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2 Dental casting alloys behaviour during power toothbrushing 3 with toothpastes with various abrasivities. Part I: wear behavior

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8 **Abstract** The purpose of this study was to evaluate the
9 long term effect of abrasivity of toothpastes normally used
10 over the surface and mechanical properties of dental casting
11 alloys. Three dental casting alloys (Ni–Cr, Co–Cr, c.p. Ti)
12 and one ceramic were chosen. Four specimens of each
13 material were immersed in artificial saliva, brushed without
14 or with one of four toothpastes of different Relative Dentine
15 Abrasivity (RDA 50, 52, 80, and 114). An electric tooth-
16 brush with a load of 250 g was used for 420 min. Mass loss
17 was determined by difference in weight, microhardness and
18 surface roughness were also measured. Two-way ANOVA
19 and non-parametric tests were used to detect significant
20 differences. Titanium specimens ($478 \mu\text{g}/\text{cm}^2$) exhibited
21 the most mass loss, whereas ceramic ($282 \mu\text{g}/\text{cm}^2$) and
22 Co–Cr ($262 \mu\text{g}/\text{cm}^2$) exhibited the least. However, ceramic
23 demonstrated the most volume loss (0.239 mm^3). The
24 abrasivity effect of the toothpaste correlated with the RDA
25 values. Slight variations in microhardness were observed
26 after toothbrushing and depended on the material but not on
27 the toothpaste used. Material surfaces were slightly
28 smoothed by toothbrushing but no significant differences

were detected. Dental casting alloys and ceramic are sus- 29
ceptible to abrasion by brushing with an electric toothbrush 30
depending on the RDA value of the toothpaste. Variations in 31
microhardness and surface roughness were not clinically 32
relevant. 33
34
35

1 Introduction 36

Brushing with toothpaste is the most common form of 37
tooth cleaning. Although manual toothbrushes are still used 38
by the majority of people, powered toothbrushes are 39
becoming increasingly popular [1]. In a comparative study 40
of three power toothbrush systems the authors concluded 41
that the action of the oscillating/rotating/pulsating tooth- 42
brushes was more effective in plaque removal than the 43
high-frequency toothbrush [2]. Moreover, a systematic 44
review of the literature concluded that powered tooth- 45
brushes with a rotation oscillation action achieve a modest 46
reduction in plaque and gingivitis compared to manual 47
toothbrushing. However, few data have been reported 48
concerning the side effects of their use [1]. To achieve 49
optimum toothbrushing, it is necessary to add organic 50
solvents and abrasive substances to the toothpaste. The 51
abrasivity of such substances should never be so high that 52
unintended damage is produced in the oral hard or soft 53
tissues, or in dental restorations. Abrasion induced by an 54
abrasive agent on a surface is influenced by a great variety 55
of the agent's properties, such as chemical composition, 56
crystal structure, cleavage, friability, hardness, particle 57
shape, surface features, and particle size distribution, sol- 58
ubility, concentration, and compatibility with other 59
ingredients of the toothpaste. The abrasivity of toothpastes 60
can be measured by a variety of methods in vitro, of which 61

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62 the Relative Dentine Abrasion (RDA) method [3] is the
63 best known, recommended by the International Standards
64 Organization [4].

65 Abrasion from brushing has been studied on enamel [5, 6],
66 dentin [5, 7, 8] composite [9] acrylic resin [8, 10–12],
67 ceramic [13], glass ionomer [14] and amalgam [14]. How-
68 ever, only a few in vitro studies have focused on the effects
69 of brushing on dental alloys [15–17]. Whereas Ni-based
70 (Ni–Cr and Ni–Cr–Be) alloys showed relatively high
71 amounts (600–800 $\mu\text{g}/\text{cm}^2$ of Ni) of elemental release after
72 brushing with toothpaste for 48 h [16], only Ni–Cr–Be alloy
73 showed a large increase in cytotoxicity under the same
74 conditions [17].

75 There are currently hundreds of alloys available for
76 prosthodontic restorations. The major factors controlling
77 alloy selection are cost, physical properties, casting tech-
78 niques, corrosion and biocompatibility [18–21]. For
79 clinical dental applications, Ni–Cr and Co–Cr alloys have
80 been developed as a cheaper alternative to gold-and pal-
81 ladium-based alloys, also featuring mechanical properties
82 [20, 21]. Titanium is a relatively new metal in cast dental
83 prostheses [22]. Whereas titanium has several advantages
84 compared to Ni–Cr and Co–Cr, including lower weight,
85 biocompatibility and low heat conductivity, it also suffers
86 from a complex casting technique, requiring high temper-
87 ature (1650°C), a special magnesium investment, and an
88 argon arc under vacuum [20, 22].

89 Brushing with toothpaste may lead to changes in the
90 surface morphology of both natural tooth and restorative
91 materials. These changes may alter plaque retention and
92 corrosion potential. The aim of this study was to evaluate
93 the long term effect of abrasivity of toothpastes normally
94 used over the surface and the mechanical properties of
95 Ni–Cr, Co–Cr, and commercially pure titanium (cpTi) after
96 power toothbrushing. In order to achieve it an in vitro
97 study was developed using rotating oscillation toothbrushes
98 and different toothpastes in artificial saliva.

