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# CENTRAL MEDITERRANEAN PHOENICIAN POTTERY IMPORTS IN THE NORTHEASTERN IBERIAN PENINSULA

E. Miguel Gascón, J. Buxeda i Garrigós, P. M. Day

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| 3                    | <b>IBERIAN PENINSULA</b>  |
| 4                    | Eva Miguel Gascón <sup>*1</sup> , Jaume Buxeda i Garrigós <sup>1</sup> and Peter M. Day <sup>2</sup>  |
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| 7<br>8               | <sup>•</sup> Cultura Material i Arqueometria UB (ARQUB, GRACPE), Dept. de Prehistòria, Història Antiga i<br>Arqueologia, Universitat de Barcelona, Montalegre 6, 08001, Barcelona, Catalonia, Spain |
| 9                    | <sup>2</sup> Department of Archaeology, University of Sheffield, Northgate House, West Street, Sheffield S1   |
| 10                   | 4ET, United Kingdom   |
| 11                   |   |
| 12                   | *Corresponding author. Tel.: +34 685532585  |
| 13                   | E-mail addresses: evamigascon@gmail.com (E. Miguel Gascón), jbuxeda@ub.edu (J. Buxeda i   |
| 14                   | Garrigós), <u>p.m.day@sheffield.ac.uk</u> (P. M. Day).  |
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| 16                   | ABSTRACT  |
| 17                   | Over recent years, there has been a growing interest in the analytical investigation of   |
| 18                   | Phoenician pottery recovered from sites in Catalonia (NE Iberian Peninsula). Studies which  |
| 19                   | integrate mineralogical, chemical and microstructural analysis have been carried out at seven   |
| 20                   | sites in the Ilercavonia and Cossetania areas, analyzing a total of 123 ceramic samples. The  |
| 21                   | characterization of these samples has confirmed the presence of Phoenician Central  |
| 22                   | The main chiestives of this paper are to determine the provenance of these products to study  |
| 23<br>24             | their mineralogical characteristics and to understand the consumption of this Phoenician Central  |
| 2 <del>4</del><br>25 | Mediterranean pottery in the context of the sites of Ilercayonia and Cossetania   |
| 26                   | All individuals have been analyzed by means of X-ray fluorescence (XRF) and X-ray   |
| 27                   | diffraction (XRD), with selected samples analyzed by means of thin-section petrography and  |
| 28                   | scanning electron microscopy (SEM). The results of this study demonstrate the presence of   |
| 29                   | Sicilian, Sardinian and Tunisian products, allowing us to see preferences of vessel types   |
| 30                   | according to source.  |
| 31                   |   |
| 32                   | KEYWORDS: Iron Age; Phoenician commerce; X-ray fluorescence; X-ray diffraction; thin-   |
| 33                   | section petrography; scanning electron microscopy   |

### 35 1. INTRODUCTION

Since 2002, analyses have been performed at the Universitat de Barcelona on the first wheelthrown pottery found in indigenous contexts of the VIII century – 575 BC in the regions of Ilercavonia and Cossetania (Catalonia). Traditionally, this pottery has been thought to have been manufactured by potters located in a variety of Western Phoenician colonies.

40 Most of this pottery comprises amphorae whose provenance has been associated with the 41 Circle of the Strait of Gibraltar, where the ancient Phoenician colonies (Aubet, 2009) of the 42 Iberian Peninsula and Atlantic Sea were located (Figure 1). In the last years, a complex 43 Phoenician cabotage system has been suggested (Rafel, 2013), starting from these Southern 44 Andalucian sites, reaching along the Iberian Peninsula coast to the indigenous communities 45 situated in Catalonia. This contact seems to have been commercial in nature, since there are no 46 Phoenican colonies in the Ilercavonia and Cossetania regions. In fact, it is in the province of 47 Málaga in Andalucia where intense archaeological work has revealed an extensive list of 48 Phoenician sites, settlements and production centers (Toscanos, Cerro del Villar, La Pancha, 49 Los Algarrobeños, Chorreras, Morro de Mezquitilla) and most studies of Phoenician pottery 50 found in Catalonia point to this area as its source (Garcia and Gracia, 2011). However, amongst the Phoenician pottery found in the Cossetania and Ilercavonia regions of Catalonia, a small 51 52 group, mainly tableware, is not related to a source area of the Circle of the Strait of Gibraltar.

53 Macroscopic study has proved of limited value in ascribing provenance to these vessels. As a 54 result one of the main objectives of the analytical research reported here is to define the 55 different Central Mediterranean Phoenician products detected at three sites in Northeast Iberia 56 (Figure 2): Sant Jaume – Mas d'en Serrà, La Ferradura (Ilercavonia) and Turó de la Font de la 57 Canya (Cossetania). In this paper we present some of the results of Miguel Gascón's 58 unpublished Doctoral Thesis (Miguel Gascón, 2014).

