## REVISITING BLOOD-BRAIN BARRIER: A CHROMATOGRAPHIC APPROACH

Xavier Subirats ${ }^{1}$, Laura Muñoz-Pascual ${ }^{1}$, Michael H. Abraham ${ }^{2}$, Martí Rosés ${ }^{1, *}$

${ }^{1}$ Institute of Biomedicine (IBUB) and Department of Chemical Engineering and Analytical Chemistry, Universitat de Barcelona, Martí i Franquès 1-11, 08028 Barcelona, Spain
${ }^{2}$ Department of Chemistry, University College London, 20 Gordon Street, London WC1H OAJ, UK
*Corresponding author

Dr. Xavier Subirats
Phone: +34 934039 119, Fax: + 34934021 233, E-mail: xavier.subirats@ub.edu

Ms. Laura Muñoz-Pascual
E-mail: lauramunozpascual1993@hotmail.com

Prof. Michael H. Abraham
Phone: +440207679 4639, Fax: +4402076797463, E-mail: m.h.abraham@ucl.ac.uk

Prof. Martí Rosés
Phone: + 34934039 275, Fax: + 34934021 233, E-mail: marti.roses@ub.edu


#### Abstract

Drugs designed to reach a pharmacological CNS target must be effectively transported across the blood-brain barrier (BBB), a thin monolayer of endothelial cells tightly attached together between the blood and the brain parenchyma. Because of the lipidic nature of the BBB, several physicochemical partition models have been studied as surrogates for the passive permeation of potential drug candidates across the BBB (octanol-water, alkane-water, PAMPA...). In the last years, biopartition chromatography is gaining importance as a noncellular system for the estimation of biological properties in early stages of drug development. Microemulsions (ME) are suitable mobile phases, because of their ease of formulation, stability and adjustability to a large number of compositions mimicking biological structures. In the present work, several microemulsion liquid chromatographic (MELC) systems have been characterized by means of the Abraham's solvation parameter model, in order to assess their suitability as BBB distribution or permeability surrogates. In terms of similarity between BBB and MELC systems (dispersion forces arising from solute non-bonded electrons, dipolarity/polarizability, hydrogen-bond acidity and basicity, and molecular volume), the passive permeability surface area product (log PS) for neutral (including zwitterions), fully and partially ionized drugs was found to be well correlated with the ME made of $3.3 \%$ SDS ( $\mathrm{w} / \mathrm{v}$; surfactant) $0.8 \%$ heptane ( $\mathrm{w} / \mathrm{v}$; oil phase) and $6.6 \% 1-$ butanol (w/v; co-surfactant) in 50 mM aqueous phosphate buffer, pH 7.4 .


## Keywords

Blood-brain barrier; LFER; $\log \mathrm{BB}$; $\log$ PS; microemulsion; liquid chromatography

## Abbreviations

BB: plasma-to-brain distribution ratio; BBB: blood-brain barrier; CNS: central nervous system; LFER: linear free energy relationships; ME: microemulsion; MELC: microemulsion liquid chromatography; PS: permeability-surface area product; SP: solute property; SDS: sodium dodecylsulfate

## 1. Blood-brain barrier

### 1.1. Experimental models: $\log \mathrm{BB}$ and $\log \mathrm{PS}$

The blood-brain barrier (BBB) plays a fundamental role in the pharmacological activity of drugs targeting the central nervous system (CNS). It is a thin monolayer of endothelial cells, tightly attached together, that separates the circulating blood and the brain parenchyma.

Two different in vivo BBB experimental models have been considered in the present work, the plasma-to-brain distribution ratio $\left(\log K_{\mathrm{p}}\right.$, also known as $\left.\log \mathrm{BB}\right)$ and the permeability-surface area product (PS). $K_{\mathrm{p}}$ accounts for the concentration of drug present in the brain at steady state in relation to that in plasma. This is, in fact, a partition coefficient between the concentrations of both bound and unbound drug in brain (intracellular and interstitial fluids) and plasma. In vivo, $\log \mathrm{BB}$ is determined at a specific time point after drug administration. It should be pointed out that bound drug molecules (for instance, to plasma and cytoplasmic proteins) are not expected to be pharmacologically active [1]. Therefore, besides BBB equilibration of unbound drug molecules, $\log \mathrm{BB}$ measures nonspecific binding to brain tissue and plasma proteins. Consequently, in the case of drug molecules significantly bound to cytoplasmic proteins in brain, $\log$ BB might fail to indicate the effective extent of BBB penetration [2]. However, $\log$ BB is a widely used parameter in BBB studies, especially for in silico predictions of BBB in vivo data $[3,4]$.

In contrast to $\log \mathrm{BB}$, in situ brain perfusion experiments, mainly performed on rodents, allow the measurement of the initial and unidirectional rate of brain penetration from blood, or usually from saline, to brain across the luminal BBB membrane, even in the case of solutes strongly bond to proteins. Perfusion time is about 30 to 180 s [5], and it ends before any equilibrium state can be reached. In this way, the clearance or $K_{\text {in }}\left(\mathrm{mL} \mathrm{g}^{-1} \mathrm{~s}^{-1}, \mathrm{~mL}\right.$ of perfusate per gram of brain tissue and second of net perfusion time) is determined. However, this parameter depends on the perfusion flow velocity and, therefore, $K_{\text {in }}$ is corrected by the flow of the perfusion fluid in brain, measured by an appropriate flow calibrant, such as radioactive iodoantypirine, microspheres or diazepam [6]. Thus, PS is obtained, by the product of luminal permeability $\left(\mathrm{cm} \mathrm{s}^{-1}\right)$ and the endothelial surface area per gram of brain tissue $\left(\mathrm{cm}^{2} \mathrm{~g}^{-1}\right)$.

### 1.2. Factors affecting the distribution and permeation between blood and brain: a LFER approach

$\log$ BB was extensively studied by Abraham and coworkers [7,8] by means of linear free energy relationships (LFER) in order to point out the factors that influence the distribution of
solutes between blood and brain. According to the solvation model for unionized molecules [9], a solute dependent variable ( $\log \mathrm{SP}$ ) is linearly related to specific interactions between solute and surrounding phase, mainly dispersion $(e \cdot E)$, dipole-dipole or dipole-induced dipole plus some polarizability interactions $(s \cdot S)$, solute hydrogen-bond acidity and basicity ( $a \cdot A$ and $b \cdot B$, respectively), and a volume term $(v \cdot V)$ related to the work of separating solvent molecules to provide a cavity of suitable size for the solute molecule and solute-solvent general dispersion interactions:

$$
\begin{equation*}
\log \mathrm{SP}=c+e E+s S+a A+b B+v V \tag{1}
\end{equation*}
$$

where $E, S, A, B$, and $V$ are solute descriptors, and $e, s, a, b$, and $v$ are system constants reflecting differences between the two condensed phases being studied, in the present case blood and brain. Thus, a set of 157 substances with directly measured and indirectly determined $\log \mathrm{BB}$ values was studied yielding the following equation [8]:

$$
\begin{array}{r}
\log \mathrm{BB}=0.044+0.511 E-0.886 S-0.724 A-0.666 B+0.861 V \\
\left(n=148, R^{2}=0.710, \mathrm{SD}=0.367, F=71\right) \tag{2}
\end{array}
$$

At the time of its publication in 2001, due to the size of the set and chemical diversity of the selected molecules, this was a good general blood-brain distribution model, which revealed the factors of brain uptake. Provided that solute descriptors are zero or positive, large and positive coefficients increase $\log \mathrm{BB}$, which means, in turn, a higher affinity for brain. Thus, according to Eq. (2), solutes interacting through $\pi$ - and n-electron pairs ( $e \cdot E>0$ ) and large molecules $(v \cdot V>0)$ show higher brain uptakes, whereas dipolar or polarizable solutes $(s \cdot S<0)$ with hydrogen-bond interactions $(a \cdot A, b \cdot B<0)$ tend to remain in the blood phase. The relatively low determination coefficient in Eq. (2) might be due to the difficulty of accurate experimental determination of $\log \mathrm{BB}$ values, and the molecular descriptors used, either experimentally measured or calculated, referred to neutral solutes.

In a later study in 2004 [10], Eq. (1) was applied to $30 \log$ PS values of neutral compounds, leading to the following equation for permeation from saline (standard deviations of the coefficients are reported in brackets):

$$
\begin{align*}
\log \mathrm{PS} & =-0.639(0.408)+0.312(0.515) E-1.009(0.158) S-1.895(0.385) A \\
& -1.636(0.410) B+1.709(0.392) V \quad\left(n=30, R^{2}=0.870, \mathrm{SD}=0.52, F=32.2\right) \tag{3}
\end{align*}
$$

It should be stressed that acidic or basic compounds that could be totally or partially ionized at the physiological pH of 7.4 were not included in that analysis, although carboxylic acids could be included in the $\log$ BB model of Eq. (2) by introduction of a correction factor [8]. In a later work, acids and bases totally ionized were also included in $\log$ PS correlations
[11]. A comparison of the coefficients in Eqs. (2) and (3) reveals that, qualitatively, bloodbrain distribution and permeation are ruled by the same factors.

### 1.3. MELC as a physicochemical method for the determination of biological activity

Beyond ethical concerns in animal experimentation, in early stages of the drug discovery process an accurate in vivo determination of biological activity for a large number of potential candidates is unaffordable. Thus, isotropic organic solvent/water partition models (octanol, hexadecane...) were studied as physicochemical surrogates of BBB [5]. However, simple partition coefficients like octanol-water were unable to model the desolvation (breaking of the hydrogen-bounds between a solute and the solvating water molecules) involved in the transfer of compound from aqueous solution into a phospholipid bilayer. The combination of partition coefficients measured in octanol-water and alkane-water allowed the inclusion of hydrogenbonding interactions, improving the prediction capacity of the model, but increasing the time required to carry out the determination. For screening purposes the measurement of several partition coefficients for a single molecule is excessively time consuming, and thus faster approaches are desirable.