99 2 Materials and methods

100 Three types of casting alloys used for full cast and porce-
101 lain-fused-to-metal restorations were selected. These were
102 Ni–Cr alloy (IPS d.SIGN 15, Ivoclar Vivadent AG, Schaan,
103 Liechtenstein) (composition % by weight; Ni:58.7, Cr:12.1,
104 Si:1.9, Mo:1.7, Fe:<1, Co:<1, Ce:<1), Co–Cr alloy (IPS
105 d.SIGN 30, Ivoclar Vivadent Inc, Amherst, NY, USA)
106 (composition % by weight; Co:60.2, Cr:30.1%, Ga:3.9,
107 Nb:3.2, B:<0.1, Fe:<0.1, Al:<0.1, Li:<0.1) and cpTi grade
108 II (Tritan[®], Dentaaurum, Ispringen, Germany) (Composition
109 % by weight; Ti:>99.5%). One ceramic material (Vintage
110 Halo Porcelain, Shofu, San Marcos, CA, USA) was chosen
111 as a control because of its abrasion resistance. 120 wax

112 patterns (18 × 18 mm and 1.5 mm thick) were cast. Eighty
113 of them were invested in phosphate-bonded invest-
114 ment (Ceramvest, Protechno, Vilamalla, Spain) to obtain
115 40 specimens of Ni–Cr and 40 of Co–Cr. For the remaining
116 40 wax patterns, a magnesia-based investment material
117 (Trinell[®], Dentaaurum, Ispringen, Germany) was used to
118 obtain the cpTi specimens by means of an appropriate
119 casting machine (Rematitan Autocast, Dentaaurum, Isprin-
120 gen, Germany). Investment heating and alloy casting
121 procedures were performed according to the manufacture's
122 instructions. All alloys were fired at 950°C to simulate
123 firing in a porcelain oven. Forty specimens of porcelain
124 were produced by means of a mold (18 × 18 mm and
125 1.5 mm thick) and fired in a high-temperature porcelain
126 oven. After casting, the alloy specimens and the porcelain
127 were polished following standard laboratory procedures
128 and then cleaned.

129 Before brushing, all specimens were weighed and their
130 surface roughness was measured using a profilometer as
131 described later. Forty specimens of each restorative mate-
132 rial were randomly distributed amongst five groups
133 ($n = 8$). Four specimens from each group were brushed
134 either without paste or with one of the following four
135 commercial toothpastes: (0) artificial saliva (Hank's
136 Balanced Salt Solution, Sigma-Aldrich, UK) without
137 toothpaste as a control, (1) Paste-50 (Sorbitol, aqua,
138 hydrated silica, PEG-12, aroma, tetrasodium pyrophos-
139 phate, cellulose gum, sodium lauryl sulfate, sodium
140 saccharin, sodium fluoride, mica, glycerin, CI 42090, CI
141 77891, RDA = 50), (2) Paste-52 (sodium fluoride, dihy-
142 drated dicalcium phosphate, glycerin, aqua, alumina,
143 sorbitol, sodium citrate, cocamidopropyl betaine, papaine,
144 sodium lauryl sulfate, carrageenan, sodium saccharin, CI
145 42090, citroxain[®], RDA = 52), (3) Paste-80 (Aqua,
146 hydrated silica, glicerin, sorbitol, PVM/MA copolymer,
147 sodium lauryl sulfate, aroma, cellulose gum, sodium
148 hydroxide, sodium fluoride, triclosan, carrageenan, sodium
149 saccharin, CI 77891, RDA = 80), and (4) Paste-114 (Aqua,
150 hydrated silica, sorbitol, glycerin, PEG-12, tetrapotassium
151 pyrophosphate, PVM/MA copolymer, aroma, sodium lau-
152 ryl sulfate, titanium dioxide, cellulose gum, carrageenan,
153 sodium fluoride, sodium hydroxide, sodium saccharin,
154 RDA = 114). The authors chose these toothpastes as they
155 are widely used and represent a wide range of RDA values.
156 The four remaining specimens from each group were
157 merely immersed in toothpaste or artificial saliva without
158 any brushing.

159 An electric toothbrush (Braun Oral-B ProfessionalCare
160 7500DLX, Braun AG, Kronberg, Germany) was chosen
161 because it is one of recent models of power toothbrushes
162 which has a pulsating action added to an oscillating/rotat-
163 ing action, and in order to realistically imitate real brushing
164 of the restorative materials. This electric toothbrush effects

165 an in-and-out movement at 40000 pulsations per minute,
 166 with simultaneous oscillations at 8800 per minute. The
 167 brush heads used were Green Oral-B® (FlexiSoft® bristles).
 168 Batteries from each of four units of this electric toothbrush
 169 were removed and a power supply with direct current was
 170 connected to obtain a voltage of 5 V. Since brushing engines
 171 need 1.2 V to work, a bridge of five diodes was installed at
 172 the power supply output and the four electric toothbrushes
 173 were connected in parallel just beyond the diode bridge
 174 (Fig. 1).

