

Master Course
in Organic Chemistry

2018-19

methods and design
in organic synthesis



Pere Romea

BAND OF BROTHERS



7. Cascade Reactions

Band of Brothers
HBO

*From this day to the ending of the world,
But we in it shall be remembered—
We few, we happy few, we band of brothers*

Henry V

William Shakespeare

Pot Economy

Remember!



Guidelines
Chapter 5

POT ECONOMY

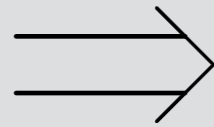
aim to complete an entire multi-step synthesis in a single pot
(see Chapter 5)

Pot Economy strives to save time and resources by avoiding purifications between individual steps within a multistep synthesis, thus minimizing the transfer of material between vessels, which increases the overall efficiency.

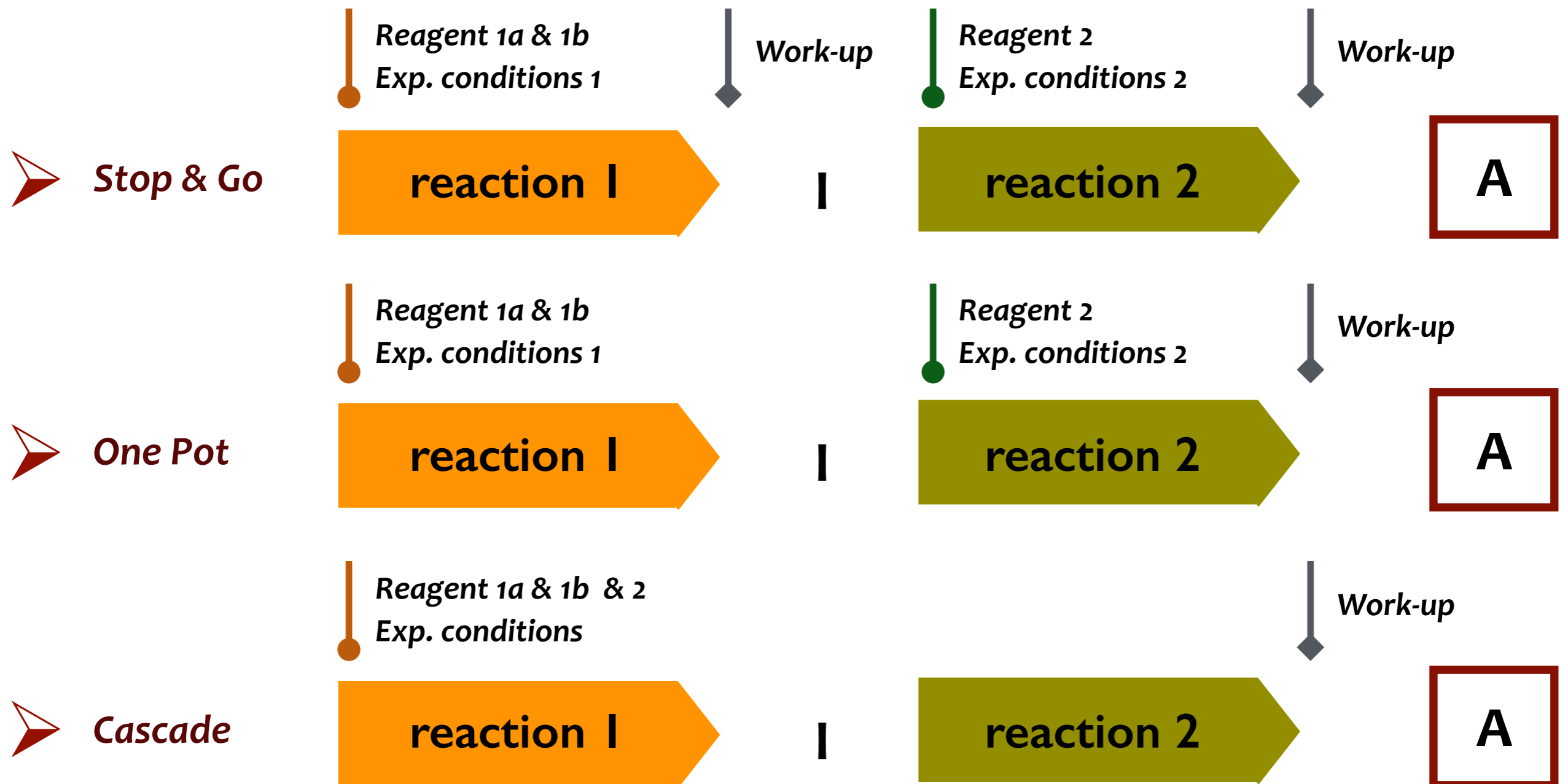
A procedure under **Pot Economy** premises can minimize chemical waste, save time, simplify practical aspects and increase the overall yield

Pot Economy

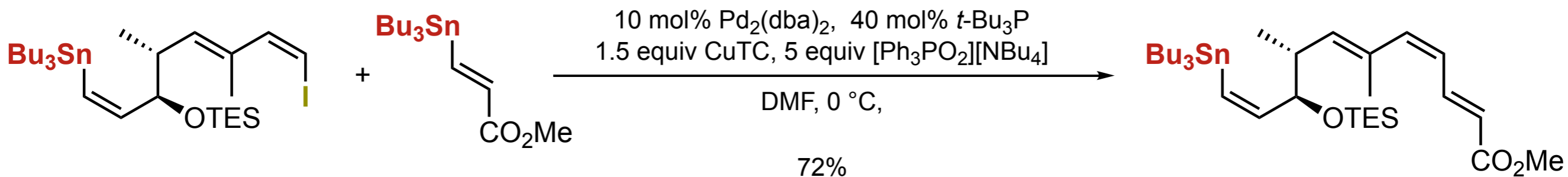
A



Reagent 1a + Reagent 1b + Reagent 2

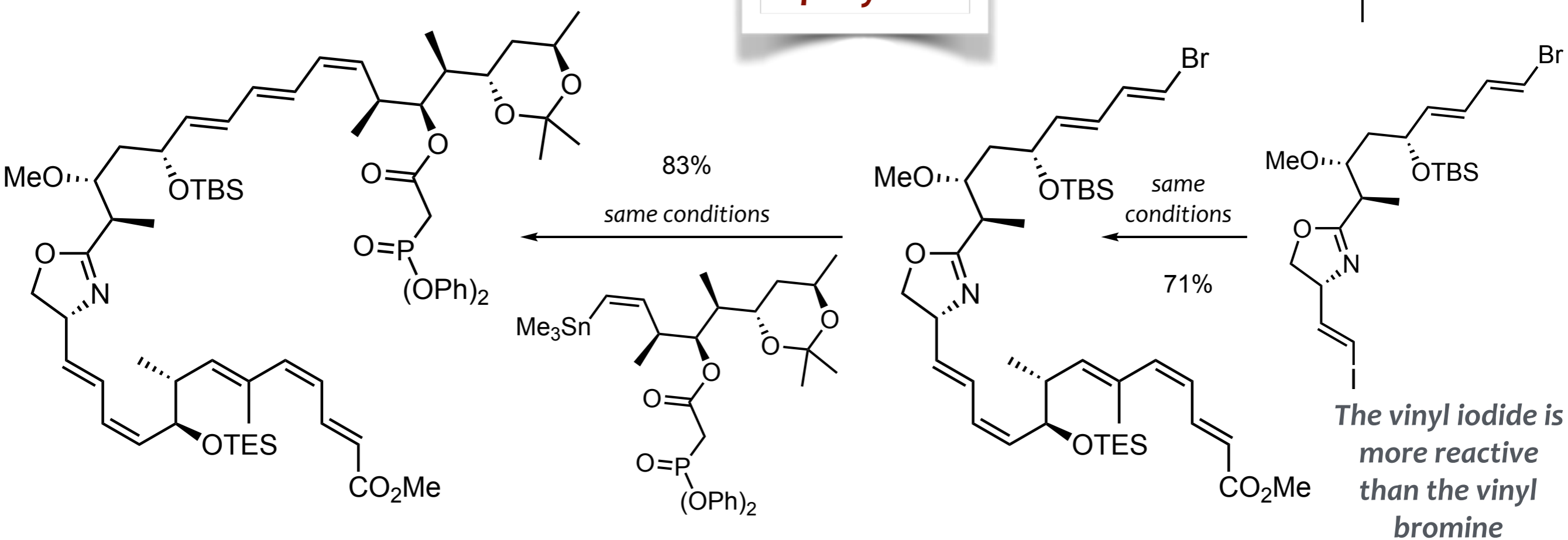


Stop&Go and One-Pot Procedures

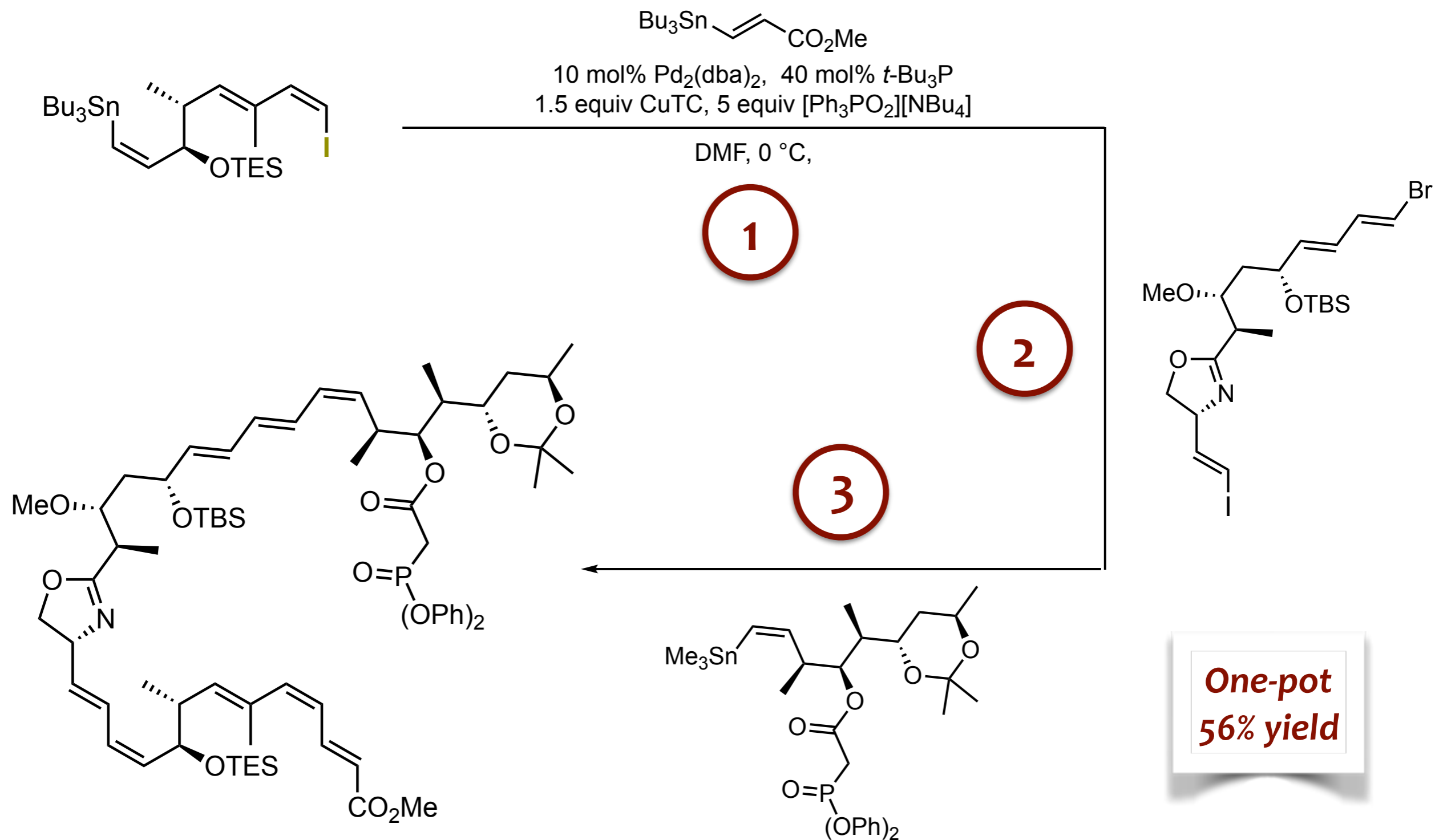


The E vinyl stannane is more reactive than the Z vinyl stannane for steric reasons

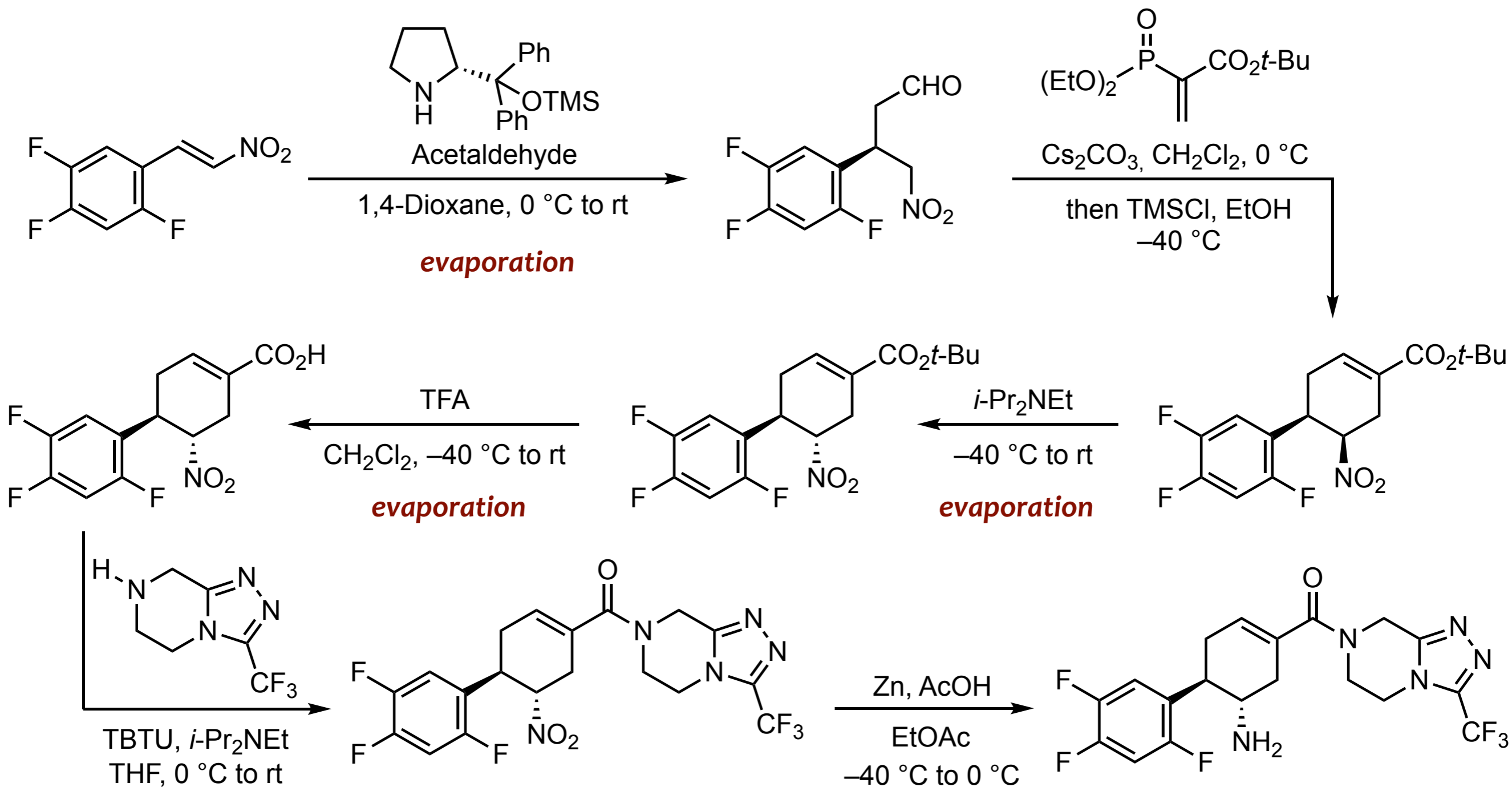
**Three steps
40% yield**



Stop&Go and One-Pot Procedures



Stop&Go and One-Pot Procedures



Six-step One-pot 63% yield

ABT-341

Concepts and Terminology



Today it is not only a question of what we can synthesize,
but **how we do it...**

The usual procedure for the synthesis of organic compounds
is the stepwise formation of the individual bonds in the target molecule.

However, it would be much more efficient if
one could form several bonds in one sequence without isolating the intermediates,
changing the reaction conditions, or adding reagents

Thus, these reactions would allow an ecologically and economically favorable production

We call this type of transformation a **DOMINO REACTION**

For this also the expression **CASCADE** has been used

Concepts and Terminology

Different authors use varying definitions as to what constitutes a cascade process.

*A variety of terms, including **cascade**, **domino**, **tandem**, and **sequential**, are used in the literature, often seemingly interchangeably and with liberal abandon ...*

*For our subjective purposes, we shall employ the term **CASCADE** to encompass all of the above descriptors*

*If further reagents are added at various points **ONE POT TRANSFORMATION***

*Although often composed solely of intramolecular transformations, cascade reactions can also occur intermolecularly, in which case they also fall under the category of **MULTICOMPONENT REACTIONS**, where three or more compounds react to form a single product that retain majority of the atoms of the starting materials*



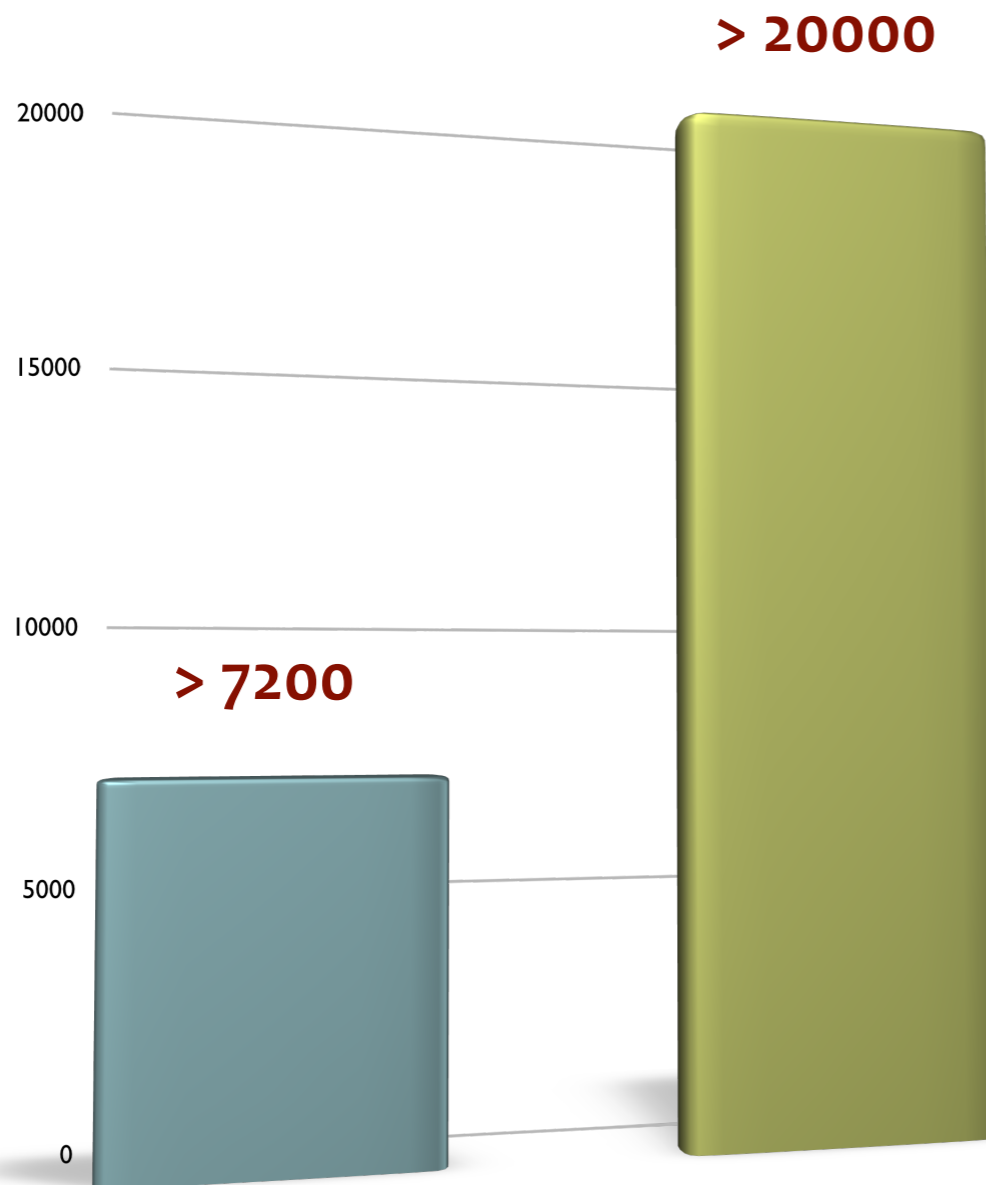
Concepts and Terminology

DOMINO

Tietze, L. F. *CR* 1996, 96, 115

CASCADE

Nicolaou, K. C. *ACIE* 2006, 45, 7134



Number of Articles Containing Term in the Title

Concepts and Terminology

Domino Reactions in Organic Synthesis

Tietze, L. F. *CR* 1996, 96, 115

Asymmetric Multicomponent Reactions (AMCRs): The New Frontier

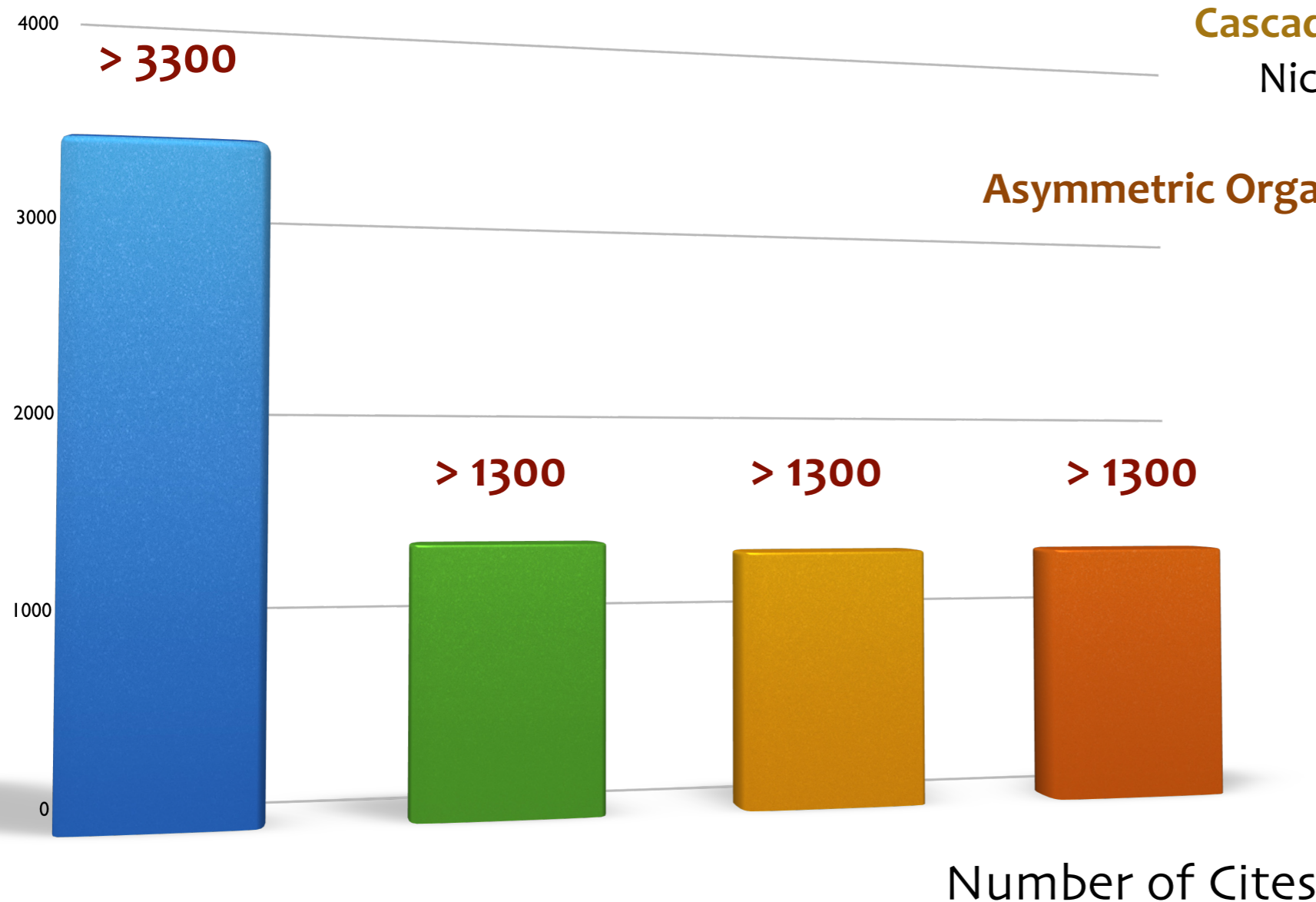
Yus, M. *ACIE* 2005, 44, 1602

Cascade Reactions in Total Synthesis

Nicolaou, K. C. *ACIE* 2006, 45, 7134

Asymmetric Organocatalytic Domino Reactions

Enders, D. *ACIE* 2007, 46, 1570



Concepts and Terminology

Domino (or **Cascade**) reaction is a process involving two or more bond forming transformations (usually C–C bonds), which take place under the same reaction conditions without adding additional reagents and catalysts and in which the subsequent reactions result as a consequence of the functionality formed in the previous step

Concepts and Terminology

The rapid generation of molecular complexity from simple starting materials is of paramount importance in synthetic chemistry

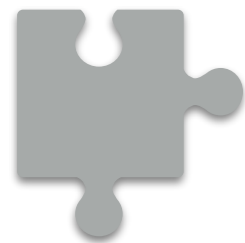
The design of cascades to provide specific targeted molecules of considerable structural and stereochemical complexity poses a significant intellectual challenge and can be one of the most impressive activities in natural product synthesis.