The site of Sant Jaume - Mas d'en Serrà (Garcia and Gracia, 2011) has different types of pottery related to Central Mediterranean production. There are three narrow-necked cylindrical jars (MOS026, MOS040 and MOS060) and a carinated red slipped bowl (MOS037). The settlement features a defensive system unique in the whole of the Northeastern Iberian Peninsula during the Early Iron Age (Garcia, 2009). It seems that in Catalonia, Sant Jaume was one of the centres with the strongest trade relations with the Phoenician world, thus up to 30% of the wheel-made pottery excavated is related to Phoenician imports (Garcia *et al.*, 2015).

66 The other Ilercavonian site included in this study is La Ferradura (Maluquer, 1983), where 67 only one indeterminate pot was related to a Central Mediterranean provenance (FER005). The 68 settlement hosts a complex of eleven rooms, some clearly related to metallurgical activity. Most

69 Iron Age sites of Ilercavonia are related to metallurgy, which seems to be closely connected 70 with Phoenician commerce.

71 In the case of Turó de la Font de la Canya (Asensio et al., 2005), four narrow-necked 72 cylindrical jars (TFC025, TFC036, TFC040, and TFC041) and one oil bottle (TFC072), were 73 identified as Central Mediterranean products. This site is a *campo de silos*, a large area full of 74 silos devoted to the storage of the product of a system of intense cereal exploitation. This type 75 of settlement is predominant in the Cossetania region.

#### 76 **2. METHODS**

#### 77 2.1 X-ray fluorescence (XRF)

78 To characterize the chemical composition of these samples (Table 1), XRF was carried out in 79 the Centres Científics i Tecnològics of the Universitat de Barcelona. Due to the long duration of the research project, two different XRF instruments were used. More detail information about 80 81 the analytical routine has been published elsewhere (Miguel Gascón and Buxeda i Garrigós, 82 2013).

83 First, in the case of samples MOS026, MOS037 and MOS040, major and minor elements (MgO, Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, P<sub>2</sub>O<sub>5</sub>, K<sub>2</sub>O, CaO, TiO<sub>2</sub>, MnO, and Fe<sub>2</sub>O<sub>3</sub> -as total Fe-) were determined by 84 85 preparing duplicates of glass beads. Trace elements (V, Cr, Co, Ni, Cu, Zn, Ga, Rb, Sr, Y, Zr, 86 Nb, Mo, Sn, Ba, Ce, W, Pb, and Th), as well as Na<sub>2</sub>O, were determined by powdered pills. The 87 quantification of the concentrations was performed using a (WDXRF) Phillips PW2400 88 spectrometer with an Rh excitation source.

Samples MOS060, FER005, TFC025, TFC036, TFC040, TFC041 and TFC072 were 89 analyzed with a different XRF machine. In this case, major and minor elements (including 90 91 Na<sub>2</sub>O) were determined by duplicates of glass beads. Trace elements were also determined by 92 powdered pills. However, this time determination of concentrations was performed using an Axios<sup>mAX</sup>-Advanced PANalytical spectrometer with an Rh excitation source. 93

94 All differences between the two XRF machines were taken into account in order to be able to 95 compare the data obtained.

#### 96 2.2 X-ray diffraction (XRD)

97 The mineralogical composition was studied by means of XRD at the Centres Científics i 98 Tecnològics of the Universitat de Barcelona.

99 Measurements for samples MOS026, MOS037 and MOS040 were made using a Siemens D-

500 diffractometer working with the Cu-K $\alpha$  radiation ( $\lambda = 1.5406$  Å) at 1.2 kW (45 kV - 30 100

mA). Measurements where taken from (4 to 70)°20, at 1°20 min<sup>-1</sup> (step size =  $0.05^{\circ}20$ ; time = 3 101 s).

In the case of samples MOS060, FER005, TFC025, TFC036, TFC040, TFC041 and TFC072 measurements were made using a Bragg-Brentano PANalytical X'Pert PRO MPD Alpha diffractometer equipped with an X'Celerator detector, working with the Cu-K $\alpha$  radiation ( $\lambda =$ 1.5406 Å) at 1.8 kW (45 kV – 40 mA). Measurements where taken from (4 to 100)°2 $\theta$ , with an step size of 0.026°2 $\theta$  and an acquisition time of 50 s per step.

## 108 **2.3 Thin-section petrography**

Samples MOS037, MOS040, FER005, TFC025, TFC036, and TFC041 were also analyzed in terms of thin-section petrography at the University of Sheffield. All specimens were cut transversely to the direction of the wheel-thrown marks using an Evans saw. Thin-sections were examined at a range of magnifications from x25 to x100 under a petrological microscope to study their mineralogy, petrography and texture. Petrographic groups were described following Whitbread's system (Whitbread, 1989, 1995).

## 115 **2.4 Scanning electron microscopy (SEM)**

Finally, some specimens (TFC041, TFC036 and MOS037) were selected for analysis by SEM attached to an energy-dispersive X-ray analyser (EDX) in order to characterize the microstructure, the sintering stage of the ceramic matrix and to identify aplastic inclusions. Thus, fresh fractures of the observed samples were coated with a carbon layer in a high vacuum atmosphere. An acceleration voltage of 20 kV, beam current of 1 nA, and 100 s per microanalysis was used.

These analysis were carried out at the Centres Científics i Tecnològics of the Universitat de
 Barcelona, using a *JEOL* JSM-6510 and Quanta 200 microscopes.