175 Specimens were fixed into a round-shaped plastic recipi-
 176 ent and were totally immersed in abrasive slurry containing
 177 10 g of toothpaste and 70 ml of artificial saliva (pH 7.0–7.4),
 178 except for the group without toothpaste which was immersed
 179 only in artificial saliva. For the specimens to be brushed, a
 180 weight was added to the base upon which each electric
 181 toothbrush was fixed in order to give the specimen a vertical
 182 load of 250 g (± 3 g) during brushing. Each experiment was
 183 allowed to run for 7 h at room temperature ($20 \pm 3^\circ\text{C}$).
 184 A new toothbrush and fresh slurry were used for each specimen.

185 After brushing, each sample was rinsed with distilled
 186 water and air dried for 10 s. Specimens were weighed
 187 using an electronic balance (Sartorius BP211D, Sartorius
 188 AG, Goettingen, Germany, accuracy of 0.01 mg) at the
 189 beginning of the experiment and after 70, 140, 280 and
 190 420 min of brushing/immersion. The area of brushing was
 191 calculated soaking the bristles of a toothbrush with ink
 192 under the experimental situation by means of an image
 193 analyzer and the result was 1.936 cm^2 . The material loss
 194 per unit area was determined by dividing the difference in
 195 weight before and after brushing/immersion by 1.936 cm^2
 196 at each time. The volume loss was calculated as the weight
 197 loss divided by the density of each material.



Fig. 1 The materials tested immersed in toothpaste slurry and placed into four plastic recipients are submitted to brushing by four units of electric toothbrush which are fixed to platforms subjected to a 250 g load

198 The surface roughness measurements were made using a
 199 profilometer (Mitutoyo SurfTest SV-512; Mitutoyo, IL,
 200 USA) with a 5 nm resolution and assisted with appropriate
 201 software (Surfpack, v 3.0, Mitutoyo, Japan) on the speci-
 202 mens before and after 420 min of brushing. Before
 203 evaluation, a Gaussian filter was used to remove errors of
 204 form and waviness. For each specimen five different lengths
 205 (sampling length 0.8 mm, and transversing length 2.5 mm)
 206 were analyzed following the ISO/JIS B0601. Ra was used to
 207 give a numerical characterization of the surface roughness.
 208 Ra is the arithmetical mean deviation of the profile and is
 209 calculated as the arithmetical mean of the absolute values of
 210 the profile deviations from the mean line. One centrally
 211 positioned surface area was analyzed and care was taken to
 212 relocate the same area during subsequent registrations. The
 213 microhardness of the materials was measured using a
 214 microhardness tester (Matsuzawa DMH1) at a load of 500 g
 215 for 15 s, before brushing and after 420 min of brushing.
 216 Five different areas for each specimen were chosen to make
 217 indentations and were averaged. Care was taken to ensure
 218 that testing was performed on a flat surface.

219 Specimens of each material were examined in a scanning
 220 electron microscope (SEM) (JEOL®JSM-6400, Jeol USA
 221 Inc, Peabody, Ma, USA) before and after brushing with
 222 different toothpastes, at a constant working distance of
 223 10 mm, with magnifications $500\times$, $1000\times$, or $2000\times$.
 224 Ceramic specimens were gold-sputtered (Balzers SCD 004,
 225 USA) for 40 s with a 10 nm layer and then viewed under
 226 vacuum in a SEM. Alloy specimens were not coated with any
 227 material due to their metallic nature. Toothpastes were
 228 examined by means of environmental scanning electron
 229 microscopy (ESEM) (JEOL 2020 electroscan®, Jeol USA
 230 Inc, Peabody, Ma, USA) with magnifications of up to $1000\times$.

231 Statistical analyses was performed using SPSS 12.0S for
 232 Windows (SPSS Inc, Chicago, IL, USA). Two-way ANOVA
 233 and *post hoc* Tukey tests were used to examine the effects of
 234 toothpaste and restorative material, as well as any interaction
 235 effects, on mass loss and microhardness. The amount of
 236 variation explained by the model was determined by the
 237 adjusted R^2 values. Volume loss and surface roughness data
 238 were analyzed using the Mann-Whitney *U* and Wilcoxon
 239 signed-rank tests. Statistical significance was set in advance
 240 at the 0.05 probability level.

3 Results

242 None of the specimens that were simply immersed in
 243 toothpaste, without brushing, exhibited mass loss. Mass
 244 loss curves for materials brushed are shown in Fig. 2. The
 245 rate of mass loss decreased as a function of time. Mass loss
 246 per unit area values for all materials and for all toothpastes
 247 after 420 min of brushing are given in Table 1. A two-way