Microemulsion liquid chromatography (MELC) is a very interesting technique, especially in the field of pharmaceutical analysis, because of the ability of the microemulsions (ME) used as mobile phases to solubilize both lipophilic and hydrophilic compounds and its separation capabilities [12,13]. Oil-in-water ME are made of oil droplets (octane, heptane...) stabilized by a surfactant (SDS, sodium cholate, Brij 35...) and a cosurfactant (a short-chain alcohol as 1-butanol, 1-pentanol...) and dispersed in an aqueous buffer. The anionic SDS is commonly used as surfactant in a concentration range of 2-3\%, and typically the amount of oil is frequently below $1 \%[12,13]$. When linear alkanes are involved in the ME, the mass ratio between SDS and the cosurfactant is suggested to be 0.5 [14]. For such systems, the oil-in-water ME strongly depends on the salt concentration and it can only exist in a relatively small water-rich range of compositions [15,16]. Once prepared, ME are stable and variations in their composition ( pH , buffer nature, surfactant type and concentration...) do not significantly change their functionality [17]. However, retention mechanisms in MELC systems are complex, since solutes are expected to partition at least between the bulk aqueous phase, the oil droplet, and the surfactant-coated stationary phase [18].

Furthermore, and this is the main point of this study, ME can be used as physicochemical surrogate models of biological processes, such as lipophilicity [19-21] or BBB [22-24], since ME mimic, to some extent, the properties of cell membranes. Liu and
coworkers [22], following a LFER approach, characterized several MELC systems and compared them to biological ones. The authors concluded that a C18 stationary phase and a ME mobile phase consisting of $3.3 \%$ SDS, $6.6 \%$ butanol, $1.6 \%$ heptane and $88.5 \% 50 \mathrm{mM}$ phosphate buffer pH 7.0 (all percentages in weight) was a good surrogate of BBB distribution, particularly $\log$ BB. However, Liu and coworkers [22] studied only 37 compounds, six of which were left out as outliers.

The purpose of this study is the comparison of several MELC systems to BBB systems by means of the Abraham model in order to find appropriate MELC systems for surrogation of BBB systems. Since in principle the Abraham model was derived for non ionic compounds, a further goal is to check the performance of MELC surrogation for drugs that should be totally or partially ionized drugs at the blood physiological pH .

## 2. Material and methods

### 2.1. Instrumentation

pH measurements were taken with a Crison (Barcelona, Spain) 5014 combination electrode (glass electrode and a reference electrode with a 3.0 M KCl solution in water as salt bridge) in a Crison GLP22 pH meter. MEs were sonicated in a J.P. Selecta (Barcelona, Spain) ultrasonic bath with a power of 360 W .

HPLC measurements were performed on a Shimadzu (Kyoto, Japan) HPLC system consisting of two LC-10ADvp pumps, a SIL-10ADvp auto-injector, an SPD-M10Avp diode array detector, a CTO-10ASvp oven at $37^{\circ} \mathrm{C}$ and a SCL-10Avp controller. A $5 \mu \mathrm{~m} 150 \times 4.6$ mm Gemini C18 column and a $4 \times 3.0 \mathrm{~mm}$ guard cartridge from Phenomenex (Torrance, CA, USA) were used at a flow rate of $1.0 \mathrm{~mL} \mathrm{~min}^{-1}$. Each compound was analyzed at least in triplicate and injection volumes were set to $10 \mu \mathrm{~L}$. Retention factors were expressed as $\log k=$ $\log \left(\left(t_{\mathrm{R}}-t_{0}\right) / t_{0}\right)$, where $t_{\mathrm{R}}$ and $t_{0}$ were the retention times of analyte and potassium bromide (Merck, for analysis) as dead timer marker, respectively.

### 2.2 Mobile phase and sample preparation

Water was deionized to a resistivity of $18.2 \mathrm{M} \Omega \mathrm{cm}$ by the Milli-Q plus system from Millipore (Billerica, MA, USA). Aqueous buffer was prepared from sodium dihydrogenphosphate (Merck, 99\%) and sodium hydrogenphosphate (J. T. Baker, 99.5\%) to a final concentration of 50 mM and pH 7.4 . Under magnetic stirring and at room temperature, $3.3 \% \mathrm{w} / \mathrm{v}$ of SDS (Sigma-Aldrich, $>99 \%$ ) was dissolved in aqueous buffer until a transparent colorless solution was obtained. Then pH was adjusted to 7.4 by the addition of small
volumes of a 3 M NaOH solution prepared shortly before use from pellets (Merck, > 99\%), followed by the addition of $6.6 \% \mathrm{w} / \mathrm{v}$ 1-butanol (Sigma-Aldrich, 99.8\%) and the desired amount of heptane ( $0 \%, 0.8 \%$ or $1.6 \% \mathrm{w} / \mathrm{v}$; Merck, for analysis). At this point, the solution became white and turbid. Magnetic stirring was maintained for 10 min and the desired ME volume was adjusted with aqueous buffer (in order to compensate the volume contraction of the mixture). Then the ME was sonicated for about 30 min until it became clear again, and finally the solution was left to stand at room temperature for at least 12 h . Immediately before use, ME was vacuum filtered using a Büchner funnel and a $0.45 \mu \mathrm{~m}$ nylon membrane (Teknokroma, Spain).

Injected compounds were provided by Abbott Laboratories (Abbot Park, IL, USA), Acros Organics (Geel, Belgium), Astrazeneca (London, UK), Baker (Center Valley, PA, USA), Boehringer Ingelheim Pharmaceuticals (Ridgefield, CT, USA), Carlo Erba (Milano, Italy), Esteve (Barcelona, Spain), Janssen (Beerse, Belgium), Merck (Billerica, MA, USA), Roche (Basel, Switzerland), Scharlau (Barcelona, Spain), Sigma-Aldrich (St. Louis, MO, USA), and Toronto Research Chemicals (Toronto, ON, Canada); all of high purity grade ( $\geq 97 \%$ ). $10 \mathrm{mg} \mathrm{mL}^{-1}$ stock solutions were prepared in methanol (Fisher, HPLC grade) and ten-fold diluted with ME before injection.

### 2.3 HPLC and column cleaning

After a working session, in order to avoid the precipitation of SDS, the HPLC instrument and column were washed at a flow rate of $1 \mathrm{~mL} \mathrm{~min}^{-1}$ with water/methanol $95: 5$ followed by water/methanol 5:95, 30 min each.

## 3. Results and discussion

### 3.1. LFER characterization of BBB permeability

A new LFER characterization study according to Eq. (1) was conducted which broadens the chemical diversity of test compounds in relation to Eq. (3). The study was based in the in situ rodent brain perfusion permeability data referred to permeation from saline at pH 7.4 and corrected for ionization, compiled by Avdeef [5]. Molecules were selected that exhibited BBB passive permeation only, avoiding carrier-mediated or actively transported processes. Therefore, the solvation property selected for this study was the so called intrinsic passive permeability $\left(\log P_{0}{ }^{\mathrm{BBB}}\right)$. In fact, $\log P_{0}{ }^{\mathrm{BBB}}$ is just a correction of $\log$ PS for ionized compounds and therefore $\log P_{0}{ }^{\mathrm{BBB}}=\log$ PS in the case of non-ionized species. Observed log $P_{0}{ }^{\mathrm{BBB}}$ values obtained from experiments with rats were correlated with measured (when
available) or calculated molecular descriptors [25] (see Table 1), according to Eq. (1) then the fitted coefficients were used to back calculate $\log P_{0}^{\mathrm{BBB}}$ values, and finally a linear regression was established between observed and predicted $\log P_{0}^{\mathrm{BBB}}$ values. In this work, compounds with residuals higher than twice the standard deviation of the regression were considered as outliers. After excluding these values from correlations, the final coefficients obtained are those in Eq. (4) and the corresponding plot is presented in Figure 1:

$$
\begin{align*}
\log P_{0}^{\mathrm{BBB}} & =-4.048(0.139)+0.213(0.133) E-0.947(0.126) S-0.438(0.150) A \\
& -1.497(0.163) B+1.953(0.133) V \quad\left(n=141, R^{2}=0.833, \mathrm{SD}=0.64, F=135\right) \tag{4}
\end{align*}
$$

It is noteworthy that Eq. (4) covers a wide range of permeability values (about $7 \log$ units) and includes molecules with different chemical properties as reflected by their descriptors (Table 1).