124

#### 125 **3. RESULTS**

126 The elemental concentrations determined by means of XRF are a special case of the 127 projective d+1-dimensional space, the simplex  $S^d$ . Projective points are represented by a d+1-128 dimensional vector of coordinates adding up to a constant k ( $k \in R^+$ ),

129

130  $\mathbf{x} = [\mathbf{x}_1, ..., \mathbf{x}_{d+1}] \mid \mathbf{x}_i \ge 0 \ (i = 1, ..., d+1), \ \mathbf{x}_1 + ... + \mathbf{x}_{d+1} = k,$ 

131 (in the present case, k = 100), a subset in the positive orthant  $R^{d+1}$ , following a multiplicative 132 model with logarithmic interval metrics (Barceló-Vidal *et al.*, 2001; Aitchison, 2005; Buxeda i 133 Garrigós, 2008). Therefore, for the statistical data treatment, raw concentrations have been ALR 134 (additive log-ratio) transformed, according to

$$\mathbf{x} \in S^{d} \rightarrow \mathbf{y} = \log(\frac{\mathbf{x}_{d}}{\mathbf{x}_{d+1}}) \in R^{d},$$

being  $S^d$  the d-dimensional simplex,  $\mathbf{x}_{-D} = [\mathbf{x}_1,...,\mathbf{x}_d]$ , and  $\mathbf{D} = d+1$ , or CLR (centered log-ratio) transformed, according to

- 140
- 141

 $\mathbf{x} \in \mathbf{S}^{d} \rightarrow \mathbf{z} = \log(\frac{\mathbf{x}}{g(\mathbf{x})}) \in \mathbf{R}^{d+1},$ 

142

being  $S^d$  the d-dimensional simplex, and  $g(\mathbf{x})$  the geometric mean of all D (D = d+1) components of  $\mathbf{x}$  (Aitchison, 1986; Buxeda i Garrigós, 1999). Moreover, several elements were discarded: Mo and Sn due to low analytical precision, and Co and W because of the possible contaminations from the tungsten carbide cell of the mill. Th was also not taken into account, on account of possible interferences due to the high Sr concentrations, that could not be corrected in the case of those analysis done with the (WDXRF) Phillips PW2400 spectrometer.

149

150 A cluster analysis was performed with R software (R Core Team, 2013) using the square Euclidian distance and the centroid agglomerative algorithm on the subcomposition Na<sub>2</sub>O, 151 152 MgO, Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, K<sub>2</sub>O, CaO, TiO<sub>2</sub>, MnO, and Fe<sub>2</sub>O<sub>3</sub>, with CLR transformation of the 218 153 Phoenician samples of the ARQUB's database (figure 3). Only major and minor elements were 154 considered in this statistical treatment since the abovementioned dataset included 178 155 specimens analyzed by other research teams. These individuals belong to Phoenician pottery or 156 to chronologically related ceramics, but only major and minor elements concentrations were 157 determined. Moreover, since the analytical routine was also different from the present project 158 and, in most cases, there are no inter-laboratory calibration studies, this comparison should be 159 considered semi quantitative. The analytical case studies of these 178 specimens are: a study 160 conducted at the Università di Palermo on the Phoenician productions of Mozia and Solunto in 161 Sicily (Alaimo et al., 1998, 2002); three studies conducted at the Insituto di Ricerche 162 Technologiche per la Ceramica – CNR (Faenza) on the Phoenician tableware of Carthage (Amadori and Fabbri, 1998a), Toscanos (Amadori and Fabrri, 1998b), Sardinia (Tharros, Santo 163 164 Antioco and Monte Sirai), and Ischia (Amadori and Fabbri, 1998c); the grey products from the 165 Iberian site of Ullastret (Pradell et al., 1995); the amphorae of the Palaià Polis of Empúries (Vendrell-Saz, 2005); and the Phoenician kilns of Cerro del Villar (Cardell et al., 1999). In the 166 167 case of Carthage, Sicily and Sardinia, control groups were formed by samples classified as local products on grounds of their typology, their chemical and mineralogical analysis by means of 168 169 XRF and thin-section petrography.

170 The statistical data treatment is summarized in the dendrogram resulting from the cluster 171 analysis in Figure 3. It shows a complex structure where we observe groups related to Central 172 Mediterranean sources, according to the comparative data bank considered. Thus, group CER 173 contains a possible Sardinian sample (MOS037), while a possible Sicilian provenance can be 174 suggested for sample FER005, which is included in group SIC. Also of interest is a possible 175 Tunisian (Carthaginian) provenance for samples TFC025, TFC036, TFC040, TFC041, TFC072, 176 MOS026, MOS040 and MOS060 in the complex structure CAR that can be divided into four 177 different groups: CARa, CARb, CARc and CARd.

178

Ther follows an assessment of the chemical and mineralogical data obtained through theXRF, XRD, SEM and petrographic analysis for each of these groups.