*Cascade reactions contribute immeasurably to both the **science and art** of total synthesis, bringing not only improved **practical efficiency** but also enhanced aesthetic appeal to synthetic planning*

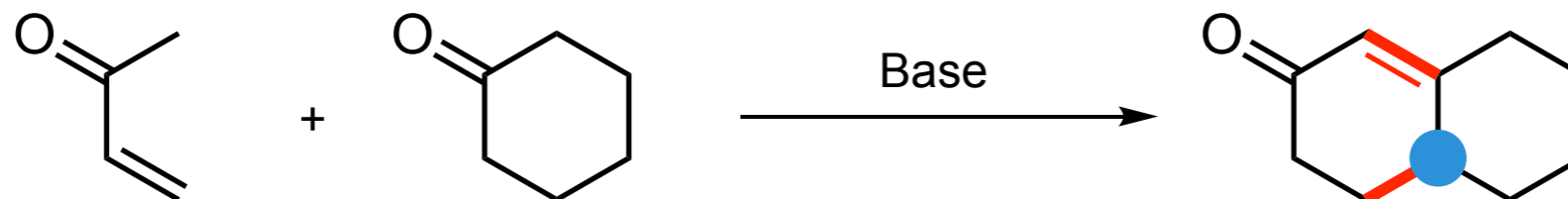
*Although often composed solely of **intramolecular transformations**, cascade reactions can also occur **intermolecularly***

*According to the mechanism of the first step, one can distinguish between a **nucleophilic, electrophilic, radical, pericyclic, transition-metal, or organocatalytic** cascades, which can be combined with reactions of the described type in a second, third, or fourth step*

Nucleophilic Cascades



Robinson Annullation

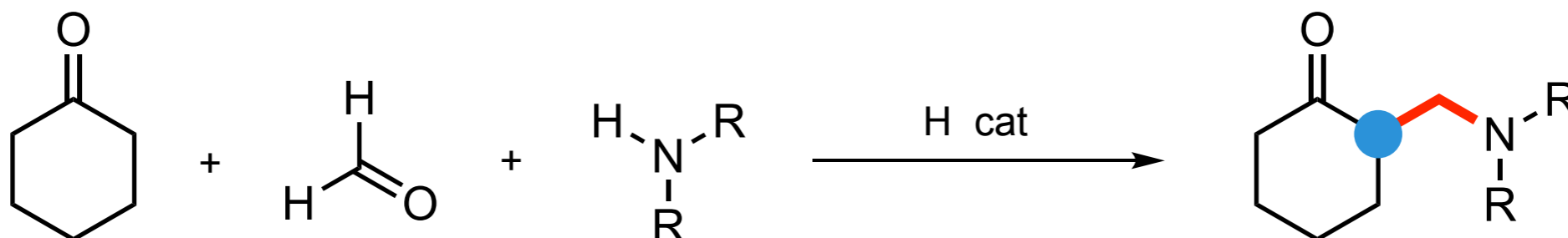


Remember!



See also
Chapter 4 & 6

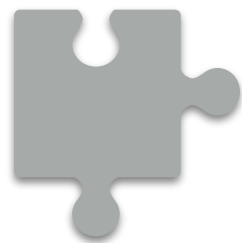
Mannich Reaction



See also
Multicomponent Reactions

2 new bonds & 1 stereocenter

Nucleophilic Cascades



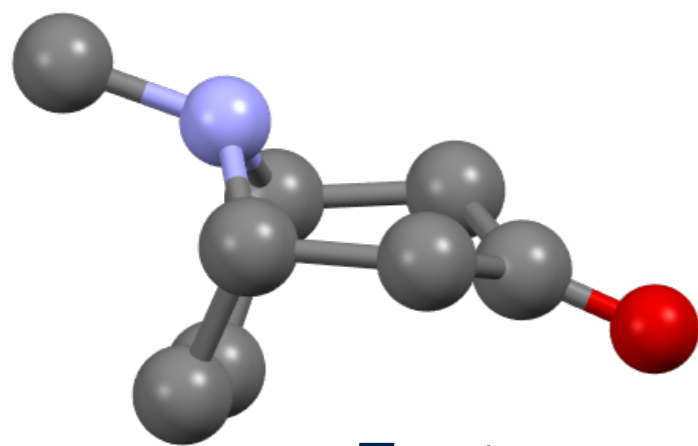
Robinson's landmark synthesis of tropinone

By imaging hydrolysis at the points indicated by the dotted lines, the substance may be resolved into succinaldehyde, methyl amine, and acetone, and this observation suggested a line of attack of the problem which has resulted in a **direct synthesis**

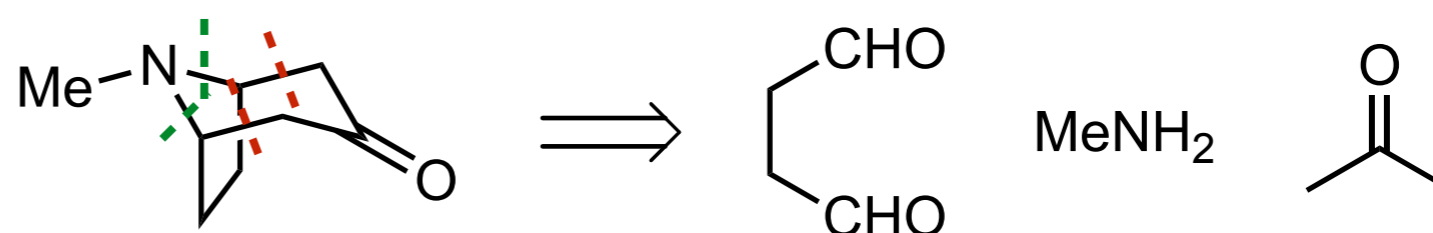
Sir Robert Robinson
(1886-1975)



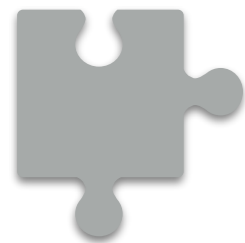
NOBEL PRIZE IN CHEMISTRY 1947



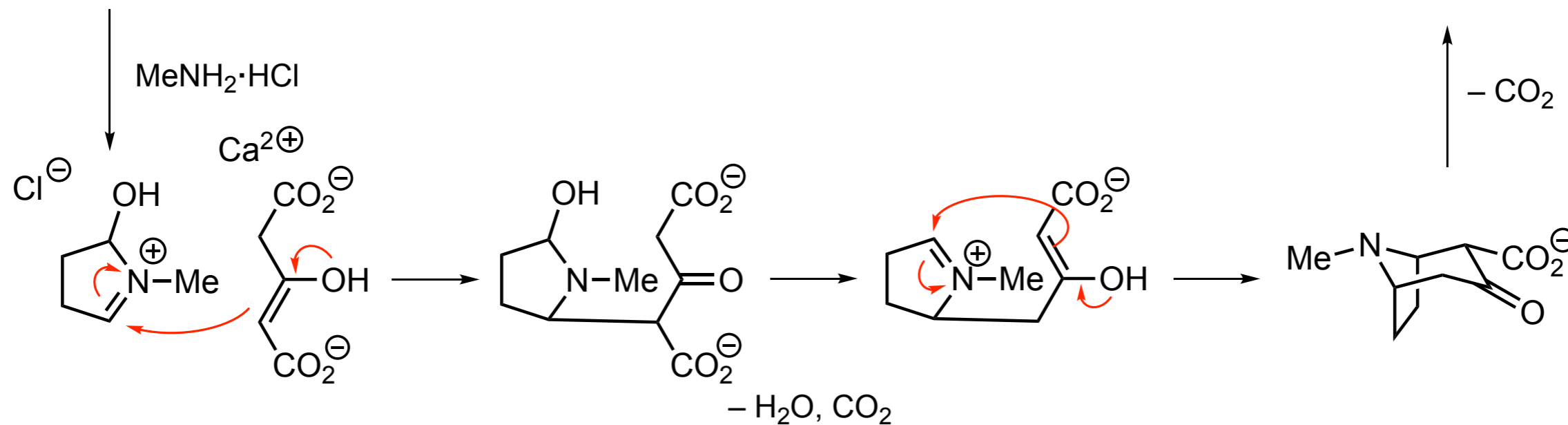
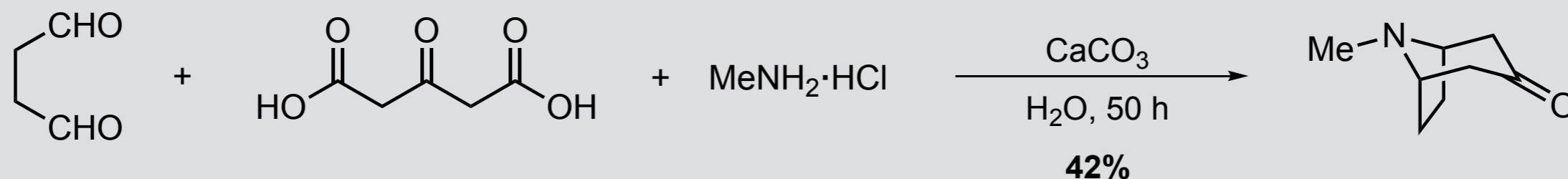
Tropinone



Nucleophilic Cascades

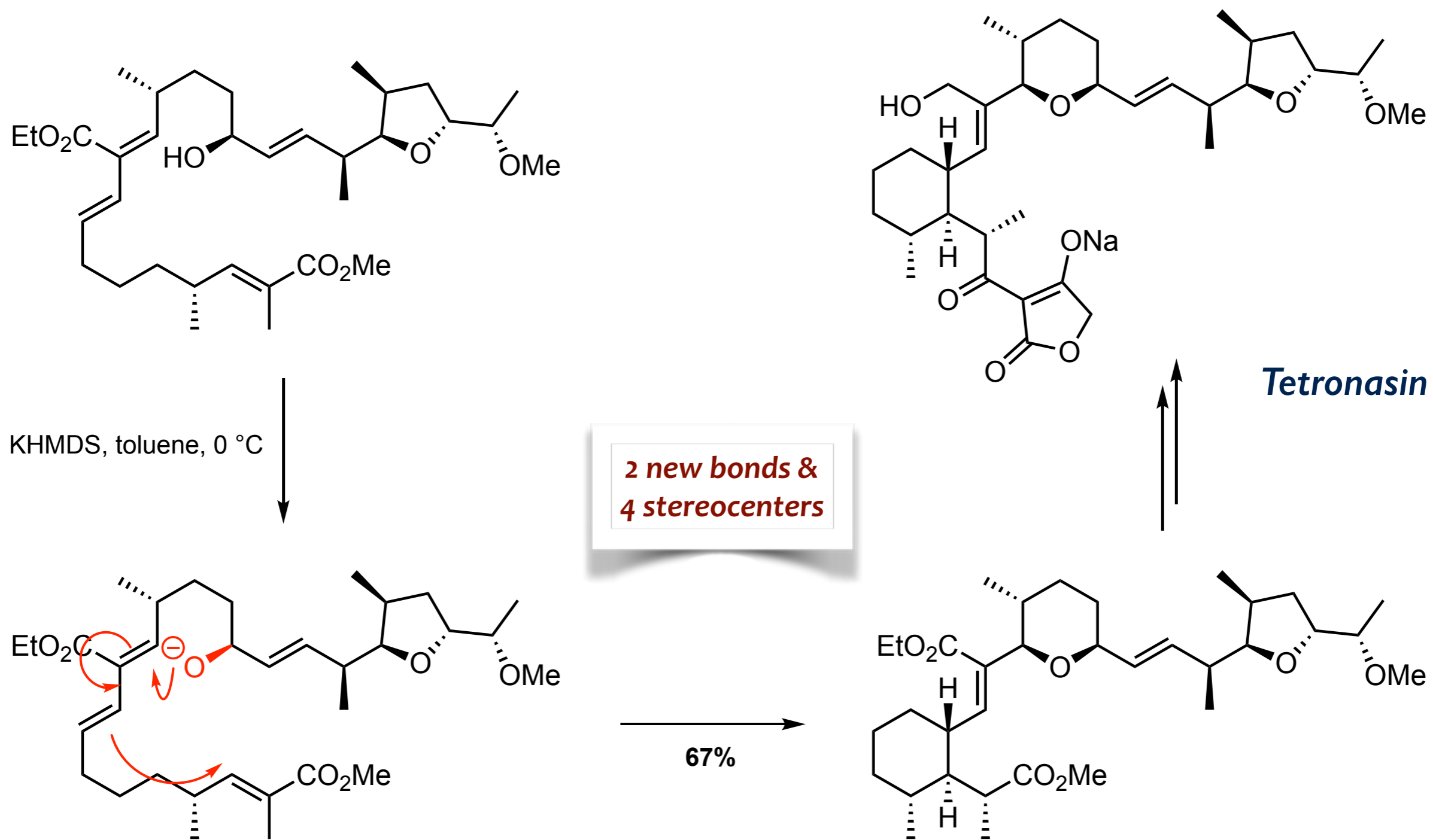


Robinson's landmark synthesis of tropinone



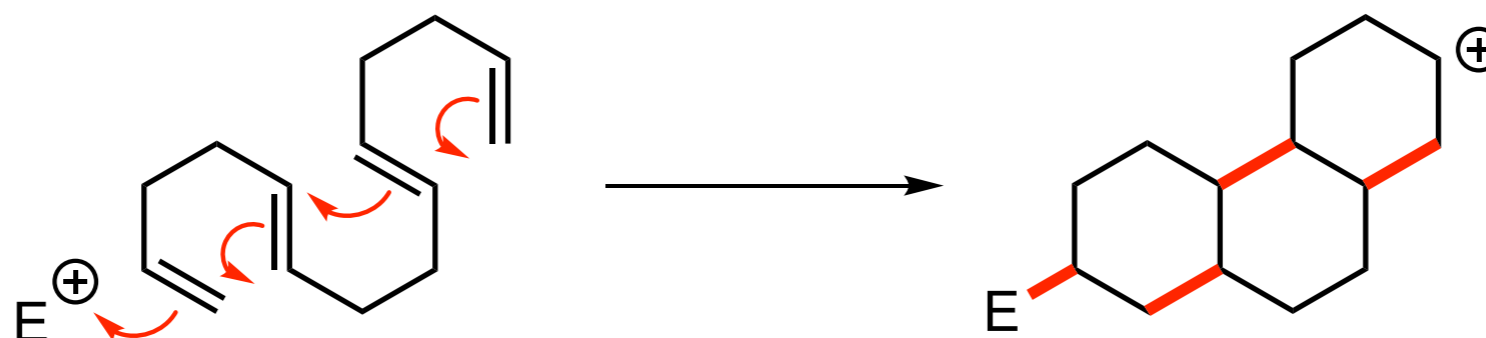
4 new bonds in a single step

Nucleophilic Cascades



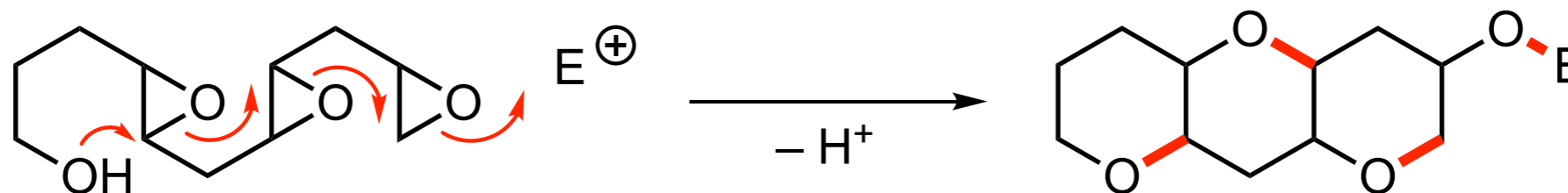
Electrophilic Cascades

➤ Polyene Cyclizations



Tietze, L. F. *CR* **1996**, 96, 115. Nicolaou, K. C. *ACIE* **2006**, 45, 7134.
Johnston, J. N. *CR* **2005**, 105, 4730. Barrett, A. G. M. *Synthesis* **2019**, 51, 67

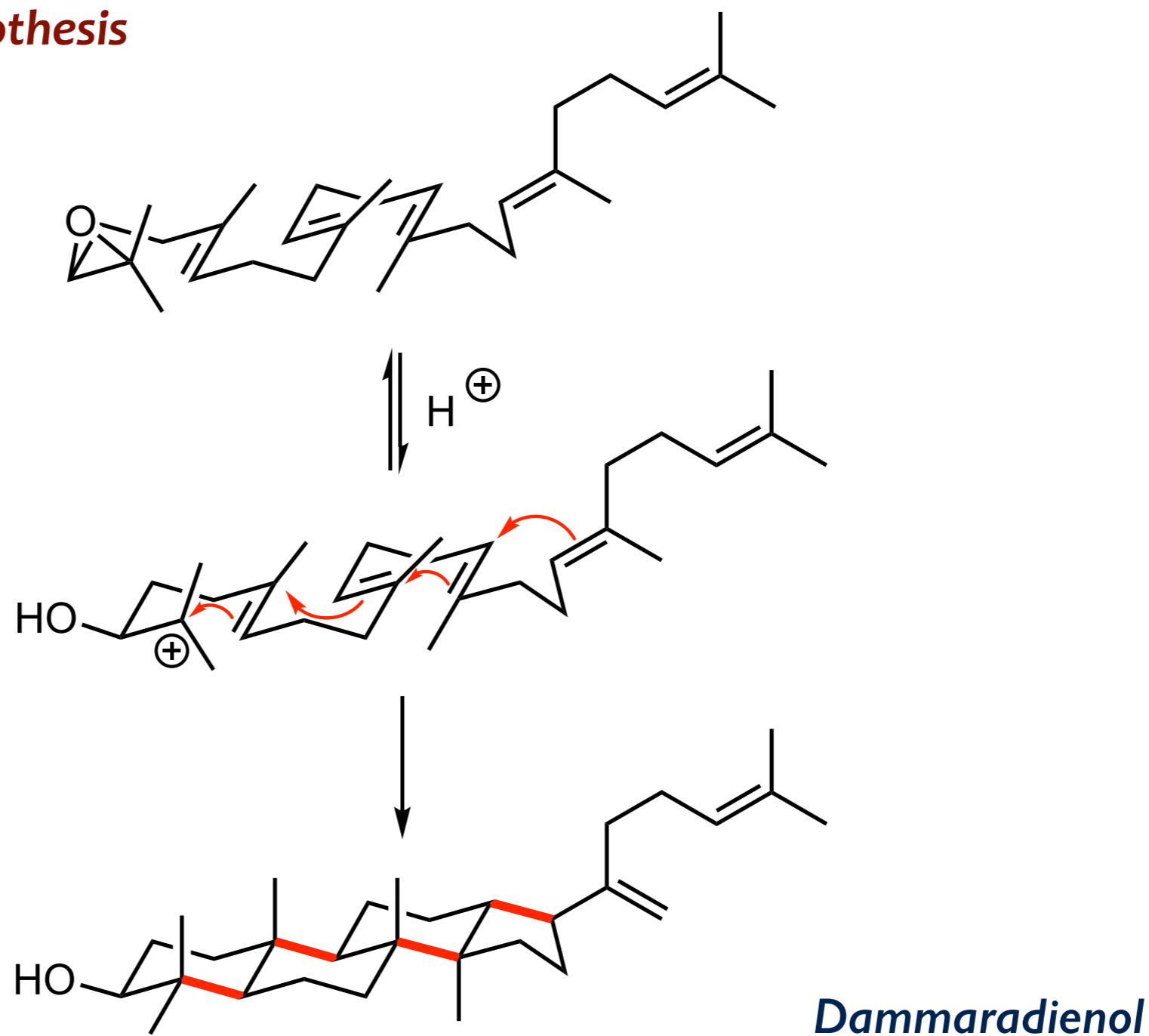
➤ Epoxide-Opening Cascades



Jamison, T. F. *CSR* **2009**, 38, 3175; *ACIE* **2009**, 48, 5250; *Mar. Drugs* **2010**, 8, 763

Electrophilic Cascades. Polyene Cyclizations

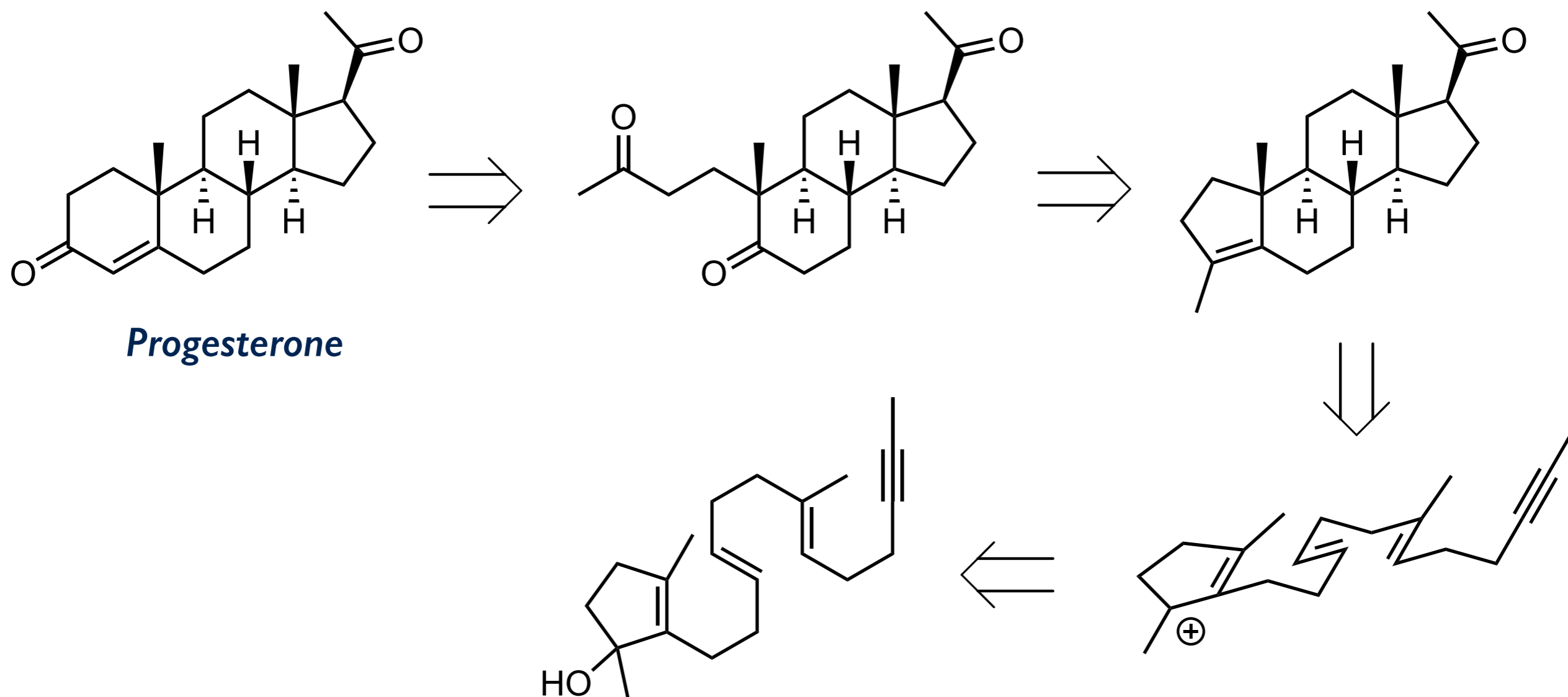
Stork-Eschenmoser hypothesis



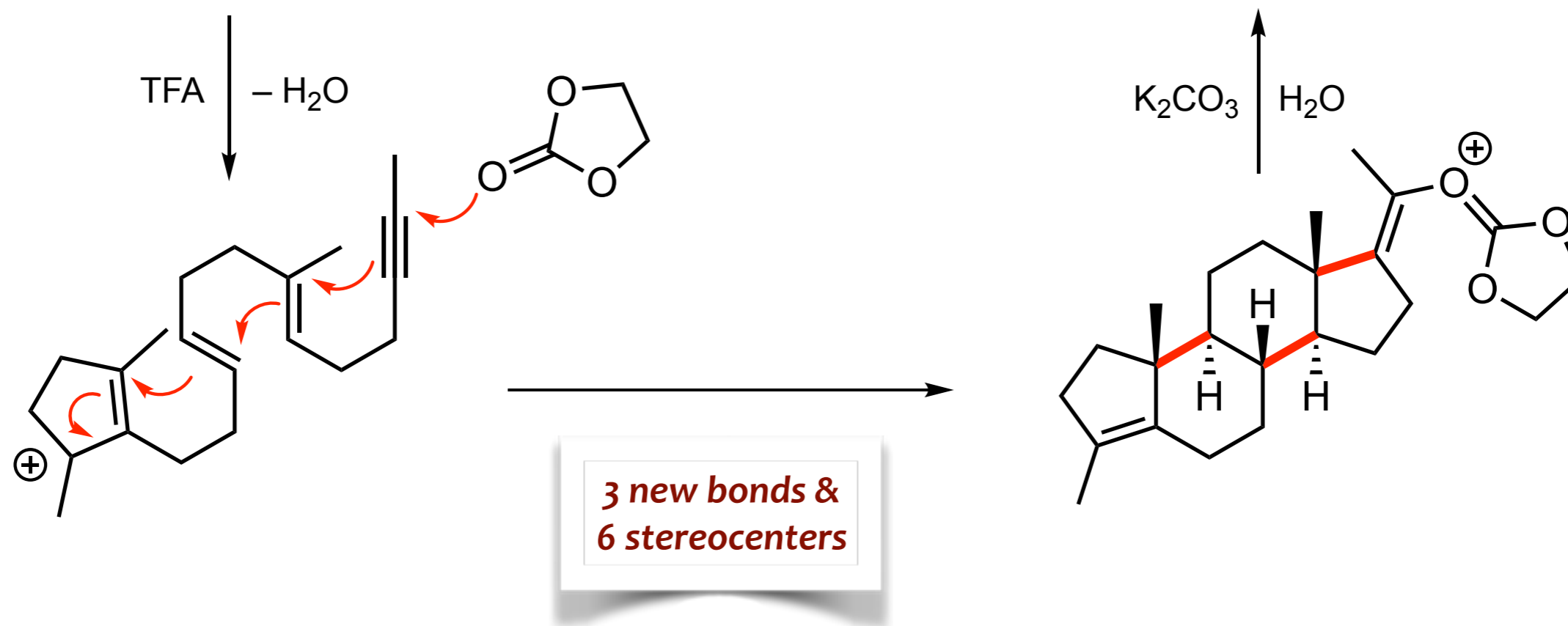
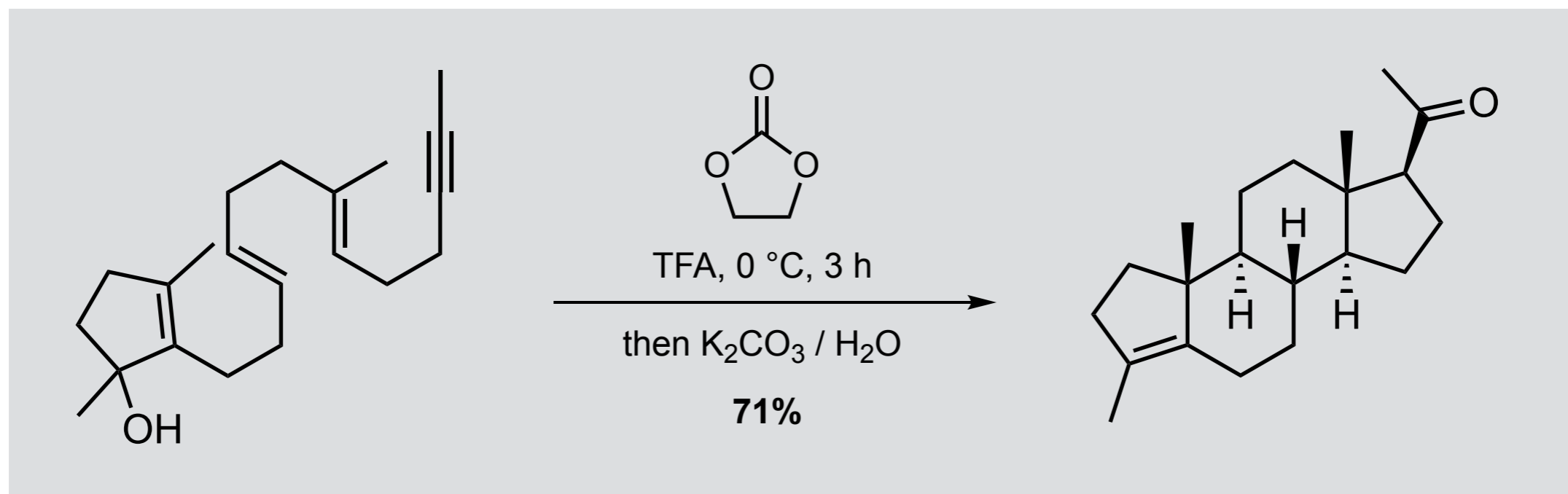
Electrophilic Cascades. Polyene Cyclizations