### 3.2. LFER characterization of MELC systems

With the aim of exploring the predictive capacity of MELC systems for the prediction of BBB distribution or permeability, three different mobile phases were prepared from 50 mM phosphate buffer pH 7.4 containing the same SDS and 1-butanol concentration ( $3.3 \%$ and $6.6 \% \mathrm{w} / \mathrm{v}$, respectively) but with different amounts of heptane ( $0,0.8$, and $1.6 \% \mathrm{w} / \mathrm{v}$ ). Test compounds (Table 2) were selected to present different chemical characteristics (hydrogenbonding interactions, dipolarity/polarizability...) and to be unionized at the desired pH in order to build the correlations between $\log k$ and neutral molecular descriptors. The column temperature was set to $37^{\circ} \mathrm{C}$ because this is the physiological temperature. One additional advantage of $37^{\circ} \mathrm{C}$ over room temperature is the higher the temperature, the lower the mobile phase viscosity and consequently the instrumental backpressure. Once outliers were excluded (Figure 2), the following equations were obtained:

$$
\begin{gather*}
\log k_{1.6 \% \text { heptane }}=0.179(0.059)-0.011(0.052) E-0.418(0.072) S-0.283(0.099) A \\
-1.148(0.074) B+1.203(0.095) V \quad\left(n=46, R^{2}=0.938, \mathrm{SD}=0.15, F=122\right) \tag{5}
\end{gather*}
$$

$$
\begin{align*}
& \log k_{0.8 \% \text { heptane }}=0.186(0.053)-0.010(0.046) E-0.411(0.062) S-0.237(0.086) A \\
& \quad-1.133(0.064) B+1.231(0.082) V \quad\left(n=45, R^{2}=0.952, \mathrm{SD}=0.13, F=153\right) \tag{6}
\end{align*}
$$

$$
\begin{array}{r}
\log k_{0.0 \% \text { heptane }}=0.197(0.052)-0.015(0.039) E-0.353(0.056) S-0.167(0.086) A \\
-1.196(0.059) B+1.202(0.081) V \quad\left(n=41, R^{2}=0.959, \mathrm{SD}=0.11, F=163\right) \tag{7}
\end{array}
$$

Interestingly, both ME (Eq. (5) and (6)) show nearly identical system coefficients despite the different concentration of heptane, and they are even similar to the micellar system without heptane (Eq. (7)). Apparently the oil phase slightly favors interactions with
dipolar/polarizable solutes with hydrogen-bonding acidity properties, whereas the micellar phase shows somewhat affinity for molecules with hydrogen-bonding basicity.

### 3.3. Comparative study

A very interesting tool for the quantification of the similarity between two systems is the euclidean distance (d) of their characteristic vectors [26]. $e, s, a, b$, and $v$ coefficients on Eq. (1) define the properties of a particular system, and they can be considered as the elements of a five-dimensional vector. When the comparison is established between vectors of different magnitudes, for instance $\log \mathrm{BB}$ and $\log k$, it is convenient to divide the elements by the length of the vector to obtain unit vectors ( $e_{\mathrm{u}}, s_{\mathrm{u}}, a_{\mathrm{u}}, b_{\mathrm{u}}$, and $v_{\mathrm{u}}$, Table 3), and then calculate the distance (Table 4). Complementarily, a plot of the two principal components (PC) obtained after a PCA analysis of the elements of unit vectors provides an approximate visual representation of similarity between systems.

In this study the comparison was performed between the biological systems of Eqs. (24) and the chromatographic surrogates of Eqs (5-7) and that reported by Liu and coworkers [22] mentioned in section 1.3, further referred as MP3 system according to the designation used in the original paper. From the data presented in Table 3, it can be concluded that all biological systems have in common that the larger the molecular volume, the more favored brain uptake, followed in a lesser extent by the capacity of interactions through $\pi$ - and $n$ electrons. The coefficients of both permeability parameters, $\log$ PS and $\log P_{0}^{\mathrm{BBB}}$, are very similar with the exception of the solute hydrogen-bonding acidity, more negative for $\log$ PS. Concerning the comparison of chromatographic systems, differences between $\log k_{0.8 \%}$ and $\log k_{\text {MP3 }}$ were larger than expected, given that both ME were prepared in a similar way.

Concerning the PCA plot shown in Figure 3, the chromatographic approaches assayed in the present work form a cluster, with the ME systems containing 0.8 and $1.6 \%$ of heptane being slightly closer to each other. Interestingly, although the physicochemical system used by Liu et al. (log kMP3) [22] was proposed as a surrogate of biological $\log \mathrm{BB}$, according to this PCA results it is much more similar to $\log$ PS, and the top left $\log$ BB seems to be far from the rest of all other systems, either biological or chromatographic. It must be pointed out that, according to the PCA loadings, the most relevant contribution to PC1 is the hydrogenbond basicity of the system $\left(-0.33 e_{u}, 0.34 s_{u}, 0.74 a_{u},-0.34 b_{u}\right.$, and $\left.0.34 v_{u}\right)$, and therefore the systems with more negative $a_{\mathrm{u}}$ values lead to negative and similar PC1 digits $(\log \mathrm{BB}, \log$ PS,
and $\left.\log k_{\mathrm{MP}}\right)$, whereas the opposite trend is obtained for the less negative ones $\left(\log P_{0 \mathrm{BBB}}, \log \right.$ $k_{0} \%, \log k_{0.8 \%}$, and $\left.\log k_{1.6 \%}\right)$.

The quantitative estimation of differences between pairs of systems shown in Table 4 confirms the significant difference between $\log k_{0.8 \%}$ and $\log k_{\mathrm{MP3}}$ observed on the PCA plot, much larger than initially expected taking into account that both ME were prepared in a similar way. The particular reasons leading to this mismatch are difficult to elucidate, but we provide here tentatively some of the possible explanations. Firstly, the representativity of the compounds used for correlations must be examined. In the present work the number of molecules included in the characterization set was larger than that of Liu (45 vs 26), and the studied $\log k$ range was wider ( $-0.848 / 1.203 \mathrm{vs}-0.365 / 1.212$ ). Another possible reason might lie in the chromatographic column used. Although both stationary phases were C 18 , the particular support and column technology might affect the retention of analytes (Gemini vs AT Chrom). Finally, the accuracy in the dead time measurement and thus in the determination of retention factors might have had an influence in the characterization (potassium bromide peak vs first significant deviation of the baseline).

In relation to the biological systems, there is nearly the same distance from the three studied MELC systems to $\log$ BB and to $\log$ PS, with the distance to the latter being slightly shorter (Table 4). $\log$ PS and $\log P_{0}{ }^{\text {BBB }}$ were initially expected to be closer to each other, since the latter is a correction of the former in order not to consider only the permeation of unionized species, which was very convenient in order to increase the number of compounds involved in the LFER characterization, but both of them are related to the BBB penetration. In order to find the possible reasons of this mismatch, a joint PCA was performed with the molecular descriptors ( $E, S, A, B$, and $V$ ) of both sets of compounds included in the correlations of Eqs. (3) and (4), and the scores of the two main PC are plotted in Figure 4. Although the 30 substances included in log PS study show a reasonably good distribution over the two PCs, the higher number of compounds used for $\log P_{0}{ }^{\mathrm{BBB}}$ characterization allow a better coverage of the chemical diversity space, including molecules that broadened the range of hydrogen-bonding properties (A, $0.00 / 0.95$ vs. $0.00 / 2.30 ; B, 0.48 / 2.55$ vs $0.45 / 4.04$ ) and $\pi$ - and $n$-electrons interactions ( $\mathrm{E}, 0.21 / 3.48$ vs. $0.18 / 4.63$ ).

When comparing the calculated distances between the chromatographic systems characterized in the present work and the biological BBB parameters, the highest similarity (i.e. the lowest distance) was obtained for the ME containing a $0.8 \%$ of heptane and $\log P_{0}^{\mathrm{BBB}}$ (0.175). In contrast, the shortest distance with $\log$ BB was found to be 0.597 in the case of the

ME with a $1.6 \%$ of oil. Therefore, according to the LFER characterization, the chromatographic systems here studied seemed to be better models of BBB permeability (log $\mathrm{PS} / \log P_{0}{ }^{\mathrm{BBB}}$ ) rather than distribution $(\log \mathrm{BB})$ measurements, particularly the ME containing a $0.8 \%$ of heptane.

### 3.4. MELC system as surrogate model for BBB

The previous section shows that MELC systems can be good surrogate sytems for brain perfusion of non ionized compounds ( $\log P_{0}{ }^{\mathrm{BBB}}$ ), but many BBB active drugs are partially or totally ionized at the physiological blood pH . Thus, it would be very convenient to test MELC surrogation for ionized drugs.

With the aim of assessing the predictive capacity of the proposed physicochemical system as a BBB model, several analytes with known $\log$ BB (Table 5) or $\log$ PS (Table 6) values were injected using as mobile phase the ME with a $0.8 \%$ of heptane. About only onefourth of the injected substances were unionized at pH 7.4 , which corresponds to saline solutions employed in the brain perfusion assays, and therefore it was the selected pH for the chromatographic mobile phase, $\log$ PS data were used instead of $\log P_{0}{ }^{\text {BBB }}$ as a measure of unidirectional brain penetration. Depending on the acid-base properties of the compounds an appropriate mobile phase pH might possibly allow an estimation of the penetration of unionized species, but these results could not be correlated with in vivo data since these experiments can be only performed at pH values close to the physiological one. Literature BBB values were plotted against obtained chromatographic retention factors (Figure 5) and after removing outliers from the correlations the following models for $\log \mathrm{BB}$ and $\log$ PS were built:

$$
\begin{align*}
\log \mathrm{BB} & =0.524(0.084) \log k_{0.8 \%}-0.072(0.058) \\
& \left(n=42, R^{2}=0.496, \mathrm{SD}=0.34, F=39\right) \tag{8}
\end{align*}
$$

$\log \mathrm{PS}=1.149(0.080) \log k_{0.8 \%}-2.286(0.061)$

$$
\begin{equation*}
\left(n=40, R^{2}=0.843, \mathrm{SD}=0.39, F=204\right) \tag{9}
\end{equation*}
$$

As expected from the LFER study, the MELC chromatographic system was not a good surrogate of $\log \mathrm{BB}$, since only $50 \%$ of the variance in $\log \mathrm{BB}$ was predictable from retention factors and the slope of the regression is relatively low. In addition, compounds with extreme $\log$ BB values, either below -1.10 (ritonavir, flurbiprofen, didanosinec, salbutamol, atenolol) or above 1.15 (metoprolol, promazine, haloperidol, fluphenazine), were considered as outliers and thus the model failed in its modeling capacity. The standard deviation of the regression might appear to be acceptable (0.34), but it must be pointed out that the amplitude between
the lowest and the highest $\log \mathrm{BB}$ values is only 2.25 units. In contrast, the chromatographic system explained log PS variance ( $84 \%$ ) better and outliers were distributed along all the biological property range. In this case the standard deviation of the fitting was slightly higher ( 0.39 ), but in relation to a wider scale of $\log$ PS values ( 3.66 units). The presence of a relatively high number of outliers might be explained not only because of differences between biological and chromatographic systems, but also as a consequence of the experimental complexity of in situ brain perfusion experiments. In fact, from single compounds significantly different log PS values can be found in the literature. For instance, this was the case of the outlier sucrose, with reported $\log$ PS values in the range between -5.4 and -3.7 , but also quercetin ( -3.8 and -2.7 ) or quinidine ( -3.7 and -2.7 ). In case of different data from single compounds, averaged $\log$ PS values were considered in the correlations, providing a rough estimate of its accuracy, but unfortunately for some solutes only single results were reported. It is also noteworthy to mention that the chromatographic system was intended to model passive permeation, and thus it should not be applied to molecules that might present any kind of active transport through the BBB.