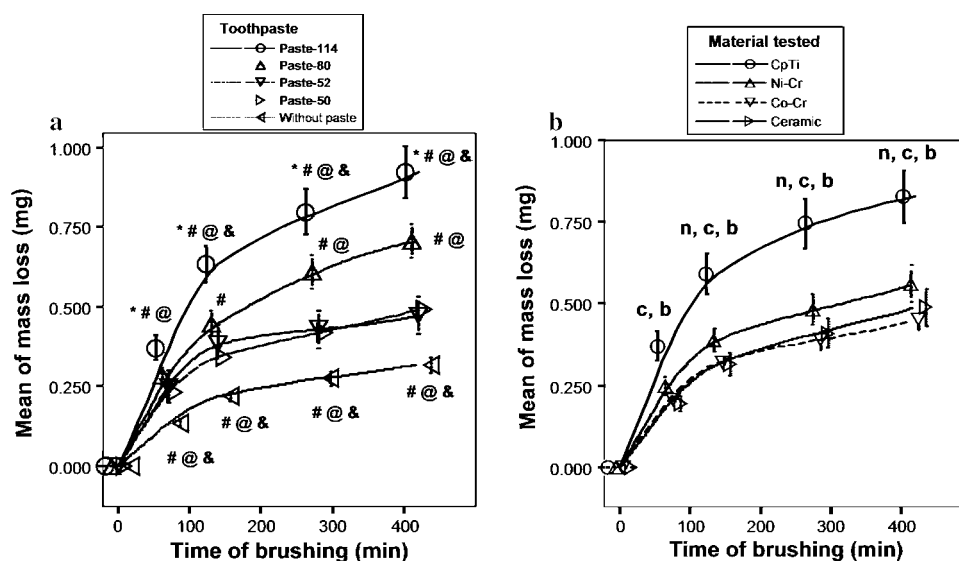


Fig. 2 (a) Mean of mass loss (mg) of the four materials tested produced by brushing without toothpaste and with the four different toothpastes over 420 min. (b) Mean of mass loss (mg) for each material tested produced by brushing over 420 min regardless of the toothpaste used. Error bars represent Standard Error of Means. Statistical

significant differences are indicated by: * $P < 0.05$ with respect to “without paste”; # $P < 0.05$ with respect to Paste-50; @ $P < 0.05$ with respect to Paste-52; & $P < 0.05$ with respect to Paste-80; ^c $P < 0.05$ with respect to ceramic; ^b $P < 0.05$ with respect to Co-Cr; ⁿ $P < 0.05$ with respect to Ni-Cr. Analyzed by Mann-Whitney U test

248 ANOVA revealed that both material and the type of
249 toothpaste had significant effects on mass loss ($R_a^2 = 0.85$;
250 $P < 0.001$). Titanium suffered the most abrasion whereas
251 Co-Cr and ceramic suffered the least. Further, the higher
252 the RDA of the toothpaste produced the higher the mass
253 loss in the materials tested. The ANOVA also revealed a
254 significant interaction between toothpaste and material
255 ($P < 0.001$). Whereas abrasion of titanium, Ni-Cr and
256 ceramic depended significantly on the RDA of the tooth-
257 paste used, Co-Cr exhibited a similar mass loss for all the

toothpastes tests. In contrast, Mann-Whitney U and
Wilcoxon signed-rank tests revealed that ceramic lost more
volume ($P < 0.001$) that Co-Cr or Ni-Cr after 420 min of
brushing (Table 2).

A slight but no significant variation in surface roughness
was observed after brushing with or without toothpaste for
all restorative materials (Table 3). Vickers hardness values
are presented in Table 4. The hardness of Ni-Cr alloy is
significantly ($P < 0.01$) lower than that of the other
materials tested, whilst Co-Cr exhibited the highest

Table 1 Means (SD) of mass per unit area ($\mu\text{g}/\text{cm}^2$) after 420 min as a result of brushing with or without toothpaste for each restorative material

	Without paste	Paste-50 Silica	Paste-52 alumina	Paste-80 Silica	Paste-114 Silica + TiO ₂	Mean toothpaste*
Ceramic	136 (21)	213 (37)	136 (62)	359 (36)	418 (75)	282 (126) ^{x,y}
Co-Cr	112 (14)	245 (58)	269 (83)	252 (41)	284 (32)	262 (53) ^x
Ni-Cr	186 (18)	241 (51)	191 (89)	332 (42)	498 (37)	316 (132) ^y
cpTi	223 (38)	314 (75)	381 (114)	516 (65)	702 (69)	478 (170) ^z
Mean	164 (50) ^a	253 (64) ^b	244 (124) ^b	365 (108) ^c	476 (165) ^d	

* Mean for 4 types of toothpaste. Different letters indicate statistical differences (two-way ANOVA, Tukey *post-hoc* test, $P < 0.05$)

Table 2 Means (SD) of volume loss (mm^3) after 420 min as a result of brushing with or without toothpaste for each restorative material

	Without paste	Paste-50 Silica	Paste-52 alumina	Paste-80 Silica	Paste-114 Silica + TiO ₂	MEAN toothpaste*
Ceramic	0.115 (0.017)	0.181 (0.032)	0.115 (0.052)	0.305 (0.031)	0.355 (0.064)	0.239 (0.107) ^x
Co-Cr	0.027 (0.003)	0.058 (0.014)	0.063 (0.020)	0.059 (0.010)	0.067 (0.008)	0.062 (0.013) ^y
Ni-Cr	0.045 (0.004)	0.058 (0.012)	0.046 (0.021)	0.080 (0.010)	0.120 (0.009)	0.076 (0.032) ^y
cpTi	0.096 (0.016)	0.135 (0.032)	0.164 (0.049)	0.221 (0.028)	0.221 (0.028)	0.205 (0.073) ^x
Mean material	0.071 (0.039) ^a	0.108 (0.058) ^b	0.0166 (0.059) ^{a,b,c}	0.166 (0.108) ^{b,c,d}	0.211 (0.128) ^d	

