvest periods are summarised in Table 4. Arsenite content is only reported for tap and waste water samples since in the
remaining samples As(III) was quantitatively oxidised to As(V) during the microwave
extraction procedure.

The total arsenic in the waste water samples collected after each substrate submersion was 3.5 and 4.6 μ g As L⁻¹ for the first and second harvest, respectively. Inorganic arsenic (as the sum of arsenite and arsenate) was the major compound, corresponding to 88% and 78% of the total As in the first and second, respectively. Furthermore, DMA and MA were determined as minor species in both cases, probably extracted from the mycelium and/or substrate.

Substrate samples were collected throughout the cultivation study and the total As content was 0.14, 0.12 and 0.15 mg As kg⁻¹ dm for the initial, medium and final substrate, respectively. The major arsenic compound in the three substrate samples was iAs and DMA was also quantified as a minor species. The results showed that the arsenic content of the substrate, either total or species, remained unchanged during the cultivation study.

484 In terms of fruiting bodies, the total arsenic content in the first and second harvest was 0.39 and 0.42 mg As kg⁻¹ dm respectively (Table 4), which is consistent 485 with the range obtained in the present study for all commercial edible Shiitake (0.11 to 486 1.44 mg As kg⁻¹ dm) (Table 3) and also within the range reported in the literature 487 (Maihara et al., 2008; Wuilloud et al., 2004). The arsenic concentrations of the fruiting 488 bodies did not differ significantly between the first and second harvest. The distribution 489 of arsenic species in Shiitake was similar to that of the purchased mushrooms and 490 revealed that iAs was the major As compound with a concentration of 0.33 mg As kg⁻¹ 491 dm (accounting for 85% of the total As) and 0.38 mg As kg⁻¹ dm (accounting for 90% 492

of the total As) in the first and second harvest, respectively. These results are consistent 493 with the range found in commercial edible samples (0.086 to 1.38 mg As kg^{-1} of iAs) 494 (Table 3). Other arsenic compounds were found as minor species and similar 495 distributions were found in each harvest: DMA 6.7% and 5.2%, MA 8.7% and 2.9% of 496 the total As for the first and second harvest, respectively. AB and TMAO were below 497 the LOQ and AC was below the LOD. Although MA was not found in the initial 498 499 substrate, it was detected in both mushroom samples. Furthermore, an unknown compound was found by the cationic column with a retention time of 380 s. This 500 501 unknown cationic arsenic species could be attributed to TETRA due to the matching of 502 the retention times when using the same chromatographic conditions (Kirby, Maher, 503 Ellwood, & Krikowa, 2004). However, it was not possible to check this attribution due to the lack of appropriate standards. This arsenic species was not found in any of the 504 substrate samples and is shown in Table 4 as 'Unknown cation'. 505

506 Few studies on arsenic species in cultivated mushrooms are available in the literature. Smith and co-authors cultivated Agaricus bisporus (Smith et al., 2007), which 507 was grown in compost amended with either arsenic-contaminated mine waste or an 508 509 arsenate solution. Surprisingly, AB was found in mushrooms and was absent from compost not inoculated with A. bisporus. The authors hypothesised that the biosynthesis 510 of AB was a product of fungal, not microbial, arsenic metabolism. In another study of 511 cultivated A. bisporus (Soeroes et al., 2005) the results showed that mycelia were 512 513 capable of taking up As(V) of the contaminated substrate. Arsenic speciation revealed 514 that the majority of the incorporated arsenic in the treated A. bisporus was present as 515 inorganic arsenic, highlighting the potential health risk posed by its consumption.

516 According to the present results, toxic inorganic arsenic was the main arsenic 517 species found in both the cultivated and purchased Shiitake products. However, it is not entirely clear whether Shiitake mushrooms accumulate inorganic arsenic from the substrate, or produce it through biotransformations. Therefore, more studies on the cultivation of Shiitake grown on different commercial substrates and under different cultivation conditions are needed to investigate the uptake and distribution of arsenic in mushroom fruiting bodies.

