

Mechanical response evaluation of microcapsules from different slurries

Jessica Giro-Paloma¹, Camila Barreneche^{1,2}, Mònica Martínez¹; Boštjan Šumiga³, Ana Inés Fernández¹, Luisa F. Cabeza²

¹ Universitat de Barcelona. Faculty of Chemistry. Department of Materials Science and Metallurgical Engineering. C/ Martí i Franquès, 1, 08028 Barcelona (Spain). Phone: 34-934021298, Fax: 34-934035438, e-mail: jessicagiro@ub.edu, c.barreneche@ub.edu, monicamartinez@ub.edu, ana_inesfernandez@ub.edu

² GREA Innovació Concurrent. Edifici CREA. Universitat de Lleida. C/ Pere de Cabrera s/n, 25001-Lleida (Spain), Phone: 34-973003576, Fax: 34-973003575, e-mail: lcabeza@diei.udl.cat

³ Iskra Mehanizmi d.o.o., Fužine 9, 1240 Kamnik. Phone: +38618309850, Fax:+38618309867, email: bostjan.sumiga@iskra-mehanizmi.si

Abstract

Thermal energy storage (TES) is one method to accumulate thermal energy. In TES, latent heat storage using phase change materials (PCM) has attracted a lot of interest, recently. Phase change slurries (PCS) consist on a carrier fluid binary system composed of water as the continuous phase and microencapsulated PCM as the dispersed phase. In this paper, two PCS to be used for TES in buildings were studied: Micronal[®] DS 5007 X, from BASF company, and PCS28, a laboratory made sample. Both samples were characterized using particle size distribution and scanning electron microscopy, to observe the regular spherical microcapsules, the surface morphology, and the wall shell thickness of the microcapsules. Atomic force microscopy was used to analyze the force needed to break the PCS microcapsules, a critical parameter when the PCS are to be used in active pumpable systems, and also to evaluate the effective Young's modulus. Both samples were studied with the microcapsules broken and unbroken. The physicochemical and thermal properties were reported in a previous paper, and it can be concluded that both are proper candidates to be used in TES building heating and cooling applications, but the acrylic shell microcapsules present better breakage resistance to be used in active systems.

32

33 *Keywords:* Phase Change Slurry (PCS); Particle Size Distribution (PSD); Scanning
34 Electron Microscopy (SEM); Atomic Force Microscope (AFM); Mechanical
35 characterization.

36

37 **1. Introduction**

38 Phase Change Materials (PCM) are well known in Thermal Energy Storage (TES)
39 because of their capacity to absorb and slowly release the latent heat involved in the
40 phase change process [1], as well as the ability to store a large amount of energy in
41 relatively small volumes. TES with PCM achieves energy conservation in buildings
42 with thermal comfort [2]. Energy storage is very important where the energy source is
43 intermittent, as in solar energy field, and it can decrease the time between energy supply
44 and energy demand. For this reason, energy storage is essential in energy conservation
45 issues. There are some reported methods to include PCM in building walls [3] such as
46 with impregnation. There is also in the literature some studies [4,5] exposing the
47 introduction of PCM in constructive materials [6], such as gypsum [7,8], concrete [9],
48 wood [10], etc. As there is leakage when mixing PCM with building materials, PCM
49 has to be encapsulated for technical use and microcapsules were considered to address
50 this issue [11]. Microencapsulated Phase Change Materials (MPCM) [12,13,14] are
51 small vessels with a hydrophobic core material and a hard covering that accepts size
52 alterations, maintains its shape and avoids corrosion problems. MPCM temperature
53 remains unchanged during the heat absorption/release process and these can be applied
54 in passive storage systems such as component in buildings envelopes [15], as well as, in
55 sandwich panels [16]. MPCM can also be added to a fluid in order to be transported or
56 pumped in an active storage system. They can be used in active storage systems such as
57 pumping slurries [17,18,19], which are microcapsules suspensions or emulsions
58 [20,21,22] using a dispersant agent in order to stabilize the distribution of the
59 microcapsules in order to enhance its thermal behaviour. These slurries are known as
60 Phase Change Slurries (PCS).

61 Different core/shell combinations that have been studied take into account the
62 external polymer and the PCM nature (inorganic or organic) [23,24]. In this particular

63 case, the core samples are saturated hydrocarbon paraffin wax as the PCM. In this
64 paper, two samples were characterized: Micronal[®] DS 5007 X provided by BASF[®] and
65 a laboratory prepared suspension of microcapsules based on analysis and experimental
66 optimization of in situ polymerisation technology for its preparation from scientific and
67 patent literature [25]. This comparison aspect between both types of MPCM is very
68 important as it is wanted to identify the best MPCM laboratory manufactured, or at
69 least, the most similar commercial MPCM in a laboratory preparation. The two samples
70 have melting temperatures around 24 - 28 °C that are considered close to the indoor
71 comfort temperature in buildings [26]. The physicochemical and thermal properties
72 were evaluated in a previous paper [27].

73 The main objective of this study is to compare the physical and mechanical
74 behaviour of two PCS samples able to be used in active systems in order to discern the
75 most proper microcapsules for using them in a TES media for heating and cooling
76 applications.

77

78 **2. Materials and methods**

79 *2.1. Materials*

80 Two types of PCS were studied in this paper. Micronal[®] DS 5007 X which has an
81 acrylate polymer as a shell in a slurry medium, with a proportion of 2:5, (slurry +
82 dispersant):water, and paraffin wax in the core. The other sample is PCS28 which is a
83 laboratory made sample, and the shell was prepared with a melamine-formaldehyde
84 (MF) resin. The paraffin wax inside the microcapsules PCS28 is *n*-octadecane. The
85 phase change temperature of PCS28 is around 28 °C and the proportional relation is 1:2,
86 (slurry + dispersant):water [27]. The main reason for choosing these two types of
87 MPCM was to compare the different mechanical behaviour related to the MPCM
88 polymeric shell (acrylic or MF) and to decide which shell is better to microencapsulate
89 the PCM.