Johnson's landmark synthesis of progesterone

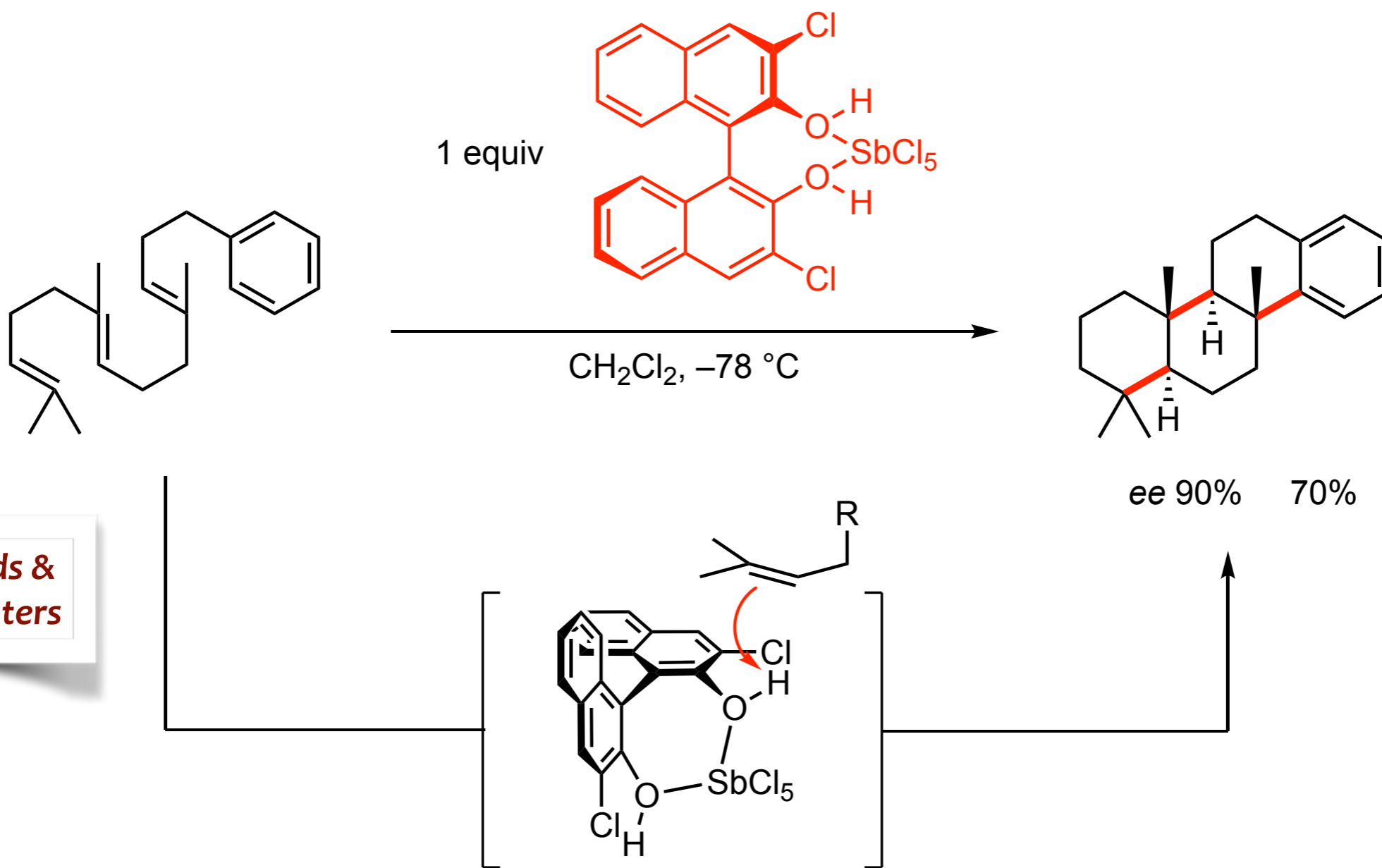


Electrophilic Cascades. Polyene Cyclizations



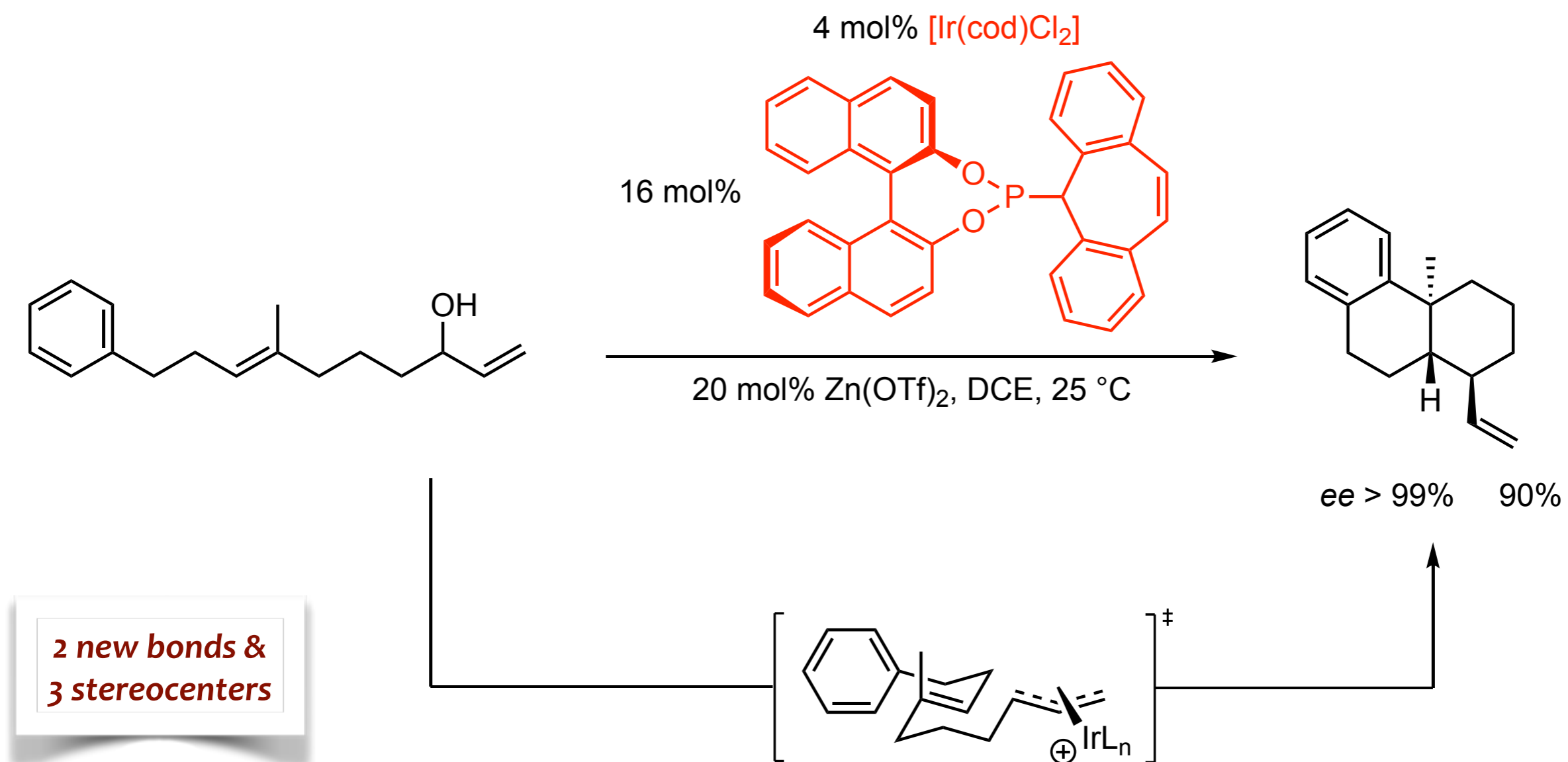
Electrophilic Cascades. Polyene Cyclizations

Polyene cyclizations controlled by chiral Lewis acid can be **enantioselective**



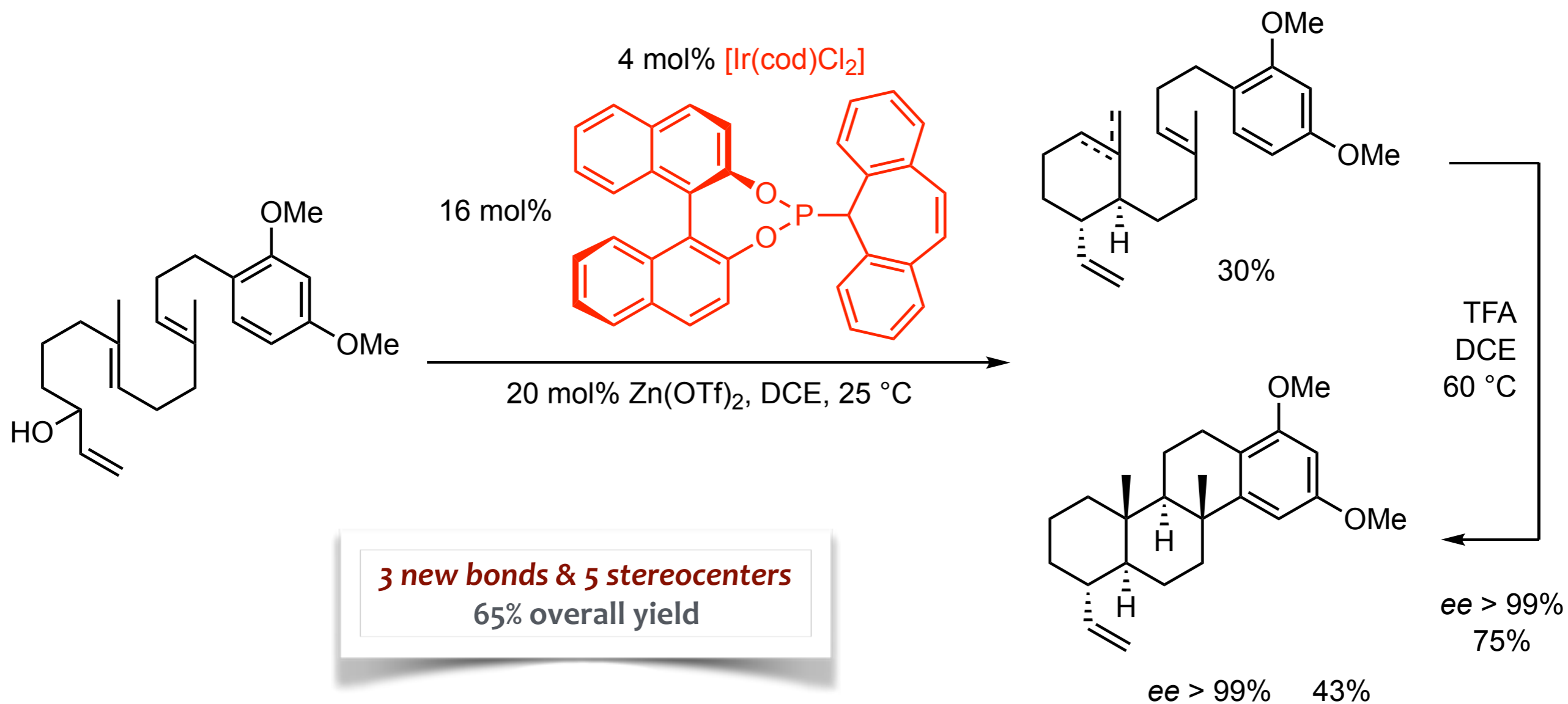
Electrophilic Cascades. Polyene Cyclizations

Polyene cyclizations controlled by chiral Lewis acid can be **enantioselective** and **catalytic**



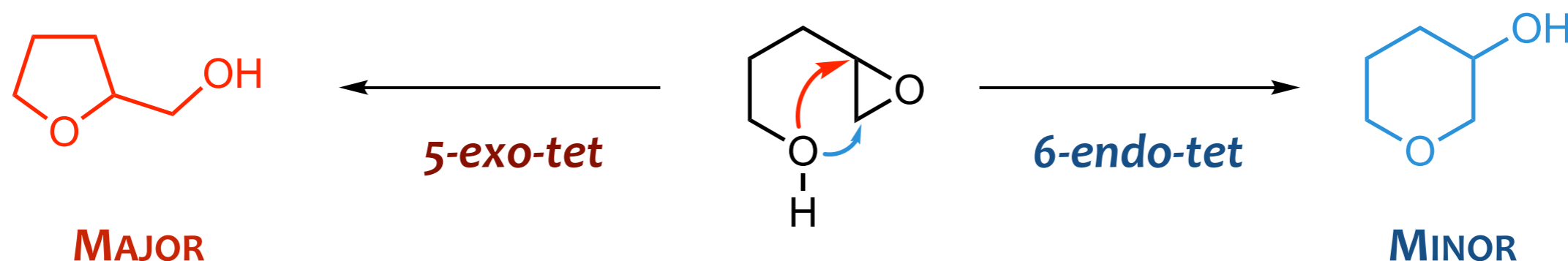
Electrophilic Cascades. Polyene Cyclizations

Polyene cyclizations controlled by chiral Lewis acid can be **enantioselective** and **catalytic**

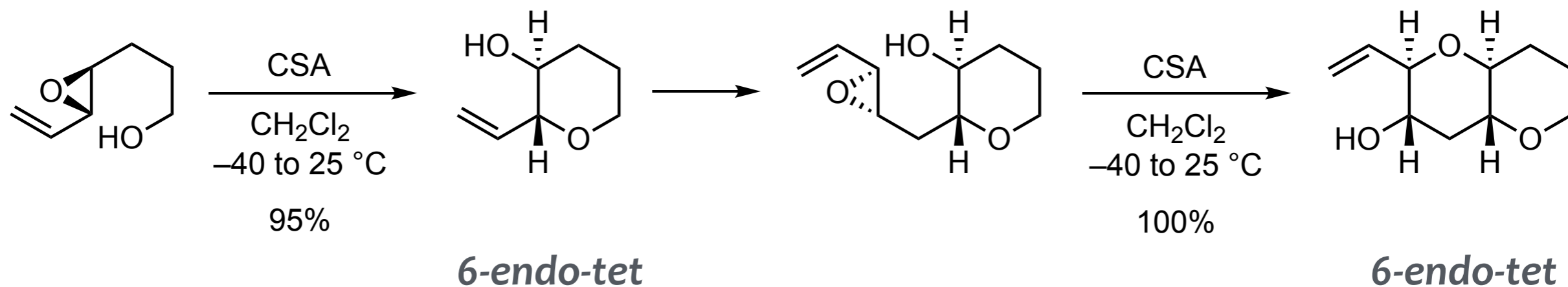


Electrophilic Cascades. Epoxide Opening

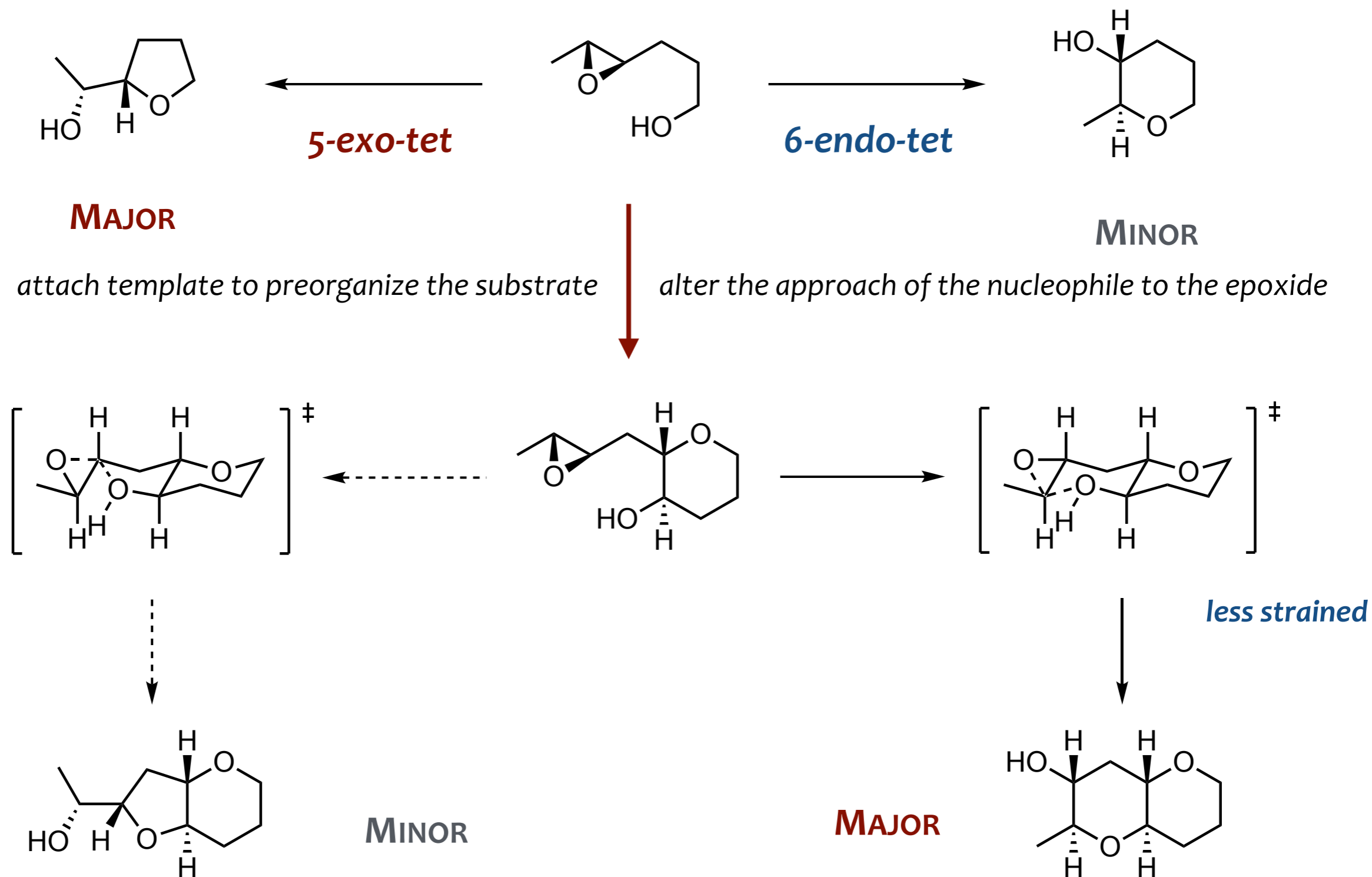
The cyclization of hydroxy epoxides often favors the **exo-tet** opening



Methods for **endo-cyclizations** rely on substituents that stabilize the partial positive charge or directing groups that deactivate the exo pathway



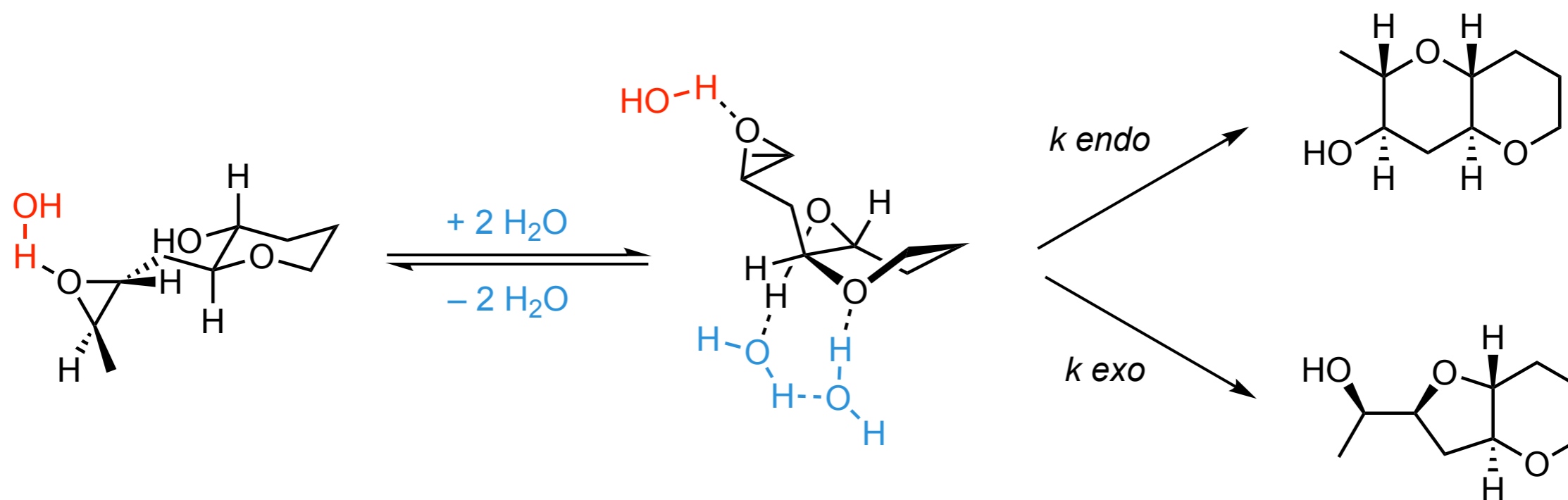
Electrophilic Cascades. Epoxide Opening



Electrophilic Cascades. Epoxide Opening

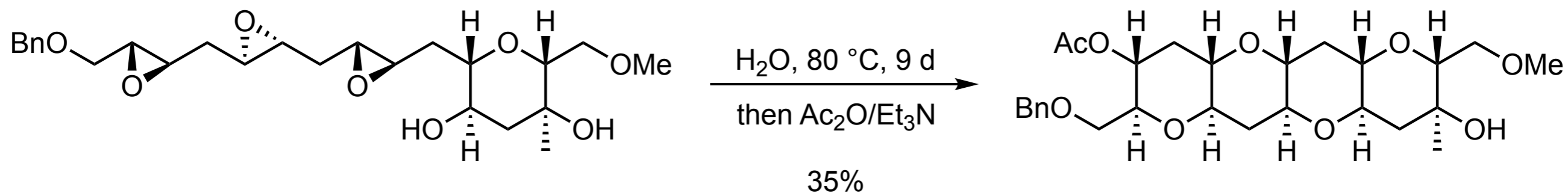
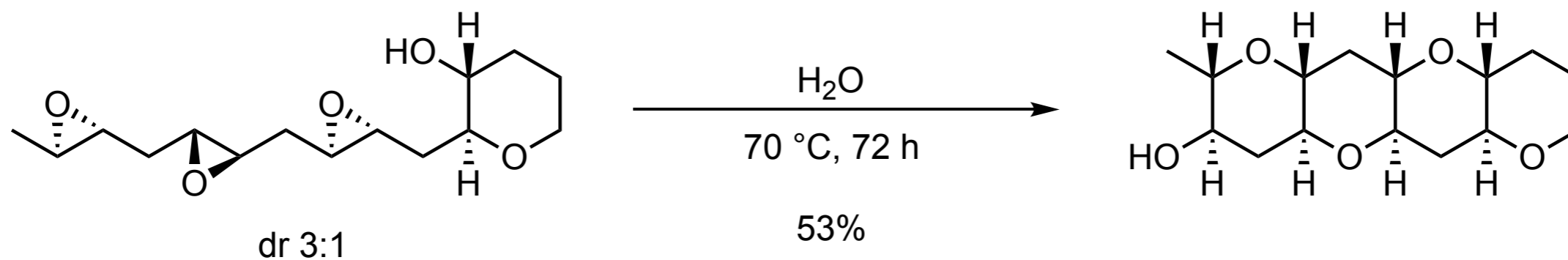
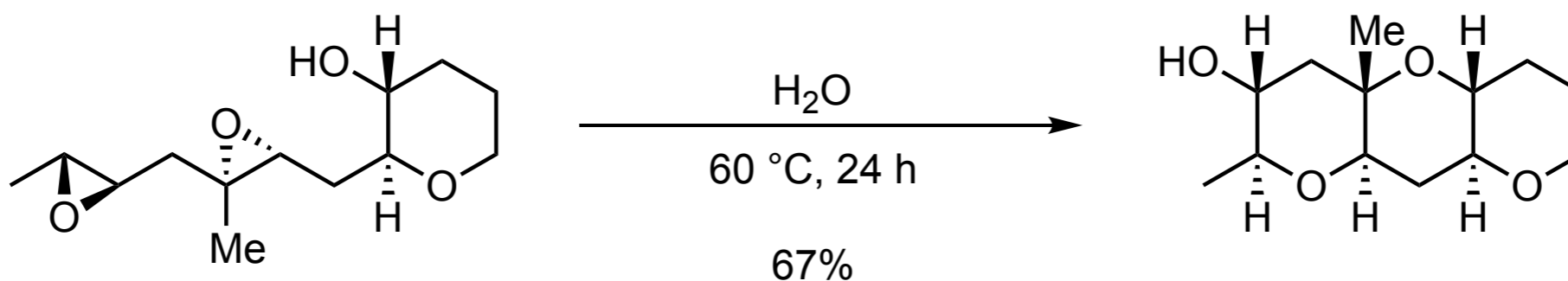
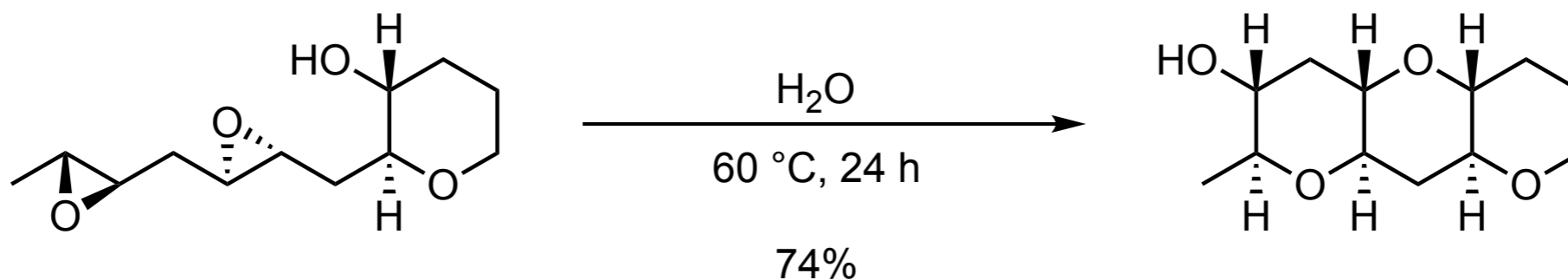
Water plays a crucial role both in the conversion and the selectivity