Application of Eq. (9) to the different forms (neutral, zwitterionic or ionized) of acid and basic drugs of diverse structure means that MELC surrogation of blood-brain perfusion can be extended to all types of drugs regardless of drug charge or structure. Since both solvent media (MELC mobile phase and blood saline plasma) are mainly similar aqueous phases, drugs exhibit similar $\mathrm{p} K_{\mathrm{a}}$ values and degrees of ionization, surrogation can be extended to partially ionized drugs. This is an additional advantage of MELC for surrogation of biological systems over other surrogating HPLC mobile phases containing organic solvents.

## 4. Conclusions

MELC systems of SDS+1-butanol+heptane at pH 7.4 have been characterized and compared to blood brain transport by the Abraham model. Increasing the heptane concentration up to $1.6 \%$ does not significantly changed the properties of the ME. The most relevant factor for solute retention was the molecular volume, suggesting a high affinity of large compounds for the C18 stationary phase. In contrast, dipolar/polarizable analytes and those with hydrogen-bonding basicity interacted preferably with the ME mobile phase, decreasing retention times. The oil concentration seemed to have a minor effect on interactions through $\pi$ - and n-electrons and solute acidity by hydrogen-bonding, reducing retention as well but to a much lesser extent.

A chromatographic system consisting of a Gemini C18 column as stationary phase and a ME made of 50 mM phosphate buffer $\mathrm{pH} 7.4,3.3 \% \mathrm{w} / \mathrm{v}$ SDS, $6.6 \% \mathrm{w} / \mathrm{v}$ of 1-butanol, and $0.8 \% \mathrm{w} / \mathrm{v}$ of heptane as mobile phase is proposed as surrogate model for the rate of BBB penetration, particularly the logarithm of the passive permeability surface area product (log PS). Chromatographic retention factors $(\log k)$ of neutral and ionized drugs are directly and linearly related to $\log$ PS, without the need of any additional correction parameter.

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## CONFLICT OF INTEREST STATEMENT

The authors declare no conflict of interest.

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Table 1. Intrinsic permeability values $\left(\log P_{0}{ }^{\mathrm{BBB}}\right.$ ) [5] and solute descriptors [25] of the compounds used on Eq. (4)

| Compound | $\log P_{0}^{\text {BBB }}$ | E | $S$ | A | $B$ | V |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1-Aminocyclohexanecarboxylic acid* | -5.99 | 0.56 | 0.98 | 0.78 | 0.93 | 1.16 |
| 3-Hydroxyanthranilic acid* | -2.72 | 1.28 | 1.38 | 1.03 | 0.83 | 1.09 |
| 3-Hydroxykyunrenine | -6.49 | 1.70 | 2.19 | 1.31 | 1.71 | 1.63 |
| 5-F-Uracil | -5.67 | 0.97 | 1.29 | 1.17 | 0.99 | 0.77 |
| Acetamide | -5.05 | 0.48 | 0.36 | 0.31 | 0.45 | 0.51 |
| Adenosine* | -4.80 | 2.69 | 2.64 | 0.97 | 2.22 | 1.75 |
| Aldosterone | -5.46 | 2.13 | 3.35 | 0.48 | 1.91 | 2.75 |
| Amantadine* | -1.20 | 0.84 | 0.68 | 0.21 | 0.64 | 1.29 |
| Aminoguanidine | -5.85 | 0.95 | 0.69 | 0.69 | 1.47 | 0.61 |
| Aminopyrine | -3.30 | 1.78 | 1.78 | 0.00 | 1.60 | 1.87 |
| Amitriptyline | -1.48 | 2.25 | 1.78 | 0.00 | 1.00 | 2.40 |
| Amoxapine | -2.75 | 2.25 | 1.68 | 0.16 | 1.43 | 2.25 |
| Anthranilic acid | -4.91 | 1.08 | 1.48 | 0.74 | 0.50 | 1.03 |
| Antipyrine | -4.00 | 1.32 | 1.50 | 0.00 | 1.48 | 1.48 |
| Arabinose | -6.63 | 0.98 | 1.55 | 0.94 | 1.52 | 1.06 |
| Ascorbic acid* | -2.54 | 1.23 | 1.68 | 1.12 | 1.65 | 1.11 |
| Atomoxetine | -1.27 | 1.37 | 1.36 | 0.13 | 0.90 | 2.19 |
| Brompheniramine | -1.70 | 1.70 | 1.57 | 0.00 | 1.02 | 2.26 |
| Bupropion | -2.09 | 1.14 | 1.30 | 0.09 | 1.02 | 1.94 |
| Butanediol | -5.03 | 0.42 | 0.71 | 0.63 | 0.62 | 0.79 |
| Butanol | -2.88 | 0.22 | 0.42 | 0.37 | 0.48 | 0.73 |
| Butyric acid* | -2.15 | 0.21 | 0.64 | 0.61 | 0.45 | 0.75 |
| Caffeine | -3.90 | 1.50 | 1.72 | 0.05 | 1.28 | 1.36 |
| Carbamazepine | -3.74 | 2.15 | 2.11 | 0.53 | 1.10 | 1.81 |
| Carmustine | -3.81 | 0.83 | 2.06 | 0.16 | 0.77 | 1.39 |
| Cetirizine* | -5.80 | 2.05 | 2.24 | 0.57 | 1.76 | 2.94 |
| Chlorambucil* | -0.80 | 1.22 | 1.60 | 0.57 | 0.80 | 2.26 |
| Chlorpheniramine | -1.84 | 1.47 | 1.34 | 0.00 | 1.35 | 2.21 |
| Chlorpromazine | -1.33 | 2.20 | 1.83 | 0.00 | 0.94 | 2.41 |
| Cimetidine | -5.92 | 1.70 | 1.73 | 0.67 | 2.21 | 1.96 |
| Citalopram | -2.07 | 1.66 | 1.87 | 0.00 | 1.08 | 2.53 |
| Clemastine | -0.96 | 1.70 | 1.55 | 0.00 | 0.97 | 2.76 |
| Clozapine | -2.66 | 2.46 | 1.82 | 0.18 | 1.44 | 2.43 |
| Colchicine* | -5.20 | 2.23 | 2.59 | 0.31 | 1.95 | 2.99 |
| Corticosterone | -4.29 | 1.86 | 3.43 | 0.40 | 1.63 | 2.74 |
| Creatinine* | -6.60 | 1.03 | 0.51 | 0.31 | 1.07 | 0.84 |
| DADLE | -6.80 | 3.01 | 5.54 | 2.30 | 3.76 | 4.41 |
| Daunomycine* | -2.40 | 3.59 | 3.53 | 0.93 | 3.06 | 3.67 |
| DDEP | -3.60 | 2.39 | 2.09 | 0.45 | 0.98 | 1.97 |
| DDMP | -3.47 | 2.39 | 2.08 | 0.45 | 0.98 | 1.83 |
| Dianhydrogalactitol | -5.60 | 0.98 | 1.09 | 0.46 | 1.18 | 0.97 |