523

524 4. Conclusions

525

526 Total arsenic and arsenic species were determined in several edible Shiitake 527 products as well as in home-cultivated fruiting bodies. Arsenic speciation analysis 528 showed that inorganic arsenic was the predominant arsenic compound in all samples, accounting for 84% of the total arsenic. Moreover, other arsenic species such as DMA, 529 MA, AB, and TMAO were found as minor compounds. Despite the low intake of 530 Shiitake products in the European population, the found inorganic arsenic contents 531 could contribute to iAs exposure and therefore Shiitake products should not be ignored 532 as possible source of iAs. 533

The analytical method used may contribute to increase the availability of reliable results on inorganic arsenic in edible mushrooms. Furthermore, the present results may be useful in ongoing discussions in the European Commission and the CODEX Alimentarius for establishing and implementing future maximum levels of inorganic arsenic in food commodities.

539

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541

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549	References
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Table 1

Operating conditions of the LC-ICPMS system

ICPMS parameters

RF power	1550 W
Make up gas flow, Ar	0.32 L min ⁻¹
Carrier gas flow, Ar	0.85 L min ⁻¹
Spray chamber (type and temperature)	Scott-type and 15 °C
Sampler and skimmer cones	Nickel
Nebuliser	BURGENER Ari Mist HP
Sampling depth	8.0 mm
Cell exit	-36 V
Masses	m/z 75 (75 As), m/z 35 (35 Cl) and m/z 77 (40 Ar 37 Cl)
Collision cell	OFF
Dwell time	2.0 s (m/z 75), 0.1 s (m/z 35 and m/z 77)
QP/OctP bias difference	3 V

0		
	Anionic exchange	Cationic exchange
Column	Hamilton PRP-X100 (250 mm x 4.1 mm, 10 μm)	Zorbax 300-SCX. (250 mm x 4.6 mm, 5 μm)
Pre column	Hamilton PRP-X100. (20 x 2.0 mm i.d., 10 μm)	Zorbax 300-SCX. (12.5 mm x 4.6 i.d., 5 μm)
Mobile phase	20 mM NH ₄ H ₂ PO ₄ , pH=5.8	20 mM pyridine, pH=2.6
Flow rate	1.5 mL min ⁻¹	1.5 mL min ⁻¹
Injection volume	100 µL	50 µL
Column temperature	Room temperature 24 °C	Room temperature 24 °C
Pressure	145 bar	152 bar
Arsenic species	As(III), DMA, MA and As(V)	AB, AC and TMAO
Elution	Isocratic, 8 min	Isocratic, 9 min

Chromatographic conditions

Dofound Material	a letoT	Total			A second second					Sum of	Extraction	Column
	1 Utal AS	extracted As			Alsellic species	coles				species	eniciency (%)	(%)
			DMA	МА	iAs	AB	AC	TMAO	TMAO Unknown cation ^d			
ERM-BC211 Rice	0.256 ± 0.009	0.252 ± 0.011	0.125 ± 0.005 0.011 ± 0.001	0.011 ± 0.001	0.122 ± 0.006	COD	<lod< th=""><th>d0.1⊳</th><th><pre><tod< pre=""></tod<></pre></th><th>0.258 ± 0.012</th><th>86</th><th>102</th></lod<>	d0.1⊳	<pre><tod< pre=""></tod<></pre>	0.258 ± 0.012	86	102
Certified value	0.260 ± 0.013 ^a		0.119 ± 0.013 ^a		0.124 ± 0.011 ^a							
NIST SRM 1570a Spinach leaves	0.069 ± 0.005	0.064 ± 0.007	<lod< th=""><th>d01≻</th><th>0.059 ± 0.005</th><th><lod<< th=""><th><lod< th=""><th>d0.1></th><th><lod<< th=""><th>0.059 ± 0.005</th><th>93</th><th>92</th></lod<<></th></lod<></th></lod<<></th></lod<>	d01≻	0.059 ± 0.005	<lod<< th=""><th><lod< th=""><th>d0.1></th><th><lod<< th=""><th>0.059 ± 0.005</th><th>93</th><th>92</th></lod<<></th></lod<></th></lod<<>	<lod< th=""><th>d0.1></th><th><lod<< th=""><th>0.059 ± 0.005</th><th>93</th><th>92</th></lod<<></th></lod<>	d0.1>	<lod<< th=""><th>0.059 ± 0.005</th><th>93</th><th>92</th></lod<<>	0.059 ± 0.005	93	92
Certified value	0.068 ± 0.012 ^a				$0.054 \pm 0.012^{\circ}$							
WEPAL-IPE-120 Agaricus bisporus	0.167 ± 0.012	0.166 ± 0.021	0.047 ± 0.004	d01>	0.033 ± 0.001	0.067 ± 0.004	<lod< th=""><th>007></th><th>$\begin{array}{c} 0.009 \pm \\ 0.001 \end{array}$</th><th>$\begin{array}{c} 0.156 \pm \\ 0.010 \end{array}$</th><th>66</th><th>94</th></lod<>	007>	$\begin{array}{c} 0.009 \pm \\ 0.001 \end{array}$	$\begin{array}{c} 0.156 \pm \\ 0.010 \end{array}$	66	94
Indicative value	$0.137\pm0.067~^b$											