Jamison suggested that epoxy alcohol cyclizations in water occur for hydrated conformations with the appropriate geometry for the reaction



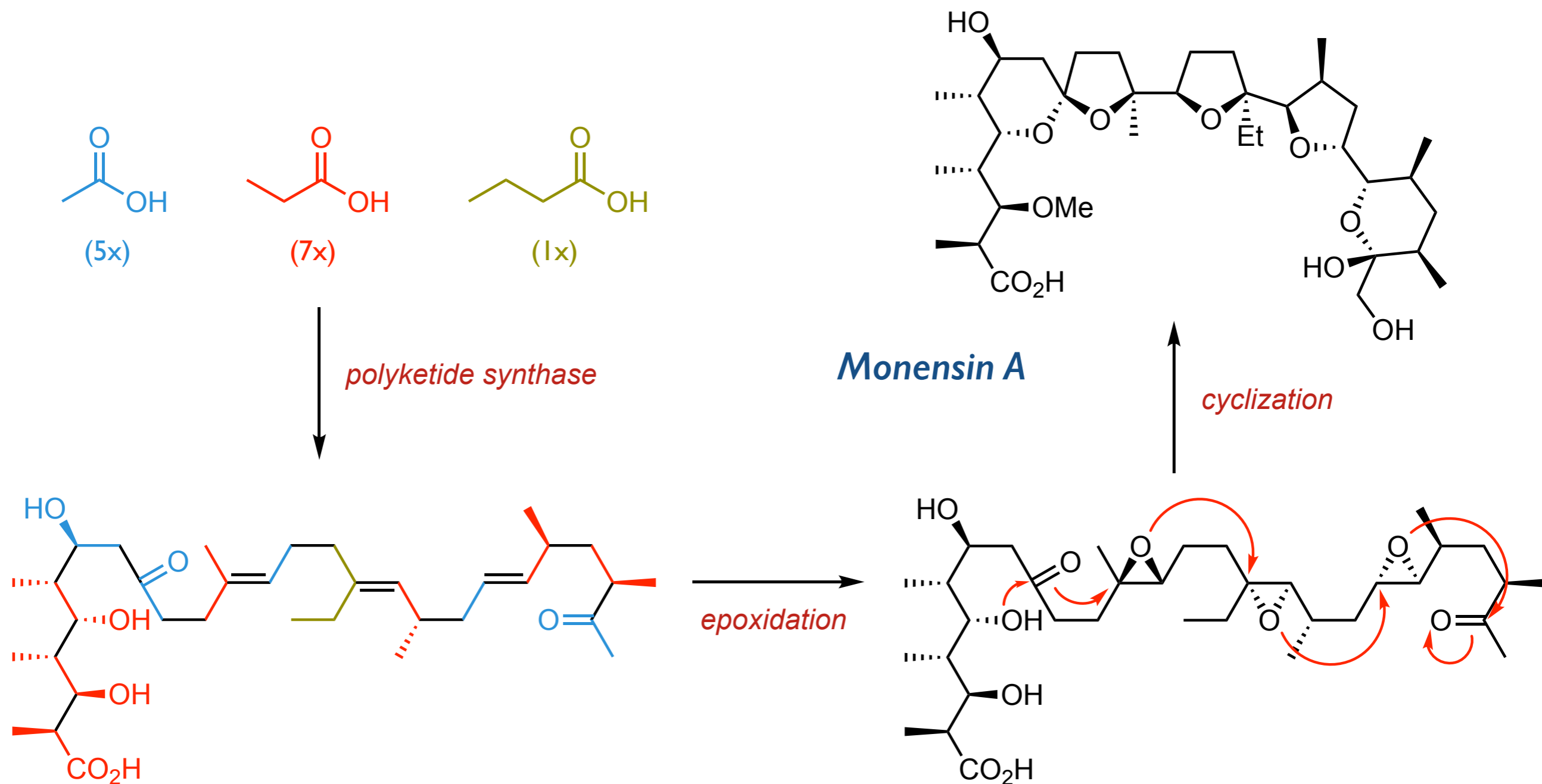
$k_{\text{endo}} \gg k_{\text{exo}}$

Electrophilic Cascades. Epoxide Opening

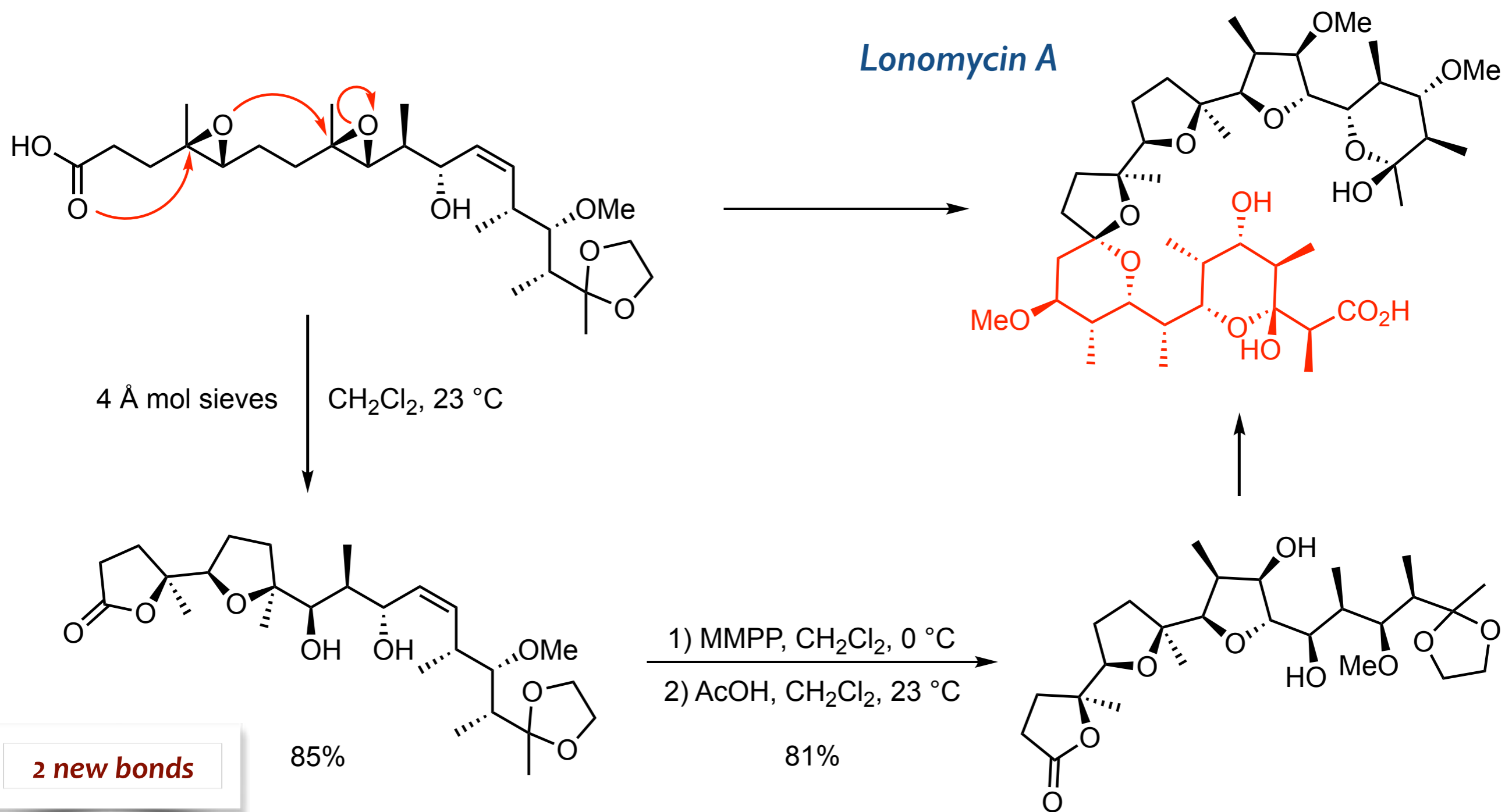


Electrophilic Cascades. Epoxide Opening

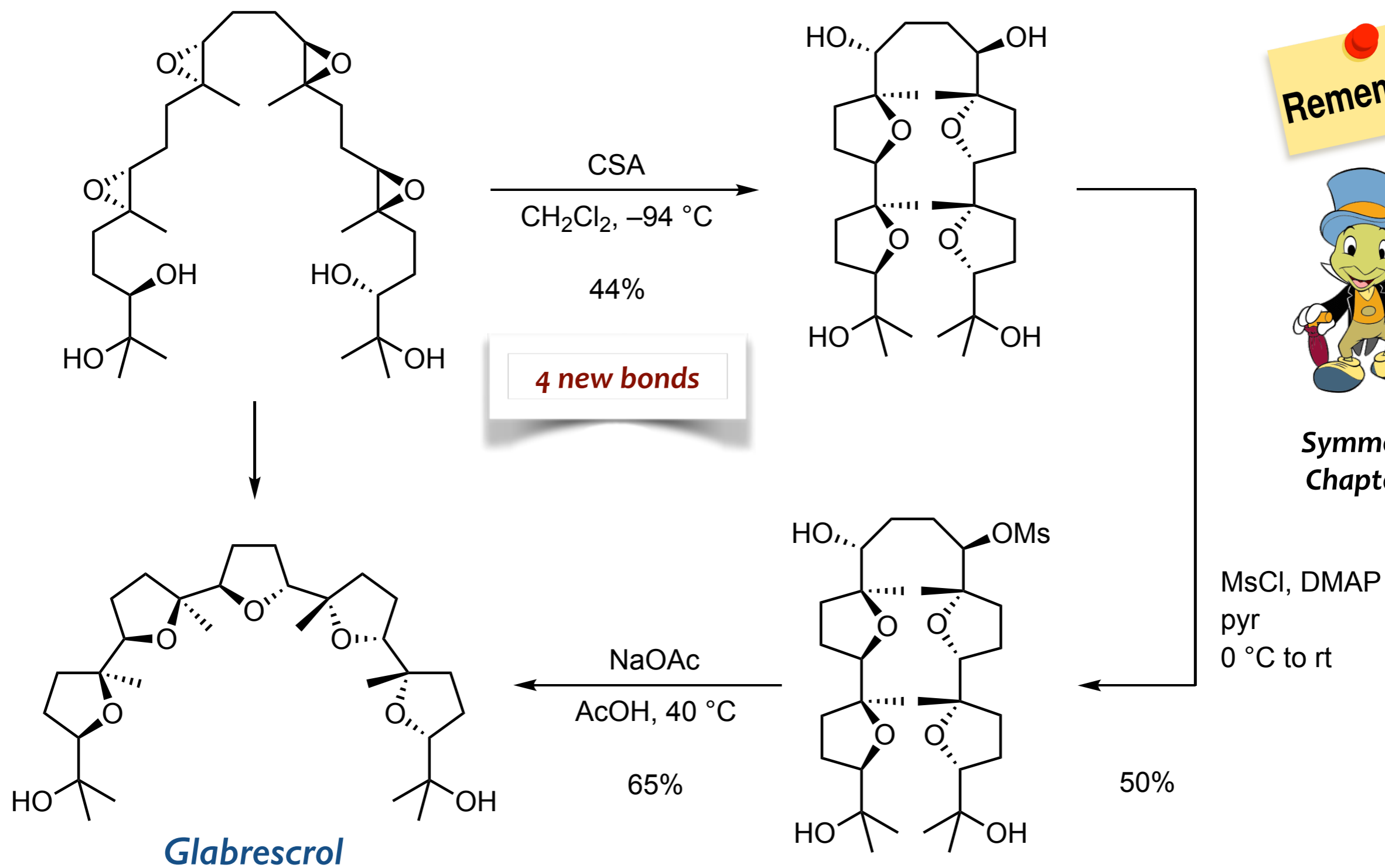
Cane-Celmer-Westley hypothesis on the polyether biogenesis



Electrophilic Cascades. Epoxide Opening



Electrophilic Cascades. Epoxide Opening



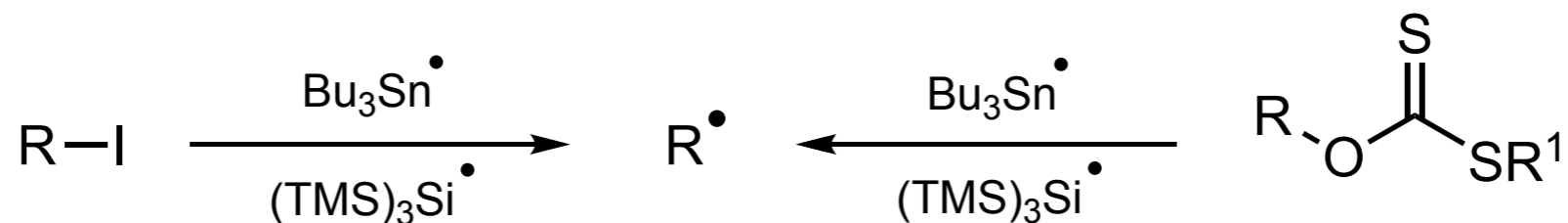
Remember!



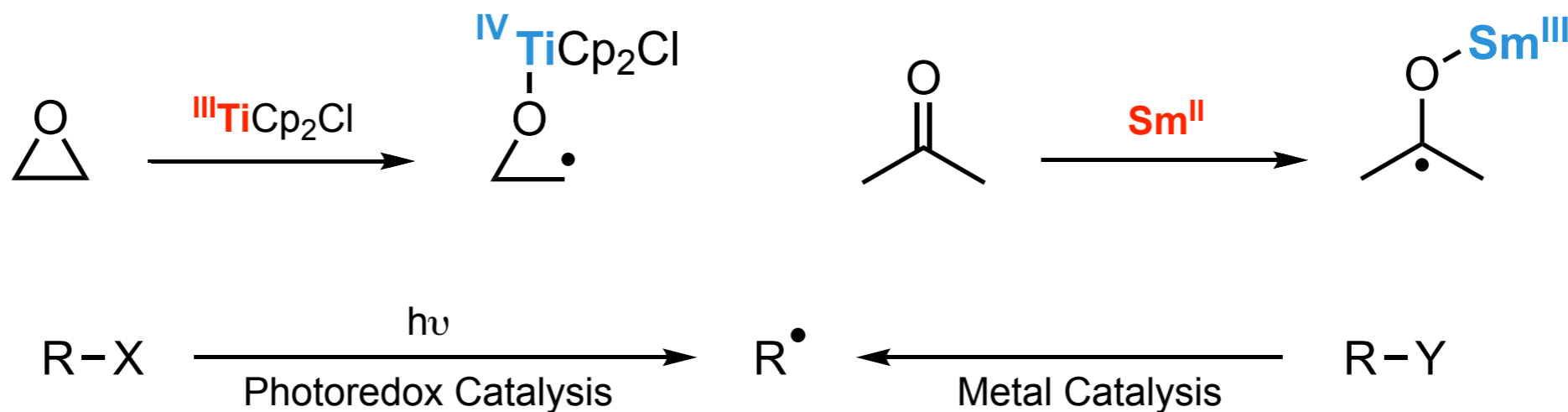
**Symmetry
Chapter 3**

Radical Cascades

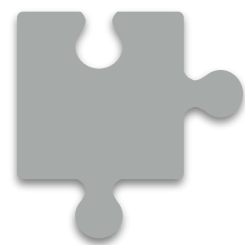
Tin & Silicon-mediated Cascades



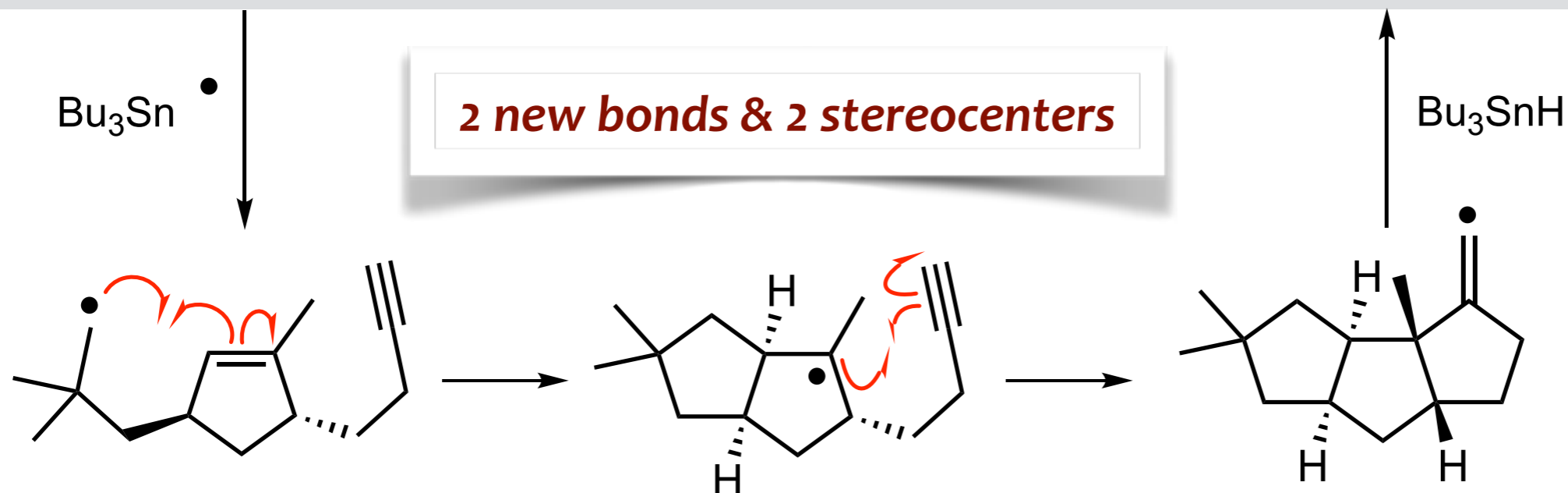
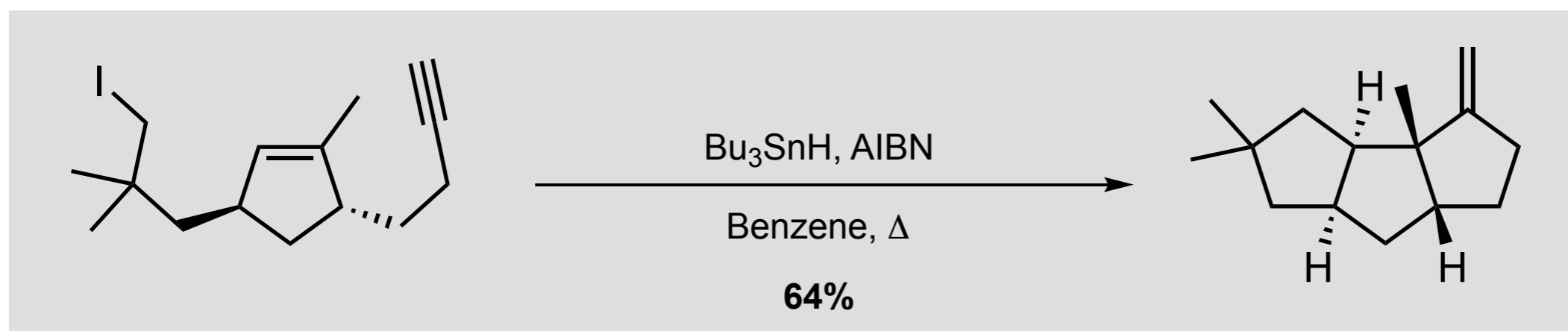
Transition metal-mediated Cascades (SET)



Radical Cascades. Tin Mediated



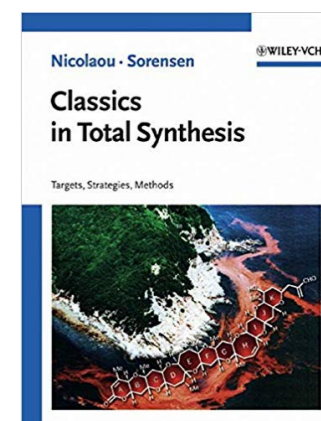
Curran's landmark synthesis of hirsutene



Remember!