| Diazepam | -3.30 | 2.08 | 1.57 | 0.00 | 1.25 | 2.07 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Dibromodulcitol | -5.72 | 1.44 | 1.65 | 1.23 | 1.26 | 1.54 |
| Diphenhydramine | -1.94 | 1.36 | 1.43 | 0.00 | 0.95 | 2.19 |
| Donepezil | -1.68 | 2.12 | 2.49 | 0.00 | 1.50 | 3.03 |
| Dopamine* | -2.68 | 1.35 | 1.46 | 1.20 | 1.04 | 1.22 |
| Doxepin | -1.24 | 1.75 | 1.46 | 0.00 | 0.98 | 2.32 |
| Doxorubicin | -4.00 | 3.75 | 3.69 | 1.17 | 3.34 | 3.73 |
| DPDPE | -5.60 | 3.87 | 5.81 | 2.30 | 4.04 | 4.77 |
| Ehylene glycol | -5.30 | 0.40 | 0.90 | 0.58 | 0.78 | 0.51 |
| Ergotamine | -3.82 | 4.63 | 3.87 | 0.85 | 3.56 | 4.21 |
| Erythritol | -6.90 | 0.62 | 1.20 | 0.83 | 1.45 | 0.91 |
| Estradiol | -3.30 | 1.80 | 1.77 | 0.86 | 1.10 | 2.20 |
| Ethanol | -3.40 | 0.25 | 0.42 | 0.37 | 0.48 | 0.45 |
| Ethosuximide | -4.46 | 0.74 | 0.94 | 0.34 | 0.93 | 1.12 |
| Fexofenadine* | -6.60 | 2.72 | 2.48 | 1.20 | 2.12 | 4.09 |
| Fluoxetine | -1.10 | 1.01 | 1.19 | 0.13 | 0.78 | 2.24 |
| Fluphenazine | -3.35 | 2.16 | 2.30 | 0.26 | 1.80 | 3.09 |
| Flurbiprofen* | -0.58 | 1.50 | 1.51 | 0.57 | 0.58 | 1.84 |
| Fluvastatin | -2.28 | 2.75 | 2.48 | 1.20 | 1.46 | 3.13 |
| Formamide | -5.72 | 0.47 | 1.30 | 0.64 | 0.57 | 0.37 |
| Fructose | -6.80 | 1.30 | 1.61 | 1.31 | 1.83 | 1.20 |
| Ftorafur | -5.02 | 1.05 | 1.66 | 0.24 | 1.14 | 1.28 |
| Gabapentin | -4.56 | 0.56 | 0.99 | 0.78 | 0.93 | 1.44 |
| Galactitol | -6.70 | 1.23 | 1.75 | 1.62 | 1.81 | 1.31 |
| Glibenclamide | -3.24 | 2.81 | 2.52 | 0.99 | 2.07 | 3.56 |
| Glucose* | -4.50 | 1.34 | 1.64 | 1.31 | 1.85 | 1.20 |
| Glycerol | -5.40 | 0.51 | 0.76 | 0.47 | 1.43 | 0.71 |
| Glycine | -5.50 | 0.37 | 0.93 | 0.78 | 0.90 | 0.56 |
| Grepafloxacin | -4.86 | 2.23 | 2.43 | 0.73 | 1.88 | 2.59 |
| Guanidine | -5.60 | 0.60 | 0.86 | 0.36 | 1.24 | 0.51 |
| Haloperidol | -2.46 | 1.90 | 1.39 | 0.40 | 1.76 | 2.80 |
| Hexanoic acid* | -1.31 | 0.17 | 0.63 | 0.62 | 0.44 | 1.03 |
| Hispidulin | -3.11 | 2.30 | 2.32 | 0.96 | 1.20 | 2.05 |
| Hydrocortisone | -5.85 | 2.03 | 3.49 | 0.71 | 1.90 | 2.80 |
| Hydroxyzine | -3.04 | 2.00 | 2.21 | 0.10 | 1.89 | 2.92 |
| Hypoxanthine | -5.46 | 1.65 | 1.68 | 0.44 | 1.04 | 0.88 |
| Inulin | -7.35 | 2.28 | 2.60 | 2.01 | 3.41 | 2.23 |
| Iodoacetamide | -4.10 | 1.03 | 1.37 | 0.49 | 0.60 | 0.76 |
| Iodoantipyrine | -3.20 | 2.01 | 1.98 | 0.00 | 1.31 | 1.74 |
| Isocarboxazid* | -3.22 | 1.61 | 2.16 | 0.39 | 1.38 | 1.74 |
| Isopropanol | -3.66 | 0.21 | 0.36 | 0.33 | 0.56 | 0.59 |
| L-Alanine | -5.44 | 0.38 | 0.92 | 0.78 | 0.93 | 0.71 |
| Lamotrigine | -4.67 | 2.27 | 2.03 | 0.35 | 0.96 | 1.65 |
| L-Arginine | -4.64 | 1.06 | 1.24 | 1.26 | 1.95 | 1.38 |

L-Aspartic acid
Levodopa*
L-Glutamic acid
L-Glutamine
L-Histidine*
Lidocaine
L-Isoleucine
L-Kynurenine
L-Lysine
L-Methionine
Lomustine
Loratidine*
L-Ornithine
Lovastatin acid
Lovastatin*
Loxapine
L-Threonine
L-Tryptophan
L-Tyrosine
L-Valine
Mannitol
Maprotiline
Melphalan*
Meprobamate
Mesoridazine*
Methanol
Methotrexate
Methylurea
Metoclopramide
Midazolam
Mirtazapine
Naproxen*
Naringenin
Nicotinamide
Octanoic acid*
Olanzapine
Oxycodone
PCNU
Pemoline
Pentazocine*
Pergolide
Perphenazine
Phenelzine
Phenytoine

| -6.66 | 0.55 | 1.37 | 1.18 | 1.26 | 2 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| -3.90 | 1.33 | 1.77 | 1.56 | 1.44 | 43 |
| -6.26 | 0.55 | 1.37 | 1.35 | 1.26 | . 06 |
| -5.28 | 0.86 | 1.12 | 1.09 | 1.35 | 10 |
| -4.28 | 1.02 | 1.74 | 1.13 | 1.41 | . 13 |
| -3.24 | 1.01 | 1.50 | 0.12 | 1.21 | 2.06 |
| -4.16 | 0.39 | 0.92 | 0.78 | 0.97 | 13 |
| -6.16 | 1.50 | 2.06 | 0.96 | 1.60 | 1.57 |
| -4.93 | 0.58 | 1.26 | 0.99 | 1.48 | 1.23 |
| -4.39 | 0.72 | 1.08 | 0.78 | 1.06 | 15 |
| -4.00 | 0.93 | 2.00 | 0.16 | 0.79 | . 72 |
| -4.00 | 2.19 | 2.09 | 0.00 | 1.14 | 2.87 |
| -4.68 | 0.58 | 1.25 | 0.99 | 1.48 | 1.09 |
| -2.53 | 1.39 | 1.84 | 1.20 | 1.62 | 3.45 |
| -3.42 | 1.38 | 2.34 | 0.31 | 1.44 | 3.29 |
| -3.36 | 2.30 | 1.67 | 0.00 | 1.49 | 2.39 |
| -5.21 | 0.61 | 1.14 | 1.03 | 1.33 | . 91 |
| -4.22 | 1.62 | 1.80 | 1.09 | 1.23 | . 54 |
| -3.90 | 1.18 | 1.60 | 1.28 | 1.29 | . 37 |
| -4.68 | 0.39 | 0.92 | 0.78 | 0.97 | 0.99 |
| -6.90 | 0.84 | 2.26 | 0.86 | 1.79 | 1.31 |
| -0.40 | 1.76 | 1.27 | 0.13 | 0.68 | 2.33 |
| -5.27 | 1.43 | 1.90 | 0.78 | 1.37 | 2.22 |
| -5.09 | 0.71 | 1.62 | 0.89 | 1.12 | . 73 |
| -1.41 | 2.87 | 2.97 | 0.00 | 1.69 | 96 |
| -3.66 | 0.28 | 0.44 | 0.43 | 0.47 | 0.31 |
| -5.40 | 3.51 | 4.23 | 1.85 | 2.82 | 3.22 |
| -5.70 | 0.53 | 1.14 | 0.59 | 0.70 | 0.61 |
| -2.86 | 1.59 | 1.57 | 0.54 | 1.50 | 2.34 |
| -3.11 | 2.57 | 2.01 | 0.00 | 1.38 | 26 |
| -2.75 | 2.08 | 1.67 | 0.00 | 1.22 | 2.11 |
| -0.77 | 1.51 | 1.98 | 0.60 | 0.68 | 78 |
| -3.96 | 2.23 | 2.19 | 1.30 | 1.14 | 1.89 |
| -4.88 | 1.01 | 1.09 | 0.63 | 1.00 | 0.93 |
| -1.14 | 0.15 | 0.65 | 0.62 | 0.45 | 1.31 |
| -2.73 | 2.30 | 1.59 | 0.13 | 1.45 | 2.37 |
| -3.40 | 2.18 | 2.28 | 0.23 | 1.80 | 2.26 |
| -4.86 | 1.47 | 2.72 | 0.50 | 1.66 | . 71 |
| -5.45 | 1.48 | 1.45 | 0.21 | 1.22 | 1.26 |
| -3.69 | 1.54 | 1.13 | 0.50 | 1.04 | 2.45 |
| -1.14 | 2.22 | 1.48 | 0.31 | 1.01 | 2.54 |
| -2.61 | 2.87 | 2.33 | 0.23 | 1.84 | 3.02 |
| -4.32 | 0.98 | 1.02 | 0.34 | 0.99 | 1.20 |
| -4.09 | 1.71 | 2.19 | 0.85 | 1.00 | 1.87 |