* Mean for 4 types of toothpaste. Different letters indicate statistical differences (Mann-Withney U and Wilcoxon signed-rank tests, $P < 0.05$)

Table 3 Means (SD) of initial surface roughness (R_a) and change of R_a after brushing

	Before brushing	Without paste	Paste-50 Silica	Paste-52 alumina	Paste-80 Silica	Paste-114 Silica + TiO ₂	MEAN toothpaste
Ceramic	0.64 (0.10)	-0.09	0	-0.15	-0.09	-0.14	-0.09
Co-Cr	0.52 (0.09)	-0.05	0	-0.05	-0.13	-0.09	-0.07
Ni-Cr	0.73 (0.09)	-0.05	-0.12	-0.06	-0.01	-0.02	-0.05
cpTi	0.41 (0.07)	-0.03	+0.16	-0.06	+0.15	-0.15	+0.03
Mean material		-0.06	+0.01	-0.08	-0.02	-0.10	

Table 4 Means (SD) of initial Vickers hardness (HVN) and change in HVN after 420 min of brushing with or without toothpaste

	Before brushing	Without paste	Paste -50 Silica	Paste-52 alumina	Paste-80 Silica	Paste-114 Silica + TiO ₂	MEAN toothpaste*
Ceramic	469.5 (22.2)	-7.2 (10.8)	-10.4 (8.8)	-12.1 (4.4)	-5.0 (10.2)	-7.1 (10.4)	-8.64 (8.4) ^x
Co-Cr	527.8 (8.5)	+5.1 (5.1)	+15.4 (9.2)	+15.4 (9.6)	+17.7 (13.3)	+18.6 (13.8)	+16.8 (10.5) ^y
Ni-Cr	312.7 (41.8)	+7.7 (2.8)	+26.9 (27.3)	+22.3 (14.6)	+22.2 (20.7)	+21.2 (20.6)	+23.2 (9.1) ^y
cpTi	480.5 (8.2)	+0.2 (3.1)	+18.5 (9.6)	+4.8 (3.6)	+15.4 (3.3)	+20.1 (12.5)	+14.7 (9.6) ^x
Mean material		+1.32 (8.2) ^a	+12.6 (2.3) ^a	+7.6 (3.1) ^a	+12.6 (2.1) ^a	+13.2 (1.3) ^a	

* Mean for 4 types of toothpaste. Different letters indicate statistical differences (two-way ANOVA, Tukey *post-hoc* test, $P < 0.05$)

268 hardness ($P < 0.01$), both before and after brushing.
 269 Whereas Co-Cr and Ni-Cr became slightly harder over
 270 brushing, the Vickers hardness values for ceramic and
 271 titanium decreased ($P < 0.05$). The type of toothpaste used
 272 did not affect these variations of microhardness.

273 Third-body wear appeared to be the abrasive wear pat-
 274 tern on the metal surfaces after brushing (Fig. 3). The
 275 morphology of the abrasive particles of the different
 276 toothpastes is characterized by a polyedric round form
 277 (Fig. 4). Particles of toothpaste with less abrasive capacity
 278 are smaller and more spherical than those with more
 279 abrasive capacity.

280 4 Discussion

281 The influence of the type of restorative material and
 282 toothpaste on brushing abrasion was evaluated in this

283 in vitro study. Among the materials tested, Co-Cr alloy
 284 exhibited the highest resistance to abrasion as measured by
 285 mass and by volume and the highest hardness value.
 286 A significant correlation between abrasion resistance and
 287 surface hardness has been observed in dental casting alloys
 288 [23] glass ionomers [14] and in resins [12]. Commercially
 289 pure titanium demonstrated the least resistance to abrasion,
 290 as has been observed in another study [24]. Although
 291 ceramic specimens exhibited higher resistance to abrasion
 292 according to weight loss, they showed the greatest amount
 293 of volume loss, due to their low density (2.28 g/cm³)
 294 compared with cpTi (4.51 g/cm³), Ni-Cr (8.05 g/cm³) and
 295 Co-Cr (8.20 g/cm³).

296 Abrasion can have clinical consequences as a result of
 297 changes to the characteristics of the restorative material, as
 298 well as through the ingestion of the elements. The surface
 299 roughness did not vary significantly, so no higher plaque
 300 accumulation can be expected. However, the higher or

Fig. 3 SEM images of a Ni-Cr alloy surface before (a) and after (b) 420 minutes of brushing with Paste-114