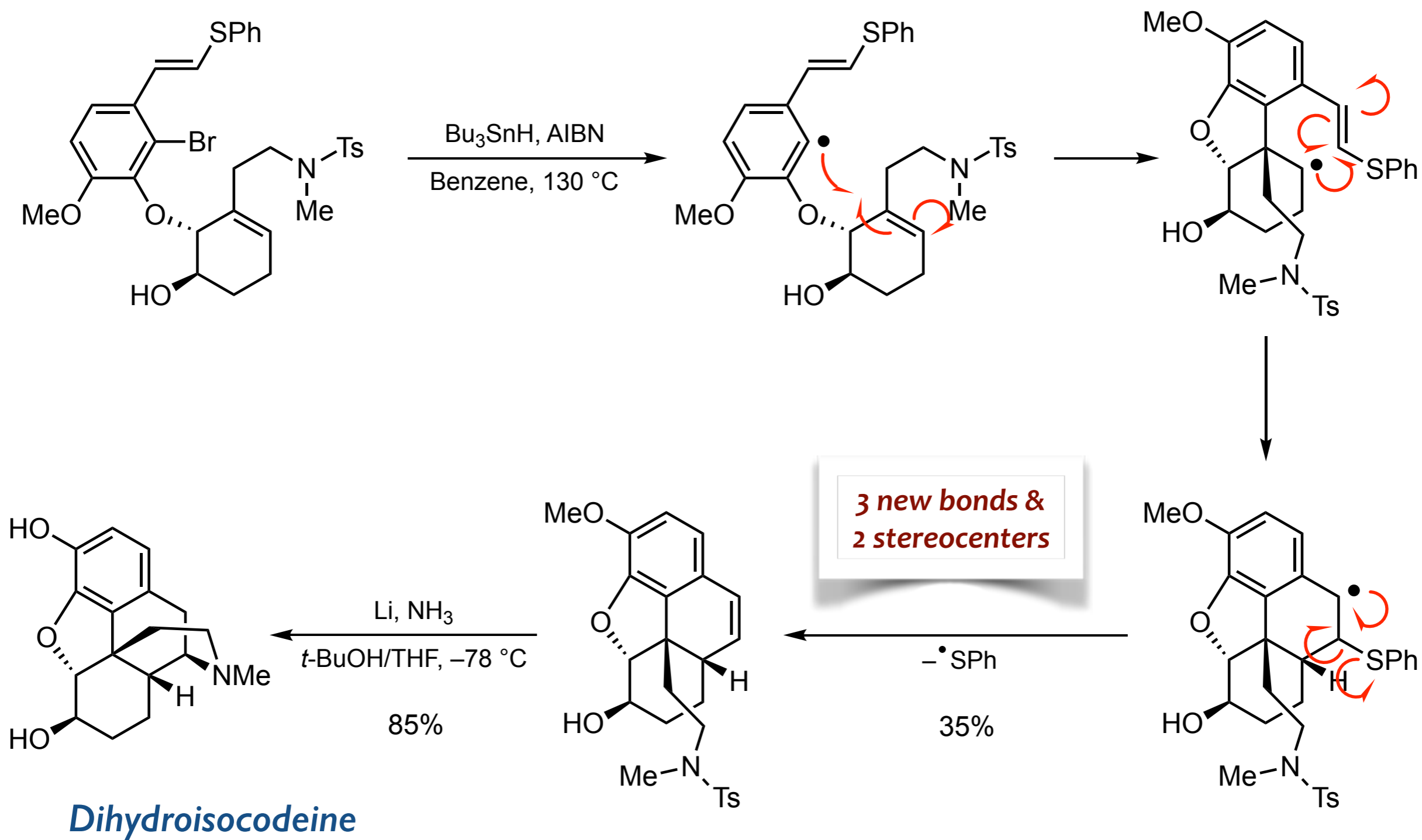


**Cyclizations
Chapter 6**



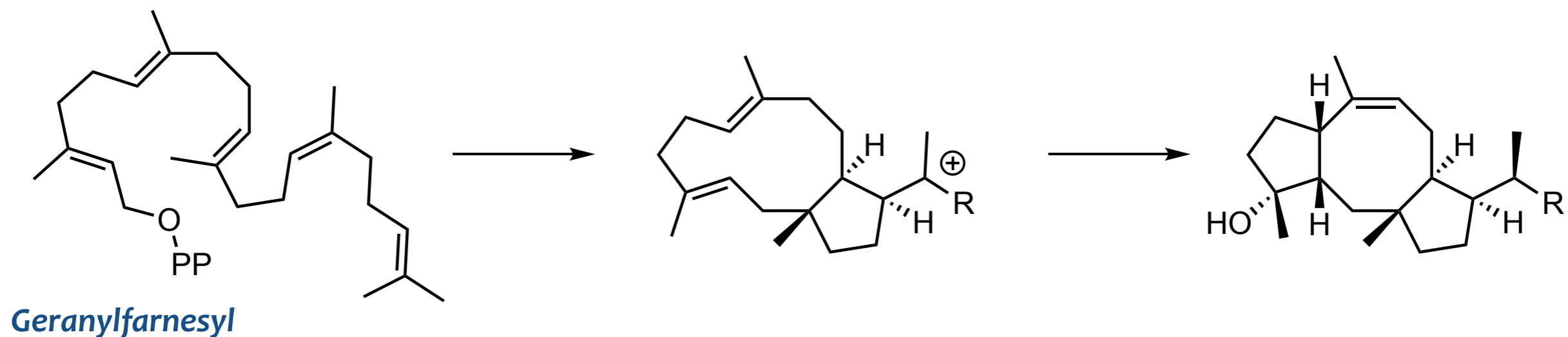
For an account, see p 381

Radical Cascades. Tin Mediated

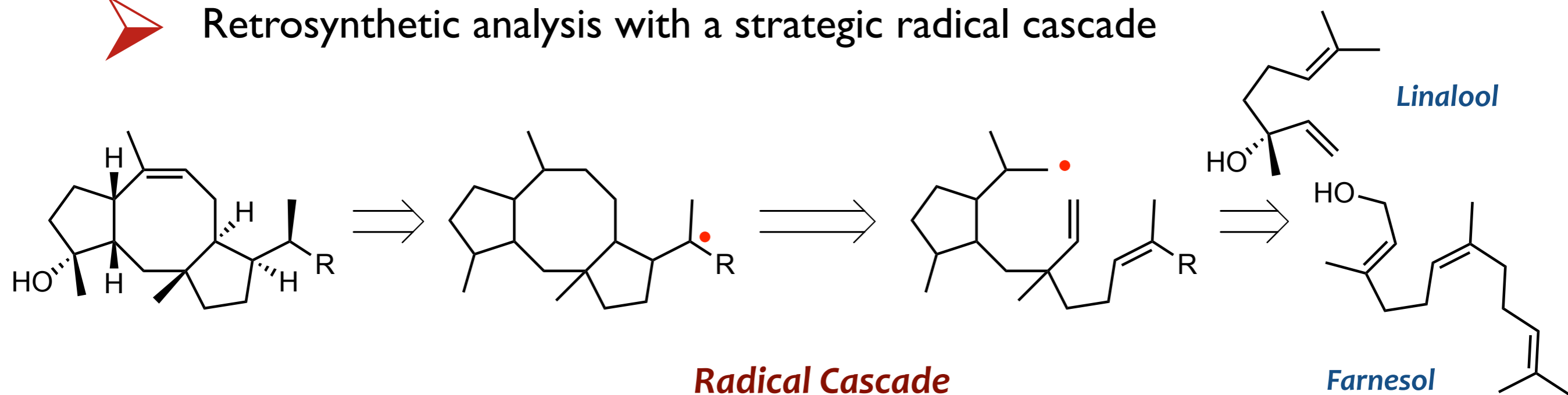


Radical Cascades. Silicon Mediated

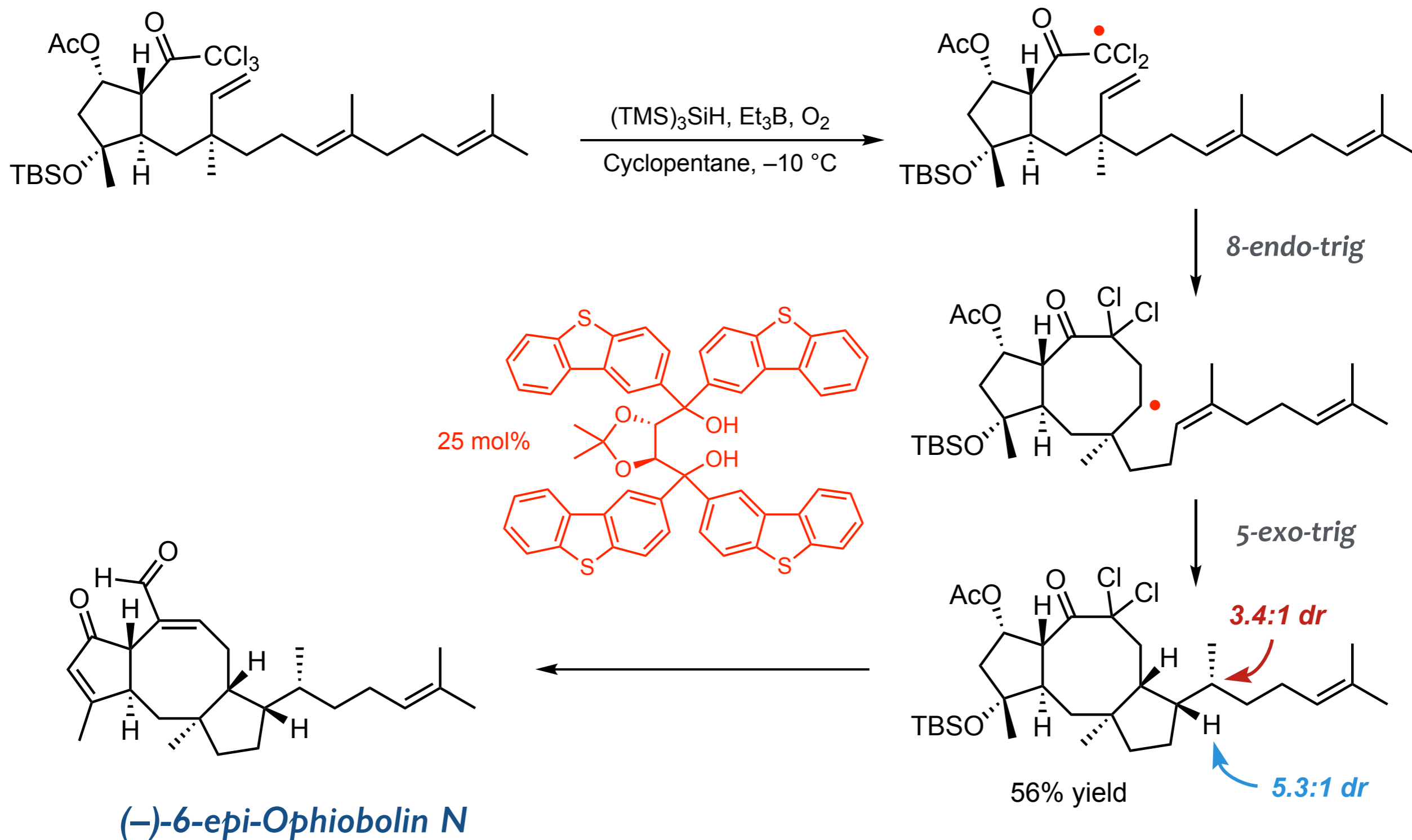
Proposed biosynthesis of ophiobiolin sesterterpenes



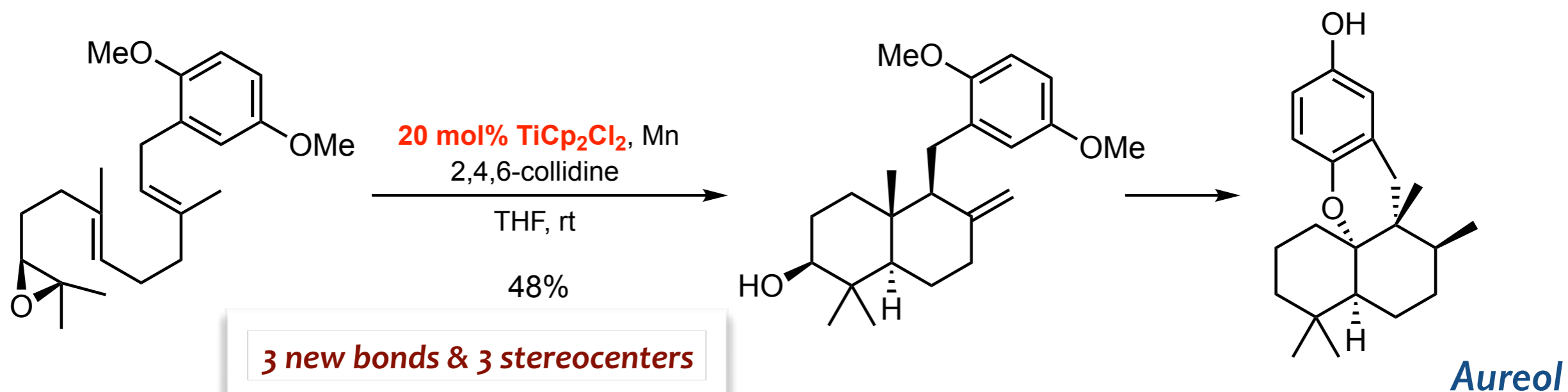
Retrosynthetic analysis with a strategic radical cascade



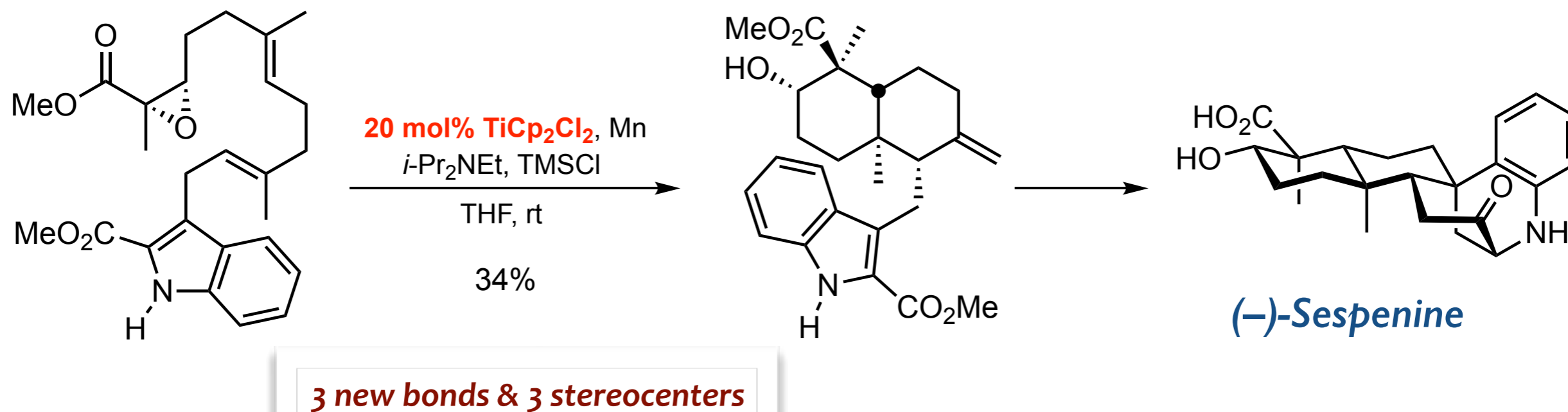
Radical Cascades. Silicon Mediated



Radical Cascades. Epoxide Opening

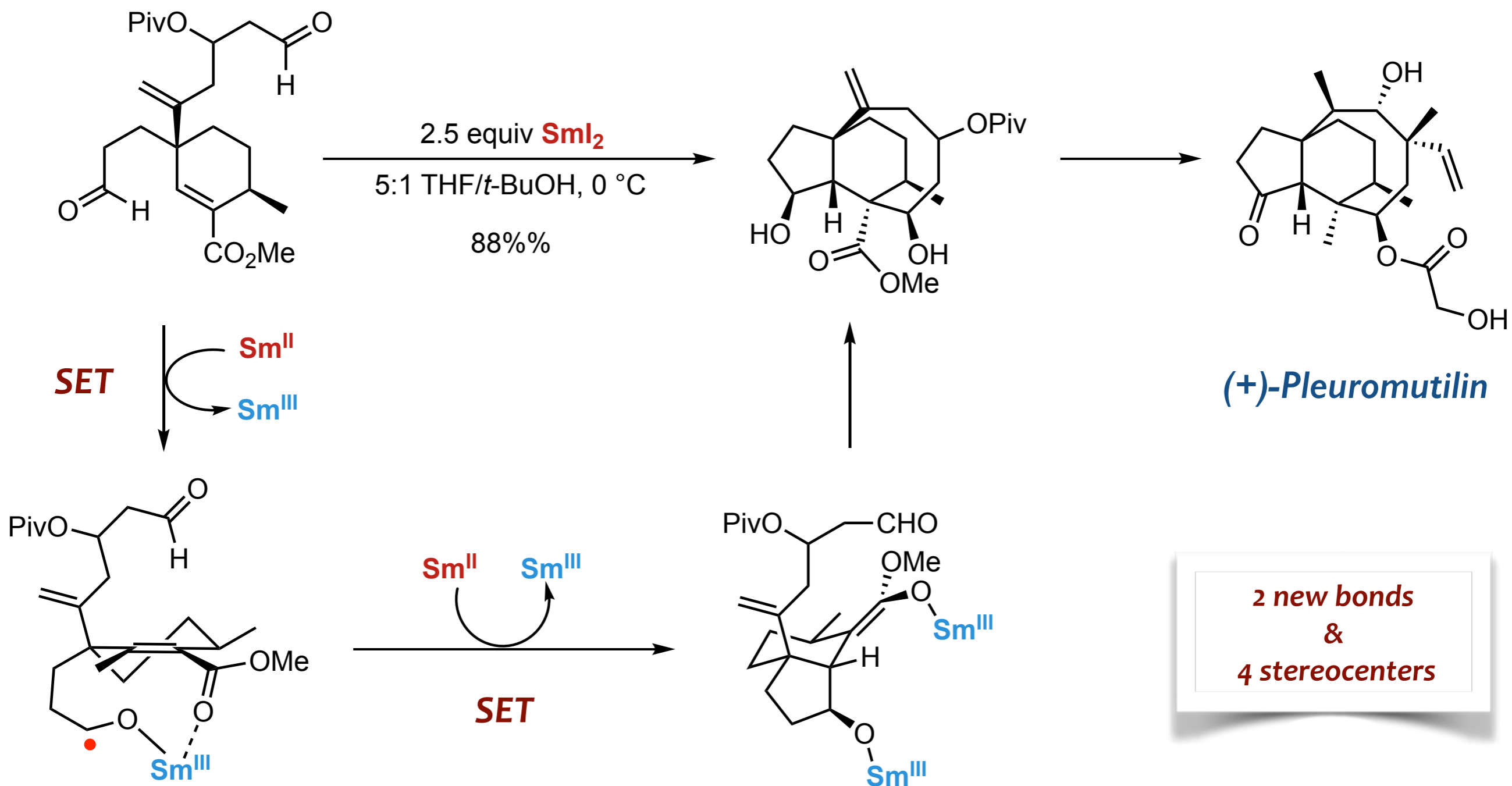


Rosales, A.; Oltra, J. E. *JOC* 2015, 80, 1866

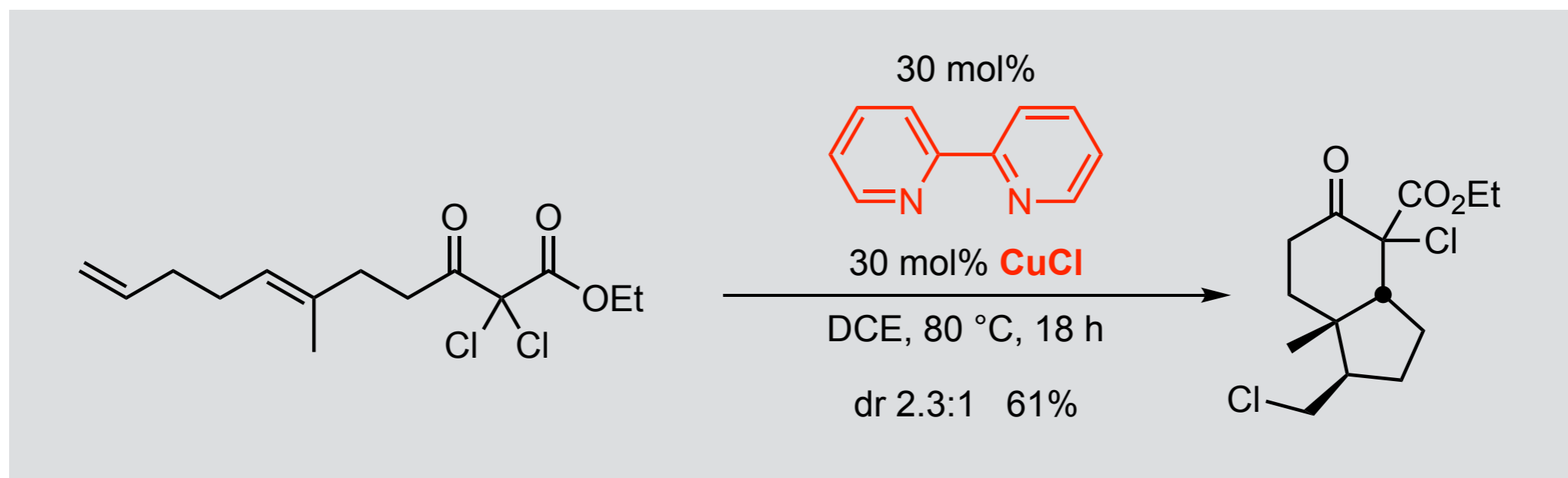


Li, A. *OCF* 2016, 3, 368

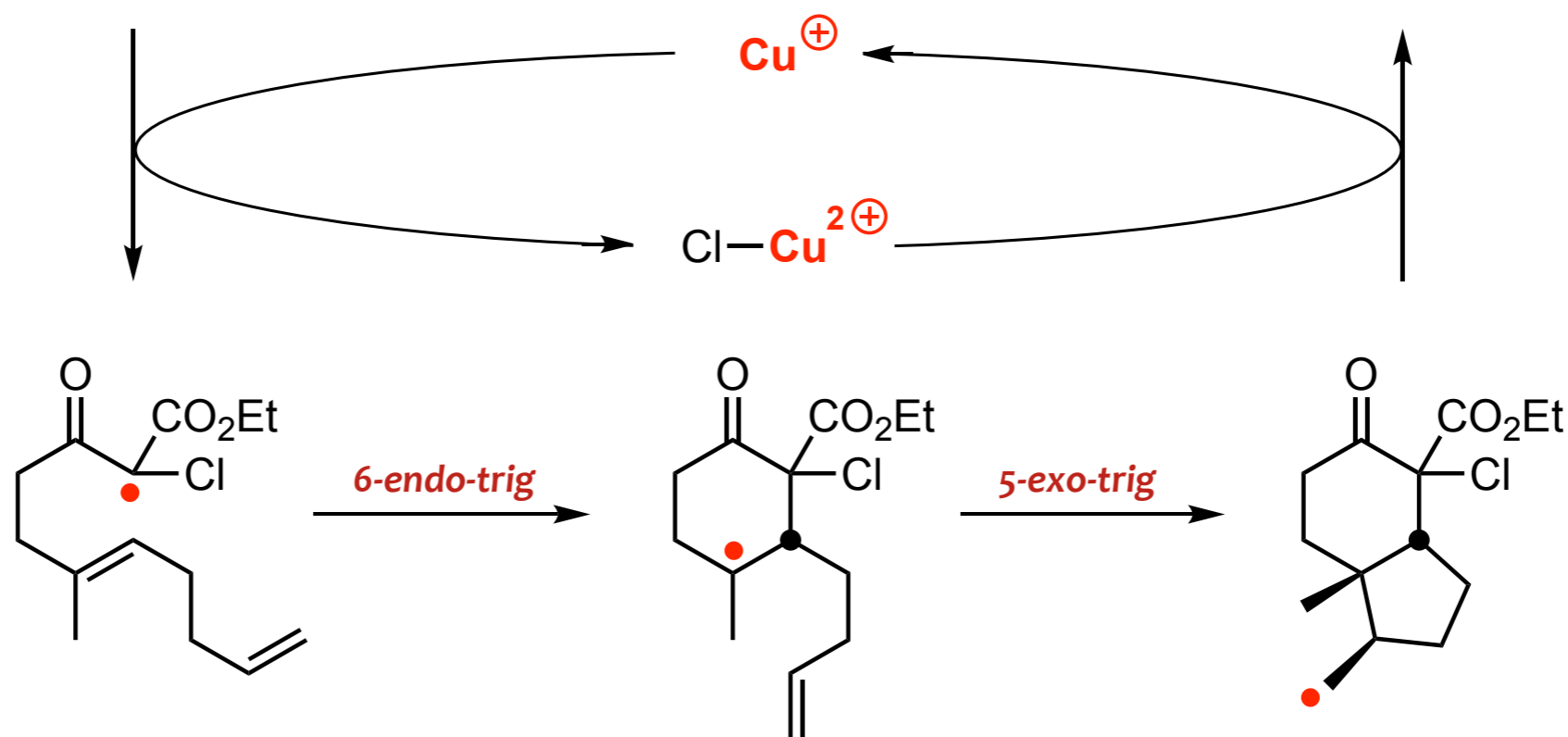
Radical Cascades. Samarium Mediated



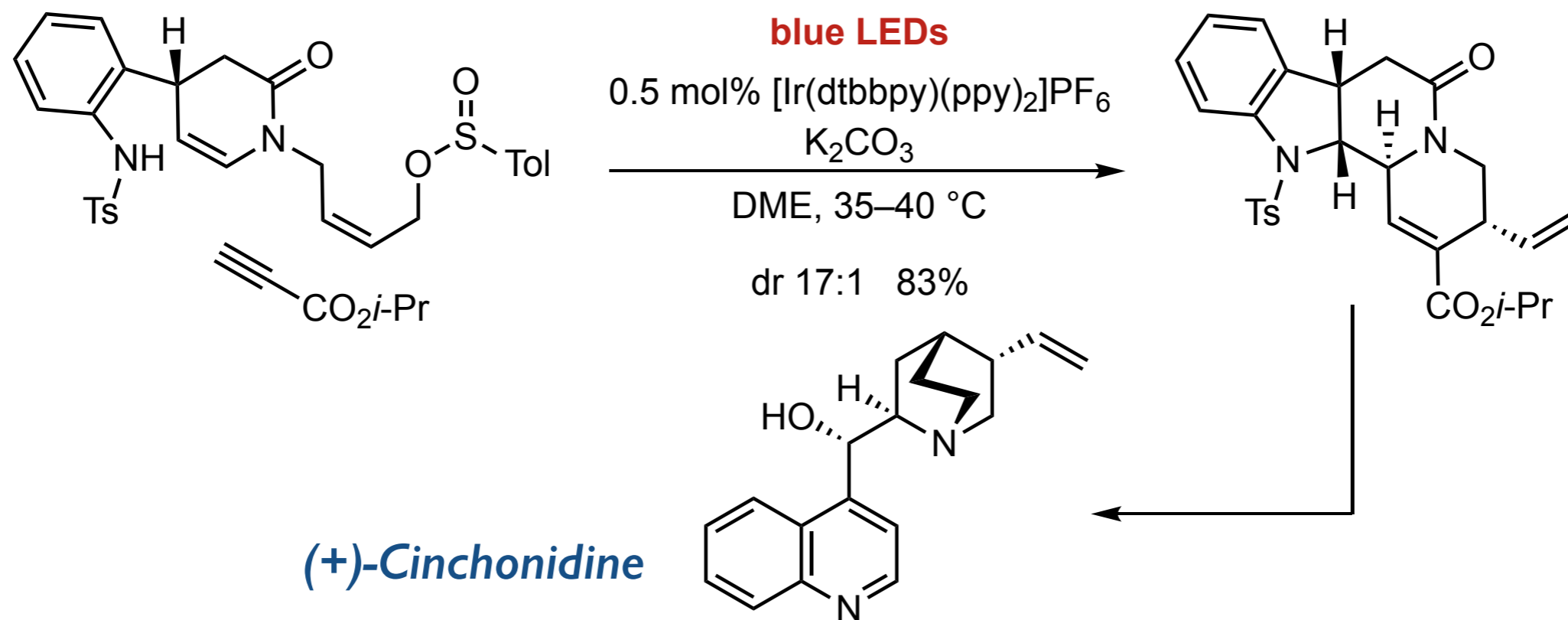
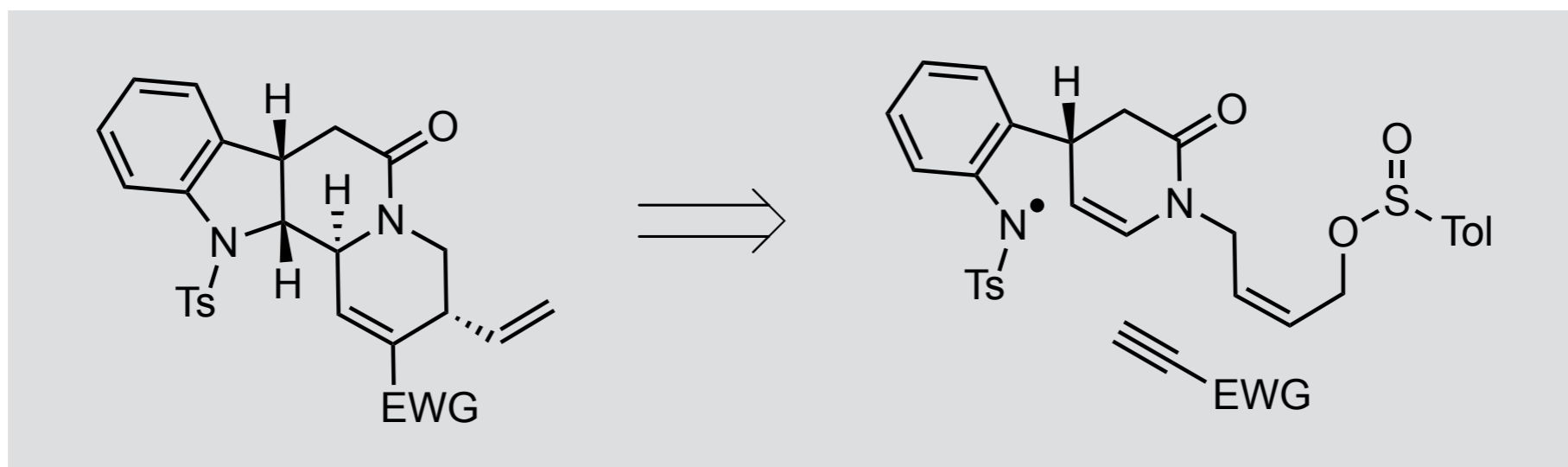
Radical Cascades. Copper Mediated



3 new bonds
&
3 stereocenters



Radical Cascades. Photoredox Catalysis



Pericyclic Cascades

Angewandte
Reviews

Molecules and Music

Toward a Symphony of Reactivity: Cascades Involving Catalysis and Sigmatropic Rearrangements

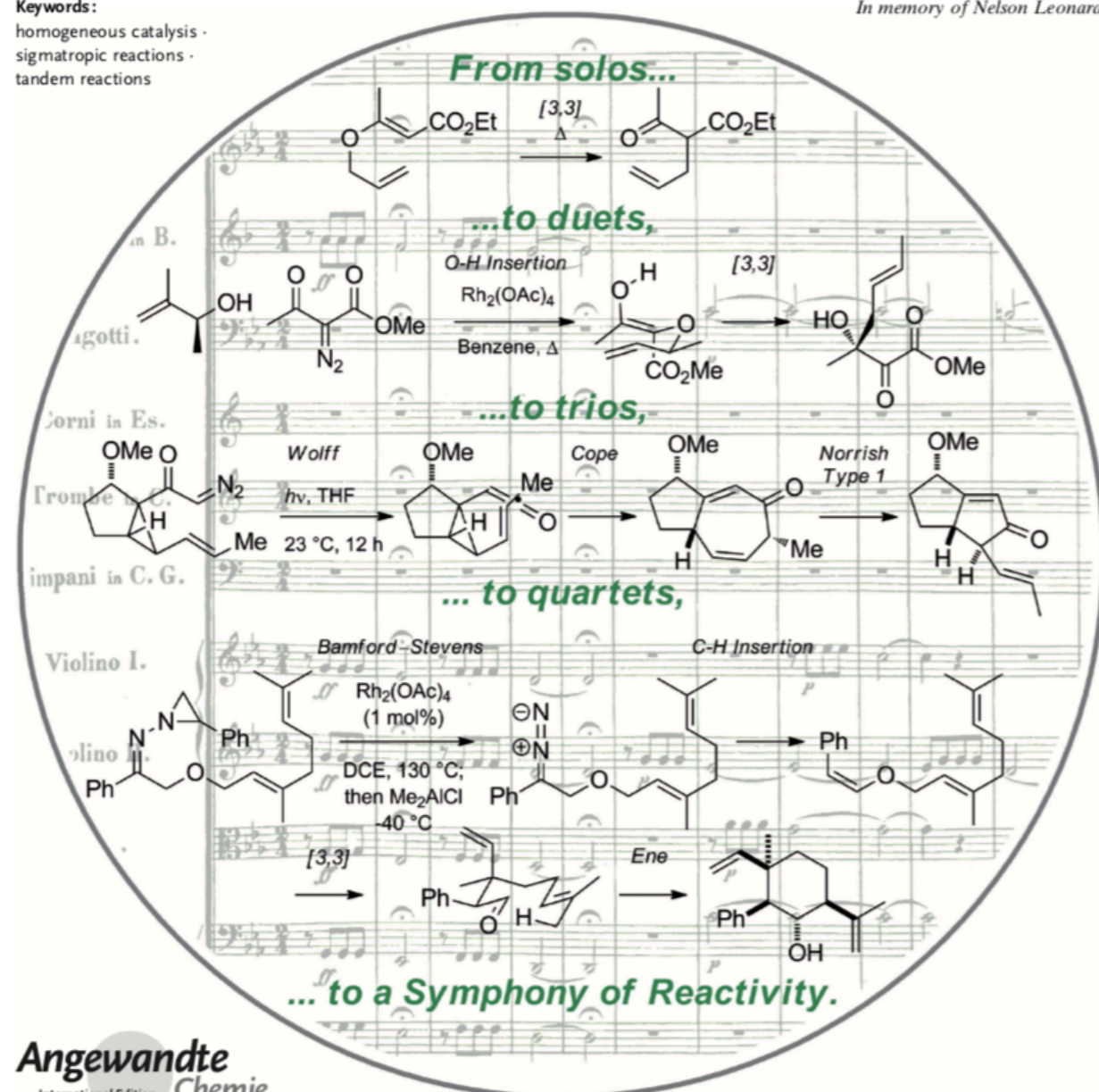
Amanda C. Jones,* Jeremy A. May, Richmond Sarpong, and Brian M. Stoltz*

A. C. Jones, B. M. Stoltz et al.

DOI: 10.1002/anie.201302572

Keywords:
homogeneous catalysis ·
sigmatropic reactions ·
tandem reactions

In memory of Nelson Leonard



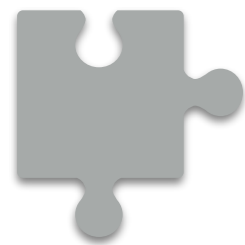
In many regards,
the inherent beauty associated with a synthetic sequence can be linked to a certain combination of the creativity with which a sequence is designed and the overall efficiency with which the ultimate process is performed. In synthesis, as in other endeavors, beauty is very much in the eyes of the beholder

Jones, A. C. ; Stoltz, B. M.