| Pramipexole | -2.70 | 1.35 | 1.35 | 0.36 | 0.97 | 1.68 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Procarbazine | -4.62 | 1.22 | 1.79 | 0.52 | 1.59 | 1.88 |
| Progesterone | -3.74 | 1.45 | 3.29 | 0.00 | 1.14 | 2.62 |
| Propranolol* | -1.30 | 1.88 | 1.43 | 0.17 | 1.42 | 2.15 |
| Propylene glycol | -4.49 | 0.37 | 0.90 | 0.58 | 0.80 | 0.65 |
| Pyrilamine | -2.90 | 1.82 | 1.92 | 0.00 | 1.59 | 2.39 |
| Pyrimethamine | -3.57 | 1.90 | 0.98 | 0.34 | 1.36 | 1.85 |
| Quercetin | -4.70 | 2.68 | 2.64 | 1.88 | 1.63 | 1.96 |
| Quetiapine | -3.06 | 2.72 | 1.93 | 0.23 | 2.01 | 2.91 |
| Quinidine | -3.90 | 2.40 | 1.71 | 0.23 | 1.81 | 2.55 |
| Quinine | -3.45 | 2.47 | 1.23 | 0.37 | 1.97 | 2.55 |
| Quinolinic acid | -6.26 | 0.99 | 1.59 | 1.14 | 1.06 | 1.11 |
| Rimantadine* | 0.13 | 0.84 | 0.67 | 0.21 | 0.68 | 1.57 |
| Rimonabant | -3.60 | 3.38 | 3.13 | 0.26 | 1.55 | 3.21 |
| Risperidone | -2.94 | 2.59 | 2.23 | 0.00 | 1.70 | 3.04 |
| Rizatriptan | -4.43 | 2.21 | 2.05 | 0.31 | 1.28 | 2.14 |
| Salicylic acid* | -1.02 | 0.89 | 0.84 | 0.71 | 0.38 | 0.99 |
| Selegiline | -3.12 | 1.00 | 1.00 | 0.09 | 0.71 | 1.72 |
| Sucrose | -6.90 | 1.97 | 2.50 | 2.10 | 3.00 | 2.23 |
| Sumatriptan | -5.06 | 1.87 | 2.28 | 0.85 | 1.88 | 2.27 |
| Tacrine | -1.51 | 1.61 | 0.94 | 0.13 | 0.61 | 1.60 |
| Temazepam | -3.35 | 2.29 | 1.55 | 0.12 | 1.70 | 2.13 |
| Terfenadine | -0.92 | 2.55 | 2.04 | 0.63 | 1.80 | 4.01 |
| Testosterone | -3.40 | 1.54 | 2.59 | 0.32 | 1.19 | 2.38 |
| Theobromine | -5.00 | 1.50 | 1.60 | 0.50 | 1.38 | 1.22 |
| Theophylline | -5.00 | 1.50 | 1.60 | 0.54 | 1.34 | 1.22 |
| Thioridazine | -1.95 | 2.70 | 2.10 | 0.00 | 1.30 | 2.90 |
| Thiothixene | -2.35 | 2.94 | 2.59 | 0.00 | 2.19 | 3.36 |
| Thiourea | -5.50 | 0.84 | 0.82 | 0.77 | 0.87 | 0.57 |
| Thymidine | -5.84 | 1.78 | 2.01 | 0.81 | 2.11 | 1.66 |
| Thymine | -3.93 | 1.09 | 1.23 | 1.00 | 1.01 | 0.89 |
| Tiagabine* | -4.45 | 1.77 | 1.60 | 0.57 | 1.02 | 2.89 |
| Tolbutamide | -2.64 | 1.44 | 1.61 | 0.68 | 1.33 | 2.06 |
| Trazodone | -3.13 | 2.64 | 2.47 | 0.00 | 1.92 | 2.73 |
| Trifluoperazine | -3.00 | 2.00 | 1.80 | 0.00 | 1.50 | 2.89 |
| Trimethylene glycol | -5.40 | 0.40 | 0.91 | 0.77 | 0.85 | 0.65 |
| TYR-MIF-1 | -5.78 | 2.59 | 4.80 | 1.71 | 3.30 | 3.48 |
| Urea | -6.00 | 0.74 | 0.57 | 0.52 | 0.87 | 0.46 |
| Valproic acid | -2.00 | 0.18 | 0.60 | 0.61 | 0.45 | 1.31 |
| Warfarin* | -1.56 | 2.30 | 2.18 | 0.35 | 1.49 | 2.31 |
| Xanthine | -5.62 | 1.50 | 1.60 | 0.97 | 1.07 | 0.94 |
| Zaleplon | -4.25 | 2.36 | 2.60 | 0.00 | 1.42 | 2.31 |
| Ziprasidone | -3.25 | 3.38 | 2.67 | 0.48 | 1.65 | 2.92 |
|  |  | 0.15 | 0.36 | 0.00 | 0.38 | 0.31 |


| Maximum | 0.13 | 4.63 | 5.81 | 2.30 | 4.04 | 4.77 |
| ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| Median | -3.90 | 1.50 | 1.63 | 0.52 | 1.28 | 1.84 |
| Average | -3.89 | 1.58 | 1.76 | 0.60 | 1.36 | 1.89 |
| SD | 1.66 | 0.85 | 0.87 | 0.50 | 0.64 | 0.92 |

Average $\log P_{0}^{\mathrm{BBB}}$ is reported in case of different literature values for the same compound. Experimental molecular descriptors marked in bold.
*Compounds excluded from correlation.

Table 2. Molecular descriptors [25] and measured retention factors of the compounds used for the characterization of the chromatographic systems containing $0,0.8$, and $1.6 \%$ of heptane ( $\mathrm{w} / \mathrm{v}$ ).

| Compound | Molecular descriptors ${ }^{\text {a }}$ |  |  |  |  | $\log k^{\text {b }}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | E | S | A | B | V | 0\% | 0.8\% | 1.6\% |
| 1,2,4-trimethylbenzene | 0.68 | 0.56 | 0.00 | 0.19 | 1.14 | 1.119 | 1.116 | 1.071 |
| 2-nitroanisole | 0.97 | 1.34 | 0.00 | 0.45 | 1.09 | 0.419 | 0.425 | 0.380 |
| 4-chloroacetanilide | 0.98 | 1.47 | 0.64 | 0.51 | 1.24 | 0.358 | 0.346 | 0.199 |
| Acetamide | 0.46 | 1.30 | 0.54 | 0.68 | 0.51 | -1.200 | -0.848 | -0.888 |
| Acetanilide | 0.90 | 1.39 | 0.48 | 0.67 | 1.11 | 0.057 | 0.070 | -0.022 |
| Acetophenone | 0.82 | 1.01 | 0.00 | 0.48 | 1.01 | 0.410 | 0.449 | 0.440 |
| Aminopyrene | 1.78 | 1.78 | 0.00 | 1.60 | 1.87 | -0.162 | -0.120 | -0.182 |
| Anisole | 0.71 | 0.75 | 0.00 | 0.29 | 0.92 | 0.788 | 0.827 | 0.832 |
| Anthracene | 2.29 | 1.34 | 0.00 | 0.28 | 1.45 | 1.144 | 1.081 | 1.027 |
| Antipyrine | 1.32 | 1.50 | 0.00 | 1.48 | 1.48 | -0.365 | -0.281 | -0.366 |
| Benzaldehyde | 0.82 | 1.00 | 0.00 | 0.39 | 0.87 | 0.390 | 0.436 | 0.442 |
| Benzamide | 0.99 | 1.50 | 0.49 | 0.67 | 0.97 | -0.120 | -0.106 | -0.192 |
| Benzene | 0.61 | 0.52 | 0.00 | 0.14 | 0.72 | 0.866 | 0.908 | 0.928 |
| Benzofuran | 0.89 | 0.83 | 0.00 | 0.15 | 0.91 | 0.934 | 0.936 | 0.919 |
| Benzyl alcohol | 0.80 | 0.87 | 0.39 | 0.56 | 0.92 | 0.104 | 0.124 | 0.046 |
| Bromobenzene | 0.88 | 0.73 | 0.00 | 0.09 | 0.89 | 1.000 | 0.991 | 0.977 |
| Butanone | 0.17 | 0.70 | 0.00 | 0.51 | 0.69 | -0.297 | -0.178 | -0.162 |
| Butylbenzene | 0.60 | 0.51 | 0.00 | 0.15 | 1.28 | 1.170 | 1.149 | 1.096 |
| Butyrophenone | 0.80 | 0.95 | 0.00 | 0.51 | 1.30 | 0.839 | 0.838 | 0.815 |
| Caffeine | 1.50 | 1.72 | 0.05 | 1.28 | 1.36 | -0.663 | -0.533 | -0.606 |
| Carbamazepine | 2.15 | 2.11 | 0.53 | 1.10 | 1.81 | 0.212 | 0.197 | 0.068 |
| Celecoxib | 2.51 | 2.43 | 0.44 | 1.22 | 2.47 | 0.732 | 0.653 | 0.447 |
| Cortisone | 1.96 | 3.50 | 0.36 | 1.87 | 2.76 | 0.008 | -0.004 | -0.105 |
| Coumarin | 1.06 | 1.76 | 0.00 | 0.43 | 1.06 | 0.219 | 0.224 | 0.163 |
| Diazepam | 2.08 | 1.57 | 0.00 | 1.25 | 2.07 | 0.459 | 0.424 | 0.274 |
| Ethylbenzene | 0.61 | 0.51 | 0.00 | 0.15 | 1.00 | 1.061 | 1.052 | 1.037 |
| Flunitrazepam | 2.10 | 2.15 | 0.00 | 1.48 | 2.14 | 0.333 | 0.314 | 0.180 |
| Hydrocortisone | 2.03 | 3.50 | 0.71 | 1.90 | 2.80 | 0.041 | 0.026 | -0.109 |
| Lamotrigine | 2.27 | 2.03 | 0.35 | 0.96 | 1.65 | 0.153 | 0.186 | 0.074 |
| Loratadine | 2.19 | 2.09 | 0.00 | 1.14 | 2.87 | 0.857 | 0.772 | 0.548 |
| $\mathrm{N}, \mathrm{N}$-dimethylacetamide | 0.36 | 1.35 | 0.00 | 0.77 | 0.79 | -0.954 | -0.710 | -0.782 |
| Naphthalene | 1.34 | 0.92 | 0.00 | 0.20 | 1.09 | 1.043 | 1.019 | 0.997 |
| Nitrobenzene | 0.87 | 1.11 | 0.00 | 0.28 | 0.89 | 0.621 | 0.631 | 0.631 |
| N -phenylurea | 1.11 | 1.33 | 0.79 | 0.79 | 1.07 | 0.005 | 0.020 | -0.084 |
| Omeprazole | 2.67 | 3.18 | 0.35 | 2.05 | 2.52 | 0.349 | 0.346 | 0.186 |
| Paracetamol | 1.06 | 1.63 | 1.04 | 0.86 | 1.17 | -0.609 | -0.461 | -0.547 |
| Pentachloronitrobenzene | 1.47 | 1.70 | 0.00 | 0.01 | 1.50 | 1.248 | 1.203 | 1.113 |
| Phenanthrene | 2.06 | 1.29 | 0.00 | 0.29 | 1.45 | 1.132 | 1.095 | 1.018 |
| Prednisolone | 2.21 | 3.10 | 0.71 | 1.92 | 2.76 | 0.066 | 0.034 | -0.083 |
| Pregnenolone | 1.36 | 3.29 | 0.32 | 1.18 | 2.67 | 0.778 | 0.700 | 0.519 |
| Progesterone | 1.45 | 3.29 | 0.00 | 1.14 | 2.62 | 0.671 | 0.606 | 0.473 |
| Propiophenone | 0.80 | 0.95 | 0.00 | 0.51 | 1.16 | 0.666 | 0.683 | 0.685 |
| Propylbenzene | 0.60 | 0.50 | 0.00 | 0.15 | 1.14 | 1.127 | 1.119 | 1.073 |