90 *2.2. Methodology*

91 The size of the microcapsules was studied analysing the Particle Size Distribution
92 (PSD). Also, by using Scanning Electron Microscopy (SEM) the dimensions of the

93 microcapsules can be measured. Finally, they were analyzed by Atomic Force
94 Microscopy (AFM) [12,28,29], which is commonly used to examine the nano- or
95 microscale properties of the surface and in close proximity surface regions. AFM is a
96 powerful tool for evaluating polymeric materials on a sub-micrometer scale because it
97 admits lesser forces and higher upper resolutions than nanoindenters [30]. In this study,
98 it was identified the maximum force that can be applied on the top of microcapsules
99 (dried PCS) to break them, analyzing the typical deflection-indentation curves. Besides,
100 it was calculated the Effective Young's modulus (E) distribution in a specific region of
101 each microsphere for the two studied samples. The tests were done at 23 °C and 45 °C
102 for both samples, which is with the core material in solid and liquid phase.

103 *2.2.1. Particle Size Measurements and Scanning Electron Microscopy*

104 The Particle Size Distribution (PSD) was analyzed using a Beckman Coulter®
105 LSTM 13 320 with Universal Liquid Module. The results were analyzed applying the
106 Mie mathematical model, because it fits well for homogenous, spherical, and
107 transparent or opaque particles with diameter below 50 μm . Afterwards, each sample
108 was circulated separately a total of 10 minutes in order to estimate sample volume
109 changes, to evaluate possible MPCM degradation or MPCM accumulation.

110 Samples morphologies were observed by Scanning Electron Microscopy (SEM)
111 using a Jeol JSM-6510 instrument under vacuum atmosphere, high voltage (15 kV), and
112 images obtained by secondary electrons. Furthermore, the microcapsules were broken in
113 purpose applying a compressive strength to study the wall shell thickness, which is
114 directly related with the mechanical properties of the microcapsules.

115 *2.2.2. Atomic Force Microscopy (AFM)*

116 The AFM equipment used to evaluate the mechanical properties of microcapsules
117 (two dried PCS samples) was Multimode 8 and Nanoscope V electronics from Bruker
118 with a Peak Force Quantitative Nanomechanics mode (QNM). The diamond probe used
119 was from Bruker, with a $388 \text{ nN}\cdot\text{nm}^{-1}$ spring constant. Therefore, the peak force
120 amplitude was 300 nm, the peak force frequency was 0.5 kHz, and finally, the
121 maximum vertical force was 500 nN. One aliquot of each PCS was diluted in water in a
122 small beaker. Then, 50 μl of this solution were poured on a freshly cleaved mica surface

123 under a N₂ stream until complete dryness. The samples were observed and mechanically
124 tested without further treatment.

125 The goal of the first assay is to quantify the highest vertical force that the samples
126 were able to resist after creating a permanent elastic deformation executing a force. It
127 was obtained microcapsules images at 23°C, and after that, it was applied a force on the
128 top of the microcapsule until breaking the shell to perform a force curve. Subsequently,
129 the particle was imaged again. This process was repeated three times at 23 °C for each
130 microcapsule (PCM in solid state); following that, PCS microcapsules temperature was
131 raised to 45 °C (PCM in liquid state), and the whole process was repeated again three
132 times for each sample. A total of 18 tests were performed. The second set of
133 experiments consists on testing the stiffness and calculate the Effective Young's
134 modulus (E) of the dried PCS microcapsules at the two different temperatures
135 performing two repetitions at 23 °C and 45 °C (PCM in solid and liquid state,
136 respectively). In all the cases, the temperature was measured with a micro-thermocouple
137 type-K in contact with the sample holder.

138

139 3. Results and discussion

140 3.1. Particle Size Measurements

141 3.1.1. Particle Size Distribution (PSD)

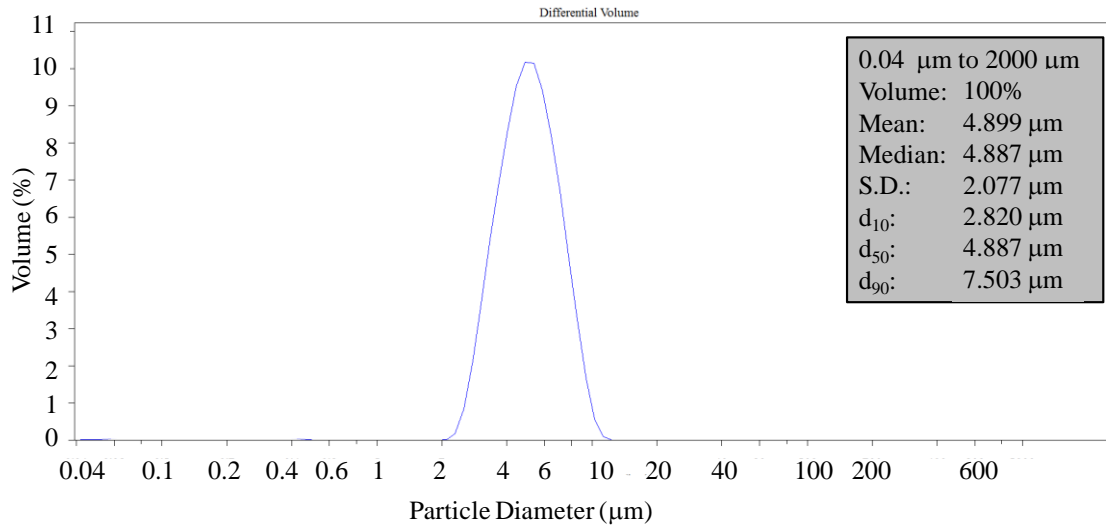
142 PSD was evaluated following the Mie mathematic model, and Table 1 lists the main
143 results obtained. The parameter d_{10} means that 10 % of the volume of the particles had a
144 diameter below the given value; for d_{50} , 50 % of the particles had a diameter below the
145 given value, and finally, 90 % of particles had a size volume below d_{90} .

146 Table 1. Particle Size Distribution results from microcapsules under study

	Micronal [®] DS	PCS28
	5007 X	
PSD d_{10} (μm)	2.79	0.14
PSD d_{50} (μm)	4.88	3.46
PSD d_{90} (μm)	7.48	6.62

PSD mean (μm)	4.88	3.14
Standard deviation (μm)	2.09	2.62

147 At the light of the results, PCS28 presents lower values than Micronal[®] DS 5007X.
 148 Thereby, the repeatability over the measurements is shown in Figure 1 for Micronal[®]
 149 DS 5007 X. The profile curves obtained are almost identical, showing a quite narrow
 150 distribution, being assured the repeatability between the different experiments.