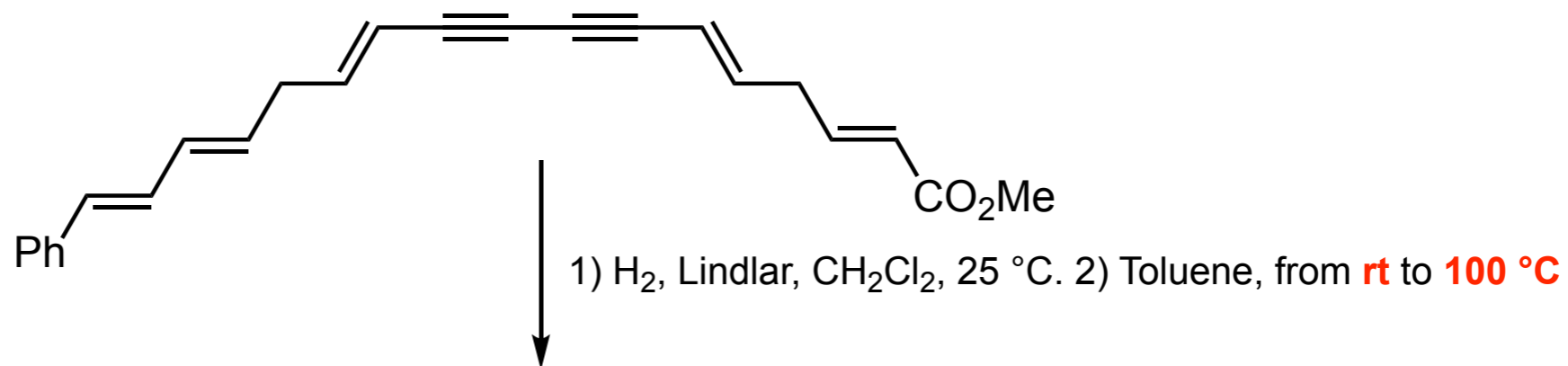
ACIE 2014, 53, 2556

Angewandte
International Edition
Chemie

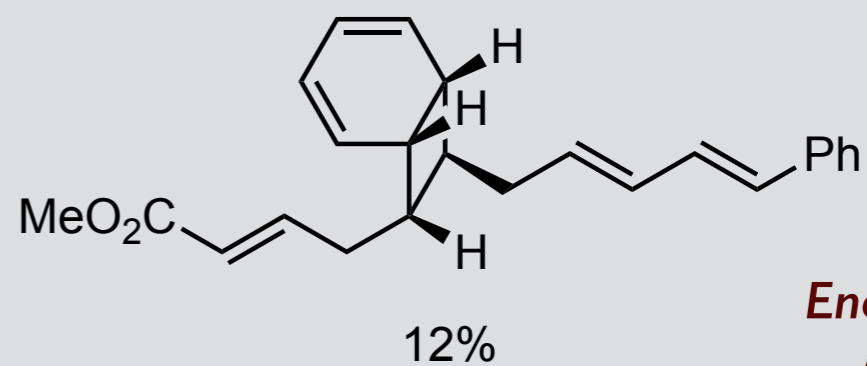
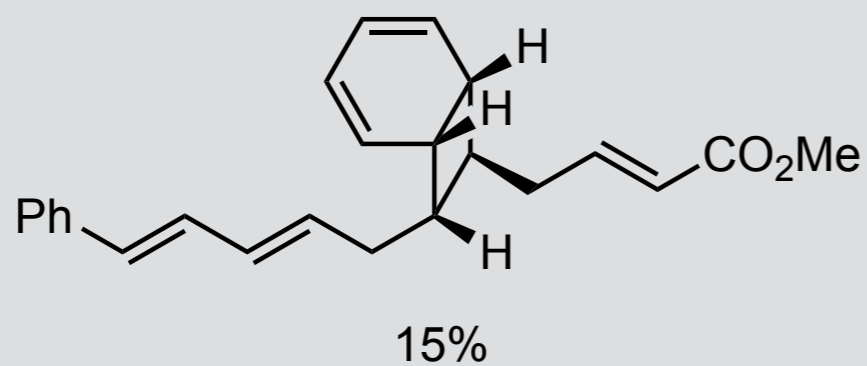
Electrocyclic Cascades



Nicolaou's landmark biomimetic synthesis of Endiandric Acid B, C, F, and G

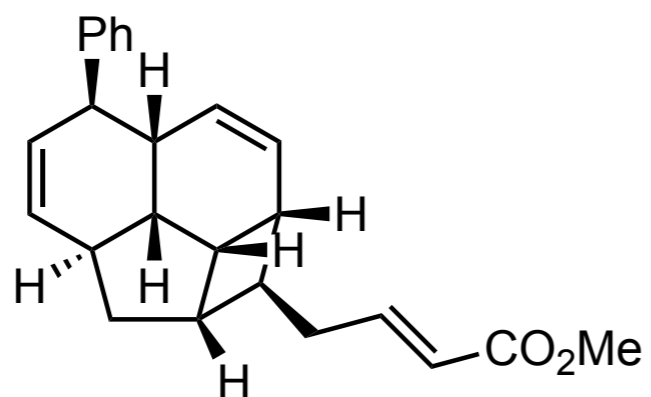


BEFORE HEATING



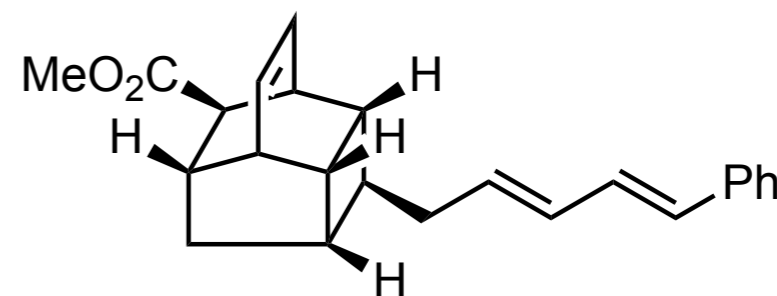
AFTER HEATING

Endiandric Acid B
Methyl Ester

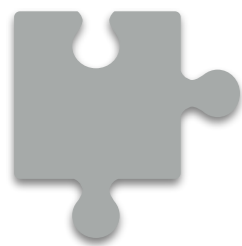


Overall yield 28%

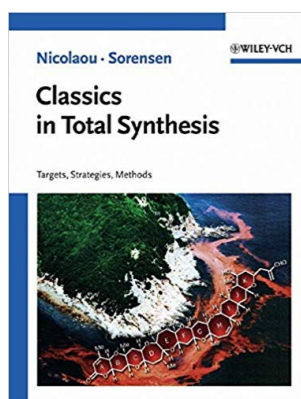
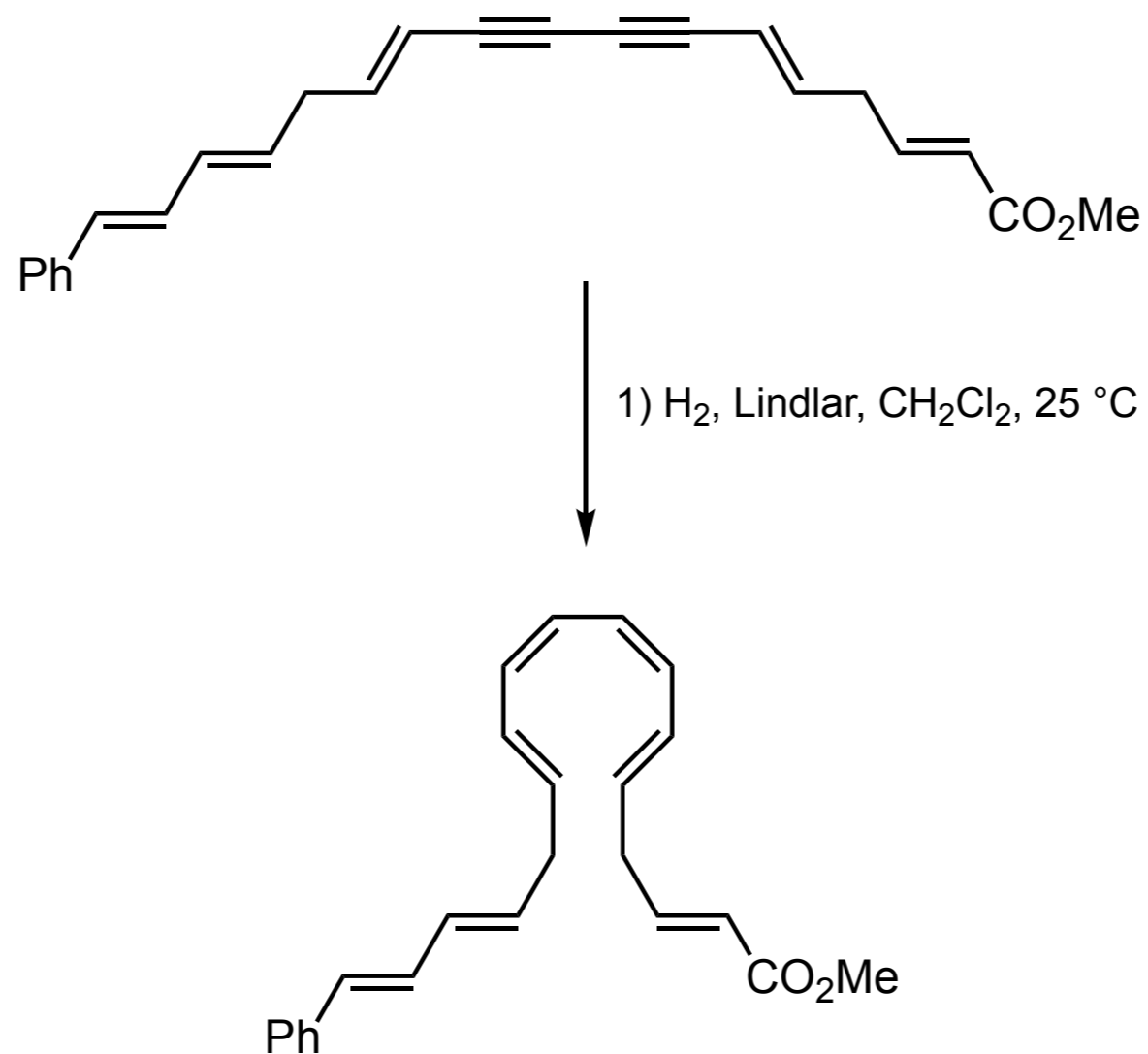
Endiandric Acid C
Methyl Ester



Electrocyclic Cascades



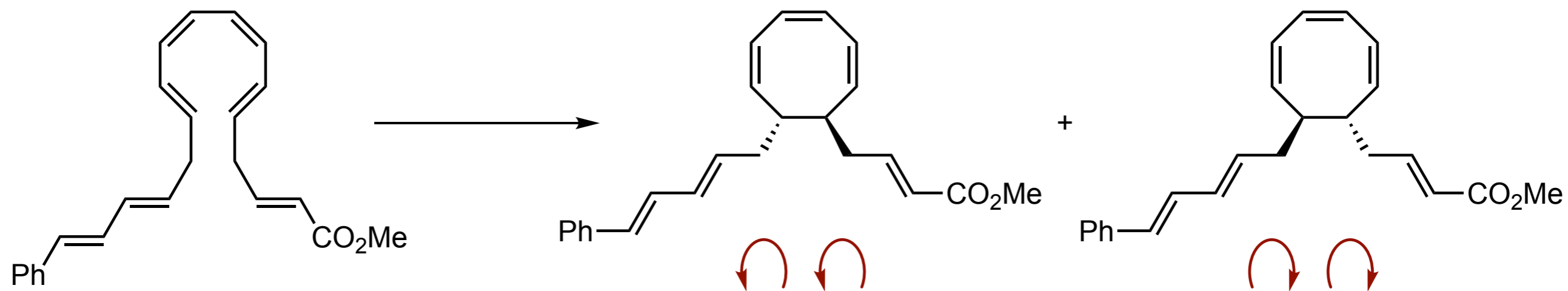
Nicolaou's landmark biomimetic synthesis of Endiandric Acid B, C, F, and G