Table 2. Quality assessment of total arsenic and arsenic species in reference materials. Concentrations are expressed as mg As kg^{-1} dry mass (mean \pm SD, n = 3).

^a Certified value: mean \pm uncertainty.

^b Indicative value: mean \pm standard deviation.

^c Reported value for iAs according to expert laboratories in IMEP-112: mean \pm expanded uncertainty (k=2) (de la Calle et al., 2012).

^d Unknown cation arsenic species with a retention time of 380 s.

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1.20 ± 0.03 $<1.0D$ 0.006^{\pm} $<1.0D$ 0.037 99 0.31 ± 0.01 $0.067 \pm$ $0.032 \pm$ $<1.0D$ 0.018 ± 0.002 $0.511 \pm$ 98 0.31 ± 0.01 0.064 $0.033 \pm$ $<1.0D$ $<0.018 \pm 0.002$ $0.51 \pm$ 98 0.10 ± 0.01 $<1.0D$ $<1.0D$ $<1.0D$ $<1.0D$ 0.010^{\pm} 98 0.10 ± 0.01 $<1.0D$ $<1.0D$ $<1.0D$ $<1.0D$ 0.010^{\pm} 96 0.90 ± 0.04 $<1.0D$ $<1.0D$ $<1.0D$ <0.02 96 0.90 ± 0.04 $<1.0D$ $<1.0D$ $<1.0D$ 0.016^{\pm} 97 0.15 ± 0.01 $<1.0D$ $<1.0D$ $<1.0D$ $<0.016^{\pm}$ 97 0.15 ± 0.01 $<1.0D$ $<1.0D$ $<1.0D$ 0.016^{\pm} 97 0.15 ± 0.01 $<1.0D$ <0.01 0.016^{\pm} 97 0.17 ± 0.01 $<1.0D$ <0.01 0.016^{\pm} 93 0.17 ± 0.01	DMA	DMA	DMA		МА	iAs	Unknown anion ^a	AB	AC	TMAO			
0.31 ± 0.01 0.067 ± 0.003 0.003 $< LOD$ 0.018 ± 0.002 0.023 98 0.10 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.010 ± 0.00 96 0.10 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.045 96 0.90 ± 0.04 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.045 96 0.38 ± 0.08 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.010 ± 0.034 97 0.15 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.016 ± 0.00 96 0.17 ± 0.01 $< LOD$ $< LOD$ $< LOD$ 0.017 ± 0.01 93 0.17 ± 0.01 $< LOD$ $< LOD$ 0.017 ± 0.01 90 90 0.007 ± 0.02 $< LOD$ $< COD$ 0.017 ± 0.01 93 93 93	$\begin{array}{ccc} 1.42 \pm 0.06 & 1.41 \pm 0.07 & \begin{array}{c} 0.070 \pm \\ 0.004 \end{array}$	1.41 ± 0.07	0.070 ± 0.004		0.025 ± 0.002	1.20 ± 0.03	COD>	$\begin{array}{c} 0.006 \pm \\ 0.001 \end{array}$	<lod <<="" td=""><td>¢100</td><td>$\begin{array}{c} 1.30 \pm \\ 0.037 \end{array}$</td><td>66</td><td>92</td></lod>	¢100	$\begin{array}{c} 1.30 \pm \\ 0.037 \end{array}$	66	92
0.10 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.010 ± 0.04 96 0.90 ± 0.04 $< LOD$ $< LOQ$ $< LOD$ $< LOD$ $< LOD$ 0.045 99 0.90 ± 0.04 $< LOD$ $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.045 99 1.38 ± 0.08 $< LOD$ $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.045 99 0.15 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.014 91 0.17 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< OO$ 91 91 0.355 ± 0.02 $< LOD$ $< LOD$ $< LOD$ $< OO$ 91 91 0.355 ± 0.02 $< LOD$ $< OO$ $< OO$ 0.017 ± 0.01 91 91	$\begin{array}{cccc} 0.58\pm 0.02 & 0.57\pm 0.03 & 0.070\pm \\ 0.003 & 0.003 \end{array}$	$\begin{array}{c} 0.57\pm 0.03 & 0.070\pm \\ 0.003 & 0.003 \end{array}$		_	0.009 ± 0.001	0.31 ± 0.01	$\begin{array}{c} 0.067 \pm \\ 0.004 \end{array}$	0.032 ± 0.003	<pre><tod< pre=""></tod<></pre>	0.018 ± 0.002	0.51 ± 0.023	98	90