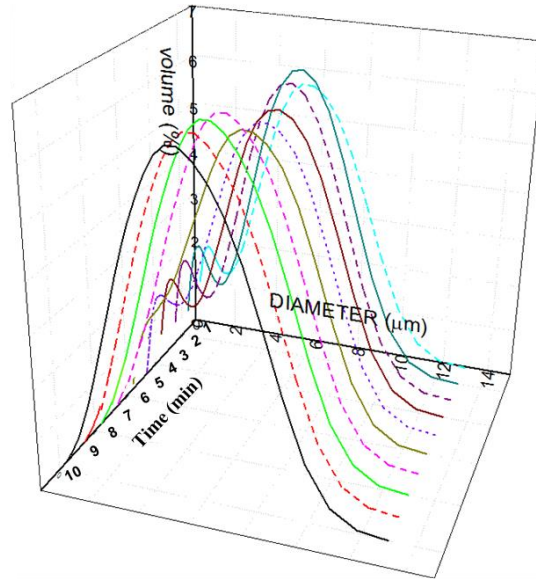


151

152 Figure 1. Particle Size Distribution of PCS Micronal[®] DS 5007 X.

153

154 The evolution over time for PCS28 results is plotted in Figure 2. This graph
 155 illustrates that at the beginning of the experiment there was a bimodal distribution
 156 showing two peaks, one corresponding to a fraction of particles with smaller size and
 157 the other one corresponding to bigger particles. After 10 minutes (1 minute per each
 158 cycle) inside the circuit, the distribution shows one peak. Therefore, an agglomeration
 159 particle process was obtained over time for this sample.



160

161

Figure 2. Particle Size Distribution progress of the PCS28 slurry in front of time.

162

163

3.1.2. Scanning Electron Microscopy (SEM)

164

165

166

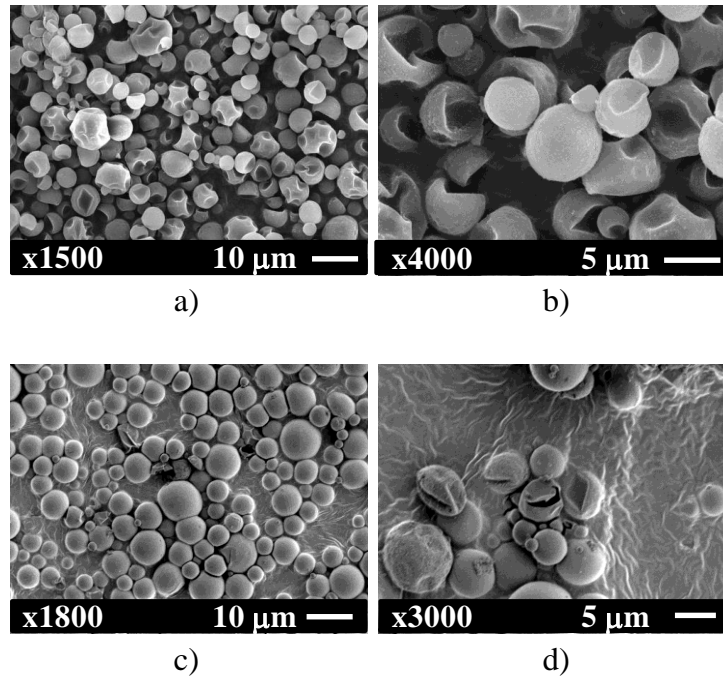
167

168

169

170

SEM images of the two samples are shown in Figure 3. It was done a drying process at room temperature to better observe the size of the PCS microcapsules, followed by a compression of the sample to crash the microcapsules. Particle sizes (left) and wall thicknesses (right) are detailed in this figure. For both samples, Micronal[®] DS 5007 X and PCS28, the particle size is variable. Note that the samples under study show approximately the same shell thickness, which is around 0.5 μm . Furthermore, the main results for microcapsules size and wall shell thickness are listed in Table 2.



171

172 Figure 3. SEM images. (a) PCS Micronal[®] DS 5007 X x 1,500, (b) PCS Micronal[®] DS
 173 5007 X x 4,000; and (c) PCS28 x 1,600, (d) PCS28 x 3,000

174

175

Table 2. SEM results from microcapsules under study

	Micronal [®] DS 5007 X	PCS28
Size (μm)	4.0 - 8.0	2.1 - 10.2
Size wall thickness shell (μm)	0.5	0.5