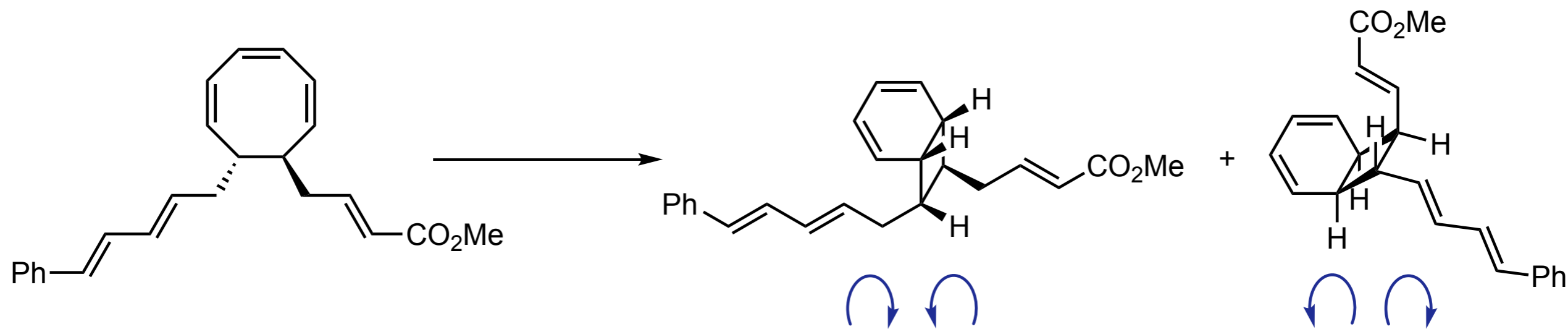
For an account, see p 265

Electrocyclic Cascades

➤ *Conrotatory 8π electron electrocyclic*

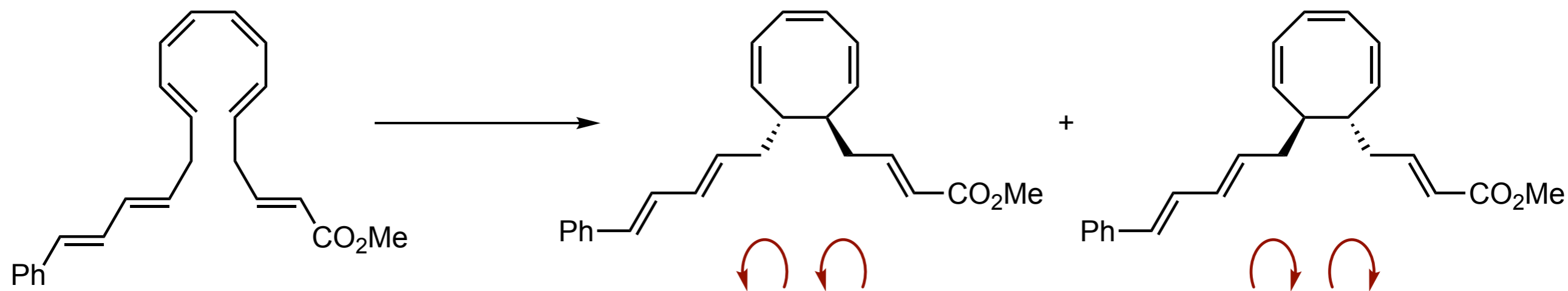


➤ *Disrotatory 6π electron electrocyclic*

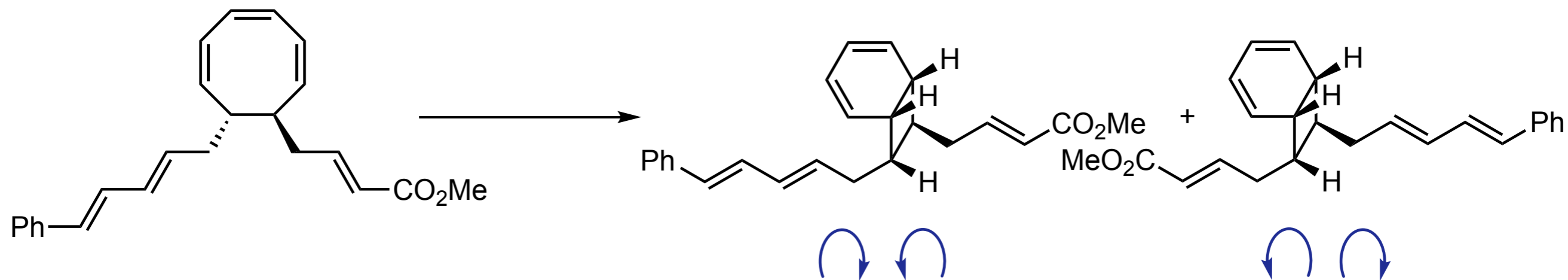


Electrocyclic Cascades

➤ *Conrotatory 8π electron electrocyclic*

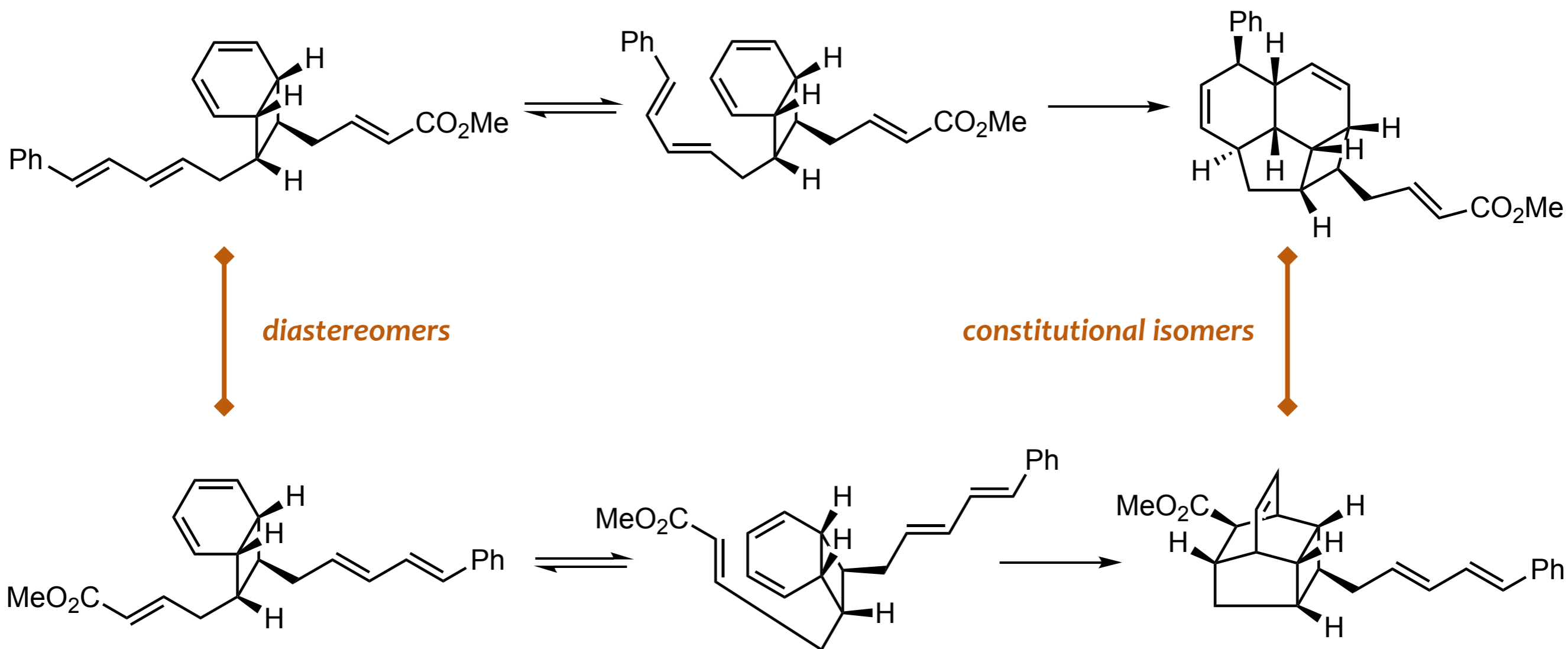


➤ *Disrotatory 6π electron electrocyclic*

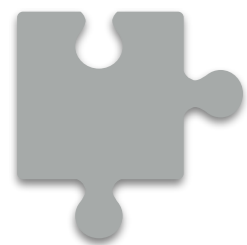


Electrocyclic Cascades

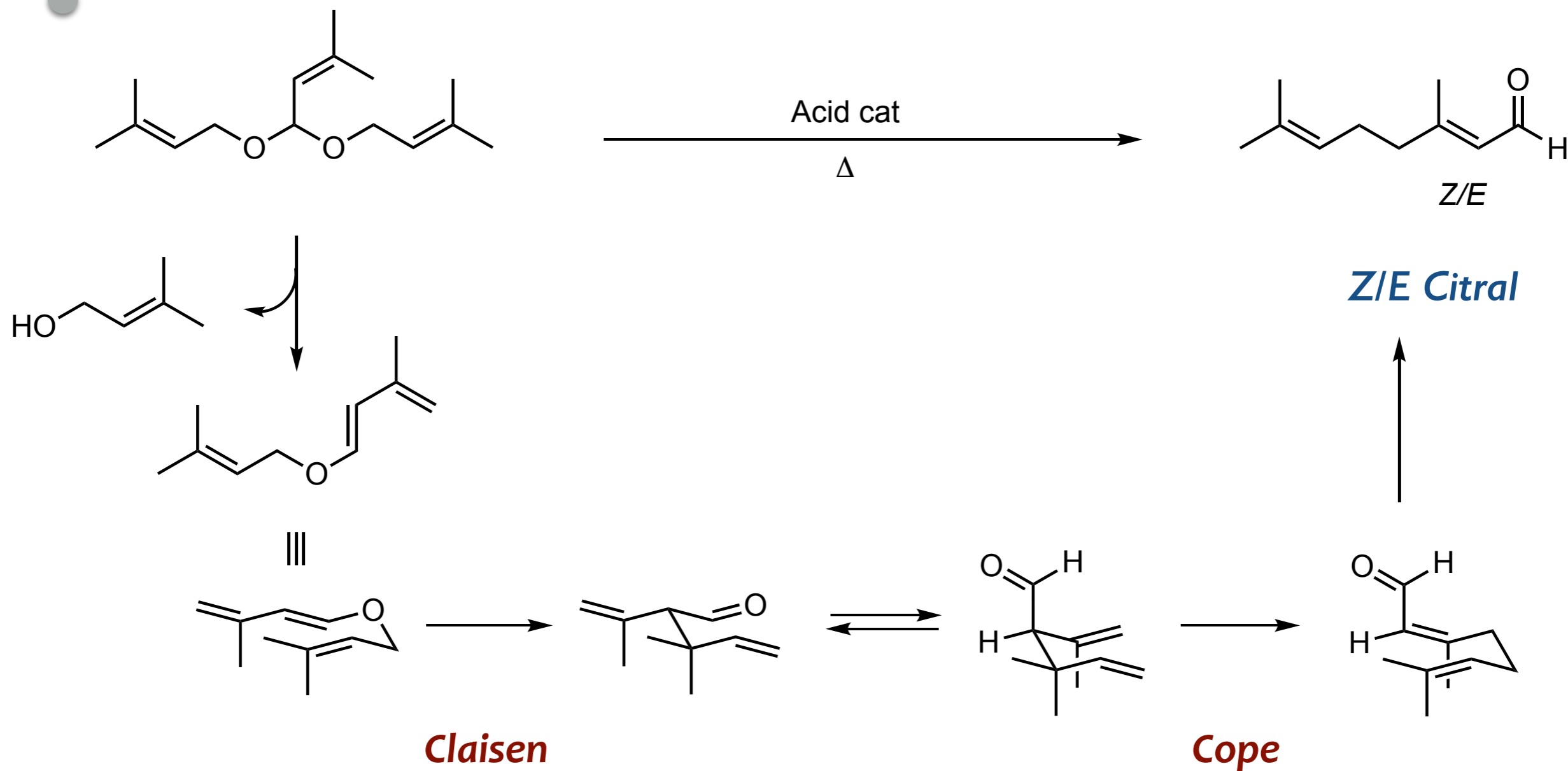
➤ Intramolecular Diels Alder



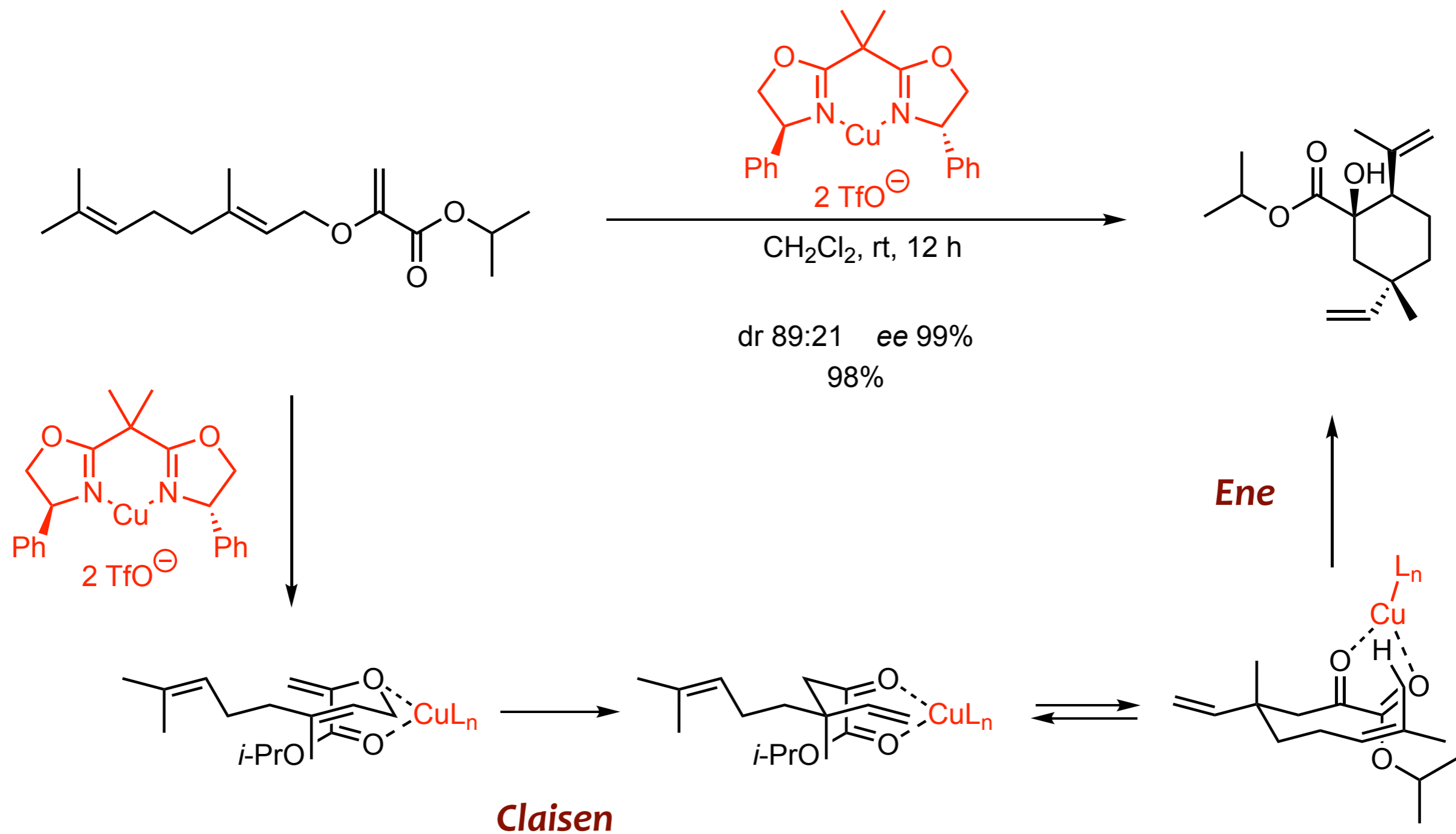
Sigmatropic Cascades



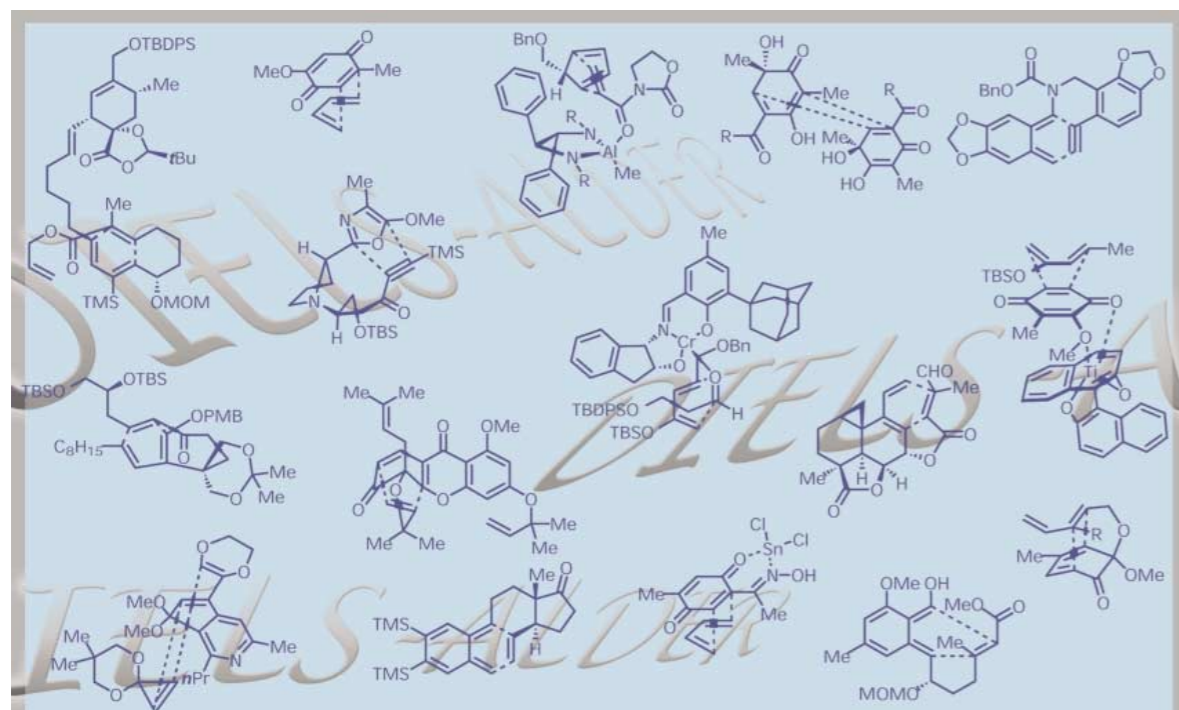
BASF Continuous Citral Process



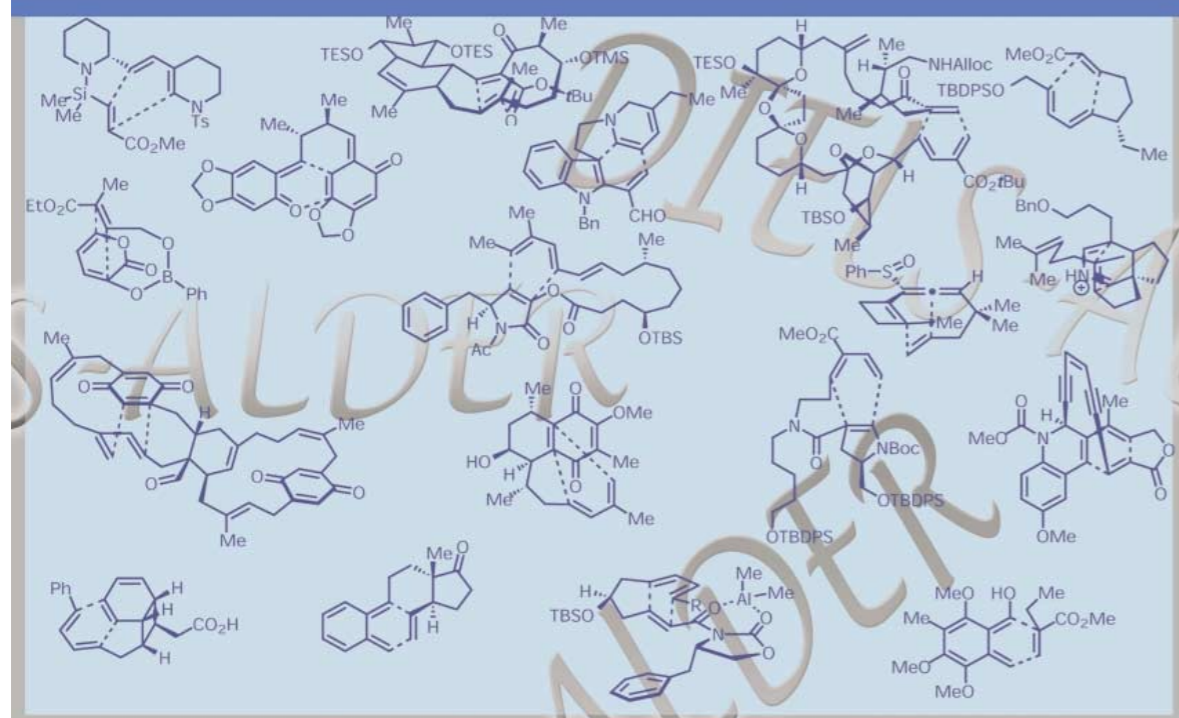
Sigmatropic Cascades



Diels Alder Cascades



The *Diels-Alder* Reaction in Action



The Diels-Alder reaction

has both enabled and shaped the art and science of total synthesis over the last few decades to an extent which, arguably, has yet to be eclipsed by any other transformation in the current synthetic repertoire. With myriad applications of this magnificent pericyclic reaction, often as a crucial element in elegant and programmed cascade sequences facilitating complex molecule construction, the Diels-Alder cycloaddition has afforded numerous and unparalleled solutions to a diverse range of synthetic puzzles provided by nature in the form of natural products.

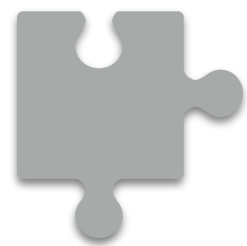
Nicolaou, K. C. *ACIE* 2002, 41, 1668



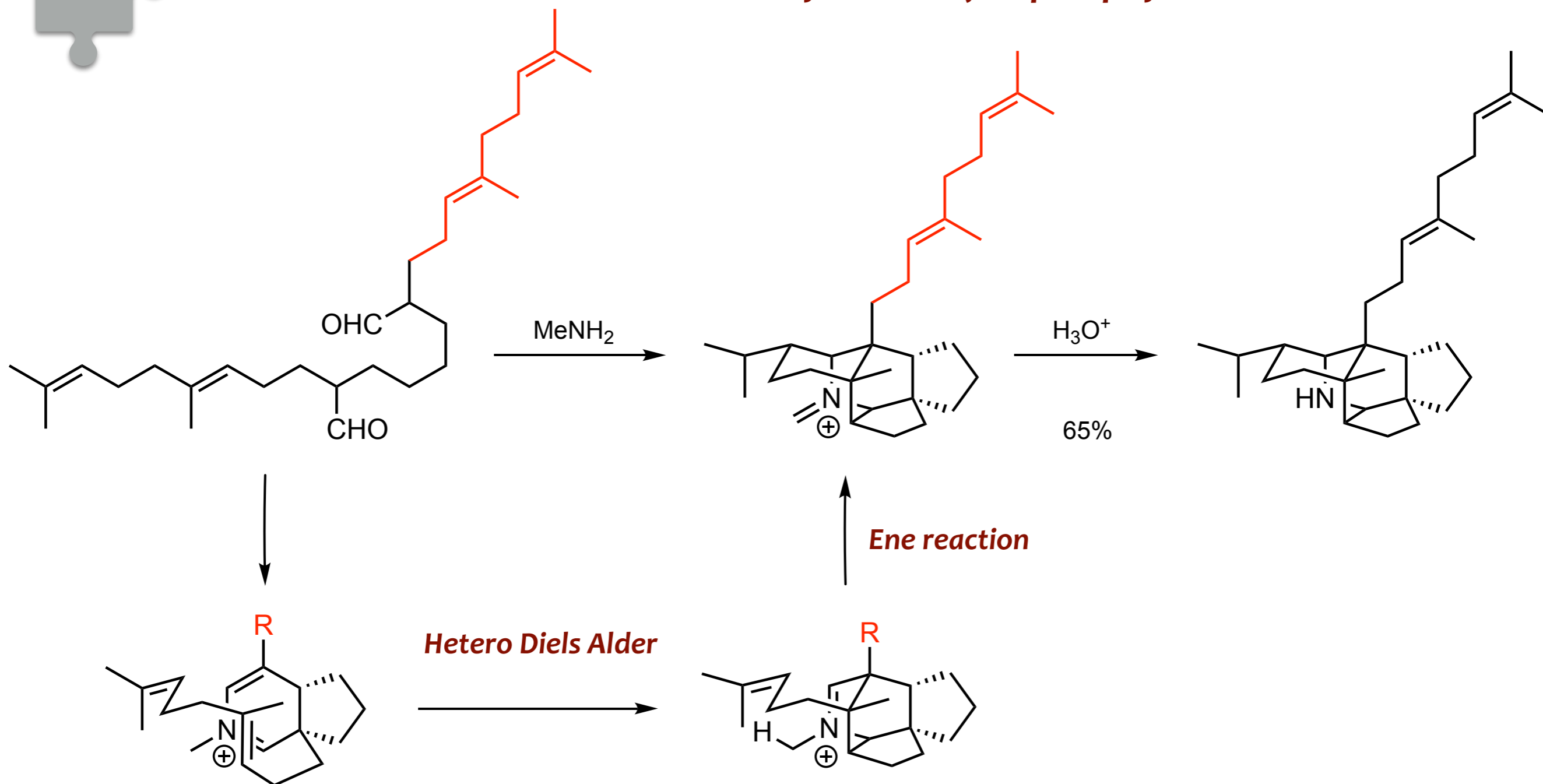
Cyclizations
Chapter 6

Remember!

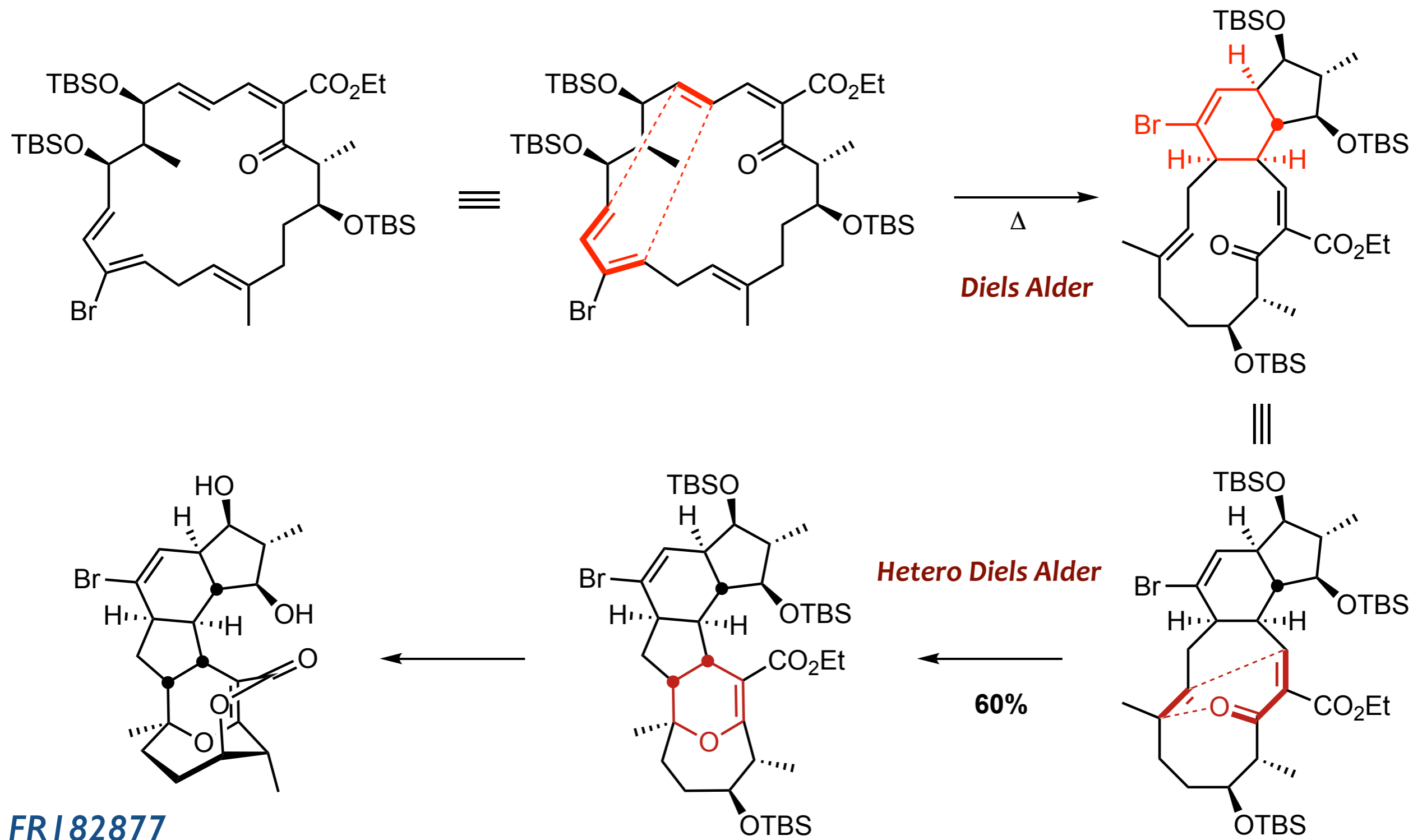
Diels Alder Cascades



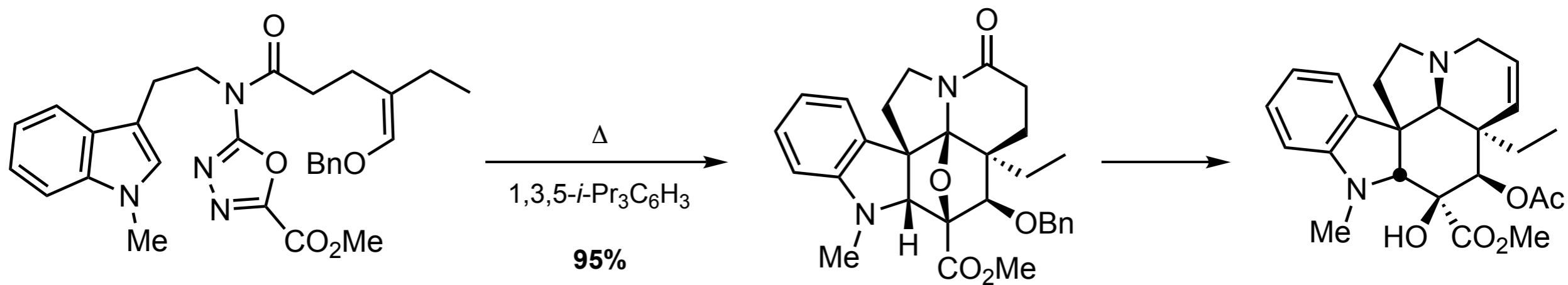
Heathcock's landmark biomimetic synthesis of Daphniphyllum alkaloids



Diels Alder Cascades

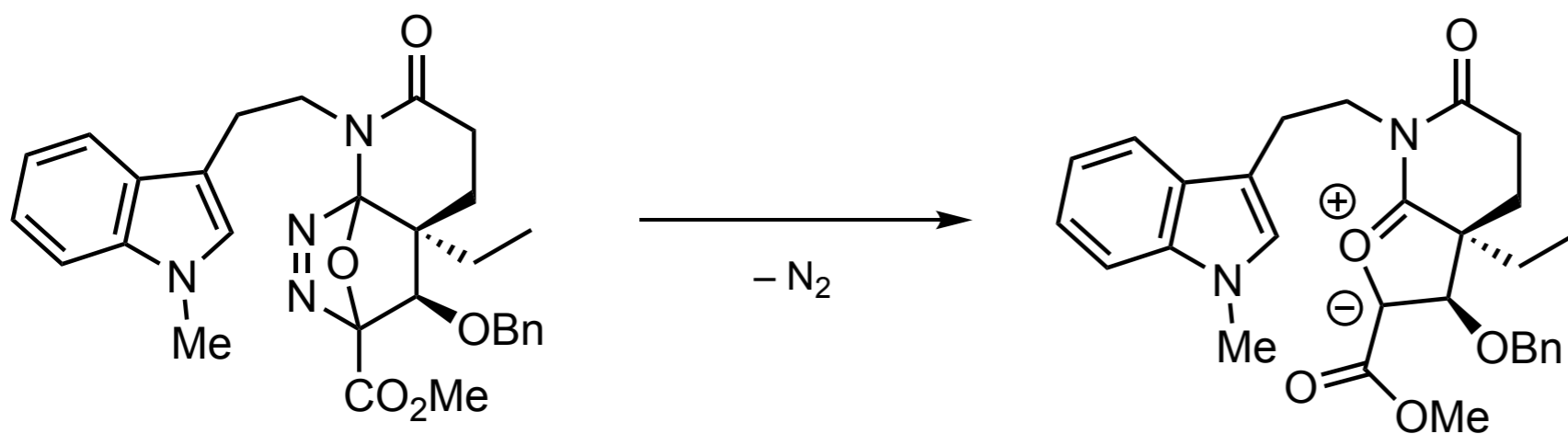


Diels Alder Cascades



Diels Alder [4 π _s + 2 π _s]

[4 π _s + 2 π _s] 1,3-Dipolar cycloaddition

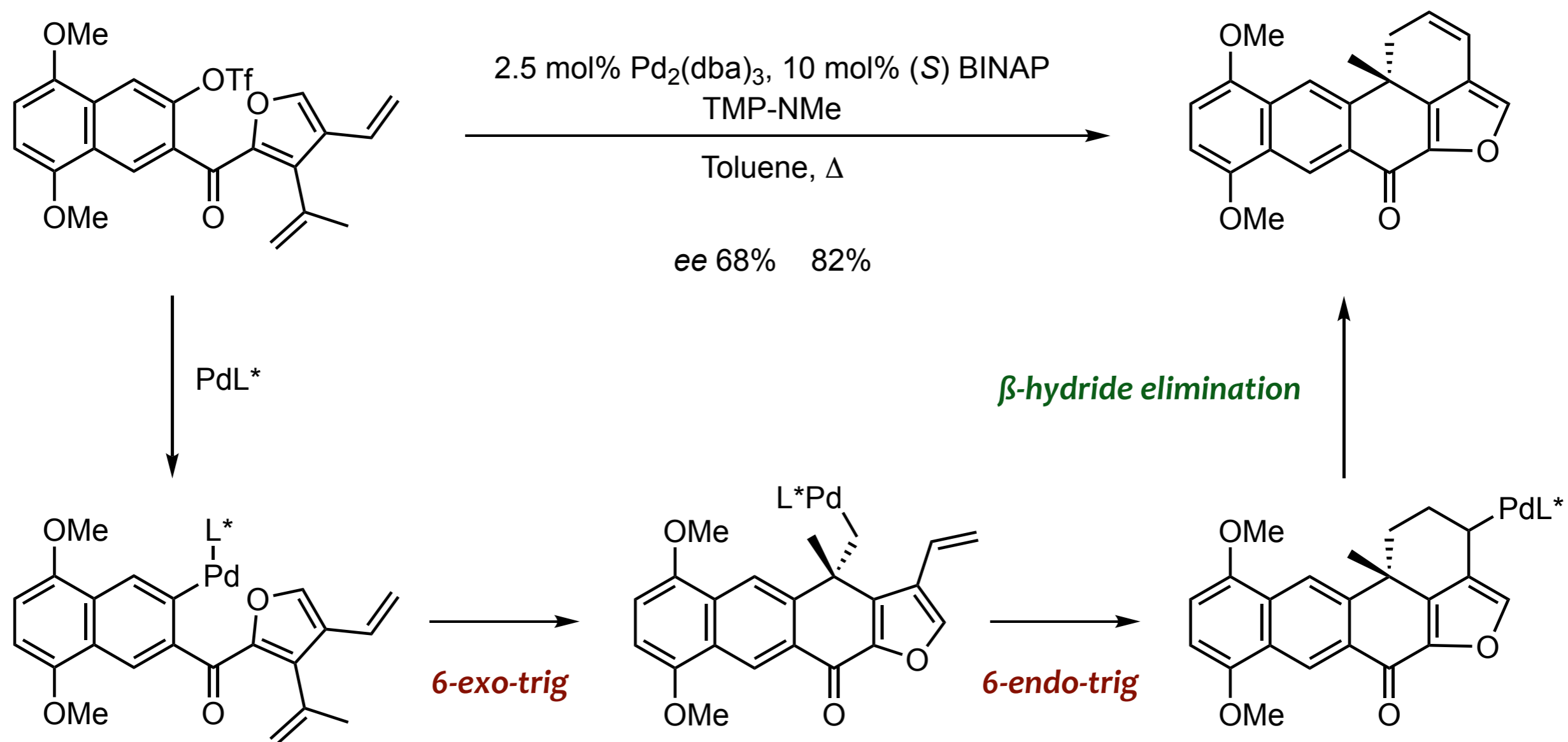


Remember!