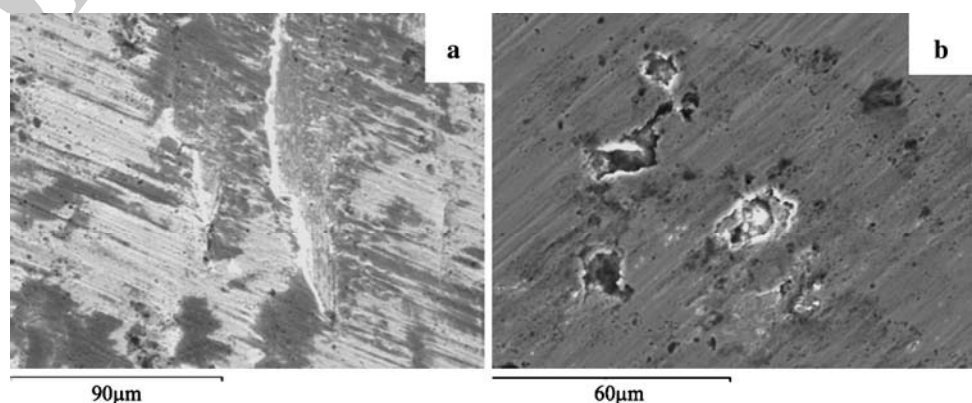
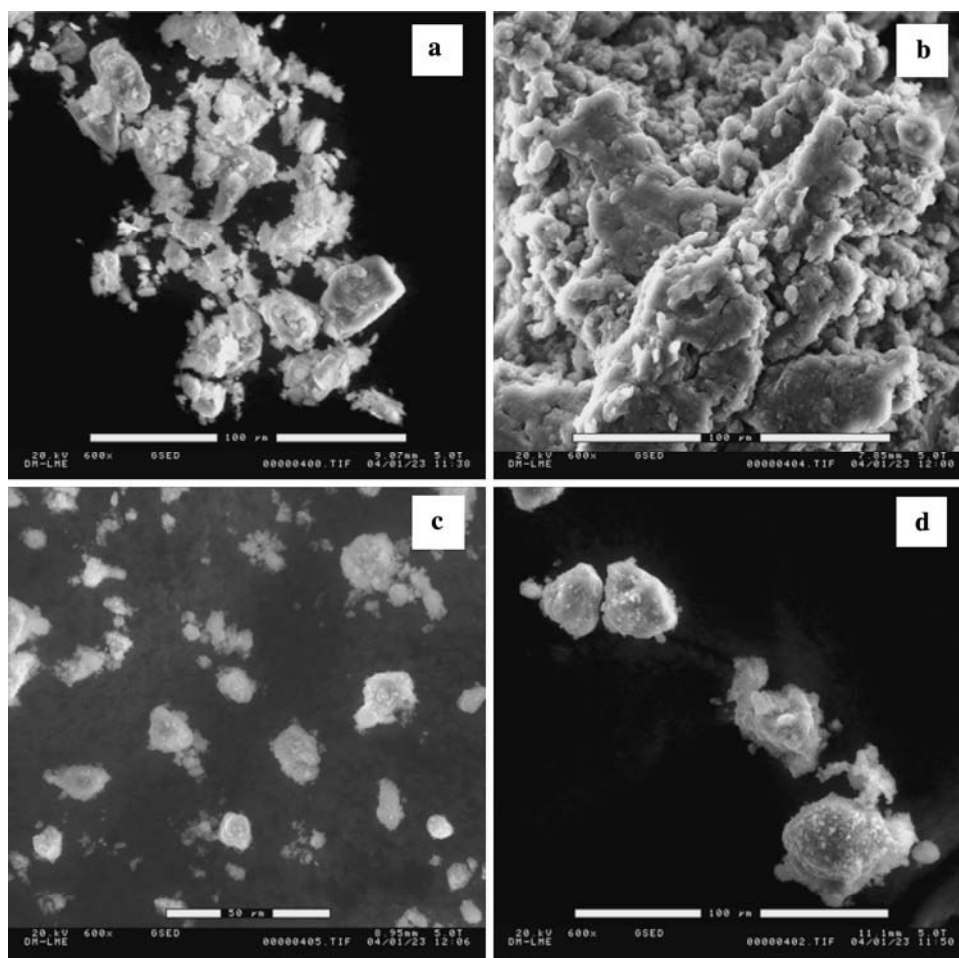


Fig. 4 ESEM micrograph of the abrasive particles of the four toothpastes: (a) Paste-50, (b) Paste-52, (c) Paste-80 and (d) Paste-114



301 lower accumulation of dental plaque depends not only on
 302 the roughness of the surface. Some other factors, as the
 303 interfacial free energy of adhesion of bacteria and pH may
 304 also be involved [25]. Although the measurement of
 305 roughness did not reflect the entire surface, five different
 306 lengths for each material were analyzed. The abrasive
 307 particles from toothpaste present higher hardness values
 308 than the dental materials and produce plastic deformation
 309 up to fracture of the peaks of the surface profile. This lack
 310 of change in roughness despite the mass loss may be due to
 311 the mechanical polished produced by the abrasive particles.
 312 In contrast, a rougher surface was observed in composite
 313 material because of the selective abrasion of the resin
 314 matrix and the dislodgement of filler particles [9]. More-
 315 over, the style of brush may wear the material evenly.
 316 Microhardness presents an increase due to the plastic
 317 deformation on the surface of the materials tested produced
 318 by the friction of the abrasive particles. Plastic deformation
 319 produced by abrasive mechanisms is well known that pro-
 320 duces a high density of dislocations in the metallic surface,
 321 increasing the surface microhardness [26]. Wataha and
 322 coworkers [16] found that 600–800 μg of Ni were released