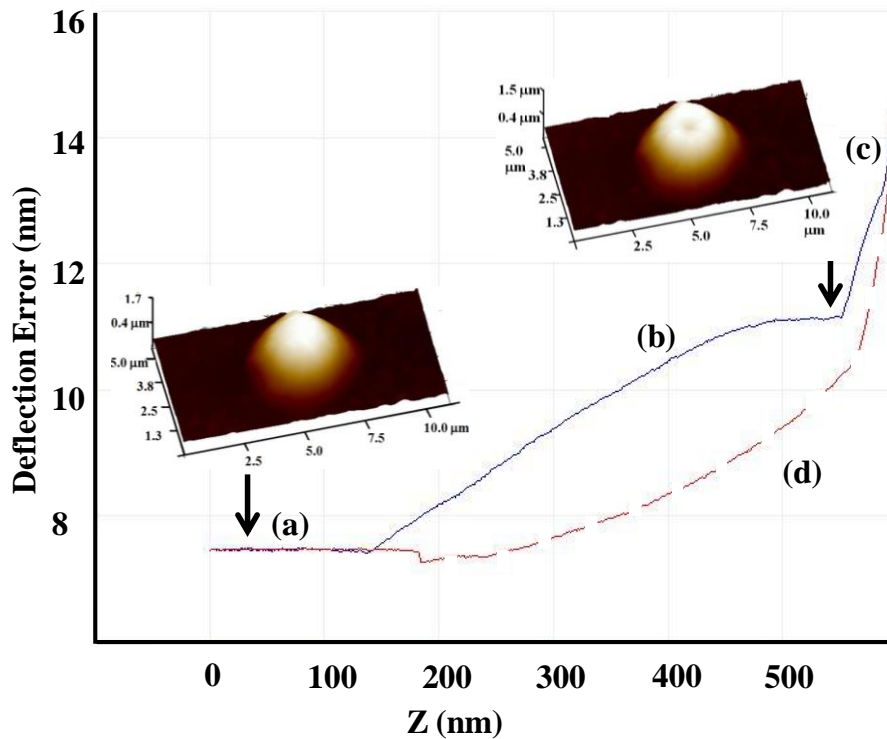
176

177 3.2. Atomic Force Microscopy

178 3.2.1. Micronal[®] DS 5007 X

179 The typical loading curve for Micronal[®] DS 5007 X at 23 °C is shown in Figure 4.
 180 The deflection error vs. Z is represented, where the deflection error is proportional to the
 181 applied force, and Z is the penetration of the tip inside the sample. At the beginning of
 182 the experiment there is no force applied (a). Subsequently, in this case, applying 11 μN
 183 of load (transition (b) to (c)) the microcapsule breaks. The tip penetrated around 0.6 μm,

184 allowing to observe the created hole in the position where the AFM probe has
185 punctured. Finally, the unloading curve shows the retracting of the tip (d).



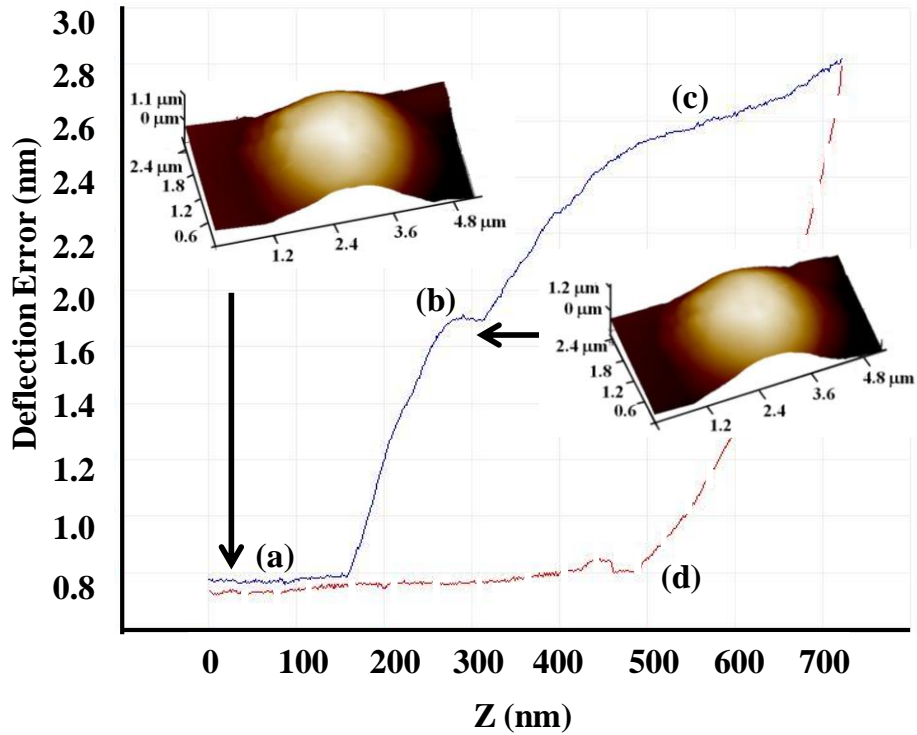
186

187 Figure 4. Mechanical testing of the elastic-plastic region by AFM of a Micronal® DS
188 5007 X microcapsule at 23 °C, (a) no contact, continuous line; (b) plastic penetration;
189 (c) deformation of the sample; and (d) retraction of the AFM probe, discontinuous line.

190

191 The same procedure of puncturing the sample was done at 45 °C, as Figure 5 shows.
192 In this case, the indentation breakthrough occurs at 1.8 μN , because the higher the
193 temperature, the softer the polymer.

194



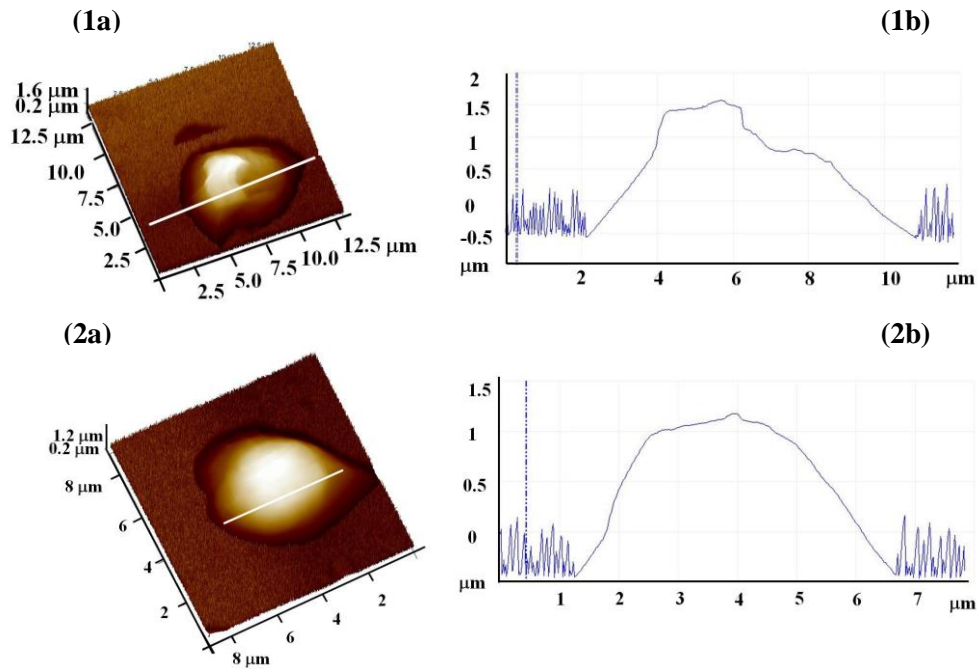
195

196 Figure 5. Mechanical testing of the elastic-plastic region by AFM of Micronal® DS 5007
 197 X microcapsule at 45 °C; (a) no contact, continuous line; (b) plastic penetration; (c)
 198 deformation of the sample; and (d) retraction of the AFM probe, discontinuous line.