0.90 ± 0.04 $< LOD$ $< LOD$ $< LOD$ $< LOD$ $< UOD$ 0.045 99 1.38 ± 0.08 $< LOD$ $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.084 97 1.38 ± 0.08 $< LOD$ $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.084 97 0.15 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.010 99 0.15 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.014 93 0.17 ± 0.01 $< LOD$ $< LOD$ $< LOD$ $< LOD$ 0.0074 $< IOD$ 0.010 93 0.35 ± 0.02 $< LOD$ $< LOD$ $< IOD$ 0.0074 $< IOD$ 0.010 93	0.11 ± 0.02 0.11 ± 0.01 <loq< th=""><td>0.11 ± 0.01 <loq< td=""><td></td><td>v</td><td><pre><tod< pre=""></tod<></pre></td><td>0.10 ± 0.01</td><td><lod< td=""><td><lod< td=""><td>Q01></td><td><lod< td=""><td>$\begin{array}{c} 0.10 \pm \\ 0.010 \end{array}$</td><td>96</td><td>95</td></lod<></td></lod<></td></lod<></td></loq<></td></loq<>	0.11 ± 0.01 <loq< td=""><td></td><td>v</td><td><pre><tod< pre=""></tod<></pre></td><td>0.10 ± 0.01</td><td><lod< td=""><td><lod< td=""><td>Q01></td><td><lod< td=""><td>$\begin{array}{c} 0.10 \pm \\ 0.010 \end{array}$</td><td>96</td><td>95</td></lod<></td></lod<></td></lod<></td></loq<>		v	<pre><tod< pre=""></tod<></pre>	0.10 ± 0.01	<lod< td=""><td><lod< td=""><td>Q01></td><td><lod< td=""><td>$\begin{array}{c} 0.10 \pm \\ 0.010 \end{array}$</td><td>96</td><td>95</td></lod<></td></lod<></td></lod<>	<lod< td=""><td>Q01></td><td><lod< td=""><td>$\begin{array}{c} 0.10 \pm \\ 0.010 \end{array}$</td><td>96</td><td>95</td></lod<></td></lod<>	Q01>	<lod< td=""><td>$\begin{array}{c} 0.10 \pm \\ 0.010 \end{array}$</td><td>96</td><td>95</td></lod<>	$\begin{array}{c} 0.10 \pm \\ 0.010 \end{array}$	96	95
1.38 ± 0.08 $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ 0.084 97 0.15 ± 0.01 $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ 0.015 ± 0.01 99 0.15 ± 0.01 $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ 0.013 ± 0.9 99 0.58 ± 0.01 $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ $< \text{LOD}$ 0.633 ± 0.9 93 0.17 ± 0.01 $< \text{LOD}$ $< \text{LOQ}$ $< \text{LOQ}$ 0.017 ± 0.01 0.033 ± 0.02 93 0.355 ± 0.02 $< \text{LOD}$ $< \text{LOQ}$ $< \text{LOQ}$ 0.077 ± 0.01 0.0072 ± 0.02 0.0072 ± 0.02 0.0072 ± 0.02 0.007 ± 0.02 $0.0010 \pm 0.038 \pm 0.02$	0.93 ± 0.01 0.93 ± 0.02 0.025 ± 0.0 0.001 0	$\begin{array}{c} 0.93 \pm 0.02 & 0.025 \pm \\ 0.001 & 0.001 \end{array}$		0.0	0.021 ± 0.004	0.90 ± 0.04	<lod< td=""><td>≥200</td><td><pre><tod< pre=""></tod<></pre></td><td><lod< td=""><td>0.95 ± 0.045</td><td>66</td><td>102</td></lod<></td></lod<>	≥200	<pre><tod< pre=""></tod<></pre>	<lod< td=""><td>0.95 ± 0.045</td><td>66</td><td>102</td></lod<>	0.95 ± 0.045	66	102
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1.44 ± 0.04 1.40 ± 0.11 <loq <math="">\begin{array}{c} 0.0\\ 0.\end{array}</loq>	1.40 ± 0.11 <loq< td=""><td></td><td>0.0 0.</td><td>$\begin{array}{c} 0.041 \pm \\ 0.004 \end{array}$</td><td>$1.38 \pm 0.08$</td><td></td><td><pre></pre></td><td><lod< td=""><td>001></td><td>1.42 ± 0.084</td><td>67</td><td>102</td></lod<></td></loq<>		0.0 0.	$\begin{array}{c} 0.041 \pm \\ 0.004 \end{array}$	1.38 ± 0.08		<pre></pre>	<lod< td=""><td>001></td><td>1.42 ± 0.084</td><td>67</td><td>102</td></lod<>	001>	1.42 ± 0.084	67	102