per cm^2 of Ni–Cr alloy after brushing with toothpaste for 323
 48 h, a similar value to the present study, taking into 324
 account that 65% of the Ni–Cr alloy composition is Ni, and 325
 that the rate of mass loss tended to decrease as a function of 326
 time. This means that the daily intake of nickel due to 327
 brushing could be about 1 μg assuming that all the surfaces 328
 brushed are Ni–Cr alloy. As the tolerable upper daily intake 329
 level of nickel is 1 mg, this represents no risk unless the 330
 patient is allergic to nickel [27]. There are concerns 331
 regarding the release of metal ions from the nickel-based 332
 alloys to surrounding tissues and their cytotoxicity to the 333
 tissue's normal function. Despite these concerns, results 334
 from the effect of Nickel-containing dental cast alloys in 335
 cell culture system are contradictory [19]. Moreover, Ni–Cr 336
 showed a small increase in cytotoxicity after being brushed 337
 under several relatively severe *in vitro* conditions, such as at 338
 pH 4 [17]. Large areas of the biological interaction of dental 339
 casting alloys are not yet understood, especially in the area 340
 of low dose exposure and individual differences in the 341
 appearance of adverse reactions [19]. 342

The amount of abrasion from brushing depends on the 343
 type of toothpaste used. In the present study a positive 344

345 correlation was observed between the RDA of the tooth-
 346 paste and abrasion levels as measured by weight or volume
 347 loss in all materials except for Co–Cr alloys. The RDA
 348 value of the toothpaste could also be useful to estimate its
 349 relative abrasivity on ceramic, Ni–Cr alloys and commer-
 350 cially pure titanium, but not on Co–Cr alloys. In addition to
 351 the RDA, the type of abrasive may also influence abrasion
 352 characteristics, explaining the slight difference in abrasivity
 353 of Paste-50 and Paste-52 on ceramic and Ni–Cr compared
 354 with cpTi and Co–Cr, despite their similar RDA values.

355 It is very difficult to compare the results of the present study
 356 with other in vitro studies because of differences in method-
 357 ology and materials tested. This study was designed to
 358 evaluate the effect of toothpaste on the abrasion of different
 359 restorative materials by means of a powered toothbrush. For
 360 this study, a load of 250 g was chosen because this is within
 361 the range of the optimum force for plaque removal with a
 362 powered toothbrush [28, 29] and according to the powered
 363 toothbrush manufacture's instructions [30]. However, most
 364 studies have used a load of 250–600 g and horizontal back-
 365 and-forth movements of the brush to simulate manual tooth-
 366 brushing [6, 7, 9–13]. To our knowledge, this is the first study
 367 that uses a real powered toothbrush to study abrasivity, possi-
 368 bly simulating better a clinical situation than a brushing
 369 machine. Most studies have evaluated abrasivity as a vertical
 370 wear profile measured by means of profilometry [6–9, 11, 14]
 371 whereas only a few studies have determined abrasivity
 372 through weight loss [10, 12, 13]. However, when a force is
 373 applied to a toothbrush without displacement, the bristles
 374 might abrade the material unevenly, meaning a single vertical
 375 wear profile may not be representative of the area brushed.
 376 The daily recommended time for brushing with toothpaste is
 377 2 min twice, meaning a given tooth surface might typically be
 378 in contact with the toothbrush for a maximum of 5 s twice
 379 daily [5]. Therefore one hour of brushing in this experiment
 380 may be equivalent to one year of life for a tooth surface.

381 It is very difficult to extrapolate the results of the present
 382 study to a clinical situation, as the oral cavity is subject to
 383 changes in pH and temperature, a continuous flow of sal-
 384 iva, microbiological activity, occlusal load, as well as
 385 many other factors. In spite of the limitations of the present
 386 study, brushing caused abrasion of dental casting alloys
 387 and ceramic and the intensity of which depended upon the
 388 restorative material and the RDA of the toothpaste [31–35].
 389 However, brushing dental casting alloys using a powered
 390 toothbrush and applying a load of 250 g with toothpaste of
 391 RDA of about 50, one need not expect significant conse-
 392 quences from the clinical point of view.

393 5 Conclusions

394 This study suggests that dental casting alloys and ceramic
 395 are susceptible to abrasion by brushing with an electric

toothbrush. In general, the amount of mass loss is pro-
 396 portional to the RDA value of the toothpaste. The increase
 397 of the microhardness was due to the plastic deformation
 398 produced by the abrasion and surface roughness do not
 399 presents apparent increase by the inclusion of the abrasive
 400 particles on the material roughness. These variations will
 401 produce an important increase on the corrosion behavior
 402 and ion release, as can be observed in Part II, Conse-
 403 quently, the toothbrushed can have were clinically relevant.
 404