199

200 It is important to note that when a microcapsule is heated up, the topographic
 201 section may change. This effect can be observed in Figure 6 where it can be seen the 3D
 202 view (Figure 6, 1a and 2a) and the profile of the white line of a microsphere (Figure 6,
 203 1b and 2b) at different temperatures, 23 °C and 45 °C.

204



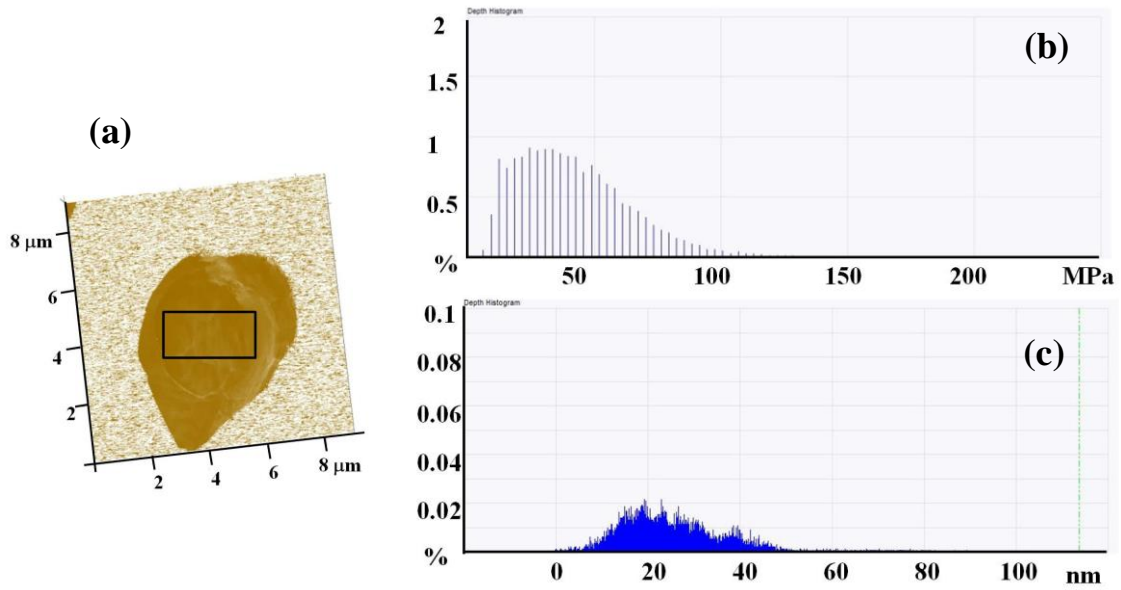
205

206 Figure 6. Micronal[®] DS 5007 X sample (1a) 3D view at 23 °C; (2a) 3D view at 45 °C;
 207 (1b) topographic image at 23 °C; (2b) topographic image at 45 °C.

208

209 Furthermore, a mechanical map at 23 °C and 0.5 KHz per second in Giro-Paloma *et*
 210 *al.* study was evaluated [17] obtaining a great dispersion of values, being between 10 to
 211 20 nm of deformation. Besides, applying a vertical force of 500 nN on the top of the
 212 microcapsule, *E* values show a great dispersion being the mean value about 200 MPa.

213 *E* mapping diagram for Micronal[®] DS 5007 X was measured also at 45 °C, and the
 214 region studied is shown in Figure 7(a). The total pixels amount of the area selected is
 215 represented the Y axis. In this case, from the deformation histogram, deformation varies
 216 around 20 nm (b), and the calculated results of the *E* histogram applying 500 nN were
 217 around 50 MPa (c), four times lower than the experiments at 23 °C due to the softening
 218 of the particle shell as well as the liquid state of PCM. The mechanical properties
 219 hastily reduced when the temperature increase and it is close to the glass transition, for
 220 Micronal[®] samples.



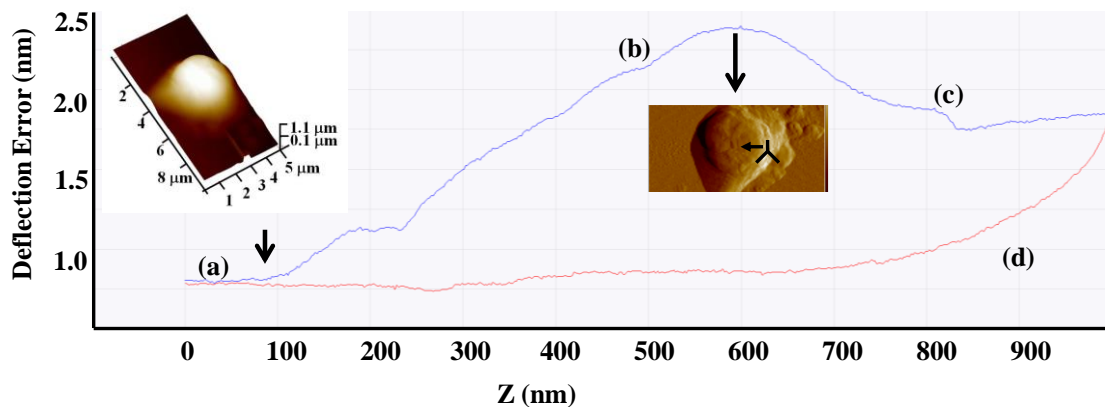
221

222 Figure 7. Effective Young's modulus mapping of Micronal[®] DS 5007 X at 45 °C; (a) 3D
 223 view; (b) Deformation histogram (nm); (c) *E* histogram (GPa).

224

225 3.2.2. PCS28

226 The experimental procedure at 23 °C was also performed for PCS28. The
 227 mechanical testing loading AFM curve of deflection error vs. *Z* is shown in Figure 8.
 228 Then, applying 2.5 μN the sample breaks and a permanent indenter print is observed in
 229 the sample (Figure 8b).