$\begin{array}{cccccccccccccccccccccccccccccccccccc$											0 15 +		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.15 ± 0.01 0.15 ± 0.01 <loq <loq<="" th=""><td>0.15 ± 0.01 <loq< td=""><td></td><td>√L(</td><td>ğ</td><td>0.15 ± 0.01</td><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>0.010 ± 0.010</td><td>66</td><td>66</td></lod<></td></lod<></td></lod<></td></lod<></td></loq<></td></loq>	0.15 ± 0.01 <loq< td=""><td></td><td>√L(</td><td>ğ</td><td>0.15 ± 0.01</td><td><lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>0.010 ± 0.010</td><td>66</td><td>66</td></lod<></td></lod<></td></lod<></td></lod<></td></loq<>		√L(ğ	0.15 ± 0.01	<lod< td=""><td><lod< td=""><td><lod< td=""><td><lod< td=""><td>0.010 ± 0.010</td><td>66</td><td>66</td></lod<></td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>0.010 ± 0.010</td><td>66</td><td>66</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>0.010 ± 0.010</td><td>66</td><td>66</td></lod<></td></lod<>	<lod< td=""><td>0.010 ± 0.010</td><td>66</td><td>66</td></lod<>	0.010 ± 0.010	66	66
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.66 ± 0.07 0.62 ± 0.01 <loq <math="">0.0</loq>	0.62 ± 0.01 <loq< td=""><td></td><td>0.0</td><td>$\begin{array}{c} 0.050 \pm \\ 0.004 \end{array}$</td><td>$0.58\pm0.01$</td><td><lod< td=""><td><pre>>COQ</pre></td><td>∂01></td><td><lod< td=""><td>0.63 ± 0.014</td><td>93</td><td>102</td></lod<></td></lod<></td></loq<>		0.0	$\begin{array}{c} 0.050 \pm \\ 0.004 \end{array}$	0.58 ± 0.01	<lod< td=""><td><pre>>COQ</pre></td><td>∂01></td><td><lod< td=""><td>0.63 ± 0.014</td><td>93</td><td>102</td></lod<></td></lod<>	<pre>>COQ</pre>	∂01>	<lod< td=""><td>0.63 ± 0.014</td><td>93</td><td>102</td></lod<>	0.63 ± 0.014	93	102
0.35 ± 0.02 <lod <math="">0.007 \pm <lod <lod="" <math="">0.38 \pm 99</lod></lod>	0.17 ± 0.02 0.17 ± 0.01 <loq <l<="" th=""><td>0.17 ± 0.01 <loq< td=""><td></td><td>$\overline{\nabla}$</td><td><pre></pre>LOD</td><td>0.17 ± 0.01</td><td><lod< td=""><td><lod< td=""><td><pre>>CLOQ</pre></td><td>001></td><td>$\begin{array}{c} 0.17 \pm \\ 0.010 \end{array}$</td><td>103</td><td>96</td></lod<></td></lod<></td></loq<></td></loq>	0.17 ± 0.01 <loq< td=""><td></td><td>$\overline{\nabla}$</td><td><pre></pre>LOD</td><td>0.17 ± 0.01</td><td><lod< td=""><td><lod< td=""><td><pre>>CLOQ</pre></td><td>001></td><td>$\begin{array}{c} 0.17 \pm \\ 0.010 \end{array}$</td><td>103</td><td>96</td></lod<></td></lod<></td></loq<>		$\overline{\nabla}$	<pre></pre> LOD	0.17 ± 0.01	<lod< td=""><td><lod< td=""><td><pre>>CLOQ</pre></td><td>001></td><td>$\begin{array}{c} 0.17 \pm \\ 0.010 \end{array}$</td><td>103</td><td>96</td></lod<></td></lod<>	<lod< td=""><td><pre>>CLOQ</pre></td><td>001></td><td>$\begin{array}{c} 0.17 \pm \\ 0.010 \end{array}$</td><td>103</td><td>96</td></lod<>	<pre>>CLOQ</pre>	001>	$\begin{array}{c} 0.17 \pm \\ 0.010 \end{array}$	103	96
	0.45 ± 0.01 0.44 ± 0.02 0.012 ± 0.0	0.44 ± 0.02 0.012 ± 0.02		0.0	$0.012 \pm$	0.35 ± 0.02	<lod></lod>	0.007 ±	<lod<< td=""><td><lod< td=""><td>$0.38 \pm$</td><td>66</td><td>87</td></lod<></td></lod<<>	<lod< td=""><td>$0.38 \pm$</td><td>66</td><td>87</td></lod<>	$0.38 \pm$	66	87