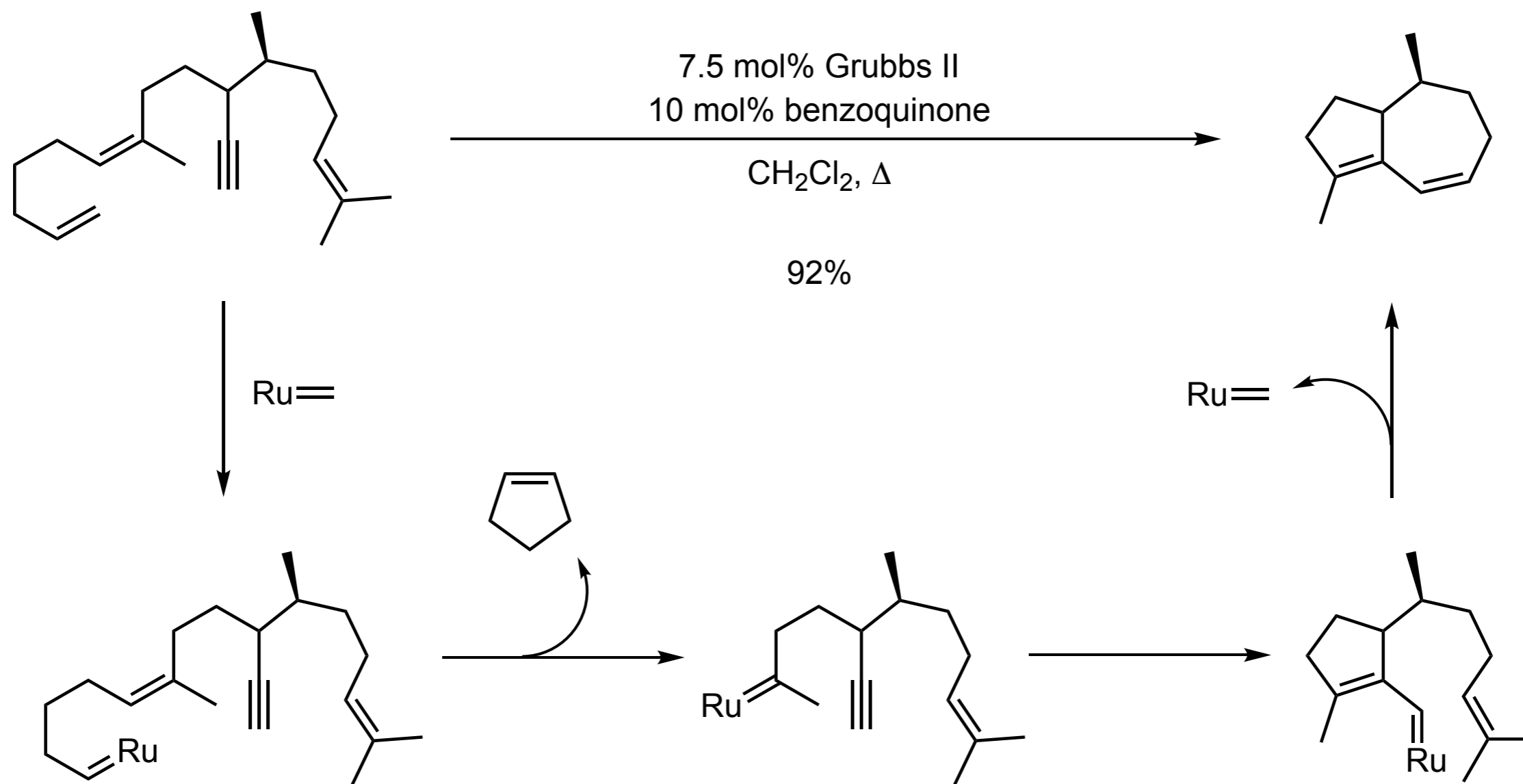


Cyclizations
Chapter 6

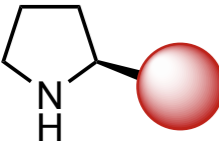



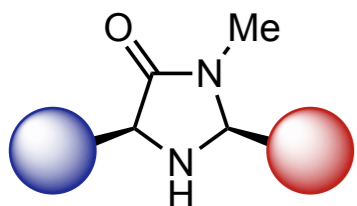



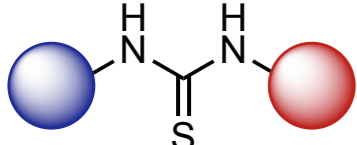


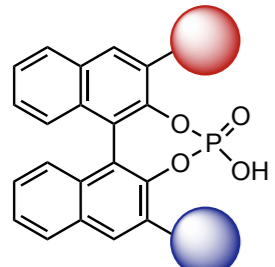


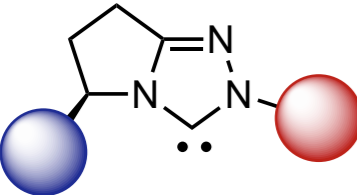


Transition Metal Cascades. Heck



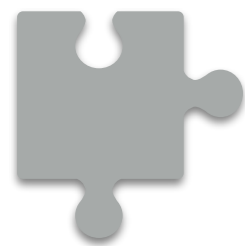
Transition Metal Cascades. Metathesis



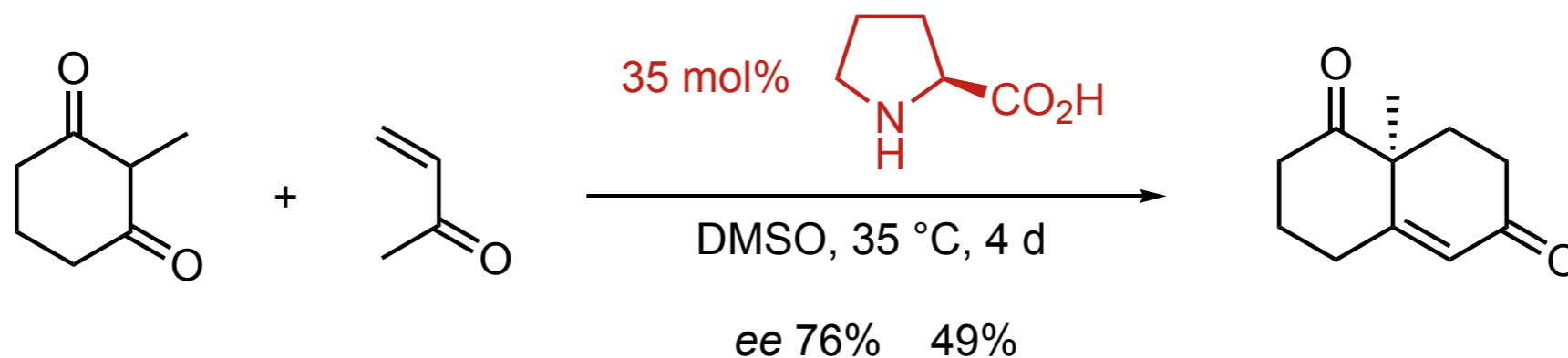
Organocatalytic Cascades

Catalyst	Activation mode	Typical reaction steps	Combination	
	Enamine activation of carbonyl compounds (HOMO raising)		Aldol reaction Michael reaction Mannich reaction	 
	Iminium activation of unsaturated aldehydes (LUMO lowering)		Michael reaction Diels-Alder reaction Friedel-Crafts reaction	 
	Hydrogen bonding (LUMO lowering)		Michael reaction Henry reaction Mannich reaction	
	Protonation (LUMO lowering)		Michael reaction Mannich reaction Friedel-Crafts reaction Reduction	
	Umpolung		Nucleophilic acylation Benzoin reaction Stetter reaction	

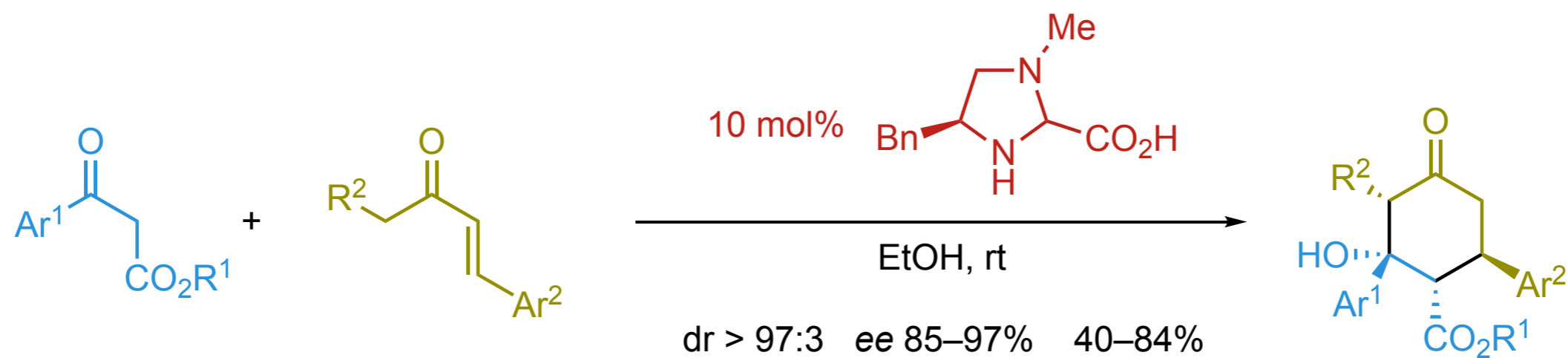
Organocatalytic Cascades



Inspirational organocatalytic cascades: Michael/Aldol reactions

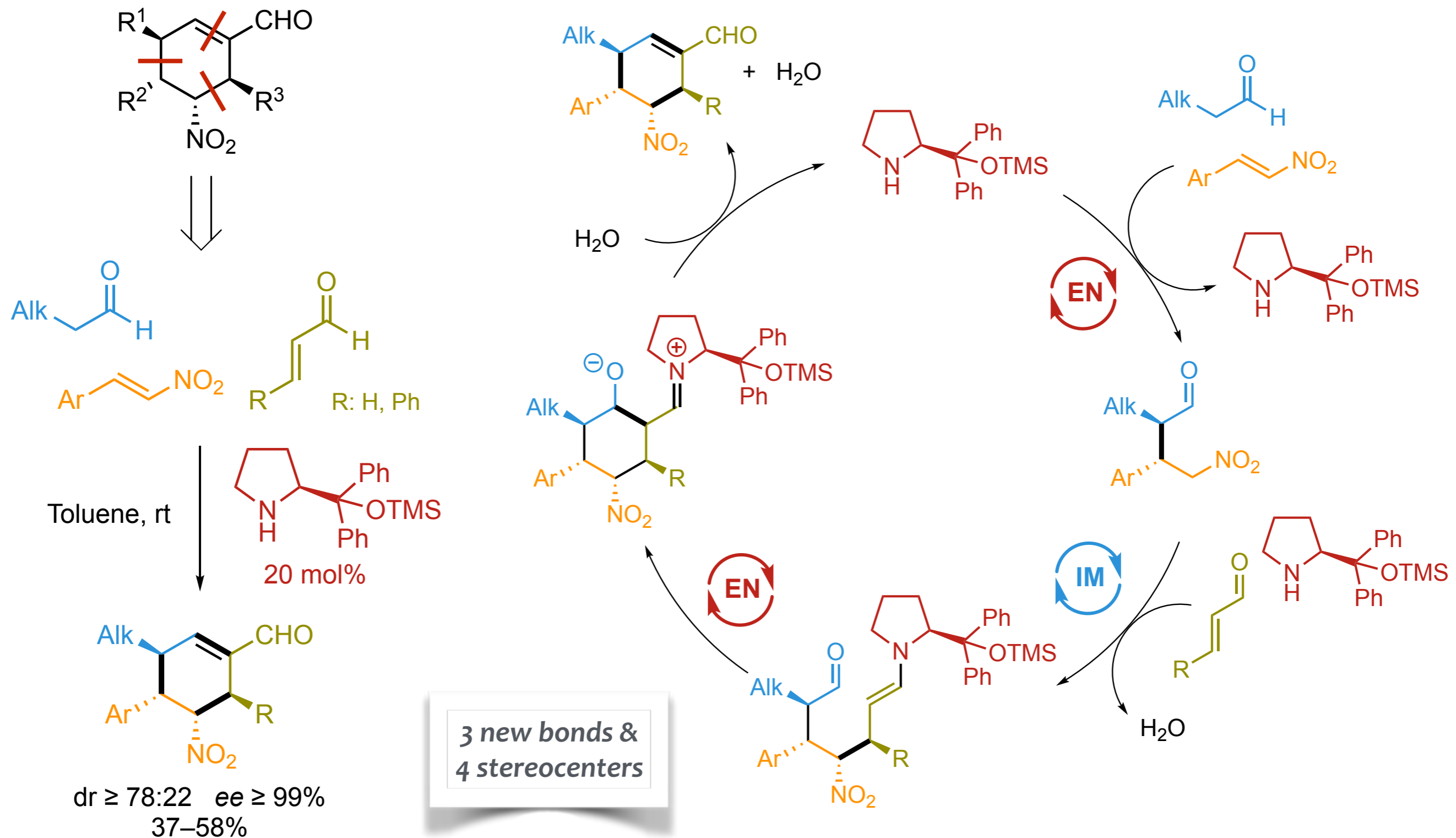


Barbas, C. *TL* **2000**, *41*, 6951



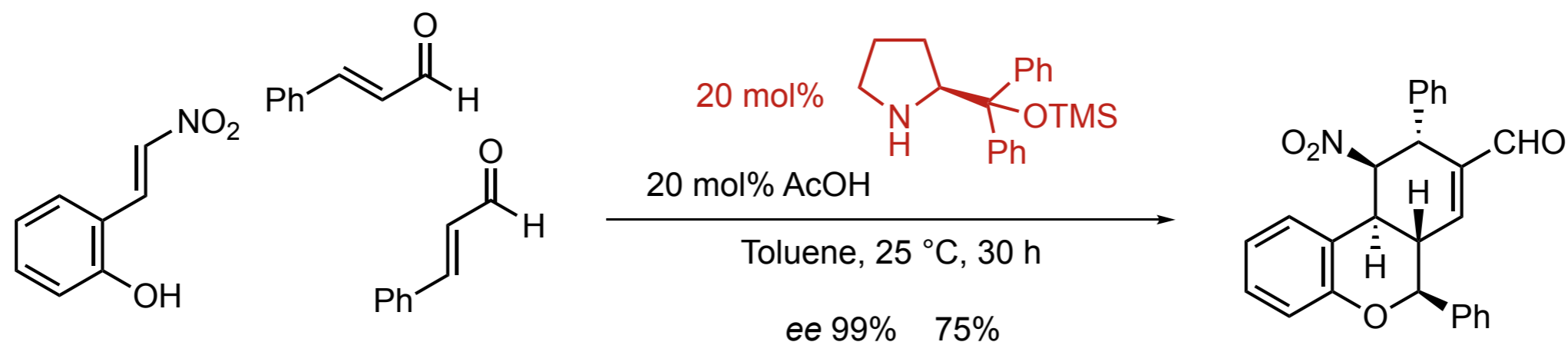
Jørgensen, K. A. *ACIE* **2004**, *43*, 1272

Organocatalytic Cascades



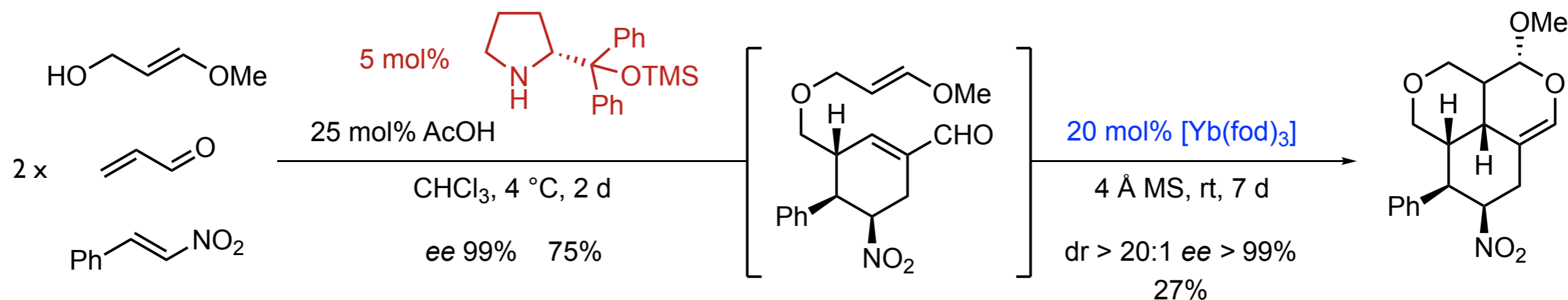
Organocatalytic Cascades

Oxa-Michael/Michael/Michael/Aldol condensation



Hong, J. E. *TL* **2009**, *80*, 1866

One pot Oxa-Michael/Michael/Michael/Aldol cond/Hetero Diels Alder



Enders, D. *ACIE* **2016**, *55*, 16153

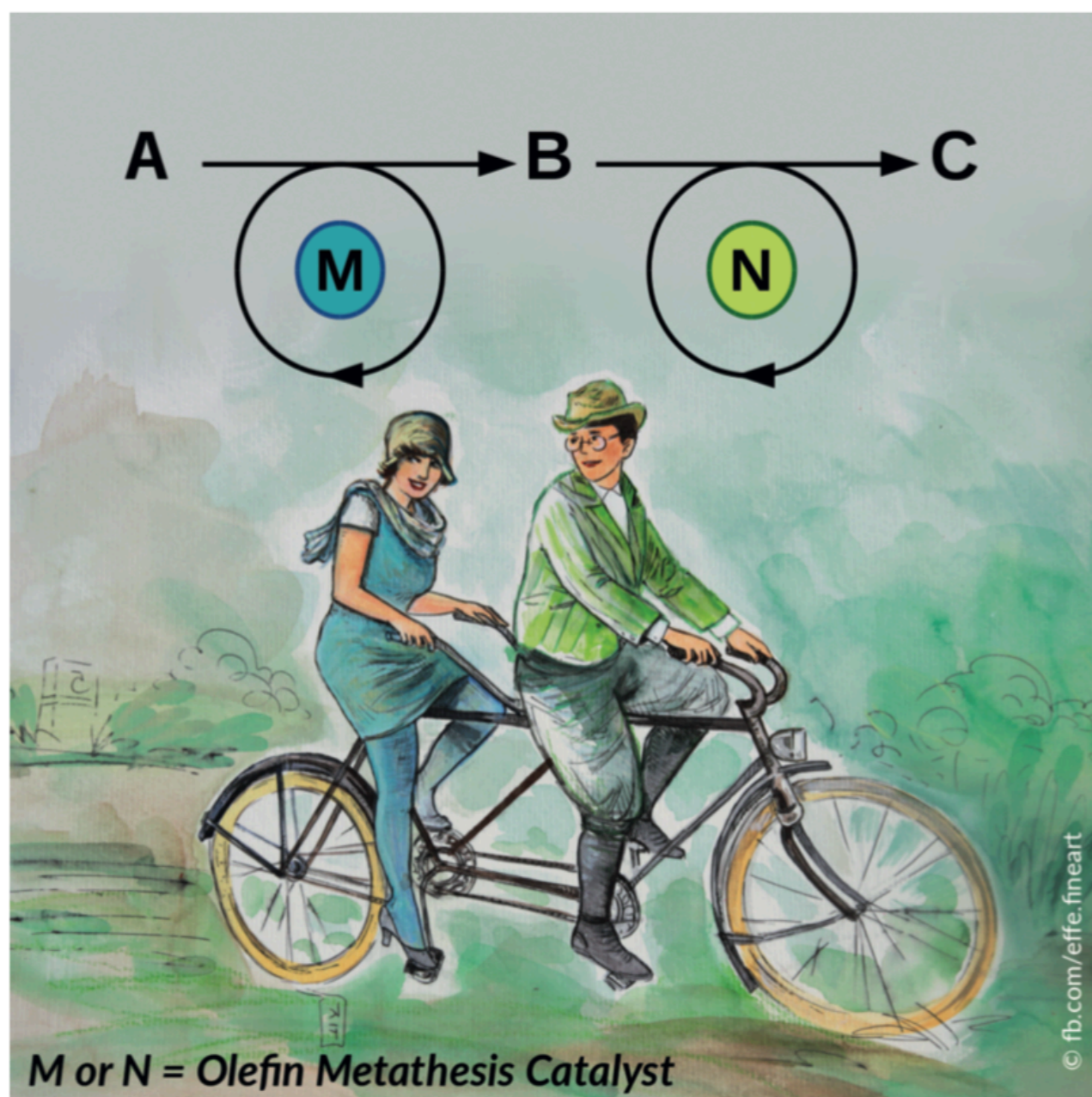
Merging Catalysts

Olefin Metathesis

Tandem Catalysis Utilizing Olefin Metathesis Reactions

Grzegorz K. Zieliński^{*(a)} and Karol Grela^{*(a, b)}

Dedicated to Professor Janusz Jurczak on the occasion of 75th birthday



Combination of catalysts
in a well orchestrated manner
produces highly efficient sequences

Fogg, D. E.; dos Santos, E. N.
Coord. Chem. Rev. **2004**, 248, 2365

Menche, D. *Nat. Prod. Rep.* **2014**, 31, 456

Marks, T. J. *Nature Chem.* **2015**, 7, 477

Zielinski, G. K.; Grela, K. *CEJ* **2016**, 22, 9440

Enzymatic Cascade Reaction in Biosynthesis

Biosynthesis

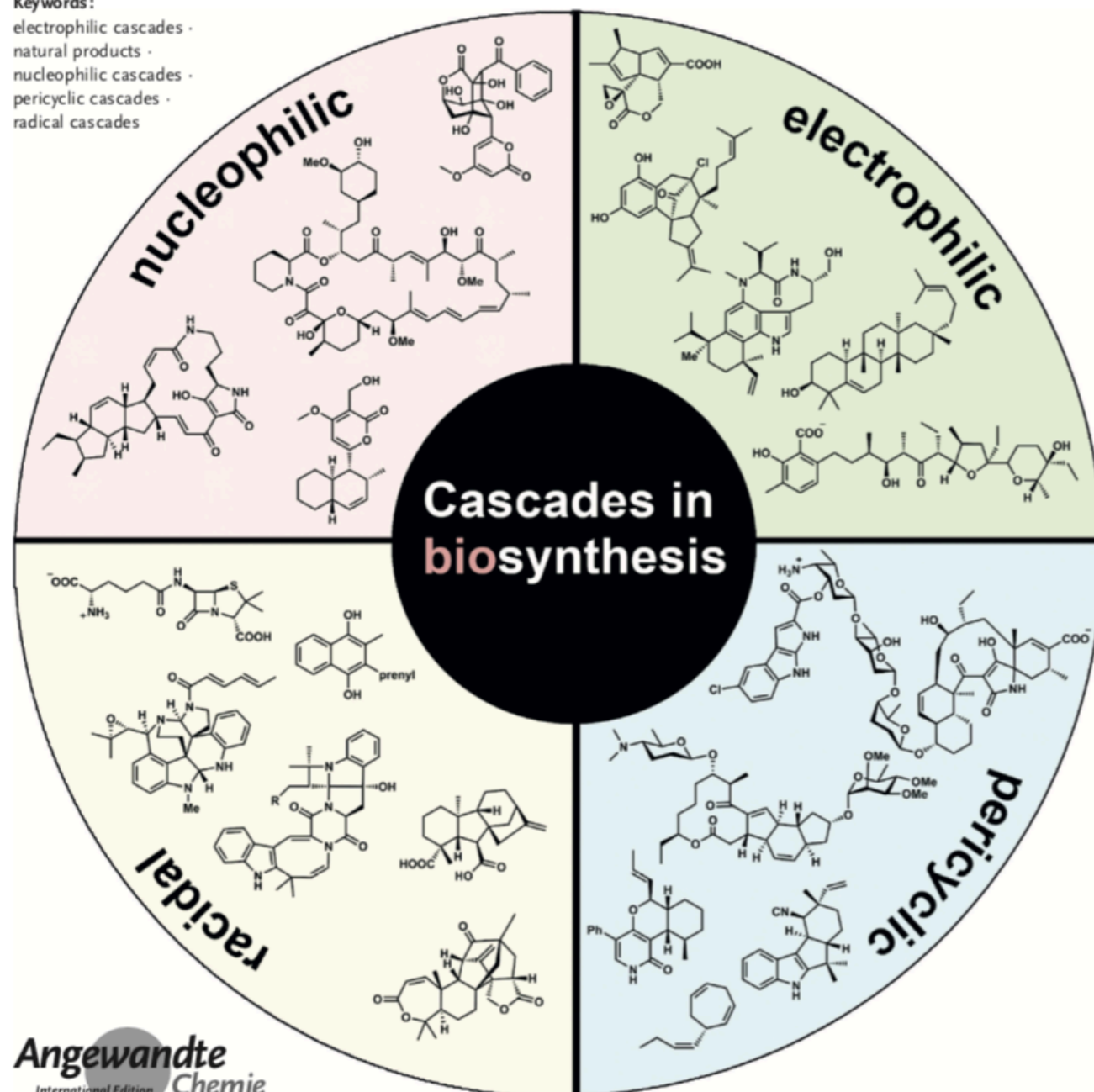
International Edition: DOI: 10.1002/anie.201807844
German Edition: DOI: 10.1002/ange.201807844

Enzymatic Cascade Reactions in Biosynthesis

Christopher T. Walsh and Bradley S. Moore*

Keywords:

electrophilic cascades ·
natural products ·
nucleophilic cascades ·
pericyclic cascades ·
radical cascades



The use of enzymes
may be an option to generate
complex molecular architectures

Moore, B. S. *ACIE* 2019, 58, 6846

Multicomponent Reactions

Multicomponent Reactions (MCRs) are one-pot reactions
employing three or more reactants,
where most of the atoms the starting materials
are incorporated in the final product generating almost no by-products

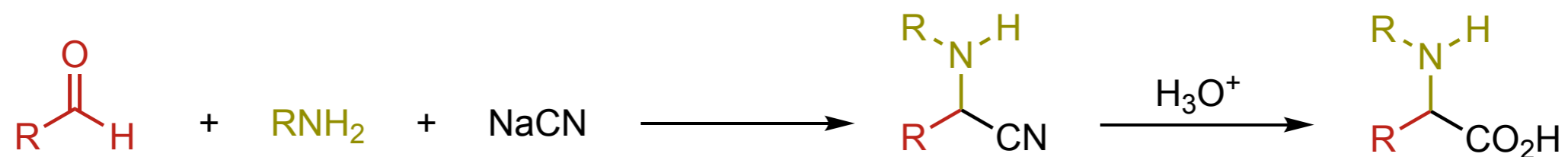
Mannich & Pauson-Khand reactions
are illustrative examples studied in former Chapters



See FG Strategies (Chap 4)
and Cyclizations (Chap 6)

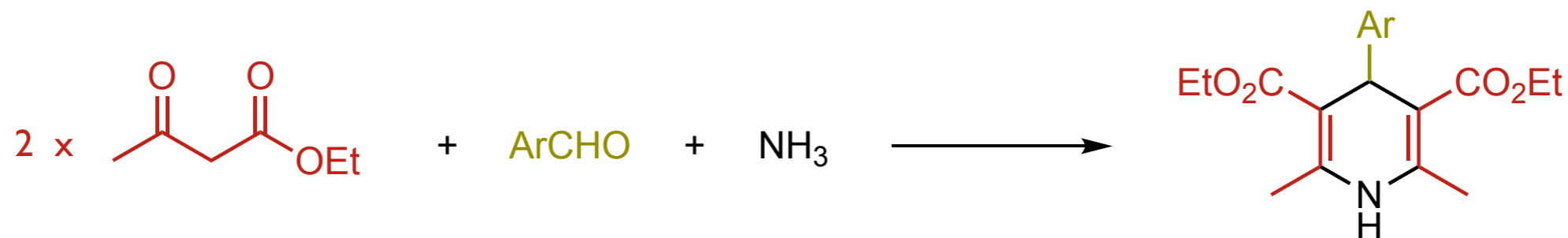
Multicomponent Reactions

➤ Strecker



Strecker, A. *Liebigs Ann. Chem.* **1850**, 75, 27

➤ Hantzsch



Hantzsch, A. *Justus Liebigs Ann. Chem.* **1882**, 215, 1

Multicomponent Reactions

➤ Ugi