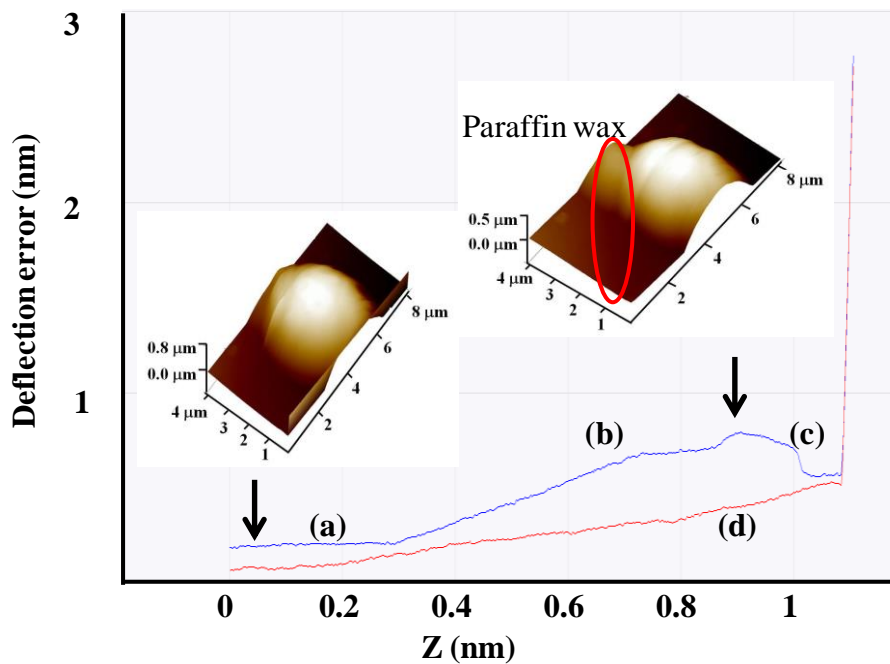


230

231 Figure 8. Mechanical testing of the elastic-plastic region by AFM of a microcapsule of
232 PCS28 at 23 °C; (a) no contact, continuous line; (b) plastic penetration;
233 of the sample; and (d) retraction of the AFM probe (red line).

234

235 Afterwards, an identical process of penetrating the microsphere with a tip was done
236 at 45 °C, as it is seen in Figure 9. This microsphere breaks at lower force (0.9 μN)
237 compared with the experiment performed at 23 °C. The particle was punctured, and the
238 inner phase change wax material spread on the mica surface, as (b) confirms.



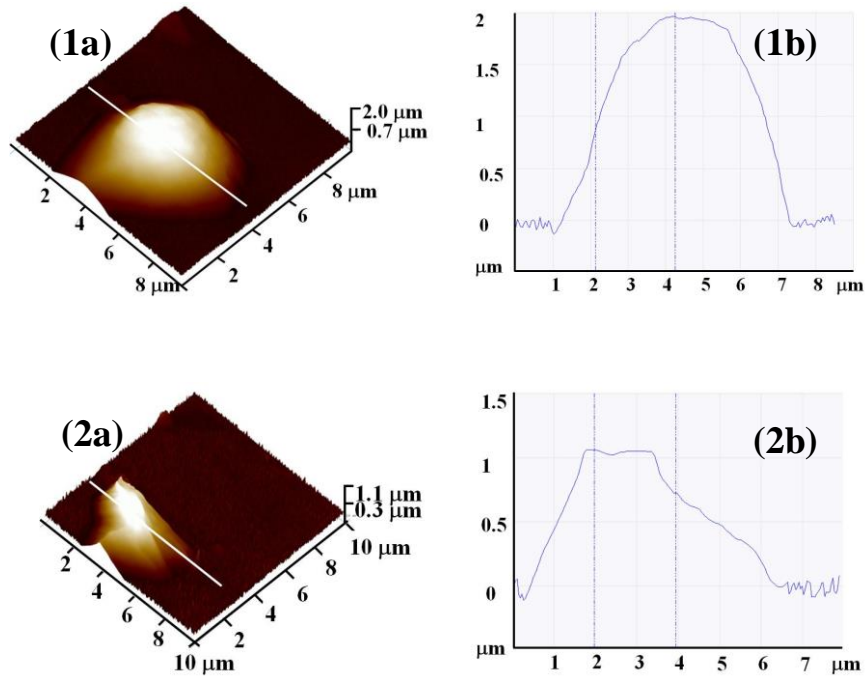
239

240 Figure 9. Mechanical testing of the elastic-plastic region by AFM of a microcapsule of
241 PCS28 at 45 °C; (a) no contact, continuous line; (b) plastic penetration;
242 of the sample; and (d) retraction of the AFM probe (red line).

243

244 When heating a microcapsule of PCS28 from 23 to 45 °C, the topographic section of
245 10 x 10 μm corresponding to the white line in the image changes. This mentioned
246 change is shown in Figure 10.

247

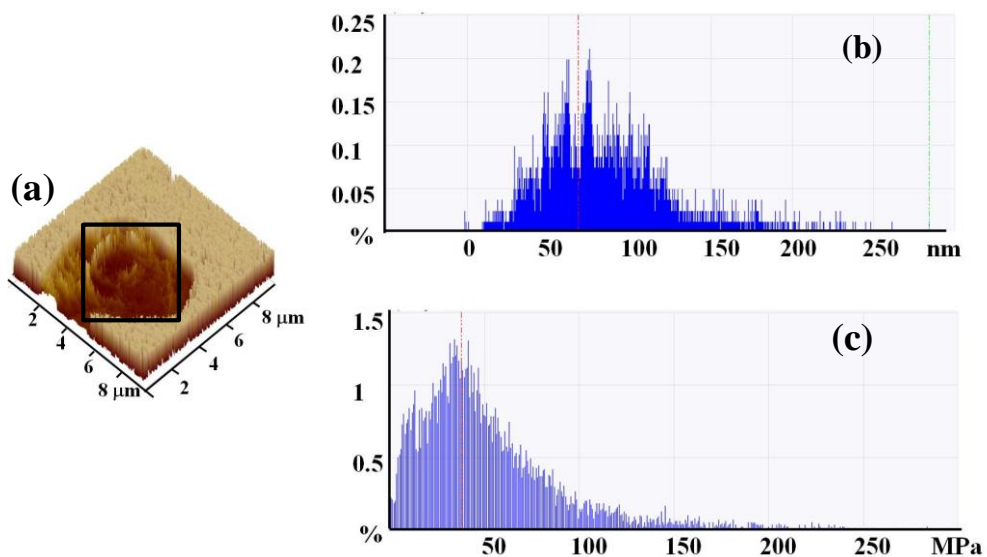


248

249 Figure 10. PCS28 sample at (1) 23 °C and, (2) 45 °C; (a) 3D view, and (b) topographic
 250 image.

251

252 The 3D view *E* map (a), the deformation histogram (b), and *E* histogram (c) of
 253 PCS28 at 23 °C are presented in Figure 11. The obtained results were 70 nm for the
 254 mean deformation and 43 MPa applying a vertical force of 500 nN. These results show
 255 that PCS28 particles are clearly softer than the Micronal[®] DS 5007 X resulting in more
 256 deformable particles.