0.12 ± 0.01 0.	0.12 ± 0.01	0.033 ± 0.001	<pre><pre>COD</pre></pre>	0.086 ± 0.011		<lod<< th=""><th><pre>clob</pre></th><th><pre>d0.1></pre></th><th>0.12 ± 0.012</th><th>66</th><th>101</th></lod<<>	<pre>clob</pre>	<pre>d0.1></pre>	0.12 ± 0.012	66	101
0.14 ± 0.01		$\begin{array}{c} 0.009 \pm \\ 0.001 \end{array}$	$\begin{array}{c} 0.006 \pm \\ 0.001 \end{array}$	0.12 ± 0.01	<pre></pre>	d01>	<lod< td=""><td><pre><tod< pre=""></tod<></pre></td><td>0.14 ± 0.012</td><td>100</td><td>96</td></lod<>	<pre><tod< pre=""></tod<></pre>	0.14 ± 0.012	100	96
0.25 ± 0.02 0	0	0.015 ± 0.001	0.010 ± 0.001	0.20 ± 0.01	<pre></pre>	<lod< td=""><td><lod< td=""><td><lod< td=""><td>$\begin{array}{c} 0.23 \pm \\ 0.012 \end{array}$</td><td>94</td><td>92</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td>$\begin{array}{c} 0.23 \pm \\ 0.012 \end{array}$</td><td>94</td><td>92</td></lod<></td></lod<>	<lod< td=""><td>$\begin{array}{c} 0.23 \pm \\ 0.012 \end{array}$</td><td>94</td><td>92</td></lod<>	$\begin{array}{c} 0.23 \pm \\ 0.012 \end{array}$	94	92
0.28 ± 0.01 0.0	0.0	0.020 ± 0.001	0.014 ± 0.001	0.22 ± 0.01	<pre></pre>	<lod< td=""><td><lod< td=""><td><pre>d0l></pre></td><td>0.25 ± 0.012</td><td>96</td><td>89</td></lod<></td></lod<>	<lod< td=""><td><pre>d0l></pre></td><td>0.25 ± 0.012</td><td>96</td><td>89</td></lod<>	<pre>d0l></pre>	0.25 ± 0.012	96	89
$0.33 \pm 0.05 \qquad \begin{array}{c} 0.0\\ 0\end{array}$	0.0	0.022 ± 0.001	$\begin{array}{c} 0.014 \pm \\ 0.001 \end{array}$	0.28 ± 0.02	<lod< td=""><td><lod< td=""><td><lod< td=""><td><pre>d01></pre></td><td>0.32 ± 0.012</td><td>66</td><td>96</td></lod<></td></lod<></td></lod<>	<lod< td=""><td><lod< td=""><td><pre>d01></pre></td><td>0.32 ± 0.012</td><td>66</td><td>96</td></lod<></td></lod<>	<lod< td=""><td><pre>d01></pre></td><td>0.32 ± 0.012</td><td>66</td><td>96</td></lod<>	<pre>d01></pre>	0.32 ± 0.012	66	96