| Pyrene | 2.60 | 1.52 | 0.00 | 0.25 | 1.59 | 1.156 | 1.111 | 1.021 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Pyrrole | $\mathbf{0 . 6 1}$ | $\mathbf{0 . 9 1}$ | $\mathbf{0 . 2 2}$ | $\mathbf{0 . 2 5}$ | 0.58 | -0.039 | 0.012 | -0.031 |
| Riluzole | 1.36 | 1.45 | 0.23 | 0.67 | 1.32 | 0.584 | 0.588 | 0.422 |
| Rofecoxib | 1.66 | 2.43 | 0.00 | 1.15 | 2.23 | 0.093 | 0.120 | 0.000 |
| Theophylline | $\mathbf{1 . 5 0}$ | $\mathbf{1 . 6 0}$ | $\mathbf{0 . 5 4}$ | $\mathbf{1 . 3 4}$ | 1.22 | -0.812 | -0.587 | -0.680 |
| Toluene | $\mathbf{0 . 6 0}$ | $\mathbf{0 . 5 2}$ | $\mathbf{0 . 0 0}$ | $\mathbf{0 . 1 4}$ | 0.86 | 0.985 | 0.998 | 0.996 |
| Valerophenone |  | $\mathbf{0 . 8 0}$ | $\mathbf{0 . 9 5}$ | $\mathbf{0 . 0 0}$ | $\mathbf{0 . 5 0}$ | 1.44 | 0.957 | 0.940 |
|  | Minimum | 0.17 | 0.50 | 0.00 | 0.01 | 0.51 | -1.200 | -0.848 |
|  | Maximum | 2.67 | 3.50 | 1.04 | 2.05 | 2.87 | 1.248 | 1.203 |

[^0]Table 3. LFER system coefficients of unit vectors.

|  | $e_{u}$ | $s_{u}$ | $a_{u}$ | $b_{u}$ | $v_{u}$ |
| :--- | :---: | :---: | :---: | :---: | :---: |
| $\log$ BB | 0.308 | -0.534 | -0.436 | -0.401 | 0.519 |
| $\log$ PS | 0.097 | -0.314 | -0.590 | -0.510 | 0.532 |
| $\log P_{0}{ }^{\text {BBB }}$ | 0.080 | -0.353 | -0.163 | -0.558 | 0.728 |
| $\log k_{\text {MP3 }}$ | 0.059 | -0.373 | -0.533 | -0.550 | 0.521 |
| $\log k_{1.6 \%}$ | -0.006 | -0.241 | -0.163 | -0.660 | 0.692 |
| $\log k_{0.8 \%}$ | -0.006 | -0.237 | -0.136 | -0.652 | 0.708 |
| $\log k_{0.0 \%}$ | -0.009 | -0.203 | -0.096 | -0.687 | 0.691 |

Table 4. Distances between pairs of studied systems.

|  | $\log B B$ | $\log$ PS | $\log P_{0}{ }^{\text {BBB }}$ | $\log k_{\text {MP3 }}$ | $\log k_{1.6 \%}$ | $\log k_{0.8 \%}$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| $\log \mathrm{BB}$ | 0 | - | - | - | - | - |
| $\log$ PS | 0.358 | 0 | - | - | - | - |
| $\log P_{0}{ }^{\text {BBB }}$ | 0.477 | 0.474 | 0 | - | - | - |
| $\log k_{\text {MP3 }}$ | 0.345 | 0.100 | 0.425 | 0 | - | - |
| $\log k_{1.6 \%}$ | 0.597 | 0.497 | 0.178 | 0.447 | 0 | - |
| $\log k_{0.8 \%}$ | 0.612 | 0.523 | 0.175 | 0.475 | 0.033 | 0 |
| $\log k_{0} \%$ | 0.661 | 0.570 | 0.230 | 0.522 | 0.082 | 0.066 |

Table 5. Biological $\log$ BB values [8] and their corresponding measured retention factors in the chromatographic system containing $0.8 \%$ of heptane ( $\mathrm{w} / \mathrm{v}$ ).

| Compounds | $\log$ BB | $\log \mathrm{k}_{0.8 \%}{ }^{\text {a }}$ | Ionization at $\mathrm{pH} 7.4{ }^{\text {b }}$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | Neutral | Zwitterionic | Negative | Positive |
| 1,2,4-trimethylbenzene | 0.16 | 1.116 | 100\% | 0\% | 0\% | 0\% |
| 1,2-dimethylbenzene | 0.30 | 1.064 | 100\% | 0\% | 0\% | 0\% |
| 1,3-dimethylbenzene | 0.29 | 1.049 | 100\% | 0\% | 0\% | 0\% |
| 1,4-dimethylbenzene | 0.31 | 1.045 | 100\% | 0\% | 0\% | 0\% |
| Acetazolamide | -0.52 | -0.934 | 43\% | 0\% | 57\% | 0\% |
| Acyclovir | -0.50 | -0.906 | 78\% | 0\% | 0\% | 22\% |
| Alprenolol | -0.23 | 0.779 | 1\% | 0\% | 0\% | 99\% |
| Aminopyrene | 0.00 | -0.120 | 100\% | 0\% | 0\% | 0\% |
| Amiodarone ${ }^{\text {c }}$ | -1.08 | 1.002 | 4\% | 0\% | 0\% | 96\% |
| Amitriptyline | 0.90 | 0.860 | 4\% | 0\% | 0\% | 96\% |
| Amprenavir | -0.56 | 0.271 | 100\% | 0\% | 0\% | 0\% |
| Antipyrine | -0.10 | -0.281 | 100\% | 0\% | 0\% | 0\% |
| Atenolol ${ }^{\text {c }}$ | -1.12 | -0.012 | 1\% | 0\% | 0\% | 99\% |
| Atropine | -0.06 | 0.383 | 1\% | 0\% | 0\% | 99\% |
| Barbital | -0.14 | -0.221 | 100\% | 0\% | 0\% | 0\% |
| Benzene | 0.37 | 0.908 | 100\% | 0\% | 0\% | 0\% |
| Butanone | -0.08 | -0.178 | 100\% | 0\% | 0\% | 0\% |
| Caffeine | -0.06 | -0.533 | 100\% | 0\% | 0\% | 0\% |
| Carbamazepine | -0.11 | 0.197 | 100\% | 0\% | 0\% | 0\% |
| Celecoxib ${ }^{\text {c }}$ | -1.00 | 0.653 | 100\% | 0\% | 0\% | 0\% |
| Clobazam | 0.35 | 0.274 | 100\% | 0\% | 0\% | 0\% |
| Codeine | 0.55 | 0.142 | 13\% | 0\% | 0\% | 87\% |
| Diazepam | 0.48 | 0.424 | 100\% | 0\% | 0\% | 0\% |
| Didanosine ${ }^{\text {c }}$ | -1.30 | -0.832 | 97\% | 0\% | 3\% | 0\% |
| Ethylbenzene | 0.20 | 1.052 | 100\% | 0\% | 0\% | 0\% |
| Flunitrazepam | 0.06 | 0.314 | 100\% | 0\% | 0\% | 0\% |
| Fluphenazine ${ }^{\text {c }}$ | 1.51 | 0.851 | 25\% | 0\% | 0\% | 75\% |
| Flurbiprofen ${ }^{\text {c }}$ | -1.68 | -0.011 | 0\% | 0\% | 100\% | 0\% |
| Haloperidol ${ }^{\text {c }}$ | 1.32 | 1.090 | 5\% | 0\% | 0\% | 95\% |
| Ibuprofen | -0.18 | 0.061 | 0\% | 0\% | 100\% | 0\% |
| Imipramine | 1.01 | 0.841 | 2\% | 0\% | 0\% | 98\% |
| Lamotrigine | 0.36 | 0.186 | 0\% | 0\% | 0\% | 100\% |
| Lidocaine | 0.34 | 0.585 | 14\% | 0\% | 0\% | 86\% |
| Metoprolol ${ }^{\text {c }}$ | 1.15 | 0.536 | 1\% | 0\% | 0\% | 99\% |
| Nevirapine | 0.00 | 0.151 | 100\% | 0\% | 0\% | 0\% |
| Nicotine | 0.56 | 0.129 | 11\% | 0\% | 0\% | 89\% |
| Paracetamol | -0.42 | -0.461 | 100\% | 0\% | 0\% | 0\% |
| Pindolol | -0.15 | 0.312 | 1\% | 0\% | 0\% | 99\% |
| Promazine ${ }^{\text {c }}$ | 1.23 | 0.786 | 2\% | 0\% | 0\% | 98\% |
| Propranolol | 0.88 | 0.685 | 1\% | 0\% | 0\% | 99\% |
| Pyrene | 0.23 | 1.111 | 100\% | 0\% | 0\% | 0\% |
| Pyrilamine | 0.49 | 0.632 | 4\% | 0\% | 0\% | 96\% |
| Quinidine | -0.32 | 0.733 | 6\% | 0\% | 0\% | 94\% |
| Riluzole | 0.30 | 0.588 | 100\% | 0\% | 0\% | 0\% |


| Ritonavir $^{\mathrm{c}}$ | -1.82 | 0.831 | $100 \%$ | $0 \%$ | $0 \%$ | $0 \%$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| Salbutamol $^{\mathrm{c}}$ | -1.17 | 0.100 | $2 \%$ | $1 \%$ | $0 \%$ | $97 \%$ |
| Salicylic acid $_{\text {Saquinavir }^{\mathrm{c}}}$ | -1.10 | -1.064 | $0 \%$ | $0 \%$ | $100 \%$ | $0 \%$ |
| Stavudine | -0.86 | 0.901 | $87 \%$ | $0 \%$ | $0 \%$ | $13 \%$ |
| Terfenadine | -0.48 | -0.793 | $93 \%$ | $0 \%$ | $7 \%$ | $0 \%$ |
| Theophylline | 1.15 | 1.300 | $1 \%$ | $0 \%$ | $0 \%$ | $99 \%$ |
| Toluene | -0.31 | -0.587 | $95 \%$ | $0 \%$ | $5 \%$ | $0 \%$ |
| Trazodone | 0.37 | 0.998 | $100 \%$ | $0 \%$ | $0 \%$ | $0 \%$ |
| Zidovudine | -0.22 | 0.484 | $83 \%$ | $0 \%$ | $0 \%$ | $17 \%$ |