## 181 **3.1 Carthaginian products: GR CARa, CARb, CARc and CARd**

182 The compositional variation matrix enabled the quantification of the total variation (vt)183 present in the data matrix and the investigation of the source of this variability. In the case 184 where all samples belonging to the CARa, CARb, CARc and CARd subgroups were considered 185 as one group (CAR), vt has a high value (vt = 1.8). This suggests that these data have a 186 polygenic character, which means that no all individuals exhibited similar compositions; so, 187 following the provenance postulate, it would be expected that they represent different units of 188 production (Buxeda i Garrigós and Kilikoglou, 2003). This chemical variability is linked to Ba, 189 MnO, CaO, Na<sub>2</sub>O, Th, Cu and Cr. The element that imposes the lowest relative variation is 190 Al<sub>2</sub>O<sub>3</sub> when used as divisor in ALR transformation and will, therefore, be used in the statistical 191 data treatment (Buxeda i Garrigós, 1999). Because of the extreme variability introduced by 192 Na<sub>2</sub>O and Ba these elements excluded from the statistical data treatment. Also Pb, Cu and  $P_2O_5$ 193 were not used because they are known to be sensitive to post-depositional perturbations.

A cluster analysis was performed, using the Euclidian distance and the centroid
agglomerative algorithm on the subcomposition MgO, SiO<sub>2</sub>, K<sub>2</sub>O, CaO, TiO<sub>2</sub>, V, Cr, MnO,
Fe<sub>2</sub>O<sub>3</sub>, Ni, Zn, Ga, Rb, Sr, Y, Zr, Nb, and Ce, ALR transformed using Al<sub>2</sub>O<sub>3</sub> as divisor.

197 Figure 4 shows the dendrogram resultant from this statistical treatment. The first group that 198 appears, CARa, includes individuals MOS040 and MOS060 that present the lowest CaO 199 concentrations among the Tunisian groups (9.60%, table 2). Next to it, another group, CARb, 200 with samples TFC025, TFC040 and TFC041 presents the highest concentrations in Rb (114 201 ppm, table 2). More separated appears CARc including a single individual, TFC072, which is 202 the most calcareous specimen of the Carthaginian products analyzed (26.1% in normalized data, 203 table 1 and table 2). Finally, CARd, presents the highest concentrations of Zn for samples 204 TFC036 and MOS026 (94 ppm, table 2).

Petrographic analysis was performed on individuals MOS040 (CARa); TFC025, TFC041
 (CARb); and TFC036 (CARd). It was possible to identify different petrographic groups:

Group 1. This group is represented by sample TFC036 (Figure 5), characterized by the presence of rounded quartz and calcite inclusions. There are very rare to rare voids forming an estimated <1% of the volume. Open-spaced common mesovughs, rare mesochannels and elongate voids, have a strong alignment to the vessel wall. Mostly equant aplastic inclusions exhibit a poor alignment parallel to the vessel wall.

- 212 It presents a fine, dense calcareous matrix homogeneous throughout the section. The colour is 213 brown to greyish in PPL and dark brown in XP (x25). The micromass is optically inactive. The 214 inclusions appear to have a bimodal grain-size distribution; the coarse fraction is set in a finer-215 grained groundmass. It is generally sub-rounded to well-rounded and well sorted. Micro fauna 216 composed by foraminifera of *globigerinida* species are present in the coarse fraction. Calcite 217 crystals have replaced the multi-chambered walls and, in very rare cases, they also contain iron 218 oxide (size < 0.08 mm). This type of microfossil is also common in the fine fraction (size < 0.04219 mm, mode = 0.025 mm). The abundance of these microfossils and sandstone indicate a 220 sedimentary environment.
- 221

Group 2. The dominant inclusion of this fabric consists of crushed calcite. It is represented by sample TFC025 (figure 6). Its microstructure presents very rare voids, with few mesovughs and very rare mesochannels. Elongate voids are open-spaced and display strong alignment to the margins of section with. Elongate and equant aplastic inclusions exhibit a moderate alignment to the vessel walls.

227

It features a fine calcareous matrix predominantly homogeneous throughout the section. It is brown/orange in PPL and brown/red in XP (x25) and optically very active. Crushed calcite is the dominant inclusion in coarse fraction, angular, euhedral and equant and well sorted (size <0.32 mm, mode = 0.15 mm).

232

Textural concentration features (Tcf) are bright red in PPL and dark red in XP (x25) with clear to merging boundaries and very high optical density. They are generally equant, rounded and have high sphericity, discordant with the micromass. The commonly display fine-grained monocrystalline quartz inclusions, with varying amounts of ferruginous matter, sometimes almost opaque. They are probably clay pellets (size 0.036 mm to 0.016 mm).

238

The angular and even regular shape of calcite inclusions present in this sample suggest that this mineral was crushed and added deliberately to the clay as temper.

Group 3. This is defined by allotriomorphic quartz and mica as predominant inclusions in the matrix (Sample MOS040 and TFC041, Figure 7). The microstructure shows rare to few voids, forming an estimated 1-3% of the volume. There are common meso- and microvughs and rare mesochannels. Elongate voids have a very strong alignment to the margins of sections and are single to double-spaced. These display partial infilling with secondary calcite. The equant, aplastic inclusions mostly exhibit a poor alignment to the vessel walls.