^a Unknown anion arsenic species with a retention time of 255 s.

ntrations are expressed as mg As kg ⁻¹ dry mass (mean	an \pm SD, n = 3).
JCe	water (mean \pm SD,
4. Total arsenic and arsenic species in cultivated Shiitake, substrate samples, and tap and waste water. Cot	\pm SD, n = 3) for Shiitake and substrate samples. Concentrations are expressed as μg As L ⁻¹ for tap and waste wa
Table 4	\pm SD

Harvest	Sample	Total As	Total extracted As			V	Arsenic species					Sum of As species	Extraction efficiency (%)	Column recovery (%)
				As (III)	DMA	МА	As (V)	AB	AC	TMAO	Unknown cation ^b			
First	Mushroom-1	0.39 ± 0.02	0.38 ± 0.02	·	0.026 ± 0.002	0.034 ± 0.002	0.33 ± 0.01	ð07≻	<lod< th=""><th>do1></th><th>$\begin{array}{c} 0.014 \pm \\ 0.001 \end{array}$</th><th>$\begin{array}{c} 0.40 \pm \\ 0.015 \end{array}$</th><th>66</th><th>105</th></lod<>	do1>	$\begin{array}{c} 0.014 \pm \\ 0.001 \end{array}$	$\begin{array}{c} 0.40 \pm \\ 0.015 \end{array}$	66	105
	Original substrate	0.14 ± 0.01	0.13 ± 0.01		0.004 ± 0.001	∂01>	0.12 ± 0.02	<pre></pre>	<pre></pre>	<pre></pre>	<lod< th=""><th>0.12 ± 0.021</th><th>92</th><th>93</th></lod<>	0.12 ± 0.021	92	93
	Medium substrate	0.12 ± 0.02	0.12 ± 0.01		0.005 ± 0.001	∂01>	0.11 ± 0.01	<pre>d0l></pre>	<pre><tod< pre=""></tod<></pre>	<pre></pre>	<pre></pre>	0.12 ± 0.011	98	96
	Tap water-1	0.85 ± 0.04	n.a ^a		<tod< th=""><th><lod< th=""><th>0.82 ± 0.05</th><th><pre></pre></th><th><pre></pre></th><th><pre></pre></th><th><lod< th=""><th>0.82 ± 0.050</th><th>ı</th><th>ı</th></lod<></th></lod<></th></tod<>	<lod< th=""><th>0.82 ± 0.05</th><th><pre></pre></th><th><pre></pre></th><th><pre></pre></th><th><lod< th=""><th>0.82 ± 0.050</th><th>ı</th><th>ı</th></lod<></th></lod<>	0.82 ± 0.05	<pre></pre>	<pre></pre>	<pre></pre>	<lod< th=""><th>0.82 ± 0.050</th><th>ı</th><th>ı</th></lod<>	0.82 ± 0.050	ı	ı
	Waste water-1	3.5 ± 0.30	n.a ^a	1.06 ± 0.09	0.17 ± 0.01	0.19 ± 0.02	2.03 ± 0.15	<pre></pre>	<tod< th=""><th><pre></pre></th><th><pre><pod< pre=""></pod<></pre></th><th>3.44 ± 0.27</th><th>ı</th><th>I</th></tod<>	<pre></pre>	<pre><pod< pre=""></pod<></pre>	3.44 ± 0.27	ı	I
Second	Mushroom-2	0.42 ± 0.03	0.42 ± 0.01	·	0.022 ± 0.001	0.012 ± 0.001	0.38 ± 0.02	∂01>	<pre><tod< pre=""></tod<></pre>	<pre>>COQ</pre>	0.013 ± 0.002	0.43 ± 0.024	66	102
	Final substrate	0.15 ± 0.01	0.15 ± 0.02	·	0.007 ± 0.001	<lod< th=""><th>0.13 ± 0.01</th><th><pre>dOl></pre></th><th><pre></pre></th><th><pre></pre></th><th><lod< th=""><th>0.14 ± 0.011</th><th>97</th><th>91</th></lod<></th></lod<>	0.13 ± 0.01	<pre>dOl></pre>	<pre></pre>	<pre></pre>	<lod< th=""><th>0.14 ± 0.011</th><th>97</th><th>91</th></lod<>	0.14 ± 0.011	97	91
	Tap water-2	0.86 ± 0.03	n.a ^a	0.79 ± 0.04	<tod< th=""><th><lod< th=""><th><pre><pod< pre=""></pod<></pre></th><th><pre>dOl></pre></th><th><pre><tod< pre=""></tod<></pre></th><th><pre></pre></th><th><pre></pre></th><th>0.79 ± 0.040</th><th>ı</th><th>ı</th></lod<></th></tod<>	<lod< th=""><th><pre><pod< pre=""></pod<></pre></th><th><pre>dOl></pre></th><th><pre><tod< pre=""></tod<></pre></th><th><pre></pre></th><th><pre></pre></th><th>0.79 ± 0.040</th><th>ı</th><th>ı</th></lod<>	<pre><pod< pre=""></pod<></pre>	<pre>dOl></pre>	<pre><tod< pre=""></tod<></pre>	<pre></pre>	<pre></pre>	0.79 ± 0.040	ı	ı
	Waste water-2	4.6 ± 0.60	n.a ^a	0.73 ± 0.04	0.31 ± 0.03	0.24 ± 0.02	2.85 ± 0.20	<pre>d01></pre>	<pre></pre>	<pre></pre>	<lod< th=""><th>$\begin{array}{c} 4.12 \pm \\ 0.29 \end{array}$</th><th>ı</th><th>ı</th></lod<>	$\begin{array}{c} 4.12 \pm \\ 0.29 \end{array}$	ı	ı