${ }^{\text {a }}$ Mean $\log k$ values obtained from triplicate injections, with SD below 0.01 in all cases.
${ }^{\mathrm{b}}$ Neutral, zwitterionic, negative and positive fraction calculated from GALAS algorithm [25].
${ }^{\text {c Excluded from correlation on Eq. (8). }}$

Table 6. Biological log PS values [5] and their corresponding measured retention factors in the chromatographic system containing $0.8 \%$ of heptane ( $\mathrm{w} / \mathrm{v}$ ).

| Compounds | $\log$ PS | $\log k_{0.8 \%}{ }^{\text {a }}$ | Ionization at $\mathrm{pH} 7.4{ }^{\text {b }}$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | Neutral | Zwitterionic | Negative | Positive |
| 5-F-Uracil | -3.77 | -0.916 | 72\% | 0\% | 28\% | 0\% |
| Acetamide | -3.05 | -0.848 | 100\% | 0\% | 0\% | 0\% |
| Aminopyrine ${ }^{\text {c }}$ | -1.30 | -0.120 | 100\% | 0\% | 0\% | 0\% |
| Amitriptyline | -1.02 | 0.860 | 4\% | 0\% | 0\% | 96\% |
| Anthranilic acid ${ }^{\text {c }}$ | -2.92 | -1.389 | 0\% | 1\% | 0\% | 99\% |
| Antipyrine | -1.94 | -0.281 | 100\% | 0\% | 0\% | 0\% |
| Caffeine ${ }^{\text {c }}$ | -1.83 | -0.533 | 100\% | 0\% | 0\% | 0\% |
| Carbamazepine | -1.74 | 0.197 | 100\% | 0\% | 0\% | 0\% |
| Corticosterone | -2.28 | 0.125 | 100\% | 0\% | 0\% | 0\% |
| Diazepam | -1.27 | 0.424 | 100\% | 0\% | 0\% | 0\% |
| Diphenhydramine | -1.24 | 0.739 | 5\% | 0\% | 0\% | 95\% |
| Estradiol | -1.08 | 0.393 | 100\% | 0\% | 0\% | 0\% |
| Fluphenazine | -1.87 | 0.851 | 25\% | 0\% | 0\% | 75\% |
| Flurbiprofen | -1.80 | -0.011 | 0\% | 0\% | 100\% | 0\% |
| Formamide | -3.72 | -0.818 | 100\% | 0\% | 0\% | 0\% |
| Glibenclamide | -2.77 | 0.117 | 7\% | 0\% | 93\% | 0\% |
| Glycine | -3.49 | -0.930 | 0\% | 100\% | 0\% | 0\% |
| Haloperidol | -1.45 | 1.090 | 5\% | 0\% | 0\% | 95\% |
| Hydrocortisone ${ }^{\text {c }}$ | -3.85 | 0.026 | 100\% | 0\% | 0\% | 0\% |
| Ibuprofen | -2.03 | 0.061 | 0\% | 0\% | 100\% | 0\% |
| Indinavir ${ }^{\text {c }}$ | -3.73 | 0.559 | 96\% | 0\% | 0\% | 4\% |
| L-Alanine ${ }^{\text {c }}$ | -3.44 | -0.109 | 0\% | 100\% | 0\% | 0\% |
| Lamotrigine | -2.68 | 0.186 | 0\% | 0\% | 0\% | 100\% |
| L-Arginine | -2.64 | -0.127 | 0\% | 3\% | 0\% | 97\% |
| L-Aspartic acid | -4.66 | -1.856 | 0\% | 0\% | 100\% | 0\% |
| L-Glutamic acid | -4.26 | -1.898 | 0\% | 0\% | 100\% | 0\% |
| L-Glutamine | -3.28 | -0.963 | 0\% | 99\% | 1\% | 0\% |
| Lidocaine | -1.90 | 0.585 | 14\% | 0\% | 0\% | 86\% |
| L-Lysine | -2.92 | -0.115 | 0\% | 0\% | 100\% | 0\% |
| L-Tryptophan | -2.23 | -0.260 | 0\% | 98\% | 2\% | 0\% |
| L-Tyrosine ${ }^{\text {c }}$ | -1.91 | -0.777 | 0\% | 98\% | 2\% | 0\% |
| L-Valine | -2.68 | -0.113 | 0\% | 100\% | 0\% | 0\% |
| Maprotiline | -1.35 | 0.889 | 0\% | 0\% | 0\% | 100\% |
| Naringenin | -1.98 | 0.053 | 59\% | 0\% | 41\% | 0\% |
| Nicotinamide | -2.88 | -0.635 | 98\% | 0\% | 0\% | 2\% |
| Perphenazine | -1.25 | 0.805 | 25\% | 0\% | 0\% | 75\% |
| Progesterone | -1.74 | 0.606 | 100\% | 0\% | 0\% | 0\% |
| Propranolol | -1.00 | 0.685 | 1\% | 0\% | 0\% | 99\% |
| Pyrilamine | -1.82 | 0.632 | 4\% | 0\% | 0\% | 96\% |
| Quercetin ${ }^{\text {c }}$ | -3.05 | 0.122 | 64\% | 0\% | 36\% | 0\% |
| Quetiapine | -1.31 | 0.543 | 99\% | 0\% | 0\% | 1\% |
| Quinidine ${ }^{\text {c }}$ | -2.92 | 0.733 | 6\% | 0\% | 0\% | 94\% |
| Salicylic acid | -3.40 | -1.064 | 0\% | 0\% | 100\% | 0\% |
| Sucrose ${ }^{\text {c }}$ | -4.52 | 0.271 | 100\% | 0\% | 0\% | 0\% |


| Terfenadine | -1.39 | 1.300 | $1 \%$ | $0 \%$ | $0 \%$ | $99 \%$ |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| Testosterone | -1.31 | 0.387 | $100 \%$ | $0 \%$ | $0 \%$ | $0 \%$ |
| Theophylline | -2.96 | -0.587 | $95 \%$ | $0 \%$ | $5 \%$ | $0 \%$ |
| Thiourea | -3.52 | -0.768 | $100 \%$ | $0 \%$ | $0 \%$ | $0 \%$ |
| Thymine | -1.93 | -0.632 | $99 \%$ | $0 \%$ | $1 \%$ | $0 \%$ |
| Trazodone | -1.46 | 0.484 | $83 \%$ | $0 \%$ | $0 \%$ | $17 \%$ |
| Verapamil | -1.76 | 0.644 | $5 \%$ | $0 \%$ | $0 \%$ | $95 \%$ |

${ }^{\mathrm{a}}$ Mean $\log k$ values obtained from triplicate injections, with SD below 0.01 in all cases.
${ }^{\mathrm{b}}$ Neutral, zwitterionic, negative and positive fraction calculated from GALAS algorithm [25].
${ }^{\text {c }}$ Excluded from correlation on Eq. (9).
PS in units of $10^{-4} \mathrm{~mL} \mathrm{~g} \mathrm{~g}^{-1} \mathrm{~s}^{-1}$

## FIGURE CAPTIONS

Figure 1. Plot of observed vs. calculated intrinsic permeability values. Empty symbols show compounds excluded from correlation.

Figure 2. Plot of observed vs. calculated retention factors of the assayed chromatographic systems. Empty symbols show compounds excluded from correlations (loratadine, N,Ndimethylacetamide, omeprazole, and rofecoxib were excluded in all systems).

Figure 3. Plot of the first two scores of the PCA of the compared biological (Eqs. 2-4) and chromatographic BBB systems (Eq. 5-7 and log $\mathrm{kmp}_{\mathrm{MP}}$ [22]).

Figure 4. Joint PCA analysis of compounds used in $\log$ PS (empty squares) and $\log P_{0}{ }^{\mathrm{BBB}}$ (full circles) correlations.

Figure 5. Plot of biological BBB distribution $(\log \mathrm{BB})$ and permeation $(\log \mathrm{PS})$ values vs. retention factors obtained for the chromatographic system containing a $0.8 \%$ of heptane at pH 7.4. Legend: ( $\bullet$ ) unionized, ( $\boldsymbol{\nabla}$ ) zwitterionic, ( $\mathbf{\square}$ ) totally or partially negatively charged, ( $\mathbf{\Delta}$ ) totally or partially positively charged, and (x) compounds excluded from correlations.

Figure 1


Figure 2




Figure 3


Figure 4


Figure 5




[^0]:    ${ }^{\text {a }}$ Experimental molecular descriptors marked in bold.
    ${ }^{\mathrm{b}}$ Mean $\log k$ values obtained from triplicate injections, with SD below 0.01 in all cases.