257

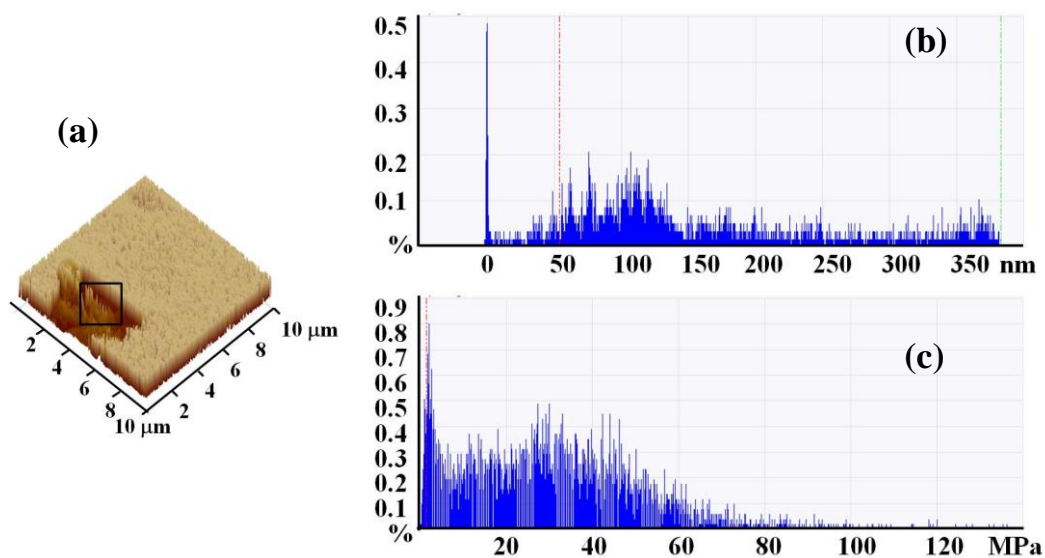
258 Figure 11. Effective Young's modulus mapping of PCS28 at 23 °C; (a) 3D view; (b)
259 Deformation histogram; (c) E histogram.

260

261 Additionally, E mapping at 45 °C for PCS28 is shown in Figure 12. As it happens
262 with Micronal® DS 5007 X, the obtained values at 45 °C are even lower compared to
263 those achieved at 23 °C. The deformation values were between 70 and 300 nm, and the
264 mean E result was 30 MPa using 500 nN of load.

265 The mechanical properties when increasing the temperature for PCS28 sample also
266 decrease because of the MF polymeric shell, since is a thermoset resin has the tendency
267 to have more fragile and pressure-sensitive walls compared with acrylic one, it is
268 susceptible to break and degrade more easily. Therefore, results demonstrate that acrylic
269 polymeric shell for MPCM for PCS is a better option compared to MF. Nevertheless,
270 the measurements were not performed in pumping conditions.

271



272

273 Figure 12. Effective Young's modulus mapping of PCS28 at 45 °C; (a) 3D view; (b)
274 Deformation histogram; (c) E histogram.

275

276 Table 3 summarizes the obtained results by means AFM technique.

277

278

Table 3. Summary of the main results.

	Micronal® DS 5007 X		PCS 28	
	23 °C	45 °C	23 °C	45 °C
Needed force to break the MPCM (μN)	11.0	1.8	2.5	0.9
Deformation (nm)	10-20	20	70	70-300
Mean Effective Young's modulus (E) value applying 500 nN (MPa)	200	50	43	30

279

280

4. Conclusions

281

282

283

284

285

286

287

288

289

290

291

292

293

294

295

296

297

298

299

300

301

302

Two different samples of PCS containing microcapsules with different polymeric shell and paraffin wax as PCM were compared in this paper. Micronal® DS 5007 X presents a bigger particle size than PCS28 sample. Furthermore, PCS28 sample shows an agglomeration over time. Additionally, the samples under study have similar wall shell thickness, around 0.5 μm . From the mechanical viewpoint, AFM technique demonstrates that it is very important to take into consideration the assay temperature, because the mechanical properties decrease abruptly when the temperature increases. The mechanical properties are reduced abruptly when the temperature is close to the glass transition for Micronal® samples. In case of PCS28 its reduction is due to nature of prepared microcapsule wall using in situ polymerisation and melamine-formaldehyde as shell material, which tend to have more brittle and pressure-sensitive walls, and were prone to cracking. Also, it is essential to consider the change in shape of the microcapsules with an increment of temperature for the studied samples. The higher the temperature, the lower the mechanical properties and this fact also is influenced by the liquid state of the paraffin wax. The needed force to break Micronal® DS 5007 X microcapsules decrease one order of magnitude when changing the temperature from 23 °C to 45 °C. Comparing Micronal® DS 5007 X with PCS28, the properties obtained from the mechanical performance are higher. In summary, from the mechanical point of view, an acrylate group to be used as a shell of microcapsules is better than a melamine-formaldehyde resin taking into account that both samples have the same thickness shell as SEM results elucidate. But, using MF as a MPCM shell, walls can be improved getting more durable, and this is also influenced by the fabrication process: laboratory

303 or industrial. Hence, results show that acrylic polymeric shell is a better choice in
304 comparison to MF, where it is difficult to change its wall brittle nature to desire more
305 elastic. However, these measurements were performed using static samples. More
306 experiments must be performed under pumping conditions.

307

308 **Acknowledgments**

309 The work is partially funded by the Spanish government (ENE2011-28269-C03-02
310 and ENE2011-22722). The authors would like to thank the Catalan Government for the
311 quality accreditation given to their research group GREA (2014 SGR 123) and research
312 group DIOPMA (2014 SGR 1543). The research leading to these results has received
313 funding from the European Union's Seventh Framework Programme (FP7/2007-2013)
314 under grant agreement n° PIRSES-GA-2013-610692 (INNOSTORAGE).