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 411

References 412

1. C. Deery, M. Heanue, S. Deacon, P.G. Robinson, A.D. Walmsley, H. Worthington, W. Shaw, A.M. Glenny, *J. Dent.* **32**, 197–211 (2004) 413
2. N.C. Sharma, C.R. Goyal, J.G. Qaqish, M.A. Cugini, M.C. Thompson, P.R. Warren, *J. Dent.* **33**, 11 (2005) 414
3. J.I. Hefferren, *J. Dent. Res.* **58**, 1535–1543 (1979) 415
4. ISO 11609. *Dentistry-toothpastes requirements, test methods and marking* (International Organisation for Standardization, Geneva, 1995) pp. 1–10 416
5. S. Hooper, N. West, M. Pickles, A. Joiner, R. Newcombe, M. Addy, *J. Clin. Periodontol.* **30**, 802–808 (2003) 417
6. T. Attin, U. Koidl, W. Buchalla, H.G. Schaller, A.M. Kielbassa, E. Hellwig, *Arch. Oral Biol.* **42**, 243–250 (1997) 418
7. M. De Menezes, C.P. Turssi, A.T. Hara At, D.C. Messias, M.C. Serra, *Clin. Oral Invest.* **8**, 151–155 (2004) 419
8. N. West, M. Addy, J. Hughes, *J. Oral Rehabil.* **25**, 885–895 (1998) 420
9. N. Tanoue, H. Matsumura, M. Atsuta, *J. Prosthet. Dent.* **84**, 93–97 (2000) 421
10. R. Richmond, T.V. Macfarlane, J.F. Mccord, *Dent. Mater.* **20**, 124–132 (2004) 422
11. C. Haselden, J.A. Hobkirk, G.J. Pearson, E.H. Davies, *J. Oral Rehabil.* **25**, 335–339 (1998) 423
12. K. Kawai, Y. Iwami, S. Ebisu, *J. Oral Rehabil.* **25**, 264–268 (1998) 424
13. N. Anil, S. Bolay, *Int. J. Prosthodont.* **15**, 483–487 (2002) 425
14. Y. Momoi, K. Hirotsaki, A. Kohno, J.F. McCabe, *Dent. Mater.* **13**, 82–88 (1997) 426
15. J.C. Wataha, P.E. Lockwood, K.B. Frazier, S.S. Khajotia, *J. Prosthodont.* **8**, 245–251 (1999) 427
16. J.C. Wataha, P.E. Lockwood, D. Mettenburg, S. Bouillaguet, *J. Biomed. Mater. Res.* **65B**, 180–185 (2003) 428
17. J.C. Wataha, P.E. Lockwood, M. Noda, S.K. Nelson, D.J. Mettenburg, *J. Prosthet. Dent.* **87**, 94–98 (2002) 429
18. W. Geurtsen, *Crit. Rev. Oral Biol. Med.* **13**, 71–84 (2002) 430
19. G. Schmalz, P. Garhammer, *Dent. Mater.* **18**, 396–406 (2002) 431
20. R.W. Wassell, A.W. Walls, J.G. Steele, *Brit. Dent. J.* **192**, 199–211 (2002) 432
21. J.C. Wataha, *J. Prosthet. Dent.* **87**, 351–363 (2002) 433
22. R. Wang, A. Fenton, *Quintessence Int.* **27**, 401–408 (1996) 434
23. I. Watanabe, C. Ohkubo, J.P. Ford, M. Atsuta, T. Okabe, *Dent. Mater.* **16**, 420–425 (2000) 435

- 455 24. C. Ohkubo, I. Watanabe, J.P. Ford, H. Nakajima, T. Hosoi, 468
456 T. Okabe, *Biomater.* **21**, 421–428 (2000) 469
457 25. J. Tagami, M. Toledano, C. Prati, (Adv. Adhesive Dent. Kuraray 470
458 Co.Ltd. Dirimido, Italy, 1999) pp. 59–73 471
459 26. J. Pena, F.J. Gil, J.M. Guilemany, *Acta. Materialia.* **50**(12), 3117– 472
460 3126 (2002) 473
461 27. J.F. Lopez-Alias, J. Martinez-Gomis, J.M. Anglada, M. Peraire, 474
462 *Dent Mater* (in press) 475
463 28. G.A. Van Der Weijden, M.F. Timmerman, P.A. Versteeg, 476
464 M. Piscaer, U. Van Der Velden, *J. Clin. Periodontol.* **31**, 620–624 477
465 (2004) 478
466 29. G.I. Mccracken, J. Janssen, M. Swan, N. Steen, M. De Jager, 468
467 P.A. Heasman, *J. Clin. Periodontol.* **30**, 409–413 (2003) 469
30. J.J. Hefferren, *Adv. Dent. Res.* **16**, 16 (2002) 470
31. F.J. Gil, J.A. Planell, *J. Biomed. Mater. Res-A.* **48**, 682–688 471
(1999) 472
32. F.J. Gil, J.M. Manero, J.A. Planell, *J. Mater. Sci.* **30**, 2526–2530 473
(1995) 474
33. J.M. Guilemany, F.J. Gil, *J. Mater. Sci.* **26**, 4626–4630 (1991) 475
34. J. Muntasell, J.L.L. Tamarit, E. Cesari, J.M. Guilemany, F.J. Gil, 476
Mater. Res. Bull. **24**, 445–452 (1989) 477
35. J. Muntasell, J.L.L. Tamarit, J.M. Guilemany, F.J. Gil, E. Cesari, 478
Mater. Res. Bull. **23**, 1585–1590 (1988)

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