248

The predominantly fine calcareous and homogeneous is brown in PPL and dark orange/brown in XP (x25). The micromass is optically active, evident in MOS040, due to a low firing temperature.

252

Quartz (size < 0.032 mm, mode = 0.018 mm) and muscovite mica (size < 0.024 mm, mode =</li>
0.015 mm) are frequent in the fine fraction. Calcite is common and there are few iron oxides
and opaques. Rare foraminifera microfossils are present in TFC041.

256 There are two different types of Tcf in sample TFC041. The first is dark orange in PL and 257 brown in XP (x25), with clear boundaries and is optically dense. This type is generally sub-258 angular, discordant with the micromass and contains non-plastic inclusions of fine, 259 monocrystalline quartz and calcite crystals (size < 0.16 mm, possible clay pellets). Another Tcf, 260 of dark red colour in PPL and XP (x25) most almost opaque, with clear, sharp boundaries. In 261 some case there is a surrounding void. Optically it is extremely dense and sub-angular, 262 discordant with the micromass, with non-plastic inclusion, such as fine-grained monocrystalline 263 quartz, distinguishable (size = 0.3 mm, possible grog).

Microfossils are more abundant and larger in MOS040, in both the coarse and fine fraction It is also possible to distinguish foraminifera and crushed molluscs (size < 0.1 mm, mode = 0.05 mm). The rounded to well-rounded monocrystalline quartz fragments present in the matrix is a diagnostic feature to relate these samples with a desert environment. It may be suggested that TFC041 is a finer version of TFC025.

269

XRD results allow the identification of different fabrics (Buxeda *et al.*, 1995) within each
chemical group of the Carthaginian products. In the case of group CARa there is just one fabric,
CARa-I, represented by samples MOS040 and MOS060. It is possible to estimate a low
equivalent firing temperature (EFT) below (800/850) °C, since no firing phases were observed
(Figure 8, top).

275

Again, only one fabric was defined for group CARb, comprising samples TFC025, TFC040
and TFC041 (CARb-I). Its EFT can be estimated below (800/850) °C.

CARc contains only one sample, TFC072, (CARc-I). The diffractogram of this oil bottle (Figure 8, centre) presents quartz, gehlenite, calcite, plagioclase and analcime, which is a sodic zeolite that appears as a perturbation in high fired calcareous ceramics (Buxeda, 1999). Its EFT can be estimated around (1000/1050) °C.

283

Two fabrics can be distinguished for CARd. Sample MOS026 represents fabric CARD-I with a low EFT, below (800/850) °C. Fabric CARd-II is represented by sample TFC036 (Figure 8, bottom) with an EFT estimated c. (950/1000-1050) °C. This EFT, similarly to the CARc-I fabric, is indicated by the total decomposition of illite-muscovite and the crystallization of plagioclase, pyroxene and gehelenite. Therefore, the presence of calcite must be associated with a secondary origin, as observed in petrographic analysis (Cau *et al.*,2002).

The study of these products was completed by SEM analysis. Fabric CARb-I was studied through sample TFC041 and it was possible to confirm its low firing (Figure 9, top), through its non vitrificatied (NV) matrix (Kilikoglou, 1994). The EFT could thus be below 750 °C. Fabrics CARa-I, CARc-I and CARd-I were not analyzed. In fabric CARd-II a stage close to total vitrification (Vc+-TV) was observed (Figure 8) and its EFT can be estimated at around (1050-1080) °C.

296 **3.2 Sardinian production: GR CER** 

Sample MOS037 can be related to a Sardinian provenance. This red slip carinated bowl is grouped together with the reference groups of the Phoenician site of Sulcis (Acquaro, 1998; Amadori and Fabbri, 1998c). It presents high concentrations in Na<sub>2</sub>O, K<sub>2</sub>O, Rb, Y, and Ce (1.92%, 4.41%, 182 ppm, 46 ppm, and 121 ppm in normalized data, respectively, table 1).

301

302 Petrographic analysis characterized MOS037 as a different group, where Tcfs are the 303 principal inclusion (Figure 10).

304

305 Group 4. There are rare to very few voids in this sample: common mesovughs, single to open 306 spaced, elongate and with a strong alignment to the section margins. Elongate and equant 307 aplastic inclusions exhibit moderate alignment parallel to the vessel walls.

The sample has a fine calcareous, largely homogeneous matrix, which is light brown to greyish-brown in PPL and orange-reddish to brown in XP (x25). Frequent Tcfs occur in both the coarse and fine fraction. They are dark red to opaque in PPL and brown-reddish to opaque in XP (x25), with sharp to clear boundaries and very high optical density. They are usually angular to sub-angular and discordant with the micromass. Some have monocrystalline quartz and calcite inclusions similar to those in the matrix. Their angularity might suggest they are grog fragments, although they may be clay pellets (size < 0.24 mm, mode = 0.112 mm in coarse fraction and size < 0.08 mm, mode = 0.032 mm in fine fraction).

The CaO concentration of MOS037 shows us that it is non-calcareous (1.64% in normalized data, Table 1 and Table 2). In XRD analysis (CER-I), no firing phases are observed (Figure 11) and the estimated EFT is below (800/850) °C.