315 **References**

[1] Mehling H, Cabeza LF. Heat and cold storage with PCM. Springer-Verlag. ISBN-13: 9783540685562, 2008.

[2] Zhou D, Zhao CY, Tian Y. Review on thermal energy storage with phase change materials (PCMs) in building applications. *Appl Energ* 2012;92:593-605.

[3] Kuznik F, David D, Johannes K, Roux JJ. A review on phase change materials integrated in building walls. *Renew Sust Energ Rev* 2011;15:379-91.

[4] Sharma A, Tyagi VV, Chen CR, Buddhi D. Review on thermal energy storage with phase change materials and applications. *Renew Sust Energ Rev* 2009;13:318-45.

[5] Agyenim F, Hewitt N, Eames P, Smyth M. A review of materials, heat transfer and phase change problem formulation for latent heat thermal energy storage systems (LHTESS). *Renew Sust Energ Rev* 2010;14:615-28.

[6] Schossig P, Henning HM, Gschwander S, Haussmann T. Micro-encapsulated phase change materials integrated into construction materials. *Sol Energ Mat Sol C* 2005;89:297-306.

-
- [7] Tyagi VV, Buddhi D. PCM thermal storage in buildings: A state of the art. *Renew Sust Energ Rev* 2007;11:1146-66.
- [8] Borreguero AM, Carmona M, Sánchez ML, Valverde JL, Rodríguez JF. Improvement of the thermal behavior of gypsum blocks by the incorporation of microcapsules containing PCMs obtained by suspension polymerization with an optimal core/coating mass ratio. *Appl Therm Eng* 2010;30:1164-9.
- [9] Ling TC, Poon CS. Use of phase change materials for thermal energy storage in concrete: An overview. *Constr Build Mater* 2013;46:55-62.
- [10] Vectstaudza J, Locs J, Bajāre D, Barreneche C, Fernández AI. Characterization of Wood/PCM Composites for Building Applications. The 2nd International Energy Storage Conference. Sustainable Energy Storage in buildings, 2013.
- [11] Bayés-García L, Ventolà L, Cordobilla R. Phase change materials (PCM) microcapsules with different shell compositions: preparation, characterization and thermal stability. *Sol Energ Mater Sol C* 2010;94:12535-40.
- [12] Giro-Paloma J, Oncins G, Barreneche C, Martínez M, Fernández AI, Cabeza LF. Physico-chemical and mechanical properties of microencapsulated phase change material. *Appl Energ* 2013;109:441-8.
- [13] Zhao CY, Zhang GH. Review on microencapsulated phase change materials (MEPCMs): Fabrication, characterization and applications. *Renew Sust Energ Rev* 2011;15:3813-32.
- [14] Salunkhe PB, Shembekar PS. A review on effect of phase change material encapsulation on the thermal performance of a system. *Renew Sust Energ Rev* 2012;16:5603-16.
- [15] Tyagi VV, Kaushik SC, Tyagi SK, Akiyama T. Development of phase change materials based microencapsulated technology for buildings: a review. *Renew Sust Energ Rev* 2011;15:1373-91.
- [16] Castellon C, Medrano M, Roca J, Cabeza LF, Navarro ME, Fernandez AI. Effect of microencapsulated phase change material in sandwich panels. *Renew Energ* 2010;35(10):2370-74.

-
- [17] Giro-Paloma J, Barreneche C, Delgado M, Martínez M, Fernández AI, Cabeza LF. Physicochemical and thermal study of a MPCM of PMMA shell and paraffin wax as a core. *Energy Procedia* 2014;48:347-54.
- [18] Delgado M, Lázaro A, Mazo J, Marín JM, Zalba B. Experimental analysis of a microencapsulated PCM slurry as thermal system and as heat transfer fluid in laminar flow. *Appl Therm Eng* 2012;36(1):370-7.
- [19] Lu W, Tassou SA. Experimental study of the thermal characteristics of phase change slurries for active cooling. *Appl Energ* 2012;91(1):366-74.
- [20] Zhang S, Niu J. Experimental investigation of effects of super-cooling on microencapsulated phase-change material (MPCM) slurry thermal storage capacities. *Sol Energ. Mat Sol C* 2010;94(6):1038-48.
- [21] Youssef Z, Delahaye A, Huang L, Trinquet F, Fournaison L, Pollerberg C, Doetsch C. State of the art on phase change material slurries. *Energ Convers Manage* 2013;65:120-32.
- [22] Delgado M, Lázaro A, Mazo J, Zalba B. Review on phase change material emulsions and microencapsulated phase change material slurries: materials heat transfer studies and applications. *Renew Sust Energ Rev* 2012;16(1):253-73.
- [23] Castellón C, Martorell I, Cabeza LF, Fernández AI, Manich AM. Compatibility of plastic with phase change materials (PCM). *Int J Energ Res* 2010;35:765-71.
- [24] Zhang GH, Zhao CY. Thermal and rheological properties of microencapsulated phase change materials. *Renew Energ* 2011;36:2959-66.
- [25] Šumiga B. Ph.D Thesis, University of Ljubljana, 2013
- [26] Su JF, Wang LX, Ren L. Preparation and characterization of double-MF shell microPCMs used in building materials. *J Appl Polym Sci* 2005;97(5):1755-62.
- [27] Giro-Paloma J, Barreneche C, Martínez M, Sumiga B, Cabeza LF, Fernández AI. Comparison of phase change slurries: Physicochemical and thermal properties. *Energy* 2015;87:223-227.

[28] VanLandingham MR, Villarrubia JS, Meyers GF. Nanoindentation of polymers: overview. *Polymer preprints* 2000;41(2):1412.

[29] Clifford CA, Seah MP. Nanoindentation measurement of Effective Young's modulus for compliant layers on stiffer substrates including the effect of Poisson's ratios. *Nanotechnology* 2009;20:1-8.

[30] Giro-Paloma J, Roa JJ, Díez-Pascual AM, Rayón E, Flores A, Martínez M, Chimenos JM, Fernández AI. Depth-sensing indentation applied to polymers: A comparison between standard methods of analysis in relation to the nature of the materials. *Eur Polym J* 2013;49:4047-53.