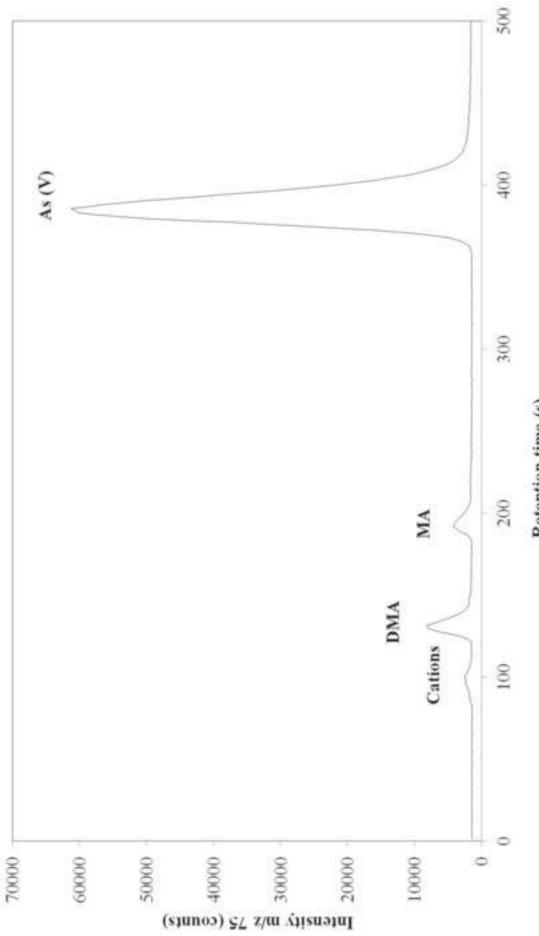
^a Not analysed, these samples were not extracted.

^b Unknown cation arsenic species with a retention time of 380 s.

Figure captions

Figure 1. Chromatogram from anion exchange by LC–ICPMS of fresh Shiitake extract.





Retention time (s)