319

320 SEM confirms the low firing suggested by the XRD spectra. In Figure 12, the microstructure
321 of the matrix displays initial to just less than continuous vitrification (IV-Vc-). Thus, the EFT is
322 in the range (750-850) °C.

## 323 **3.2 Sicilian production: GR SIC**

Sample FER005, from La Ferradura, is related by chemistry to the Phoenician production
centre of Solunto (Sicily), and thus was labelled group SIC (Acquaro, 1998; Alaimo *et al.*,
1998). It is illustrated in thin section in Figure 13.

327

Group 5. Sandstone inclusions and serpentinite are the most common features of this sample.
There are very rare voids, comprising common mesovughs, open-spaced, elongate with a strong
alignment to the margins of sections. Some have traces of secondary calcite infilling. Elongate
and equant aplastic inclusions exhibit a poor alignment to the vessel walls.

332 It presents an optically active, fine calcareous matrix homogeneous through the section, 333 greyish to brown/orange in PPL and dark brown to orangish-red in XP (x25).

334

There is dominant sub-rounded to rounded monocrystalline quartz in the coarse fraction (size < 0.176 mm), sub-rounded to rounded calcite (size < 0.16 mm), with microfossils composed of foraminifera and mollusc fragments (size < 0.07 mm). More rarely mica-schist (size < 0.108mm).

339

Tcf are dark in colour in PPL and red in XP sub-angular to sub-rounded with clear boundaries (size < 0.25 mm).

342

343 XRD analysis of this calcareous fabric (12% of CaO in normalized data, Table 1 and
344 Table 2) (designated fabric SIC-I) suggests a low EFT, c. (800/850) °C. It contains illite345 muscovite, quartz, calcite, hematite, potassium feldspars and plagioclase (figure 14). No SEM
346 analysis was performed because of the lack of specimen.

347

### 348 4. DISCUSSION

349 The Carthaginian imported vessels identified in this study are calcareous (CARa) to highly 350 calcareous (CARb, CARc and CARd). They contain very well rounded quartz, along with 351 possible indications of clay mixing and sand tempering. These technological features have been 352 observed in the Phoenician production of Tyre (Miguel Gascón, 2014). Moreover, in the case of 353 TFC025 (CARb), crushed calcite is used as temper, a technique also observed at the Phoenician 354 metropolis. XRD has also revealed that most of these fabrics exhibit low EFT (below 800 °C), 355 similar to that of the pottery produced at Tyre (Miguel Gascón and Buxeda i Garrigós, 2013). It 356 should be emphasized that the agreement between data from XRF and thin section petrography 357 is not complete and that it remains to be shown whether these represent only Carthaginian 358 products, or those of other Tunisian production centres.

The Sardinian imports, group CER, seems to be related to a different tradition especially since grog may be used as temper. Moreover, it is also a non-calcareous fabric and this lies in contrast with the technology of Carthaginian or Tyrian production.

Finally, group SIC is calcareous, low-fired (EFT below (750/800) °C), with inclusions compatible with the geology of the Solunto area in Sicily. It presents more similarities with Tyrian ceramics than the group CER.

365 It is of interest that the Ilercavonia and Cossetania sites do not have amphorae from these 366 Central Mediterranean sites, only tableware. Amongst the 220 samples examined in this study, 367 it was clear that most of the amphorae analyzed have an Andalusian origin, especially from the 368 areas of Málaga and Granada (Miguel Gascón, 2014). Also while the archaic Phoenician 369 colonies in Southern Iberia produced red slip tableware and jars, none arrived in the study area. 370 Indeed, while generally rare, the Phoenician tableware at these Northern regions is clearly from 371 the Central Mediterranean production centres. This may indicate an important consumption 372 preference. However, further research is needed in other sites of the early Iron Age in the 373 Ilercavonia and Cossetania regions to confirm that pattern.

374

#### 375 **5. CONCLUSIONS**

The appearance of Phoenician imports in the early Iron Age contexts of the Cossetania and Ilercavonia regions is notable not only for the first evidence of wheel-made pottery, but also in terms of the novel consumption of the products transported in these jars. This latter change in consumption and contact may have had deep social implications. The rarity of tableware in the sites studied is notable, but, when it is present, the choice of Central Mediterranean tableware and the total absence of Andalusian red slip demands explanation and clearly has meaning in socio-economic terms.

383 Moving to the production centres of these imported Central Mediterranean vessels, it seems 384 that there is a clearer link between the Carthaginian products and those of Tyre, than those 385 produced in Sardinia and Sicily. However, clearly such judgements on the nature and cultural 386 context of production at these centres require larger sample numbers and the detailed study of 387 production technology in the source assemblages in Tunisia, Sicily and Sardinia.

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|                                  | MOS026 | MOS040 | MOS060 | MOS037 | FER005 | <b>TFC025</b> | <b>TFC036</b> | <b>TFC040</b> | <b>TFC041</b> | <b>TFC072</b> |
|----------------------------------|--------|--------|--------|--------|--------|---------------|---------------|---------------|---------------|---------------|
| Fe <sub>2</sub> O <sub>3</sub> % | 5.06   | 5.02   | 5.04   | 4.30   | 6.88   | 5.09          | 4.89          | 4.83          | 4.97          | 5.47          |
| $AI_2O_3\%$                      | 12.26  | 13.99  | 13.76  | 19.66  | 17.63  | 14.03         | 12.49         | 14.58         | 14.61         | 16.38         |
| MnO %                            | 0.03   | 0.03   | 0.03   | 0.05   | 0.08   | 0.06          | 0.04          | 0.03          | 0.05          | 0.02          |
| P <sub>2</sub> O <sub>5</sub> %  | 0.22   | 0.24   | 0.21   | 0.11   | 0.23   | 0.48          | 0.29          | 0.45          | 0.39          | 0.47          |
| TiO₂ %                           | 0.61   | 0.87   | 0.88   | 0.74   | 0.80   | 0.74          | 0.63          | 0.79          | 0.81          | 0.88          |
| MgO %                            | 1.85   | 1.63   | 1.62   | 1.39   | 1.66   | 1.59          | 1.48          | 1.40          | 1.39          | 2.15          |
| CaO %                            | 19.49  | 9.62   | 9.57   | 1.64   | 12.03  | 21.47         | 18.05         | 17.49         | 18.24         | 26.15         |
| Na₂O %                           | 0.43   | 0.63   | 0.77   | 1.92   | 0.63   | 0.42          | 0.43          | 0.43          | 0.39          | 0.87          |
| K <sub>2</sub> O %               | 1.32   | 2.40   | 2.40   | 4.41   | 2.27   | 2.41          | 1.40          | 2.67          | 2.60          | 1.83          |
| SiO <sub>2</sub> %               | 58.60  | 65.41  | 65.54  | 65.61  | 57.64  | 53.39         | 60.11         | 57.05         | 56.33         | 45.48         |
| Ba (ppm)                         | 171    | 544    | 584    | 514    | 516    | 2105          | 591           | 1463          | 1274          | 1460          |
| Rb (ppm)                         | 57     | 95     | 96     | 182    | 94     | 109           | 54            | 116           | 117           | 87            |
| Th (ppm)                         | 9      | 12     | 9      | 27     | 10     | 12            | 4             | 11            | 12            | 11            |
| Nb (ppm)                         | 18     | 20     | 18     | 21     | 18     | 17            | 16            | 18            | 19            | 19            |
| Pb (ppm)                         | 21     | 17     | 23     | 77     | 39     | 27            | 23            | 29            | 29            | 29            |
| Zr (ppm)                         | 156    | 291    | 288    | 189    | 183    | 213           | 169           | 227           | 222           | 180           |
| Y (ppm)                          | 19     | 30     | 32     | 46     | 30     | 31            | 20            | 31            | 32            | 35            |
| Sr (ppm)                         | 460    | 206    | 206    | 131    | 236    | 382           | 493           | 317           | 299           | 675           |
| Ce (ppm)                         | 58     | 82     | 69     | 121    | 69     | 76            | 60            | 73            | 71            | 83            |
| Ga (ppm)                         | 19     | 14     | 14     | 26     | 17     | 17            | 16            | 17            | 18            | 21            |
| V (ppm)                          | 93     | 83     | 88     | 105    | 108    | 99            | 123           | 102           | 105           | 98            |
| Zn (ppm)                         | 95     | 78     | 63     | 121    | 82     | 78            | 94            | 71            | 69            | 144           |
| Cu (ppm)                         | 15     | 16     | 16     | 39     | 19     | 19            | 16            | 39            | 26            | 36            |
| Ni (ppm)                         | 31     | 31     | 30     | 26     | 51     | 30            | 31            | 29            | 30            | 61            |
| Cr (ppm)                         | 85     | 95     | 85     | 71     | 101    | 59            | 98            | 63            | 62            | 142           |

|                                  | CARa    |      | CARb    |      | CARc    | CARd    |      | CER     | SIC     |
|----------------------------------|---------|------|---------|------|---------|---------|------|---------|---------|
|                                  | (n = 2) |      | (n = 3) |      | (n = 1) | (n = 2) |      | (n = 1) | (n = 1) |
|                                  | m       | sd   | m       | sd   |         | m       | sd   |         |         |
| Fe <sub>2</sub> O <sub>3</sub> % | 5.03    | 0.01 | 4.96    | 0.13 | 5.47    | 4.97    | 0.12 | 4.30    | 6.88    |
| Al <sub>2</sub> O <sub>3</sub> % | 13.87   | 0.16 | 14.40   | 0.33 | 16.38   | 12.37   | 0.16 | 19.66   | 17.63   |
| MnO %                            | 0.03    | 0.00 | 0.05    | 0.01 | 0.02    | 0.04    | 0.01 | 0.05    | 0.08    |
| P <sub>2</sub> O <sub>5</sub> %  | 0.22    | 0.03 | 0.43    | 0.05 | 0.47    | 0.25    | 0.05 | 0.11    | 0.23    |
| TiO₂ %                           | 0.87    | 0.01 | 0.78    | 0.03 | 0.88    | 0.62    | 0.02 | 0.74    | 0.80    |
| MgO %                            | 1.63    | 0.01 | 1.46    | 0.11 | 2.15    | 1.65    | 0.26 | 1.39    | 1.66    |
| CaO %                            | 9.60    | 0.03 | 18.99   | 2.11 | 26.15   | 18.76   | 1.02 | 1.64    | 12.03   |
| Na₂O %                           | 0.70    | 0.10 | 0.41    | 0.02 | 0.87    | 0.43    | 0.00 | 1.92    | 0.63    |
| K₂O %                            | 2.40    | 0.01 | 2.56    | 0.14 | 1.83    | 1.36    | 0.06 | 4.41    | 2.27    |
| SiO <sub>2</sub> %               | 65.48   | 0.09 | 55.57   | 1.94 | 45.48   | 59.35   | 1.07 | 65.61   | 57.64   |
| Ba (ppm)                         | 564     | 28   | 1577    | 436  | 0.146   | 318     | 297  | 514     | 516     |
| Rb (ppm)                         | 95      | 1    | 114     | 4    | 87      | 55      | 2    | 182     | 94      |
| Th (ppm)                         | 10      | 2    | 12      | 1    | 11      | 6       | 4    | 27      | 10      |
| Nb (ppm)                         | 19      | 1    | 18      | 1    | 19      | 17      | 1    | 21      | 18      |
| Pb (ppm)                         | 20      | 4    | 28      | 1    | 29      | 22      | 1    | 77      | 39      |
| Zr (ppm)                         | 289     | 2    | 221     | 7    | 180     | 162     | 9    | 189     | 183     |
| Y (ppm)                          | 31      | 1    | 31      | 1    | 35      | 19      | 1    | 46      | 30      |
| Sr (ppm)                         | 206     | 0    | 331     | 44   | 675     | 476     | 23   | 131     | 236     |
| Ce (ppm)                         | 75      | 9    | 73      | 3    | 83      | 59      | 1    | 121     | 69      |
| Ga (ppm)                         | 14      | 0    | 17      | 1    | 21      | 17      | 2    | 26      | 17      |
| V (ppm)                          | 85      | 4    | 102     | 3    | 98      | 107     | 21   | 105     | 108     |
| Zn (ppm)                         | 70      | 11   | 73      | 5    | 144     | 94      | 1    | 121     | 82      |
| Cu (ppm)                         | 16      | 0    | 27      | 10   | 36      | 15      | 1    | 39      | 19      |
| Ni (ppm)                         | 30      | 1    | 30      | 1    | 61      | 31      | 0    | 26      | 51      |
| Cr (ppm)                         | 90      | 7    | 61      | 2    | 142     | 91      | 9    | 71      | 101     |





Figure 3 Click here to download high resolution image



# Figure 4 Click here to download high resolution image









# Figure 8 Click here to download high resolution image

















Table 1 Normalized concentrations of the chemical compositions of the analyzed samples

Table 2 Mean (m), standard deviation (sd) and normalized concentrations of the chemical composition group cited in the text

Figure 1 Western Phoenician colonies and Phoenician production centres located at the Circle of the Strait of Gibraltar

Figure 2. Location of the sites sampled where Central Mediterranean Phoenician products are found

Figure 3 Dendrogram illustrating cluster analysis of the 218 Phoenican individuals of ARQUB's database, along with the 178 published samples.

Figure 4. Dendrogram of the eight Phoenician individuals of possible Carthaginian provenance after cluster analysis.

Figure 5 Group 1: Photomicrograph, showing rounded quartz and calcite inclusions. TFC036 (fabric CARd-II) in XP (x25).

Figure 6 Group 2: Photomicrograph illustrating crushed calcite inclusions. TFC025 (fabric CARb-I) in XP (x25).

Figure 7 Group 3: Photomicrographs showing allotriomorphic quartz and mica matrix. TFC041 (fabric CARb-I) in XP (upper) and MOS040 (fabric CARa-I) in XP (lower)(x25)

Figure 8 XRD spectra. Top: MOS040, fabric CARa-I. Centre: TFC072, fabric CARc-I. Bottom: TFC036, fabric CARd-II. Cal: calcite; Qz: quartz; Ilt: illite-moscovite; Pl: plagioclase; Gh: gehlenite; Di: diopside (pyroxene); Hem: hematite; Anl: analcime. Abbreviations after Whitney and Evans (2010).

Figure 9 SEM microphotographs. Top: TFC041 (fabric CARb-I). Bottom: TFC036 (fabric CARd-II)

Figure 10 Group 4: Photomicrograph illustrating extural concentration features/grog temper. MOS037 (fabric CER-I) in XP (x25)

Figure 11 XRD spectrum. MOS037 (fabric CER-I). Qz: quartz; Ilt: illite-moscovite; Pl: plagioclase

Figure 12 SEM photomicrograph of MOS037 (fabric CER-I)

Figure 13 Group 5: Photomicrograph showing sandstone inclusions and serpentinite. FER005 (fabric SIC-I) in XP (x25)

Figure 14 XRD spectrum of FER005 (FabricSIC-I). Qz: quartz; Ilt: illite-muscovite; Pl: plagioclase; Hem: hematite; Kfs: potassium feldspars.